

Design, Synthesis and Cancer Cell Line Activities of Pyrazolo[3,4-*b*]pyridine Derivatives

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Received June 12, 2012; revised July 15, 2012; accepted July 26, 2012

ABSTRACT

Starting from 4,6-dimethyl-2-oxo-(1*H*)-3-pyridinecarbonitrile 1 and 3-aminopyrazolopyridine 4, a series of cyanopyridine derivatives 3a-i, Schiff bases 5a-f, urea and thiourea derivatives 6a-b, amide derivatives 7a-h, pyridopyrazolopyrimidine 8a-b and pyridopyrazolotriazine 10a-b were synthesized. Activities of eleven representative compounds were evaluated against A-549 (lung), HEPG2 (liver) and HCT-116 (colon) cancer cell lines. The findings revealed that some of the synthesized compounds showed remarkable anticancer activities, especially 8b which displayed the highest activity among the tested compounds with IC₅₀ equal to 2.9, 2.6 and 2.3 μ mol. In addition to synthesis and biological activities, we present discussion about the rationale of the design and activity of the potent compound 8b using structure-based modeling tools.

Keywords: Pyridine; Pyrazolopyridine; Anticancer Activity; CDK Inhibitors; Kinase Inhibitor

1. Introduction

Fighting cancer is a non-stop duty for drug researchers [1]. The ultimate triumph is realized when scientists bring to markets oral pills that cure cancer with toxicity as low as aspirin or tylenol (for example). In efforts to help achieving this goal, medicinal chemists and their biology collaborators scored important points in their battles with cancer during the past century. Historically, early chemotherapeutic agents such as classic cytotoxic drugs, antimetabolites, platinum therapy and antibiotics produced adverse effects (AEs) much greater than most of other medications. “Chemo” remained unpleasant experience to the most of cancer patients while the outcome was not always fruitful as chemotherapy just added few months or years to patient life’s expectancy [2, 3].

The introduction of kinase inhibitors in the beginning of the 21st century is considered a milestone victory in cancer chemotherapy as safe anticancer oral medications was first made available in world markets [4-6]. The kinase era began upon the introduction of imatinib (Glivec[™]) in 2001 as selective inhibitor of c-ABL tyrosine kinase (TK) [7,8] (**Figure 1**). Traditionally, drug discovery programs put emphasis on selectivity issue on the early discovery stage to avoid potential AEs during lead optimization stage and hence, failure of the project.

however, search for “laser” selective TK inhibitors has rarely achieved any success, simply because ATP is one ligand for many TKs which all belong to one gene family [8]. Moreover, it was discovered that imatinib hits other kinases such as PDGFR and c-KIT [9]. The reality is that when early TK programs followed the golden principle of the traditional medicinal chemistry “seek high selectivity”, it was slow to deliver new anticancer agents. Therefore, most of small molecule kinase programs shifted their strategies to “kill cancer cells” without wasting efforts in solving selectivity dilemma unless it is related to side effects [10,11]. The more recent research belongs to new wave of multiple target kinase inhibitors “dirty drugs” represented by Sorafenib, Dasatinib, Sunitinib and Pazopanib (**Figure 1**) [12]. Still, medicinal chemists must start their projects with designing inhibitors against specific TK targets even though they end-up with significant drug promiscuity [13].

Cyclin-Dependent Kinases (CDKs) are proved to be viable target for development of effective and safe anticancer drugs [14]. For instance, Roscovitine is a CDK-2 inhibitor undergoing clinical trials to treat cancers [15]. Structurally, Roscovitine and other potent CDK inhibitors contained either purine or its bicyclic congeners such as pyrazolopyrimidine and pyrazolopyridazine [16]. For instance, the compound A exhibited excellent CDK2 inhibition but it did not perform well in cancer cell line testing indicating that cell penetration is a challenge to

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the utility of this species of compounds [17]. In the present work, we are introducing compounds that are related to compound A with their inhibitory effect on lung, colon and liver cell lines (**Figure 1**).

2. Chemistry

Synthesis of target compounds were accomplished *via*

key intermediates 3-aminopyrazolopyridine 4 which was prepared in three steps starting from 1, 3-dione derivative and ethyl cyanoacetamide (**Scheme 1**) according to published method [18]. The intermediate 4 was used as diversity precursor to prepare final compounds belonging to several chemotypes. Therefore, key intermediate 4 was reacted with different appropriate aldehydes to obtain the corresponding imines 5a-f, respectively (**Scheme 2**).

