

New Probability Distributions in Astrophysics: III. The Truncated Maxwell-Boltzmann Distribution

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Abstract

The Maxwell-Boltzmann (MB) distribution for velocities in ideal gases is usually defined between zero and infinity. A double truncated MB distribution is here introduced and the probability density function, the distribution function, the average value, the r th moment about the origin, the root-mean-square speed and the variance are evaluated. Two applications are presented: 1) a numerical relationship between root-mean-square speed and temperature, and 2) a modification of the formula for the Jeans escape flux of molecules from an atmosphere.

Keywords

05.20.-y Classical Statistical Mechanics, 05.20.Dd Kinetic Theory

1. Introduction

The *Maxwell-Boltzmann* (MB) distribution, see [1] [2], is a powerful tool to explain the kinetic theory of gases. The range in velocity of this distribution spans the interval $[0, \infty]$, which produces several problems:

- 1) The maximum velocity of a gas cannot be greater than the velocity of light, c .
- 2) The kinetic theory is developed in a classical environment, which means that the involved velocities should be smaller than $\approx 1/10c$.

These items point toward the hypothesis of an upper bound in velocity for the MB. We will now report some approaches, including an upper bound in velocity: the ion velocities parallel to the magnetic field in a low density surface of a ionized plasma [3]; propagation of longitudinal electron waves in a collisionless, homogeneous, isotropic plasma, whose velocity distribution function is a trun-

cated MB [4]; fast ion production in laser plasma [5]; the release of a dust particle from a plasma-facing wall [6]; an explanation of an anomaly in the Dark Matter (DAMA) experiment [7]; a distorted MB distribution of epithermal ions observed associated with the collapse of energetic ions [8]; and deviations to MB distribution that could have observable effects which can be measured through the vapor spectroscopy at an interface [9]. However, these approaches do not clearly cover the effect of introducing a lower and an upper boundary in the MB distribution, which is the subject that will be analyzed in this paper.

This paper is structured as follows. Section 2 reviews the basic statistics of the MB distribution and it derives a new approximate expression for the median. Section 3 introduces the double truncated MB and it derives the connected statistics. Section 4 derives the relationship for root-mean-square speed versus temperature in the double truncated MB. Finally, Section 5.2 derives a new formula for Jeans flux in the exosphere.

2. The Maxwell-Boltzmann Distribution

Let V be a random variable defined in $[0, \infty]$; the MB probability density function (PDF), $f(v; a)$, is

$$f(v; a) = \frac{\sqrt{2}v^2 e^{-\frac{1}{2}\frac{v^2}{a^2}}}{\sqrt{\pi}a^3}, \quad (1)$$

where a is a parameter and v denotes the velocity, see [1] [2]. Conversion to the physics is done by introducing the variable a , which is defined as

$$a = \sqrt{\frac{kT}{m}}, \quad (2)$$

where m is the mass of the gas molecules, k is the Boltzmann constant and T is the thermodynamic temperature. With this change of variable, the MB PDF is

$$f_p(v; m, k, T) = \frac{\sqrt{2}v^2 e^{-\frac{1}{2}\frac{v^2 m}{kT}}}{\sqrt{\pi} \left(\frac{kT}{m}\right)^{\frac{3}{2}}}, \quad (3)$$

where the index p stands for physics. The distribution function (DF), $F(x; a)$, is

$$F(v; a) = \frac{\sqrt{2}a^2 \left(a\sqrt{\pi}\sqrt{2}\operatorname{erf}\left(\frac{1}{2}\frac{\sqrt{2}v}{a}\right) - 2ve^{-\frac{1}{2}\frac{v^2}{a^2}} \right)}{2\sqrt{\pi}a^3} \quad (4)$$

$$F_p(v) = \frac{\sqrt{2} \left[\left(\frac{kT}{m}\right)^{\frac{3}{2}} \sqrt{\pi}\sqrt{2}\operatorname{erf}\left(\frac{1}{2}\sqrt{2}v\frac{1}{\sqrt{\frac{kT}{m}}}\right) m - 2ve^{-\frac{1}{2}\frac{v^2 m}{kT}} kT \right]}{2\sqrt{\pi} \left(\frac{kT}{m}\right)^{\frac{3}{2}} m}. \quad (5)$$

