

# Synthesis and Biological Evaluation of Cyclic Imides with Coumarins and Azacoumarins

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Received August 1, 2012; revised September 3, 2012; accepted September 17, 2012

## ABSTRACT

A series of new N-substituted cyclic imides of coumarins and azacoumarins have been synthesized. All the compounds were screened for their antibacterial and antifungal activities.  $\alpha$ -Amylase enzyme inhibition activities of some N-substituted phthalimide derivatives have been explored. All the compounds were characterized by IR, <sup>1</sup>H-NMR, and GC-MS.

**Keywords:**  $\alpha$ -Amylase Enzyme; Coumarin and 1-Azacoumarin

## 1. Introduction

Exploration of new chemical entities (NCE) for biological applications has become an important area of research in the field of drug discovery [1,2]. Heterocyclic frameworks are present in majority of the clinically accepted drug molecules. An important feature of heterocycles is their ability to sustain enormous structural diversity which is very much necessary to establish lead compounds for a variety of pharmacological activities [3]. Nitrogen heterocycles have been recognized as privileged structures to bind the receptor sites of macromolecules, leading to the perturbation of their metabolic functions which is a characteristic feature of biologically active molecules [4]. Cyclic imides constitute an important class of compounds possessing bis-amide linkages with common nitrogen. They have been useful building blocks in the synthesis of natural products [5] and other heterocycles [6] as well. Their ability to cross biological membranes *in-vivo*, due to their hydrophobicity is well established [7]. In view of the favorable pharmacokinetic properties, derivatives of cyclic imides have been found to be exhibiting wide ranging biological properties (**Figure 1**) [8-10]. Cyclic imides with a *para*-sulphonamide group have been potential anti tubercular agents [11], incorporation of heterocyclic moiety in the imide portion or linkage with nitrogen has resulted in molecular matrices with potential anti inflammatory [12] and anti tumor activities [13].

Bis-amide linkage have been employed as core structures in the design and synthesis of peptidomimetics

which inhibit amino transferases [14,15]. In view of their biological importance, cyclic imides have been synthesized from diols [16], under microwave irradiation [17] and other routes [18]. Recent study demonstrated that chamomile plant extract suppresses the growth of human cancer cells and inhibits the pancreatic  $\alpha$ -amylase enzyme inhibitor, major components of *M. chamomilla* belongs to three different chemical classes: sesquiterpenes, coumarins, and flavonoids [19]. Trans-chalcone, a biphenolic core structure of flavonoids precursor exhibited  $\alpha$ -amylase inhibitory activity [20]. Survey of literature indicated that succinimide linked coumarins at the allylic position with respect to the C3 - C4 double bond have been employed in imaging of protein kinase activity in living cells [21]. In view of the diverse biological properties associated with coumarins and 1-aza coumarins (carbostyrils) [22], it was thought of considerable interest to synthesize cyclic imides linked to these moieties. In the light of these observations the present paper reports the synthesis of cyclic imides with general structure with three points of structural diversity in **Figure 2**.

$\alpha$ -Amylase, a salivary or pancreatic enzyme plays an important role in early breakdown of starch into glucose and maltose. Modulation of  $\alpha$ -amylase activity affects the utilization of carbohydrates as an energy source and stronger is this modulation, more significant is the reduction in the breakdown of complex carbohydrates. Some of the N-substituted phthalimide derivatives of coumarins and 1-azacoumarins have been subjected to  $\alpha$ -amylase enzyme inhibition activity. Retro synthetic analysis for the target compounds (**Figure 2**) indicates that crucial C-N bond forming step can be achieved using

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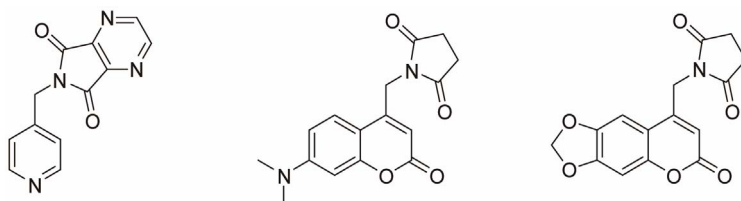
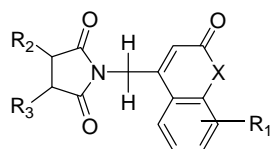


Figure 1. Structurally related biologically active compounds.



X= O, R<sub>1</sub>=6-CH<sub>3</sub>, 7-CH<sub>3</sub>, 7,8-diMe, 7-Cl, 6-OCH<sub>3</sub> R<sub>2</sub>=R<sub>3</sub>= H/1,2-Benzo

X=NH, R<sub>1</sub>=H, 7Cl, 5,8-diMe, 8-Me R<sub>2</sub>=R<sub>3</sub>= H/1,2-Benzo

Figure 2. Cyclic imides with three points of structural diversity.

either aminomethyl compounds and the corresponding dicarboxylic acids/anhydrides by an amidation/ring insertion reaction or by using 4-halomethyl heterocycles and cyclic imides by a nucleophilic displacement at the allylic carbon. We have demonstrated the leaving group ability of 4-bromomethyl coumarins and 1-aza-4-bromomethyl coumarins using a variety of O, N, S, nucleophiles leading to biologically active compounds [23-28]. In view of this we have employed the later route involving halomethyl heterocycles and cyclic imides.

## 2. Materials and Methods

**Experimental:** The melting points of the products were determined by open Capillary on Buchi apparatus. The IR spectra were recorded on a Nicole Impact-410 FT-I Spectrophotometer, using KBr pallets. <sup>1</sup>HNMR spectra were recorded on a Bruker-300 F 300 MHz spectrometer in DMSO using TMS as an internal standard. Mass data were obtained from Shimadzu GC-MS. All the reagents were of analytical reagent grade and chemically pure. Solvents were distilled and dried before use.