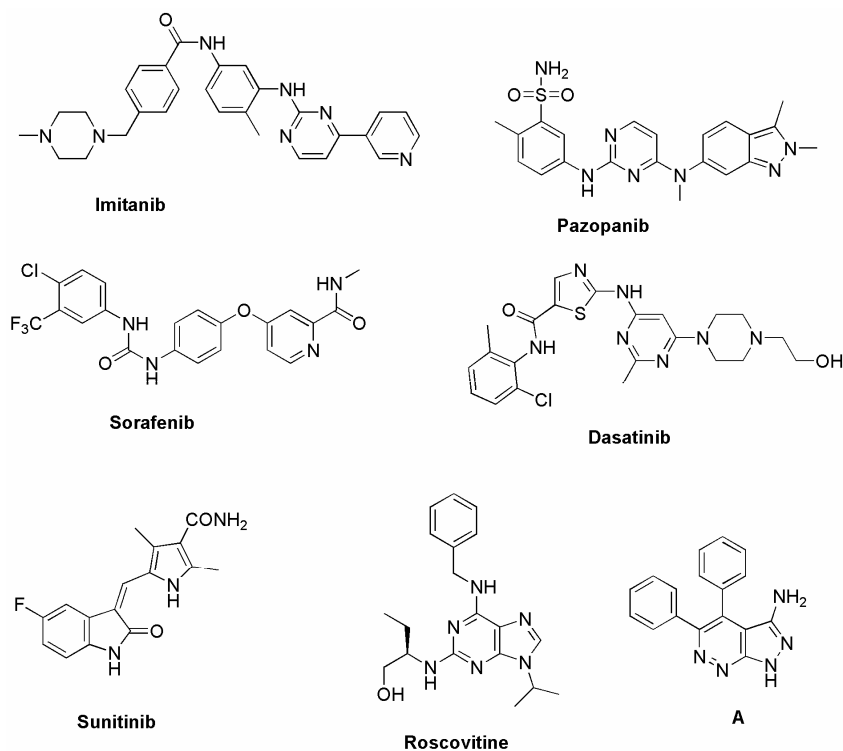
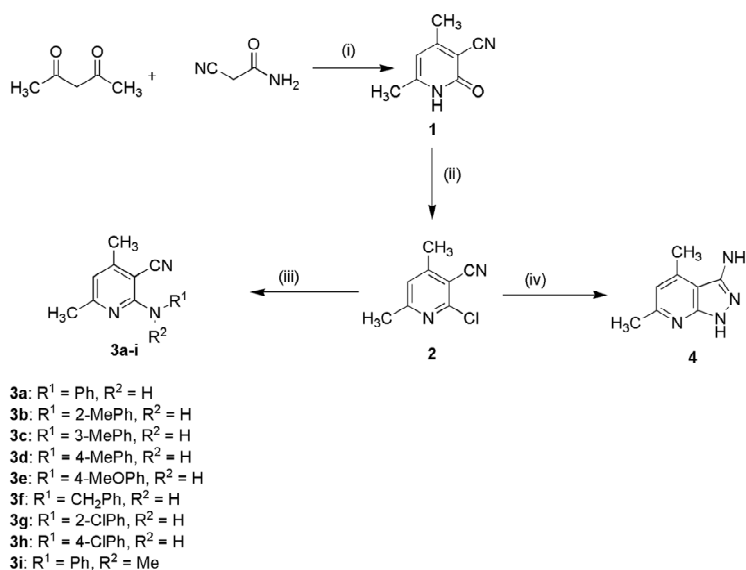
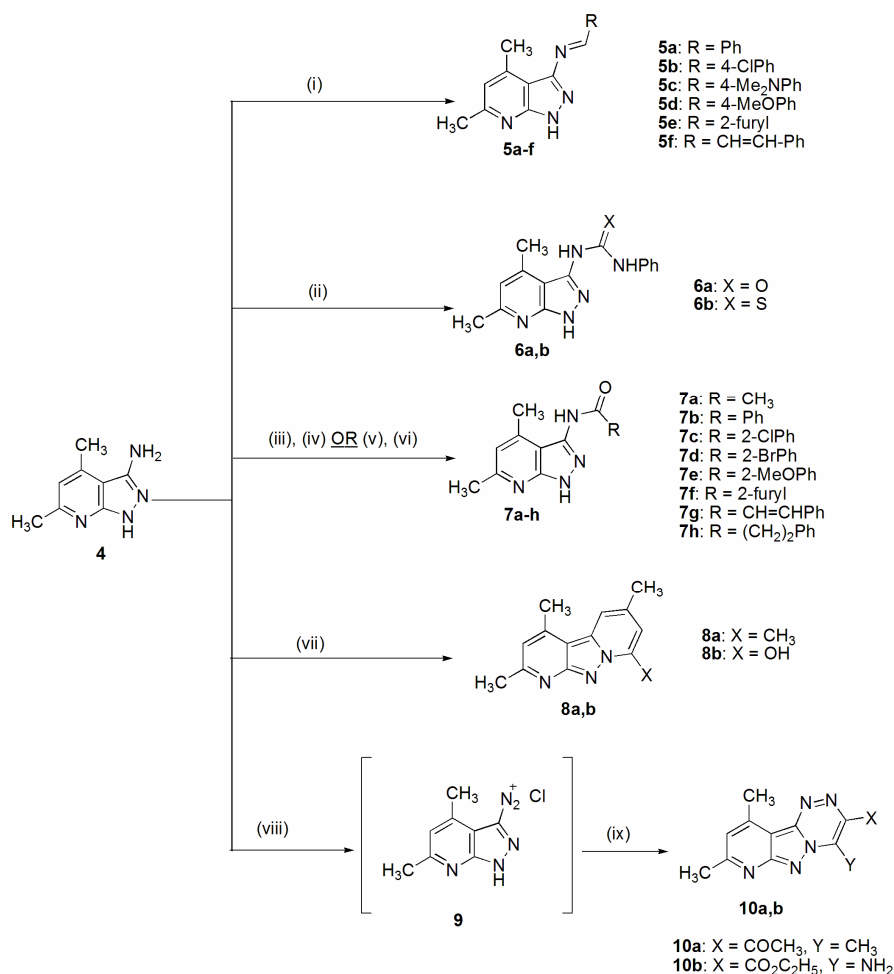