The average value or mean, μ , is

$$\mu(a) = 2 \frac{\sqrt{2}a}{\sqrt{\pi}}, \quad (6)$$

$$\mu(m, k, T)_p = 2 \frac{\sqrt{2}}{\sqrt{\pi}} \sqrt{\frac{kT}{m}}, \quad (7)$$

the variance, σ^2 , is

$$\sigma^2(a) = \frac{a^2(-8 + 3\pi)}{\pi} \quad (8)$$

$$\sigma^2(m, k, T)_p = \frac{kT(-8 + 3\pi)}{m\pi}. \quad (9)$$

The r th moment about the origin for the MB distribution is, μ'_r , is

$$\mu'_r(a) = \frac{2^{r/2+1} a^r \Gamma\left(r/2 + \frac{3}{2}\right)}{\sqrt{\pi}} \quad (10)$$

$$\mu'_r(m, k, T)_p = \frac{2^{r/2+1} \left(\sqrt{\frac{kT}{m}}\right)^r \Gamma\left(r/2 + \frac{3}{2}\right)}{\sqrt{\pi}}, \quad (11)$$

where

$$\Gamma(z) = \int_0^\infty e^{-t} t^{z-1} dt, \quad (12)$$

is the gamma function, see [10]. The root-mean-square speed, v_{rms} , can be obtained from this formula by inserting $r = 2$

$$v_{rms}(a) = \sqrt{3}a \quad (13)$$

$$v_{rms}(m, k, T)_p = \sqrt{3} \sqrt{\frac{kT}{m}}, \quad (14)$$

see Equations (7-10-16) in [11]. This equation allows us to derive the temperature once the root-mean-square speed is measured

$$T = \frac{1}{3} \frac{v_{rms}^2 m}{k}. \quad (15)$$

The coefficient of variation (CV) is

$$CV = \frac{\sigma(a)}{\mu(a)} = \sqrt{\frac{3}{8} \pi - 1}, \quad (16)$$

which is constant. The first three r th moments about the mean for the MB distribution, $\mu_r(a)$, are

$$\mu_2(a) = \frac{a^2(-8 + 3\pi)}{\pi} \quad (17)$$

$$\mu_3(a) = -2 \frac{a^3 \sqrt{2} (5\pi - 16)}{\pi^{3/2}} \quad (18)$$

$$\mu_4(a) = \frac{a^4 (15\pi^2 + 16\pi - 192)}{\pi^2}. \quad (19)$$

The mode is at

$$v(a) = \sqrt{2}a \quad (20)$$

$$v(m, k, T)_p = \sqrt{2} \sqrt{\frac{kT}{m}}. \quad (21)$$

An approximate expression for the median can be obtained by a Taylor series of the DF around the mode. The approximation formula is

$$v(a) = -\frac{1}{4}a \left(-6 + e \left(\operatorname{erf}(1) - \frac{1}{2} \right) \sqrt{\pi} \right) \sqrt{2}, \quad (22)$$

$$v(m, k, T)_p = -\frac{1}{4} \sqrt{\frac{kT}{m}} \left(-6 + e \left(\operatorname{erf}(1) - \frac{1}{2} \right) \sqrt{\pi} \right) \sqrt{2}, \quad (23)$$

which has a percent error, δ , of $\delta \approx 0.04\%$ in respect to the numerical value. The entropy is

$$\ln(\sqrt{2}\sqrt{\pi}a) - \frac{1}{2} + \gamma, \quad (24)$$

$$\ln\left(\sqrt{2}\sqrt{\pi}\sqrt{\frac{kT}{m}}\right) - \frac{1}{2} + \gamma, \quad (25)$$

where γ is the Euler-Mascheroni constant, which is defined as

$$\gamma = \lim_{n \rightarrow \infty} \left(1 + \frac{1}{2} + \frac{1}{3} + \dots + \frac{1}{n} - \ln n \right) = 0.57721\dots, \quad (26)$$

see [10] for more details. The coefficient of skewness is

$$\frac{(-10\pi + 32)\sqrt{2}}{(-8 + 3\pi)^{\frac{3}{2}}} \approx 0.48569, \quad (27)$$