**Antibacterial activity:** Agar diffusion method was employed [29] for inhibition activity against *E. coli* and *Staphylococcus aureus* bacteria, agar media was prepared by adding 10 g of peptone, 10 g of NaCl, 5 g of Yeast extract and 20 g of Agar into 1000 mL of distilled water. Initially the stock culture bacteria were revived and wells were made in the plate. Each plate was inoculated with 18 h old cultures and spread evenly on the plate. After 20 min, the wells were filled with compound at different concentrations. The control wells of *Gentamycin* were also prepared. All the plates were incubated at 37°C for 24 h and the diameter of inhibition zone was noted **Table 1**.

**Antifungal Activity:** Potato Dextrose Agar (PDA) media

was employed [30] and was prepared by adding 250 g of boiled and peeled potato squeezed and filtered. To this filtrate 20 g of dextrose was added and the volume was made up to 1000 ml by distilled water. Initially, the stock cultures of *Aspergillus niger* and *Cladosporium* were revived by inoculating in broth media grown at 27°C for 48 h. The agar plates of the above media were prepared and wells were made in the plate. Each plate was inoculated with 18 h old cultures and spread evenly on the plate. After 20 min, the wells were filled with compounds of different concentrations (100 µg/mL, 200 µg/mL and 300 µg/mL). The control plates of *Amphotericin* were also prepared. All the plates were incubated at 27°C for 48 h and the diameter of inhibition zone was noted and shown in **Table 2**.

***α*-Amylase inhibitory activity:** A stock sample concentration of 10 mg/mL concentration was prepared by using DMSO solvent. Reagent solution without test samples was used as control, activity of amylase was assayed with different concentrations of sample with control, DNS solution was prepared by dissolving 1 g of dinitro salicylic acid in 2N NaOH solution, 30 g of sodium potassium tartrate was added to the above solution and whole volume was diluted to 100 mL. The test tubes with different concentrations of sample solutions were prepared and 1ml of sodium phosphate buffer (50 mM. pH 7.0 - pH 7.3), 0.5 mL of starch, 0.1 mL of amylase enzyme (1 mg per mL sample in buffer) were added to each test tubes. The test tubes were incubated at 37°C for 10 min and 1 mL of DNS solution was added to each test tubes and the absorbance was measured at 540 nm against blank. Concentration of maltose liberated was determined by using standard maltose curve [31]. Enzyme activity was calculated and percentage of inhibition shown in the **Table 3**.

**Table 1. Anti-bacterial activity of N-substituted cyclic imide derivatives.**

S. No	Compound	<i>E. coli</i> (Gram – ve ) Inhibition zones in cm (%)			<i>S. aureus</i> (Gram + ve) Inhibition zones in cm (%)		
		300 µg	200 µg	100 µg	300 µg	200 µg	100 µg
1	2a	0.35 (13)	0.15 (7)	-	0.40 (17)	0.26 (12)	0.21 (12)
2	2b	0.65 (25)	0.45 (20)	0.28 (13)	0.67 (29)	0.45 (21)	-
3	2c	0.44 (17)	0.15 (7)	-	0.36 (16)	0.24 (11)	-
4	2d	0.94 (36)	0.75 (33)	0.30 (14)	0.82 (36)	0.61 (29)	0.35 (19)
5	2e	1.20 (46)	0.88 (38)	0.35 (17)	0.65 (28)	0.48 (23)	0.36 (20)
6	2f	0.61 (23)	0.56 (24)	0.25 (12)	0.71 (31)	0.45 (21)	-
7	2g	0.74 (28)	0.66 (29)	0.35 (17)	0.78 (34)	0.57 (27)	0.24 (13)
8	2h	0.34 (13)	-	-	0.27 (12)	0.21 (10)	-
9	2i	0.42 (16)	0.25 (11)	-	0.78 (34)	0.45 (21)	0.33 (18)
10	3a	0.51 (20)	0.32 (14)	0.20 (10)	0.50 (22)	0.35 (17)	-
11	3b	0.67 (26)	0.45 (20)	0.31 (15)	0.34 (15)	0.30 (14)	0.24 (13)
12	3c	0.45 (17)	0.22 (10)	-	0.41 (18)	0.25 (12)	-
13	3d	1.44 (55)	0.95 (41)	0.53 (25)	1.30 (57)	0.95 (45)	0.54 (30)
14	3e	0.95 (37)	0.65 (28)	0.40 (19)	1.11 (48)	0.88 (42)	0.51 (28)
15	3f	0.95 (37)	0.60 (26)	0.38 (18)	0.92 (40)	0.85 (40)	0.42 (23)
16	3g	1.13 (43)	0.74 (32)	0.61 (29)	1.25 (54)	0.97 (46)	0.57 (32)
17	3h	0.30 (12)	0.15 (7)	-	0.46 (20)	0.30 (14)	0.22 (12)
18	3i	0.45 (20)	0.20 (9)	-	0.71 (31)	0.41 (20)	-
19	<i>Gentamycin</i>	2.6	2.3	2.1	2.3	2.1	1.8

Control: DMSO.