Figure 1. Potent anticancer compounds acting by inhibition of various TKs.



Scheme 1. Reagent and conditions: (i) Et₃N, EtOH, reflux, 5 h; (ii) POCl₃, reflux, 5 h; (iii) Amines, Et₃N, reflux, 5 h; (iv) NH₂NH₂, reflux, 6 h.



Scheme 2. Reagent and conditions: (i) R-CHO, EtOH, AcOH, reflux, 5h; (ii) isocyanate or isothiocyanate in pyridine, reflux, 5 h; (iii) Acetic anhydride, reflux, 6 h; (iv) PhCOCl, Et₃N, CH₂Cl₂, r.t., 5 h; (v) R-COOH, oxalyl chloride, DMF (cat.), CH₂Cl₂, r.t., 12 h; (vi) Et₃N, CH₂Cl₂, rt, 3 h; (vii) Acetylacetone or ethyl acetoacetate, DMF, reflux, 5 hr; (viii) NaNO₂, HCl, 0°C, 3 hr; (ix) Acetylacetone or ethyl cyanoacetate, NaOAc, EtOH, 0°C, 1 hr.

In addition, condensation of the amino pyrazolopyridine 4 with phenyl iso/isothiocyanate in pyridine provided the corresponding phenylurea/thiourea derivatives 6a-b. Another type of compounds was prepared by acylation of 4 to give amides 7a-h.

Tricyclic derivatives were prepared by condensation of amino pyrazolopyridine 4 with active methylene reagents, namely: acetylacetone or ethyl acetoacetate gave the corresponding pyridopyrazolopyrimidine 8a-b, respectively. The other tricyclic species 10a-b was prepared using a method that involved diazotization of 4 followed by coupling with active methylene-containing reagents [19].

3. Cancer Cell Line Assays

3.1. Materials and Methods

3.1.1. Cell Culture

A-549 (human lung carcinoma), HePG2 (human Liver carcinoma) and HCT-116 (Human Colon carcinoma)

were obtained from KarolinskaInstitutet, Stockholm, Sweden. All cells were maintained in RPMI 1640 (Lonza-Biowahittkar) medium. All the media were supplemented with 1% antibiotic-antimycotic mixture (10,000 U/ml Potassium Penicillin (Biowest), 10,000 µg/ml Streptomycin Sulfate (Biowest) and 25 µg/ml Amphotericin B (Biowest) and 1% L-glutamine (Biowest).