and the coefficient of kurtosis is

$$\frac{15\pi^2 + 16\pi - 192}{(-8 + 3\pi)^2} \approx 3.10816. \quad (28)$$

According to [12], a random number generation can be obtained via inverse transform sampling when the distribution function or cumulative distribution function, $F(x)$, is known: 1) a pseudo number generator gives a random number R between zero and one; 2) the inverse function $x = F^{-1}(R)$ is evaluated; and 3) the procedure is repeated for different values of R . In our case, the inverse function should be evaluated in a numerical way by solving for v the following nonlinear equation

$$F(v; a) - R = 0, \quad (29)$$

$$F(v; m, k, T)_p - R = 0, \quad (30)$$

where $F(v)$ and $F_p(v)$ are the two DF represented by Equations (4) and (5). As a practical example, by inserting in Equation (29) $a=1$ and $R=0.5$, we obtain in a numerical way $v=1.538$.

3. The Double Truncated Maxwell-Boltzmann Distribution

Let V be a random variable that is defined in $[v_l, v_u]$; the *double truncated* version of the Maxwell-Boltzmann PDF, $f_t(v; a, v_l, v_u)$, is

$$f_t(v; a, v_l, v_u) = v^2 e^{-\frac{1}{2} \frac{v^2}{a^2}}, \quad (31)$$

where

$$C = \frac{-2}{CD}, \quad (32)$$

where

$$CD = a^2 \left(-a\sqrt{\pi}\sqrt{2} \operatorname{erf} \left(\frac{1}{2} \frac{\sqrt{2}v_u}{a} \right) + a\sqrt{\pi}\sqrt{2} \operatorname{erf} \left(\frac{1}{2} \frac{\sqrt{2}v_l}{a} \right) + 2v_u e^{-\frac{1}{2} \frac{v_u^2}{a^2}} - 2v_l e^{-\frac{1}{2} \frac{v_l^2}{a^2}} \right), \quad (33)$$

and $\operatorname{erf}(x)$ is the error function, which is defined as

$$\operatorname{erf}(x) = \frac{2}{\sqrt{\pi}} \int_0^x e^{-t^2} dt, \quad (34)$$

see [10]. The physical meaning of a is still represented by Equation (2); however, due to the tendency to obtain complicated expressions, we will omit the double notation. The DF, $F_t(v; a, v_l, v_u)$, is

$$F_t(v; a, v_l, v_u) = \frac{Ca^2 \left(\sqrt{\pi}\sqrt{2}a \operatorname{erf} \left(\frac{1}{2} \frac{\sqrt{2}v}{a} \right) - 2ve^{-\frac{1}{2} \frac{v^2}{a^2}} \right)}{2}. \quad (35)$$

The average value $\mu_t(a, v_l, v_u)$, is

$$\mu_t(a, v_l, v_u) = Ca^2 \left(2e^{-\frac{1}{2} \frac{v_l^2}{a^2}} - 2e^{-\frac{1}{2} \frac{v_u^2}{a^2}} + e^{-\frac{1}{2} \frac{v_l^2}{a^2}} v_l^2 - e^{-\frac{1}{2} \frac{v_u^2}{a^2}} v_u^2 \right). \quad (36)$$

The r th moment about the origin for the double truncated MB distribution is, $\mu'_{r,t}(a, v_l, v_u)$,

$$\mu'_{r,t}(a, v_l, v_u) = \frac{MN}{r+3} \quad (37)$$

where

$$MN = C2^{\frac{r+5}{4}} a^2 \times \left(\left(\frac{v_u^2}{a^2} \right)^{\frac{r-1}{4}} v_u^{r+1} e^{-\frac{1}{4} \frac{v_u^2}{a^2}} M_{\frac{r+1}{4}, \frac{r+3}{4}} \left(\frac{1}{2} \frac{v_u^2}{a^2} \right) - v_l^{r+1} e^{-\frac{1}{4} \frac{v_l^2}{a^2}} M_{\frac{r+1}{4}, \frac{r+3}{4}} \left(\frac{1}{2} \frac{v_l^2}{a^2} \right) \right) \quad (38)$$

where $M_{\mu,\nu}(z)$ is the Whittaker M function, see [10]. The root-mean-square speed, $v_{rms,t}(a, v_l, v_u)$, can be obtained from this formula by inserting $r = 2$, and is