**Table 2. Anti-fungal activity of cyclic imide derivatives.**

S. No	Compound	<i>Asperigillus niger</i> Zone of Inhibition in cm (%)			<i>Cladosporium</i> Zone of Inhibition in cm (%)		
		300 µg	200 µg	100 µg	300 µg	200 µg	100 µg
1	2a	0.21(16)	-	-	0.15 (14)	0.10 (10)	-
2	2b	0.28 (22)	0.20 (21)	-	0.22 (20)	0.18 (19)	-
3	2c	0.44 (34)	0.31 (33)	0.19 (24)	0.18 (16)	-	-
4	2d	0.57 (44)	0.40 (43)	0.23 (29)	0.48 (43)	0.30 (32)	0.20 (29)
5	2e	0.64 (49)	0.44 (47)	0.37 (46)	0.39 (35)	0.22 (23)	0.14 (20)
6	2f	0.30 (23)	0.13 (14)	-	0.25 (23)	0.15 (16)	-
7	2g	0.42 (32)	0.30 (32)	0.23(29)	0.52 (47)	0.35 (37)	0.22 (31)
8	2h	0.35 (27)	0.25 (27)	-	0.28 (25)	0.15 (16)	-
9	2i	0.48 (37)	0.30 (32)	-	0.35 (32)	0.16 (17)	0.10 (14)
10	3a	0.41 (32)	0.31 (28)	-	0.25 (23)	0.15 (16)	-
11	3b	0.36 (28)	0.27 (29)	0.21 (26)	0.34 (31)	0.22 (23)	0.10 (14)
12	3c	0.24 (18)	0.16 (17)	-	0.23 (21)	0.16 (17)	-
13	3d	0.71 (55)	0.45 (48)	0.32 (40)	0.42 (38)	0.28 (29)	0.15 (21)
14	3e	0.47 (36)	0.31 (33)	0.25 (31)	0.38 (35)	0.26 (27)	0.23 (33)
15	3f	0.36 (28)	0.24 (25)	0.15 (18)	0.36 (33)	0.18 (19)	0.10 (14)
16	3g	0.45 (35)	0.27 (29)	0.17 (21)	0.48 (44)	0.30 (32)	0.21 (30)
17	3h	0.32 (25)	0.24 (26)	-	0.28 (25)	0.15 (16)	-
18	3i	0.37 (28)	-	-	0.30 (27)	0.22 (23)	0.15 (21)
19	<i>Amphotericin</i>	1.30	0.94	0.80	1.1	0.95	0.70

Control: DMSO.

### 3. Results and Discussion

A brief synthetic scheme employed during the present work for the title compounds is depicted in scheme 1. The required 4-bromomethyl coumarins **1** (X=O) were prepared by Pechmann cyclization using various Phenols and 4-bromoethyl acetoacetate [32]. Reaction of succinimide using acetone and activated anhydrous potassium carbonate under reflux conditions resulted in cyclic imides **2a-e** (X=O). Formation of compounds **2a-e** was confirmed by <sup>1</sup>H-NMR. The methylene protons in **1** (X=O, R=6-CH<sub>3</sub>) appeared as a singlet at 4.4 ppm where as in cyclic imides **2a**, the methylene protons were deshielded and resonated at 5.0 ppm. The 1-aza-4-bromomethylcoumarin was prepared by the bromination of acetoacetanilide and cyclising the resulting *o*-bromoacetoacetanilide using sulphuric acid [33]. Under identical conditions cyclic imides **2f-i** (X=NH) were obtained. Under these optimized conditions of a number of cyclic imides **2a-i** were obtained. Finally the ORTEP diagram of the compound **2c** (X=O, R=7,8-diMe) obtained by diffraction studies [34] has confirmed the structures of the products obtained in this reaction. The present methodology was extended for the reaction of phthalimide with various 4-bromomethyl coumarins and azacoumarins to obtain compounds **3a-i** in good yields. The various N-substituted cyclic imides of coumarins and azacoumarins have been shown in **Scheme 1**.

### 4. Antibacterial Activity

Among the compounds **3a-i**, 2-((7-chloro-2-oxo-2H-chromen-4-yl) methyl) isoindoline-1,3-dione (**3d**) has exhibited 41% inhibition and 55% inhibition against *E. coli* at concentration of 200 µg/mL and 300 µg/mL respectively, It was also found to be active against *S. aureus* with inhibition of 45% and 57% at above concentration, other compounds have shown moderate activity against both bacteria. 2-((7-chloro-2-oxo-1,2-dihydroquinolin-4-yl) methyl) isoindoline-1,3-dione (**3g**) was found to be good antibacterial agent against *S. aureus* with inhibition activity of 46% and 54% at 200 µg/mL and 300 µg/mL concentrations, rest of the compounds exhibited moder-

ate antibacterial activity. N-substituted succinimide derivatives of (**2a-e**) have not shown significant antibacterial activity except 1-((6-methoxy-2-oxo-2H-chromen-4-yl) methyl) pyrrolidine-2,5-dione (**2e**). All other compounds of this series have shown moderate antibacterial activity.

### 5. Antifungal Activity

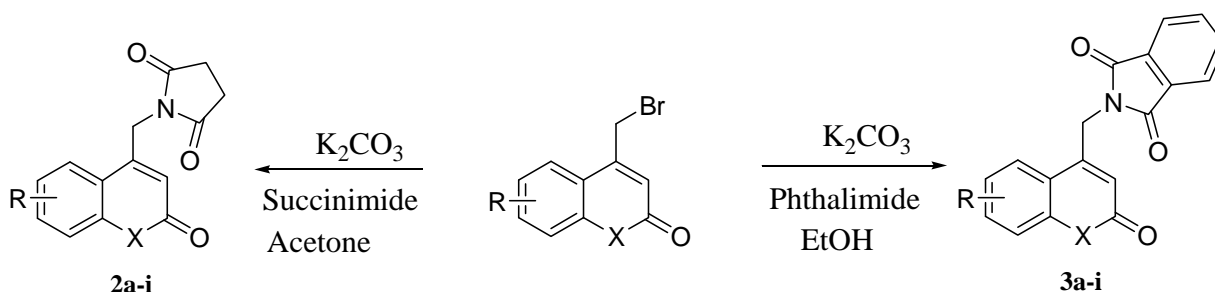
Among the series **2a-i** both the compounds **2d** and **2e** exhibited significant inhibition activity against *A. niger* at 200 µg/mL and 300 µg/mL. Compound **2g** exhibited 47% inhibition against *Cladosporium* at 300 µg/mL concentration, rest of the compounds have shown moderate activity. Apart from **3a-i** series, the compound **3d** has shown remarkable inhibition against *A. niger* with 48% and 55% inhibition at 200 µg/mL and 300 µg/mL concentration respectively. All other compounds have exhibited moderate activity.