3.1.2. MTT Cytotoxicity Assay

The cell viability was investigated using MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] assay. This reaction utilizes the mitochondrial-dependent reduction of yellow MTT into purple formazan as indication of cell viability.

All the preceding steps were carried out in sterile laminar flow cabinet biosafety class II level (Baker, SG403INT, Sanford, ME, USA). All incubations were done at 37°C in a humidified atmosphere of 5% CO₂ incubator (Sheldon, TC2323, Cornelius, OR, USA).

Cells were seeded into 96-well microtiterplates at the concentration of 10×10^3 and allowed to adhere for 24 hours. Medium was aspirated and a fresh medium (without serum) was added to the cells with various concentrations of compounds (100, 50, 10, 5, 2.5 & 1.25 $\mu\text{g}/\text{mL}$) and incubated for 48 hours. For the untreated cells (negative control), sterile deionized water was added instead of tested compounds. A positive control was set using adriamycin (doxorubicin) as a known cytotoxic natural agent.

Medium was aspirated and 40 μL MTT salt (2.5 $\mu\text{g}/\text{mL}$) (Biobasic Inc.) was added to each well and incubated for further four hours. To stop the reaction and dissolve any formed formazan crystals, 200 μL of 10% Sodium Dodecyl Sulfate (SDS) was added to each well and incubated overnight at 37°C. The amount of formazan produced was measured at 595 nm and a reference wave length of 620 nm as a background using a microplate reader (Bio-Rad Laboratories Inc., model 3350, Hercules, California, USA). The percentage of change in viability was calculated according to the formula: $[(\text{Absorbance of compound}/\text{Absorbance of negative control}) - 1] \times 100$.

A statistical significance was tested between samples and negative control (cells with vehicle) using independent t-test using SPSS 11 software. We have to denote that dimethylsulfoxide (DMSO) is the vehicle used for dissolution of compounds but its final concentration on the cells was less than 0.2%. A probit analysis was carried for IC_{50} determination using SPSS 11 program.

3.2. Results

Results of cancer cellular assays (as summarized in **Table 1**) illustrate that most of tested compounds showed variable cytotoxic activities. The tricyclic analogue 8b showed the highest potency at low $\mu\text{g}/\text{mL}$ level against all cell lines. Other derivatives had lower activities comparable to reference compound as well the tricyclic derivative 8b.

4. Discussion

4.1. Design and Rationale

In the preceding work by Braña et al. on pyrazolo [3,4] pyridazine derivative A and its analogues, it was observed that the cellular activities are not quite parallel to *in vitro* enzyme inhibition. The free primary amino group secured good potency in enzyme inhibition but no cellular activities observed for most of compounds. The only compounds in this series that showed activities in both isolated enzyme and cell lines are those with acylated amino group (compounds B, **Figure 2**); a feature that can be attributed to cell penetration ability of these compounds. However, these two compounds were less potent than the lead compound A in isolated enzyme assay [17].

Our present work is based on pyrazolo[3,4-*b*]pyridine scaffold which is a close isoster to Brana's pyrazolo [3,4-*c*]pyridazine but it has one less nitrogen which was hinted to be less important for binding according to the authors' docking studies.

It is broadly agreed that CDK2 inhibitors are characterized by at least one hydrophobic face of ligands that interact with nonpolar residues patented to this enzyme family [20]. The most frequent nonpolar residues in this region (as appeared in crystal structures) are Leu10, Phe80, Phe82 and Ile83 [21,22]. These residues form a hydrophobic pocket that surround the aminopyrimidine portion (adenine face) of ATP to help binding by reducing the entropic factors of H-bond formation. In our design, the adenine face's hydrophobic nature has been maintained but small methyl substituents on the pyridine ring were considered (*c.f.* compounds A and B). The change aimed at sliding the pyridine portion inside the hydrophobic pocket thus allowing more space for accommodating C-3 substituent (**Figure 2**). In addition, it was envisaged that this movement in ligand will give the pyrazole nitrogens and the C3 substituents opportunity to make hydrogen bonds with some residues that accommodate the ATP's ribose.