$$v_{rms,t}(a, v_l, v_u) = \sqrt{\frac{NV}{5 \left(\frac{v_u^2}{a^2}\right)^{3/4} \left(\frac{v_l^2}{a^2}\right)^{3/4}}}, \tag{39}$$

where

$$NV = 2C2^{3/4} a^2 \left(v_u^3 e^{-1/4 \frac{v_u^2}{a^2}} M_{3/4,5/4} \left(1/2 \frac{v_u^2}{a^2} \right) \left(\frac{v_l^2}{a^2} \right)^{3/4} - v_l^3 e^{-1/4 \frac{v_l^2}{a^2}} M_{3/4,5/4} \left(1/2 \frac{v_l^2}{a^2} \right) \left(\frac{v_u^2}{a^2} \right)^{3/4} \right). \tag{40}$$

The variance $\sigma_t^2(a, v_l, v_u)$ is defined as

$$\sigma_t^2(a, v_l, v_u) = \mu'_{2,t}(a, v_l, v_u) - (\mu'_{1,t}(a, v_l, v_u))^2 \tag{41}$$

and has the following explicit form

$$\begin{aligned} &\sigma_t^2(a, v_l, v_u) \\ &= 4 \left(\left((v_l + 2v_u)a^2 + v_l v_u \left(v_l + \frac{1}{2} v_u \right) \right) \left(a^2 + 1/2 v_u^2 \right) C^2 a^4 e^{-\frac{1}{2} \frac{v_l^2 + 2v_u^2}{a^2}} \right. \\ &\quad - 2 \left(\left(v_l + \frac{1}{2} v_u \right) a^2 + \frac{1}{4} v_l v_u (v_l + 2v_u) \right) C^2 a^4 \left(a^2 + \frac{1}{2} v_l^2 \right) e^{-\frac{1}{2} \frac{2v_l^2 + v_u^2}{a^2}} \\ &\quad + \left(a^2 + \frac{1}{2} v_u^2 \right) \left(\operatorname{Cerf} \left(\frac{1}{2} \frac{\sqrt{2} v_l}{a} \right) a^3 \sqrt{2} \sqrt{\pi} \right. \\ &\quad \left. - \operatorname{Cerf} \left(\frac{1}{2} \frac{\sqrt{2} v_u}{a} \right) a^3 \sqrt{2} \sqrt{\pi} + 4 \right) C a^2 \left(a^2 + \frac{1}{2} v_l^2 \right) e^{-\frac{1}{2} \frac{v_l^2 + v_u^2}{a^2}} \\ &\quad + C^2 a^4 \left(a^2 + \frac{1}{2} v_l^2 \right)^2 v_l e^{-\frac{3}{2} \frac{v_l^2}{a^2}} - \left(a^2 + \frac{1}{2} v_u^2 \right)^2 C^2 a^4 v_u e^{-\frac{3}{2} \frac{v_u^2}{a^2}} \\ &\quad + \left(\frac{3}{4} a^2 v_l + \frac{1}{4} v_l^3 \right) e^{-\frac{1}{2} \frac{v_l^2}{a^2}} + \left(-\frac{3}{4} a^2 v_u - \frac{1}{4} v_u^3 \right) e^{-\frac{1}{2} \frac{v_u^2}{a^2}} \\ &\quad - \frac{1}{2} \left(\left(\operatorname{Cerf} \left(\frac{1}{2} \frac{\sqrt{2} v_l}{a} \right) a^3 \sqrt{2} \sqrt{\pi} \right. \right. \\ &\quad \left. \left. - \operatorname{Cerf} \left(\frac{1}{2} \frac{\sqrt{2} v_u}{a} \right) a^3 \sqrt{2} \sqrt{\pi} + 4 \right) C \left(a^2 + \frac{1}{2} v_l^2 \right)^2 e^{-\frac{v_l^2}{a^2}} \right. \\ &\quad \left. + \left(a^2 + \frac{1}{2} v_u^2 \right)^2 \left(\operatorname{Cerf} \left(\frac{1}{2} \frac{\sqrt{2} v_l}{a} \right) a^3 \sqrt{2} \sqrt{\pi} \right. \right. \\ &\quad \left. \left. - \operatorname{Cerf} \left(\frac{1}{2} \frac{\sqrt{2} v_u}{a} \right) a^3 \sqrt{2} \sqrt{\pi} + 4 \right) C e^{-\frac{v_u^2}{a^2}} \right. \\ &\quad \left. + \frac{3}{4} \sqrt{\pi} \sqrt{2} \left(-\operatorname{erf} \left(\frac{1}{2} \frac{\sqrt{2} v_u}{a} \right) + \operatorname{erf} \left(\frac{1}{2} \frac{\sqrt{2} v_l}{a} \right) \right) a \right) C a^2. \tag{42} \end{aligned}$$

Although the coefficients of skewness and kurtosis for the truncated MB exist, they have a complicated expression.