#### 5.1. $\alpha$ -Amylase Inhibitory Activity

Among the N-substituted phthalimide derivatives of azacoumarin compounds, **3f** and **3h** exhibited amylase enzyme inhibition activity at different concentration but **3g** and **3i** enacted as supporters of hydrolyzing activity of  $\alpha$ -amylase enzyme, therefore, the concentration of maltose was found to be more than that of the control. All the N-substituted phthalimides of coumarins were found to exhibit  $\alpha$ -amylase enzyme inhibitory activity except compound **3e**.

The relative enzyme inhibition activity of various N-substituted phthalimides, at different concentrations has been shown in **Figure 3**. The decline in enzyme activity at different concentrations of **3d** is compared with control and relative inhibition activity is shown in **Figure 4**.

Remarkable amylase inhibitory activity was observed by 2-((7-chloro-2-oxo-2H-chromen-4-yl)methyl) isoindoline-1,3-dione (**3d**) at 10 µg, suppressed the enzyme activity by 6.15%, at 50 µg by 10.99% and at 100 µg the enzymatic activity was declined by 13.41%. Various ions have also been shown to affect the activity of the



**Scheme 1.** Synthesis of N-substituted cyclic imides **2a-i** & **3a-i**. X=O, R=6-CH<sub>3</sub>, 7-CH<sub>3</sub>, 7,8-diMe, 7-Cl, 6-OMe, X=NH, R=H,7-Cl,5,8-diMe,8-Me.

**Table 3.  $\alpha$ -amylase inhibitory activity of some of the N-substituted phthalimide derivatives.**

Sample	Quantity	OD at 540 nm	Concentration of Maltose liberated ( $\mu\text{g}$ )	Activity ( $\mu\text{moles/ml/min}$ )	% Activity
3a	100 $\mu\text{g}$	1.98	160	0.0441	97.10
	200 $\mu\text{g}$	1.91	146	0.0405	89.01
	500 $\mu\text{g}$	1.74	139	0.0385	84.62
3b	100 $\mu\text{g}$	1.98	159	0.0422	97.30
	200 $\mu\text{g}$	1.89	153	0.0424	93.19
	500 $\mu\text{g}$	1.79	143	0.0396	87.03
3c	100 $\mu\text{g}$	1.99	160	0.0444	97.58
	200 $\mu\text{g}$	1.92	166	0.0460	94.11
	500 $\mu\text{g}$	1.83	144	0.0399	87.69
3e	100 $\mu\text{g}$	2.10	168	0.0466	102.42
	200 $\mu\text{g}$	1.98	173	0.0480	105.49
	500 $\mu\text{g}$	1.94	169	0.0469	103.08
3f	100 $\mu\text{g}$	1.99	160	0.0444	97.58
	200 $\mu\text{g}$	1.92	154	0.0427	93.85
	500 $\mu\text{g}$	1.89	151	0.0419	92.09
3g	100 $\mu\text{g}$	1.99	160	0.0444	97.58
	200 $\mu\text{g}$	2.10	168	0.0466	102.42
	500 $\mu\text{g}$	1.96	171	0.0474	104.18
3h	100 $\mu\text{g}$	1.94	157	0.0437	96.05
	200 $\mu\text{g}$	1.90	152	0.0421	92.53
	500 $\mu\text{g}$	1.86	149	0.0413	90.77
3i	100 $\mu\text{g}$	2.10	168	0.0466	102.42
	200 $\mu\text{g}$	2.10	168	0.0466	102.42
	500 $\mu\text{g}$	1.96	171	0.0474	104.18
3d	100 $\mu\text{g}$	1.81	142	0.0394	86.59
	50 $\mu\text{g}$	1.86	146	0.0405	89.01
	10 $\mu\text{g}$	1.91	154	0.0427	93.85
Control	-	2.04	164	0.0455	100.00

Solvent used: DMSO.

inhibitor. Lajolo *et al.* [35] reported the increase in the activity of the inhibitor against salivary amylase mediated by ions in the order Nitrate > Chloride > Bromide > Iodide > Thiocyanate. Gibbs and Alli reported that chloride ions are important for maximum activity while Ca ions increase the rate of initial binding of the inhibitor to the amylase. They also reported that K, Mg, sulfate and Na ions did not have any effects on the amylase inhibitor activity [36].

**The starch-blocking mechanism of  $\alpha$  amylase inhibitor:**  $\alpha$ -amylase inhibitor action shows that the inhibitor is effective in preventing starch digestion by blocking access to the active site of the enzyme. During inhibition, several components of the inhibitor molecule, amylase molecule and the whole system have been reported to play important role in the mechanism. The chloride binding active sites of non-loop residues are Asp 197, Glu 233, Asp 300 and Arg 74 in amylase [35], This might be the reason that 2-((7-chloro-2-oxo-2H-chro-

men-4-yl) methyl) isoindoline-1,3-dione (**3d**) was found to be the most competent as amylase enzyme inhibitor even at 10  $\mu\text{g/mL}$  concentration.