Although the high potency of our compound (8b) was surprisingly pleasant to us, it was important to study binding features of this compound using computational tools.

Table 1. Results of cancer cell line inhibition tests.

Compound	IC_{50} ($\mu\text{g}/\text{mL}$)		
	A-549	HepG-2	HCT-116
2	106.9	131.7	101.3
3e	72.5	22.3	66.2
4	158.4	79.7	117.5
5d	96.4	97.4	NA*
5f	85.6	253.1	NA
6a	44.7	51.04	101.3
6b	105.01	128.2	NA
7a	69.9	90.4	NA
7e	64.3	96.5	NA
8b	2.9	2.6	2.3
10a	60.2	61.5	82.4
Doxorubicin	2.1	2.7	4.26

*NA= Compound is not active at tested range.

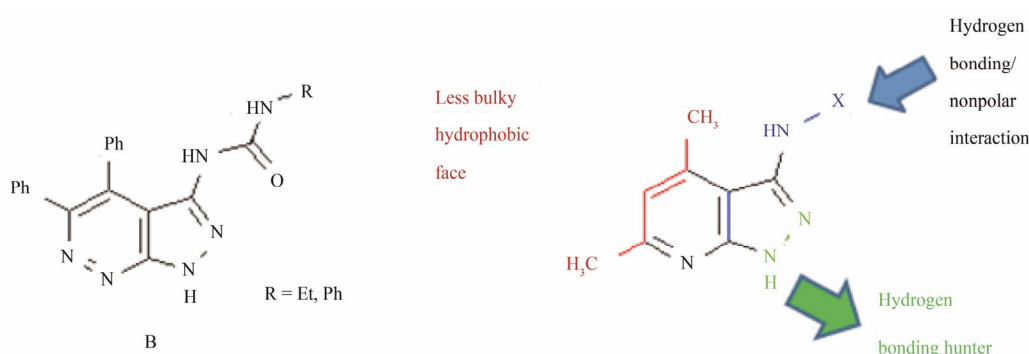


Figure 2. Compounds B are reported as antiproliferative/CDK-2 inhibitor and a sketch for basis of our design.

4.2. Structure-Based Molecular Modeling of Compound 8b

Crystal structure of Cyclin-Dependent Kinase 2 (CDK-2) (pdb entry: 1HCK) containing ATP as a ligand was used in modeling experiments without further processing.

Compound 8b was constructed on ChemDraw 8.0. Structure was copied to Discovery Studio v. 2.5.5 and saved as Sybyl.mol2 file. The structure was imported to Sybyl-X v. 1.1 and subjected to geometry optimization by running energy minimization.

Parameters of Energy Minimization as follows: Charges: Gasteiger-Marsili Charges; Force Filed: Tripos; Termination: Gradient energy change, 0.01 kcal/Å-mol; rms displacement: 0.001 Å; Non-bonded cutoff: 8.000 Å; Dielectric function: distant dependent; Dielectric constant: 1.00; Iteration: until convergence.

In the process, ATP and Magnesium were removed from active site and no further optimization was performed because this active site doesn't change significantly upon binding of the ATP. Initial docking of the new ligand 8b was in a way that placed pyrazole and pyridine rings of 8b completely overlapped with ATP's purine previous position. It was noticed that initial docking is generally acceptable as a starting structure with the exception of some unfavorable approach of one of the methyl groups to the carbonyl of Phe82 and the side chain of Asp145 (**Figure 3**).

After docking, the new complex was subjected to molecular mechanics (MM) geometry optimization using the above parameters except iterations are limited to 100. This parameter was enough to optimize the energy of the active site without unnecessary deformations of the protein. As a result, Van der Waal radii became relaxed after minimization without major movement in residues around especially Phe82 and Asp145. However, contacts of the ligand with the active site were not satisfactory as they didn't describe the exceptional activity of 8b. Therefore, the initial docking complex was subjected to unrestrained molecular dynamics (MD) 100 ps simulations at 298 K and 1atm. Rest of parameters were set to SYBYL-X default.