4. A Laboratory Application

The temperature as a function of root-mean-square speed for the MB is given by Equation (15). In the truncated MB distribution, the temperature can be found by solving the following nonlinear equation

$$v_{rms,t}(k, m, T, v_l, v_u) = v_{rms,m}, \quad (43)$$

where $v_{rms,m}$ is not a theoretical variable but is the root-mean-square speed measured in the laboratory and $v_{rms,t}$ is given by Equation (39). The laboratory measures of $v_{rms,m}$ started with [13], where a $v_{rms,m} = 388$ m/s at 400°C was found for a metallic vapor. In the truncated MB distribution, there are three parameters that can be measured in the laboratory from a kinematical point of view, as follows: the lowest velocity, v_l ; the highest velocity, v_u ; and the root-mean-square speed, $v_{rms,m}$. Setting for simplicity $v_l = 0$, we will now explore the effect of the variation of v_u on the root-mean-square speed; see **Figure 1**. The *first* example of the influence of the upper limit in velocity on the temperature is given by potassium gas [14] [15], in which molecular mass is $6.492429890 \times 10^{-26}$ kg. In **Figure 2**, we evaluate in a numerical way the temperature when $v_l = 0$ and v_u is variable in the case of a measured value of $v_{rms,m}$.

The *second* example is given by diatomic nitrogen, N_2 , in which molecular mass is $4.651737684 \times 10^{-26}$ kg. In **Figure 3**, we evaluate the temperature when $v_l = 0$ and v_u is a variable in the case of a measured value of $v_{rms,m}$.

5. The Jeans Escape

The standard formula for the escape of molecules from the exosphere is reviewed in the framework of the MB distribution. A new formula for the Jeans escape is derived in the framework of the truncated MB.

5.1. The Standard Case

In the exosphere, a molecule of mass m and velocity v_e is free to escape when

$$\frac{1}{2}mv_e^2 - G\frac{Mm}{R_{ex}} = 0, \quad (44)$$

where G is the Newtonian gravitational constant, M is the mass of the Earth, $R_{ex} = R + H$ is the radius of the exosphere, R is the radius of the Earth and H is the altitude of the exosphere. The flux of the molecules that are living in the exosphere Φ_j is

$$\Phi_j = \frac{1}{4}N_{ex}\mu_e, \quad (45)$$

where N_{ex} is the number of molecules per unit volume and μ_e is the average velocity of escape. In the presence of a given number of molecules per unit volume, the standard MB distribution in velocities in a unit volume, f_m , is

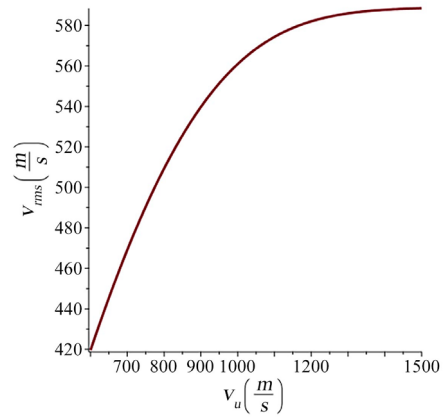


Figure 1. The theoretical root-mean-square speed as a function of the upper limit in velocity (continuous line) and standard value of the temperature (dotted line) when $a = 340$ and $v_i = 0$.