**General procedure for the preparation of compounds 2a-i and 3a-i:** A mixture of substituted 4 bromomethyl coumarins/1-azacoumarins 1 (4 mmol), succinimide/phthalimide (4 mmol) and powdered anhydrous potassium carbonate (4 mmol) in dry acetone/absolute alcohol (25 mL) was refluxed for 6 h, completion of the reaction was monitored by TLC. After the completion of the reaction, the separated solid was filtered and washed with 10% HCl and water, dried and recrystallized from suitable solvent.

**1-((6-methyl-2-oxo-2H-chromen-4-yl)methyl)pyrrolidine-2, 5-dione (2a):** White solid from acetic acid, yield 67%, m.p. 185°C - 187°C. Anal. Calc. for  $\text{C}_{15}\text{H}_{13}\text{NO}_4$  (271.27): C, 66.41; H, 4.83; N, 5.16; Found: C, 66.29; H, 4.79; N, 5.10. IR ( $\nu/\text{cm}^{-1}$ ) = 1716  $\text{cm}^{-1}$  (CO).  $^1\text{H}$  NMR ( $\delta$  ppm) 2.38 (s, 3H, C6-CH<sub>3</sub>), 2.79 (s, 2H, C14-H), 2.80(s,

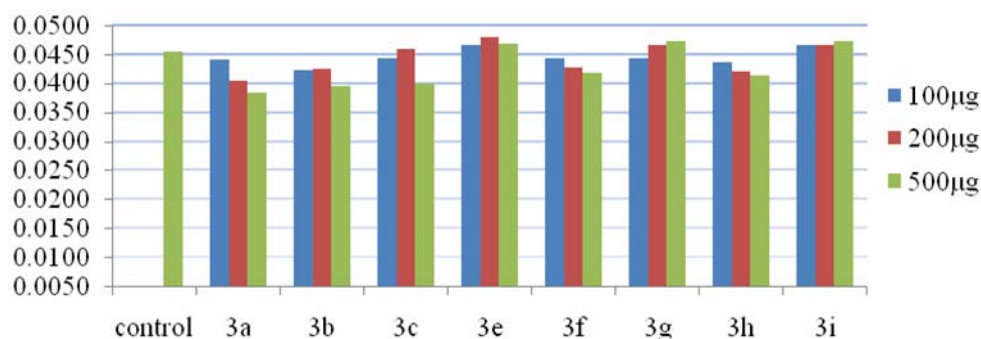


Figure 3. Graph showing the comparative analysis of samples with control.

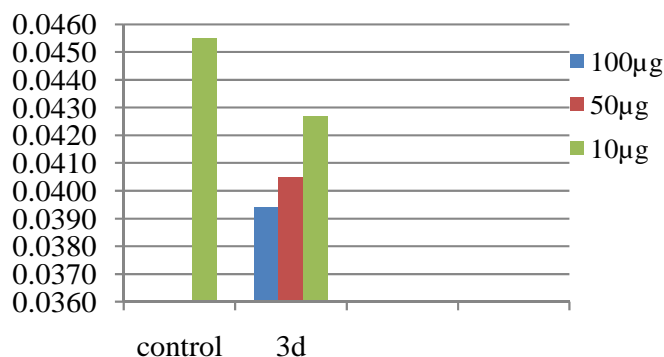


Figure 4. Graph showing the comparative activity of control and sample 3d.

2H, C15-6H), 4.88 (*s*, 2H, C4-CH<sub>2</sub>), 6.32 (*s*, 1H, C3-H), 7.27 (*d*, 1H, C7-H), 7.41 (*d*, 1H, C8-H), 7.56 (*s*, 1H, C5-H); GC-MS: *m/z* 271 (M<sup>+</sup>, 100%).

**1-((7-methyl-2-oxo-2H-chromen-4-yl)methyl)pyrrolidine-2,5-dione (2b):** White solid from acetic acid, yield 72%, m.p. 192°C - 194°C. Anal. Calc. for C<sub>15</sub>H<sub>13</sub>NO<sub>4</sub> (271.27): C, 66.41; H, 4.83; N, 5.16; Found: C, 66.31; H, 4.80; N, 5.11. IR (*v/cm*<sup>-1</sup>) = 1716 cm<sup>-1</sup> (CO). <sup>1</sup>H NMR (*δ* ppm): 2.42 (*s*, 3H, C7-CH<sub>3</sub>), 2.79 (*s*, 2H, C14-H, CH<sub>2</sub>), 2.80 (*s*, 2H, C15-CH<sub>2</sub>), 4.78 (*s*, 2H, C4-CH<sub>2</sub>), 6.34 (*s*, 1H, C3-H), 7.35 (*s*, 1H, C8-H), 7.41 (*d*, 1H, C6-H), 7.71 (*d*, 1H, C5-H); GC-MS: *m/z* 271 (M<sup>+</sup>, 100%).

**1-((7,8-dimethyl-2-oxo-2H-chromen-4-yl)methyl)pyrrolidine-2,5-dione (2c):** White crystals from acetic acid yield 68%, m.p. 212°C. Anal. Calc. for C<sub>16</sub>H<sub>15</sub>NO<sub>4</sub> (285.29): C, 67.36; H, 5.30; N, 4.91; Found: C, 67.28; H, 5.26; N, 4.89. IR (*v/cm*<sup>-1</sup>) = 1718 cm<sup>-1</sup> (CO). <sup>1</sup>H NMR (*δ* ppm): 2.29 (*s*, 3H, C7-CH<sub>3</sub>), 2.38 (*s*, 3H, C8-CH<sub>3</sub>), 2.79 (*s*, 2H, C14-H), 2.80 (*s*, 2H, C15-H), 4.75 (*s*, 2H, C4-CH<sub>2</sub>), 6.24 (*s*, 1H, C3-H), 7.25 (*d*, 1H, C6-H), 7.67 (*d*, 1H, C5-H); GC-MS: *m/z* 285.29 (M<sup>+</sup>, 100%).