Analyzing MD snapshots led to observation of stabilized complex that features important change in the positioning and hence, changes occurred in contacts of the ligand 8b with the active site (**Figure 4(a)**). The ligand 8b responded to MD simulations positively as it established hydrophobic and hydrogen bonding attraction forces with residues around. The most important change is the new bifurcate H-bond between carbonyl of the terminal pyrimidine ring with backbone carbonyl of Gln131 and side chain amide of Asn132. It was also noticed that residues of adenine pocket approached the dimethylpyridine portion of the ligand in suitable VdW distances. In addition, some favorable hydrophobic contacts of the methyl group on the pyrimidine ring with Val18 and Phe80 appeared. To confirm the validity of the MD algorithms, we extracted the ligand 8b from the MD snapshot and redocked it into CDK-2 active site without energy optimizations (**Figure 4(b)**). It was noticed that interactions of the ligand with residues especially Gln131 and Asn132 are identical to those obtained from MD simulations.

5. Conclusion

Compounds belonging to this class of compounds exhibited moderate to potent anticancer activities in cell line tests. The safety of these compounds was preliminarily investigated by acute mice toxicity test. None of the tested compounds caused death of the animals in doses up to 500 mg/kg. The most potent compound 8b is characterized by restrained side chain indicating the small space around this area in kinase active sites. In addition, the loss of potency occurred upon changing the OH group by amino either due to decrease of cellular penetration or inability of making effective electrostatic contacts with the active site as in 8b. Further SAR studies are needed for 8b especially on the hydrophobic side.

6. Experimental

General Methods

All melting points were uncorrected and measured using

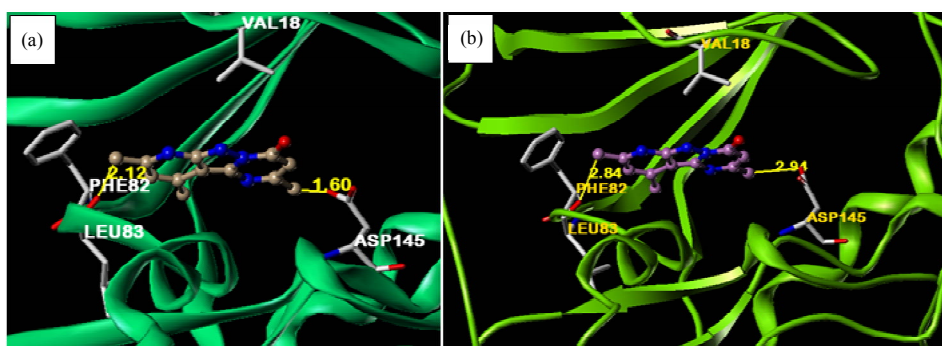


Figure 3. (a) Initial docking of 8b by complete overlap with ATP purine bicyclic structure. Note unfavorable contacts with Phe82 and Asp145; (b) 8b docked to active site after energy minimization (applying of MM).

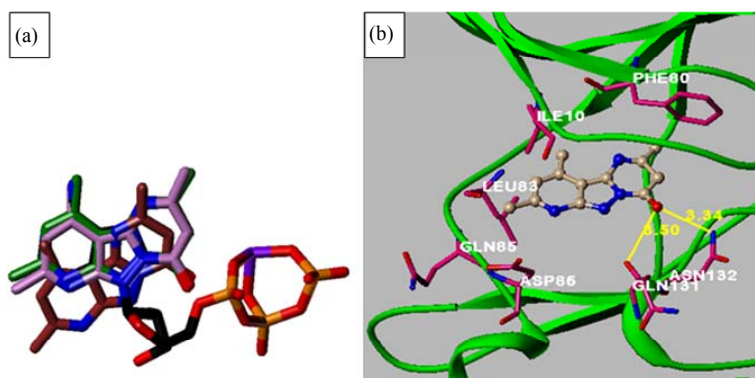


Figure 4. (a) Alignment of ATP (black) extracted from CDK-2 with 8b after docking. Initial docking position of 8b is shown in green; after applying MM is shown in violet; and after MD imulations is shown in brown. position from docking position (magenta) to position after applying molecular mechanics (green) and position after applying molecular dynamics (light brown); (b) Contacts of 8b with active site after second docking experiments based on results of MD simulations.