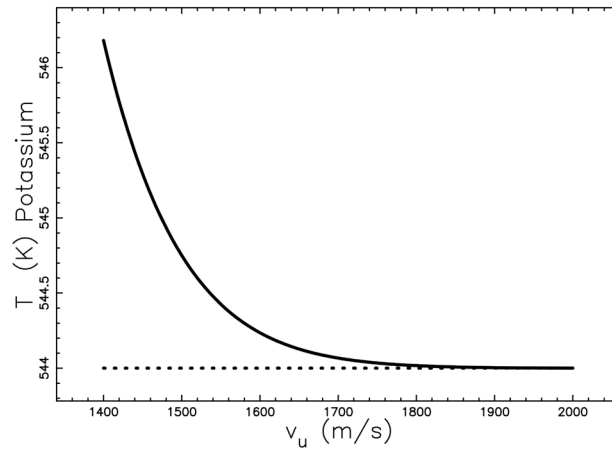


Figure 2. Temperature as a function of the upper limit in velocity for Potassium (continuous line) and standard value of the temperature (dotted line) when $v_i = 0$ and $v_{rms,m} = 589.111511$ m/s.

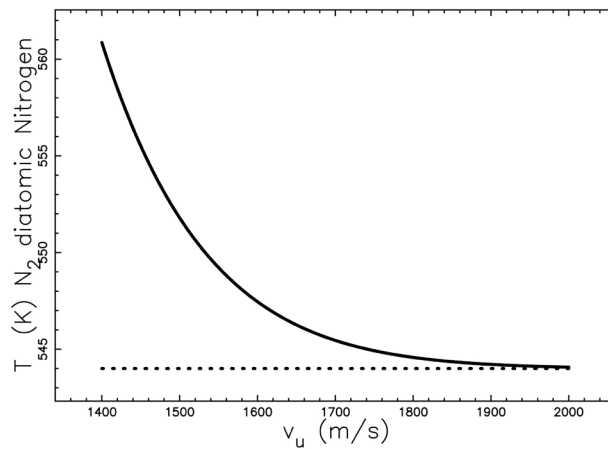


Figure 3. Temperature as a function of the upper limit in velocity for diatomic nitrogen, N_2 , (continuous line) and standard value of the temperature (dotted line) when $v_i = 0$ and $v_{rms,m} = 695.9756308$ m/s.

$$f_m(v; m, k, T, N_{ex}) = N_{ex} \frac{\sqrt{2} v^2 e^{-\frac{1}{2} \frac{v^2 m}{kT}}}{\sqrt{\pi} \left(\frac{kT}{m}\right)^{\frac{3}{2}}}. \quad (46)$$

The average value of escape is defined as

$$\mu_e = \frac{\int_{v_e}^{\infty} v f_m(v; m, k, T, N_{ex}) dv}{\int_0^{\infty} f_m(v; m, k, T, N_{ex}) dv}. \quad (47)$$

In this integral, the following changes are made to the variables

$$\lambda = \frac{1}{2} \frac{mv^2}{kT}. \quad (48)$$

Therefore,

$$\mu_e = 2(\lambda_e + 1) e^{-\lambda_e} \sqrt{2} \sqrt{\frac{kT}{\pi m}}, \quad (49)$$

with

$$\lambda_e = 2 \frac{GM}{R_{ex} v_0^2}, \quad (50)$$

where v_0 is the mode as represented by Equation (21). The flux is now

$$\Phi_j = \frac{N_{ex} (\lambda_e + 1) e^{-\lambda_e} v_0}{2\sqrt{\pi}}. \quad (51)$$

For more details see [16] [17] [18] [19]. On adopting the parameters of **Table 1** the Jeans escape flux for hydrogen is

$$\Phi_j = 3.98 \times 10^{11} \text{ molecules} \cdot \text{m}^{-2} \cdot \text{s}^{-1}, \quad (52)$$

and

$$\lambda_e = 7.78. \quad (53)$$

The Jeans escape flux for Earth at $T = 900$ K varies between $\Phi_j \approx 2.7 \times 10^{11} \text{ molecules} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$; see [20] or **Figure 1** in [21]. and $\Phi_j \approx 4 \times 10^{11} \text{ molecules} \cdot \text{m}^{-2} \cdot \text{s}^{-1}$, see [22]. Therefore, our choice of parameters is compatible with the suggested interval in flux.