**1-((7-chloro-2-oxo-2H-chromen-4-yl)methyl)pyrrolidine-2,5-dione (2d):** White solid from ethanol and DMF, yield 54%, m.p. 223°C - 224°C. Anal. Calc. for C<sub>14</sub>H<sub>10</sub>ClNO<sub>4</sub> (291.03): C, 57.65; H, 3.46; N, 4.80; Found: C, 57.39; H, 3.40; N, 4.74. IR (*v/cm*<sup>-1</sup>) = 1718 cm<sup>-1</sup>. <sup>1</sup>H NMR (*δ* ppm): 2.80 (*s*, 2H, C14-CH<sub>2</sub>), 2.81 (*s*, 2H, C15-CH<sub>2</sub>), 4.83 (*s*, 2H, C4-CH<sub>2</sub>), 6.38 (*s*, 1H, C3-H), 7.30 (*s*, 1H, C8-H),

7.48 (*d*, 1H, C6), 7.83 (*d*, 1H, C5-H); GC-MS: 291 (M<sup>+</sup>, 100%), 293 (M<sup>+</sup>2, 28%).

**1-((6-methoxy-2-oxo-2H-chromen-4-yl)methyl)pyrrolidine-2,5-dione (2e):** Pale yellow solid from acetic acid, yield 67%, m.p. 229°C - 231°C. Anal. Calc. for C<sub>19</sub>H<sub>13</sub>NO<sub>5</sub> (287.27): C, 68.06; H, 3.91; N, 4.18; Found: C, 67.77; H, 3.86; N, 4.13. IR (*v/cm*<sup>-1</sup>) = 1718 cm<sup>-1</sup> (CO). <sup>1</sup>H NMR (*δ* ppm): 2.78 (*s*, 2H, C14-CH<sub>2</sub>), 2.79 (*s*, 2H, C15-CH<sub>2</sub>), 3.78 (*s*, 3H, C6-OCH<sub>3</sub>), 6.26 (*s*, 3H, C3-H), 7.54 (*d*, 1H, C7-H), 7.69 (*s*, 1H, C5-H), 7.81 (*d*, 1H, C8-H). GC-MS: *m/z* 287 (M<sup>+</sup>, 100%).

**1-((1,2-dihydro-2-oxoquinolin-4-yl)methyl)pyrrolidine-2,5-dione (2f):** White solid from acetic acid, yield 71%, m.p. 216°C - 218°C. Anal. Calc. for C<sub>14</sub>H<sub>12</sub>N<sub>2</sub>O<sub>3</sub> (255.26): C, 65.62; H, 4.72; N, 10.93; Found: C, 65.25; H, 4.67; N, 10.89. IR (*v/cm*<sup>-1</sup>) = 3438 cm<sup>-1</sup> (NH), 1706 cm<sup>-1</sup> (CO), 1663 cm<sup>-1</sup> (lactam CO). <sup>1</sup>H NMR (*δ* ppm): 2.79 (*s*, 4H, pyrrolidine-CH<sub>2</sub>), 4.77 (*s*, 2H, C4-CH<sub>2</sub>), 6.24 (*s*, 1H, C3-H), 7.23 - 7.87 (*m*, 4H, Ar-H), 11.89 (*s*, 1H, NH). GC-MS: *m/z* 256 (M<sup>+</sup>, 100%).

**1-((7-chloro-1,2-dihydro-2-oxoquinolin-4-yl)methyl)pyrrolidine-2,5-dione (2g):** White solid from a ethanol+dioxane, yield 67%, m.p. 228°C - 230°C. Anal. Calc. for C<sub>14</sub>H<sub>11</sub>ClN<sub>2</sub>O<sub>3</sub> (290.05): C, 57.84; H, 3.81; N, 9.64; Found: C, 57.49; H, 3.76; N, 9.61. IR (*v/cm*<sup>-1</sup>) = 3353 cm<sup>-1</sup> (NH), 1714 cm<sup>-1</sup> (CO). <sup>1</sup>H NMR (*δ* ppm): 2.80 (*s*, 4H, pyrrolidine-CH<sub>2</sub>), 4.82 (*s*, 2H, C4-CH<sub>2</sub>), 6.29 (*s*, 1H, C3-H), 7.31 (*s*, 1H, C8-H), 7.45 (*d*, 1H, C6-H),

7.71 (*d*, 1H, C5-H), 11.91 (*s*, 1H, NH). GC-MS: *m/z* 290 ( $M^+$ , 100%). 292 ( $M + 2$ , 30%).

**1,2-dihydro-5,8-dimethyl-2-oxoquinolin-4-yl)methylpyrrolidine-2,5-dione (2h):** Pale green solid from acetic acid, yield 64%, m.p. 234°C - 236°C. Anal. Calc for  $C_{16}H_{16}N_2O_3$  (284.31): C, 67.59; H, 5.67; N, 9.85; Found: C, 67.24; H, 5.61; N, 9.80; IR ( $\nu/cm^{-1}$ ) = 3338  $cm^{-1}$  (NH), 1713  $cm^{-1}$  (CO). 1665  $cm^{-1}$ ,  $^1H$ NMR ( $\delta$  ppm): 2.37 (*s*, 3H, C8-CH<sub>3</sub>), 2.42 (*s*, 3H, C5-CH<sub>3</sub>), 2.78 (*s*, 4H, pyrrolidine-CH<sub>2</sub>), 4.73 (*s*, 2H, C4-CH<sub>2</sub>), 6.27 (*s*, 1H, C3-H), 7.48 (*d*, 1H, C7-H), 7.65 (*d*, 1H, C6-H), 11.78 (*s*, 1H, NH). GC-MS: *m/z* 270 ( $M^+$ , 100%).

**1-((1,2-dihydro-8-methyl-2-oxoquinolin-4-yl)methyl)pyrrolidine-2,5-dione (2i):** White solid from acetic acid, yield 58%, m.p. 251°C - 253°C. Anal. Calc for  $C_{15}H_{14}N_2O_3$  (270.28): C, 66.66; H, 5.22; N, 10.36; Found: C, 66.29; H, 5.17; N, 10.31. IR ( $\nu/cm^{-1}$ ) = 3312  $cm^{-1}$  (NH), 1715  $cm^{-1}$  (CO), 1660  $cm^{-1}$  (lactam CO).  $^1H$  NMR ( $\delta$  ppm): 2.37 (*s*, 3H, C8-CH<sub>3</sub>), 2.78 (*s*, 4H, pyrrolidine-CH<sub>2</sub>), 4.78 (*s*, 2H, C4-CH<sub>2</sub>), 6.32 (*s*, 1H, C3-H), 7.27 (*t*, 1H, C6-H), 7.44 (*d*, 1H, C7-H), 7.66 (*d*, 1H, C5-H), 11.87 (*s*, 1H, NH). GC-MS: *m/z* 270 ( $M^+$ , 100%).

**2-((6-methyl-2-oxo-2H-chromen-4-yl)methyl)isoindoline-1,3-dione (3a):** White solid from acetic acid, yield 73%, m.p. 263°C - 265°C. Anal. Calc. for  $C_{19}H_{13}NO_4$  (319.3): C, 71.47; H, 4.10; N, 4.39; Found: C, 71.16; H, 4.05; N, 4.33. IR ( $\nu/cm^{-1}$ ) = 1775  $cm^{-1}$  (cyclic CO), 1716  $cm^{-1}$  (C=O).  $^1H$ NMR ( $\delta$  ppm): 2.42 (*s*, 3H, C6-CH<sub>3</sub>), 5.02 (*s*, 2H, C4-CH<sub>2</sub>), 6.36 (*s*, 1H, C3-H), 7.37 - 7.78 (*m*, 7H, Ar-H), GC-MS: *m/z* 271 ( $M^+$ , 100%).

**2-((7-methyl-2-oxo-2H-chromen-4-yl)methyl)isoindoline-1,3-dione (3b):** White solid from acetic acid, yield 78%, m.p. 271°C - 273°C. Anal. Calc for  $C_{19}H_{13}NO_4$  (319.3): C, 71.47; H, 4.10; N, 4.39; Found: C, 71.18; H, 4.06; N, 4.33. IR ( $\nu/cm^{-1}$ ) = 1776  $cm^{-1}$  (cyclic CO), 1716  $cm^{-1}$  (CO).  $^1H$ NMR ( $\delta$  ppm): 2.44 (*s*, 3H, CH<sub>3</sub>), 5.01 (*s*, 2H, C4-CH<sub>2</sub>), 6.32 (*s*, 2H, C3-H), 7.26 - 7.75 (*m*, 7H, Ar-H). GC-MS: *m/z* 319.3 ( $M^+$ , 100%).

**2-((7,8-dimethyl-2-oxo-2H-chromen-4-yl)methyl)isoindoline-1,3-dione (3c):** White solid from acetic acid, yield 68%, m.p. 225°C - 227°C. Anal. Calc for  $C_{20}H_{15}NO_4$  (333.4): C, 72.06; H, 4.54; N, 4.20; Found: C, 71.74; H, 4.49; N, 4.17. IR ( $\nu/cm^{-1}$ ) = 1774  $cm^{-1}$  (cyclic CO), 1718  $cm^{-1}$  (CO).  $^1H$  NMR ( $\delta$  ppm): 2.37 (*s*, 3H, C7-CH<sub>3</sub>), 2.42 (*s*, 3H, C8-H), 5.02 (*s*, 2H, C4-CH<sub>2</sub>), 6.34 (*s*, 1H, C3-H), 7.35 - 7.72 (*m*, 6H, Ar-H). GC-MS: *m/z* 333 ( $M^+$ , 30%).

**2-((7-chloro-2-oxo-2H-chromen-4-yl)methyl)isoindoline-1,3-dione (3d):** White solid from ethanol + DMF, yield 71%, m.p. 241°C - 243°C. Anal. Calc for  $C_{18}H_{10}ClNO_4$  (339.03): C, 63.64; H, 2.97; N, 4.12; Found: C, 63.31; H, 4.08; N, 4.09. IR ( $\nu/cm^{-1}$ ) = 1776  $cm^{-1}$  (cyclic CO), 1718  $cm^{-1}$  (CO).  $^1H$ NMR ( $\delta$  ppm): 4.98 (*s*, 2H, C4-CH<sub>2</sub>), 6.34 (*s*, 1H, C3-H), 7.27 - 7.76 (*m*, 7H, Ar-H). GC-MS: *m/z* 339 ( $M^+$ , 100%), 341 ( $M + 2$ , 29%).

**2-((6-methoxy-2-oxo-2H-chromen-4-yl)methyl)isoindoline-1,3-dione (3e):** Pale yellow solid from Acetic acid, yield 76%, m.p. 210°C - 212°C. Anal. Calc for  $C_{19}H_{13}NO_5$  (335.30): C, 68.06; H, 3.91; N, 4.18; Found: C, 67.72; H, 3.88; N, 4.14. IR ( $\nu/cm^{-1}$ ) = 1774  $cm^{-1}$  (cyclic CO), 1719  $cm^{-1}$  (CO).  $^1H$ NMR ( $\delta$  ppm): 3.79 (*s*, 3H, C6-OCH<sub>3</sub>), 5.02 (*s*, 2H, C4-CH<sub>2</sub>), 6.36 (*s*, 1H, C3-H), 7.28 - 7.73 (*m*, 7H, Ar-H). GC-MS: *m/z* 335 ( $M^+$ , 100%).