5.2. The Truncated Case

The average value of escape for a truncated MB distribution, $\mu_{e,t}$, is

$$\mu_{e,t} = \frac{\int_{v_e}^{\infty} v f_t(v; m, k, T, N_{ex}, v_l, v_u) dv}{\int_0^{\infty} f_m(v; m, k, T, N_{ex}, v_l, v_u) dv}. \quad (54)$$

This integral can be solved by introducing the change of variable as given by Equation (48)

$$\mu_{e,t} = -2 \frac{((\lambda_u + 1) e^{-\lambda_u} - e^{-\lambda_e} (\lambda_e + 1)) \sqrt{2}}{2\sqrt{\lambda_l} e^{-\lambda_l} - 2\sqrt{\lambda_u} e^{-\lambda_u} - \sqrt{\pi} \operatorname{erf}(\sqrt{\lambda_l}) + \sqrt{\pi} \operatorname{erf}(\sqrt{\lambda_u})} \sqrt{\frac{kT}{m}}, \quad (55)$$

where λ_l is the lower value of λ and λ_u is the upper value of λ . The flux

of the molecules that are living the exosphere in the truncated MB, $\Phi_{j,t}$, is

$$\Phi_{j,t} = \frac{N_{ex} \left((\lambda_u + 1)e^{-\lambda_u} - e^{-\lambda_e} (\lambda_e + 1) \right) \sqrt{2}}{4\sqrt{\lambda_u} e^{-\lambda_u} + 2\sqrt{\pi} \operatorname{erf}(\sqrt{\lambda_l}) - 2\sqrt{\pi} \operatorname{erf}(\sqrt{\lambda_u}) - 4\sqrt{\lambda_l} e^{-\lambda_l}} \sqrt{\frac{kT}{m}}. \quad (56)$$

The increasing flux of molecules is outlined when one parameter, λ_l , is variable; see **Figure 4**. In other words, an increase in λ_l produces an increase in the flux of the molecules. The dependence of the flux when two parameters are variable, λ_l and λ_u , is reported in **Figure 5**.

Table 1. Adopted physical parameters for the exosphere.

Parameter	Value
R_{ex}	6900 km
T	900 K
N_{ex}	10^{11} m^{-3}

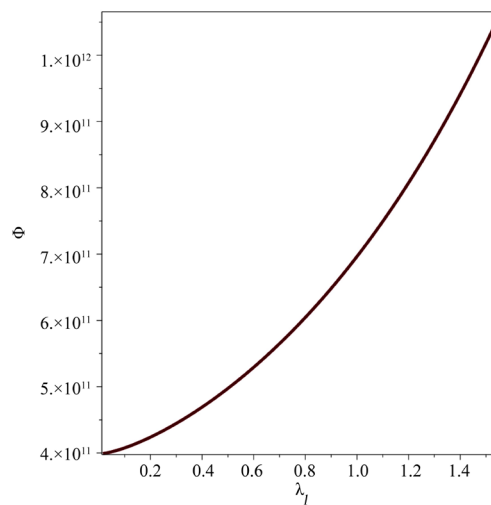


Figure 4. The flux of molecules as a function of λ_l with parameters as in **Table 1**, $\lambda_e = 7.78$ and $\lambda_u = 1000\lambda_e$.

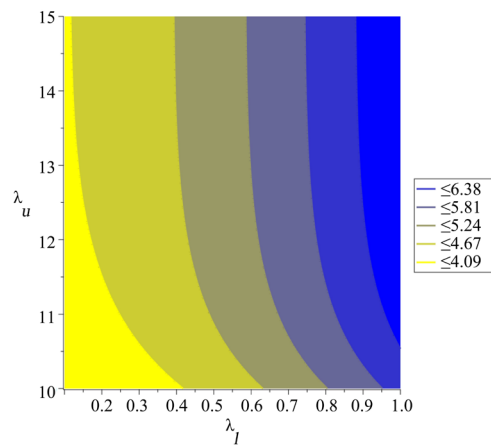


Figure 5. The flux of molecules as a function of λ_l and λ_u with parameters as in **Table 1**.

These Jeans escape fluxes for Earth are compatible with the observed values that were reported in Section 5.1.

6. Conclusion

This paper derived analytical formulae for the following quantities for a double truncated MB distribution: the PDF, the DF, the average value, the r th moment about the origin, the root-mean-square speed and the variance. The traditional correspondence between root-mean-square speed and temperature is replaced by the nonlinear Equation (43). The new formula (56) for the Jeans escape flux of molecules from an atmosphere is now a function of the lower and upper boundary in velocity.

Conflicts of Interest

The author declares no conflicts of interest regarding the publication of this paper.

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