**2-((1,2-dihydro-2-oxoquinolin-4-yl)methyl)isoindoline-1,3-dione (3f):** White solid from acetic acid, yield 74%, m.p. 287°C - 289°C. Anal. Calc for  $C_{18}H_{12}N_2O_3$  (304.08): C, 71.05; H, 3.97; N, 9.21; Found: C, 70.78; H, 3.92; N, 9.16. IR ( $\nu/cm^{-1}$ ) = 3315  $cm^{-1}$  (NH), 1776  $cm^{-1}$  (cyclic CO), 1716  $cm^{-1}$  (CO), 1655  $cm^{-1}$  (lactam CO).  $^1H$  NMR ( $\delta$  ppm): 4.85 (*s*, 2H, CH<sub>2</sub>), 6.41 (*s*, 1H, C3-H), 7.34 - 7.87 (*m*, 8H, Ar-H), 11.88 (*s*, 1H, NH). GC-MS: *m/z* 304 ( $M^+$ , 100%).

**2-((7-chloro-1,2-dihydro-2-oxoquinolin-4-yl)methyl)isoindoline-1,3-dione (3g):** White solid from acetic acid, yield 65%, m.p. 276°C - 278°C. Anal. Calc for  $C_{18}H_{11}ClN_2O_3$  (338.05): C, 63.82; H, 3.27; N, 8.27; Found: C, 63.47; H, 3.23; N, 8.22. IR ( $\nu/cm^{-1}$ ) = 3355  $cm^{-1}$  (NH), 1774  $cm^{-1}$  (cyclic CO) 1718  $cm^{-1}$ , 1663  $cm^{-1}$  (CO),  $^1H$  NMR ( $\delta$  ppm): 4.92 (*s*, 2H, CH<sub>2</sub>), 6.31 (*s*, 1H, C3-H), 7.39 - 7.88 (*m*, 7H, Ar-H), 11.91 (*s*, 1H, NH). GC-MS: *m/z* 338 ( $M^+$ , 100%), 340 ( $M + 2$ , 28%).

**2-((1,2-dihydro-5,8-dimethyl-2-oxoquinolin-4-yl)methyl)isoindoline-1,3-dione (3h):** White solid from acetic acid, yield 74%, m.p. 291°C - 293°C. Anal. Calc for  $C_{20}H_{16}N_2O_3$  (332.12): C, 72.28; H, 4.85; N, 8.43; Found: C, 71.93; H, 4.79; N, 8.40. IR ( $\nu/cm^{-1}$ ) = 3291  $cm^{-1}$  (NH), 1775  $cm^{-1}$  (cyclic CO) 1718  $cm^{-1}$ , 1661  $cm^{-1}$  (CO).  $^1H$ NMR ( $\delta$  ppm): 2.37 (*s*, 3H, C8-CH<sub>3</sub>), 4.42 (*s*, 3H, C5-CH<sub>3</sub>), 4.88 (*s*, 2H, CH<sub>2</sub>), 6.45 (*s*, 1H, C3H), 7.33 - 7.78 (*m*, 6H, ArH), 11.86 (*s*, 1H, NH). GC-MS: *m/z* 332 ( $M^+$ , 36%).

**2-((1,2-dihydro-8-methyl-2-oxoquinolin-4-yl)methyl)isoindoline-1,3-dione (3i):** White solid from ethanol + dioxane, yield 68%, m.p. 279°C - 281°C. Anal. Calc for  $C_{19}H_{14}N_2O_3$  (318.10): C, 71.69; H, 4.43; N, 8.80; Found: C, 71.27; H, 4.40; N, 8.75. IR ( $\nu/cm^{-1}$ ) = 3310  $cm^{-1}$  (NH), 1776  $cm^{-1}$  (cyclic CO) 1715  $cm^{-1}$ , 1659  $cm^{-1}$  (CO),  $^1H$ NMR ( $\delta$  ppm): 2.44 (*s*, 3H, C8-CH<sub>3</sub>), 4.86 (*s*, 2 H, CH<sub>2</sub>), 6.45 (*s*, 1H, C3H), 7.327.78 (*m*, 6H, ArH), 11.87(*s*, 1H, NH). GC-MS: *m/z* 318 ( $M^+$ , 100%).

## 6. Conclusion

The present study has shown that antimicrobial activity of N-substituted phthalimide derivatives of coumarins and 1-azacoumarins are relatively more than N-substituted succinimide derivatives the introduction of methyl groups into the coumarin and 1-azacoumarin have declined their antimicrobial activity whereas the chloro and methoxy substituted coumarin and 1-azacoumarins have enhanced their antimicrobial activity. The substituted

phthalimide derivatives of coumarins have shown amylase enzyme inhibition activity except ((6-methoxy-2-oxo-2H-chromen-4-yl)methyl) isoindoline-1,3-dione **3e**, but some 1-azacoumarin cyclic imides exhibited enzyme supporting activity and enhanced the hydrolyzing ability of amylase enzyme. The compound 2-((7-chloro-2-oxo-2H-chromen-4-yl)methyl) isoindoline-1,3-dione **3d** was found to be most efficient amylase inhibitor expected to be active in controlling hyperglycemia and useful in the management of type II diabetes and this compound could be used to design effective inhibitors of alpha-amylase. Simple synthetic method can be employed to synthesis wide range of cyclic imide derivatives by inexpensive readily available ecofriendly reagents.

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