Electro-thermal IA 9100 apparatus (Shimadzu, Japan); IR spectra were recorded as potassium bromide pellets on a Perkin-Elmer 1650 spectrophotometer (USA), Faculty of Science, Cairo University, Cairo, Egypt. ^1H -NMR spectra were determined on a Varian Mercury (500 MHz) spectrometer (Varian UK) and chemical shifts were expressed as ppm against TMS as internal reference (Faculty of Science, Cairo University, Cairo, Egypt). Mass spectra were recorded on 70 eV (EI Ms-QP 1000 EX, Shimadzu, Japan), Faculty of Science, Cairo University, Cairo, Egypt. Microanalyses were operated using Vario, Elmentar apparatus (Shimadzu, Japan), Organic Microanalysis Unit, Cairo University, Giza, Egypt. Column Chromatography was performed on (Merck) Silica gel 60 (particle size 0.06 mm - 0.20 mm). All compounds prepared in this paper are new and confirmed with spectral data.

4,6-Dimethyl-2-oxo-(1H)-3-pyridinecarbonitrile (1)

A mixture of cyanoacetamide (0.01 mol) and acetyl acetone (0.01 mol) in ethanol (20 mL) and triethylamine (1 mL) was heated under the reflux for 6 h, poured, onto cold water. The obtained precipitate was filtered off, washed with ethanol and re-crystallized from methanol to

afford 1.

Yield 89%, m.p. 290°C (Lit. mp. 290°C - 292°C [18]).

2-Chloro-4,6-dimethyl-3-pyridinecarbonitrile (2)

A mixture of 2-pyridone 1 (0.01 mol) and POCl_3 (20 mL) was heated under the reflux for 6 h, cooled, poured onto crushed ice with stirring to give a precipitate which was filtered off, air dried, and re-crystallized from methanol to afford 2.

Yield 91%, m.p. 90°C (Lit. mp. 90°C - 92°C [18]).

General procedure for the synthesis of compounds (3a-i)

A mixture of 2-chloro pyridine 2 (0.01 mol) and appropriate amine (0.04 mol), namely: aniline, 2-toluidine, 3-toluidine, 4-toluidine, 4-anisidine, benzylamine, 2-chloro-aniline, 4-chloroaniline, *N*-methylaniline in ethanol (30 mL) was stirred at 100°C for 48 h in a sealed tube. The reaction mixture gave a precipitate which was filtered, washed with EtOH and dried under vacuum to afford 3a-i.

4,6-Dimethyl-2-(phenylamino)pyridine-3-carbonitrile (3a)

Yield 87%, m.p. 277°C - 279°C, IR (KBr) ν (cm^{-1}): 3350 (NH), 2230 ($\text{C}\equiv\text{N}$). MS (EI) m/z : 223 (M^+ , 91%).

^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.59 (s, 3 H, CH₃), 2.65 (s, 3 H, CH₃), 4.85 (s, 1 H, NH, D₂O exchangeable), 6.8 (s, 1 H, C5-H), 7.1 - 7.6 (m, 5 H, Ar-H). Anal. Calcd for C₁₄H₁₃N₃ (223.27): C, 75.31; H, 5.87; N, 18.82. Found: C, 75.54; H, 6.10; N, 18.58.

4,6-Dimethyl-2-(2-methylphenylamino)-pyridine-3-carbonitrile (3b)

Yield 83%, m.p. 220-222°C, IR (KBr) ν (cm⁻¹): 3270 (NH), 2225 (C≡N). MS (EI) m/z : 237 (M⁺, 89%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.35 (s, 3 H, CH₃), 2.65 (s, 3 H, CH₃), 2.75 (s, 3 H, CH₃), 4.5 (s, 1 H, NH, D₂O exchangeable), 6.8 (s, 1 H, C5-H), 7.2 - 7.8 (m, 4 H, Ar-H). Anal. Calcd for C₁₅H₁₅N₃ (237.3): C, 75.92; H, 6.37; N, 17.71. Found: C, 75.70; H, 6.59; N, 17.48.

4,6-Dimethyl-2-(3-methylphenylamino)-pyridine-3-carbonitrile (3c)

Yield 75%, m.p. 110°C - 112°C, IR (KBr) ν (cm⁻¹): 3320 (NH), 2235 (C≡N). MS (EI) m/z : 237 (M⁺, 67%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.4 (s, 3 H, CH₃), 2.55 (s, 3 H, CH₃), 2.66 (s, 3 H, CH₃), 4.6 (s, 1 H, NH, D₂O exchangeable), 6.7 (s, 1 H, C5-H), 7.4 - 7.9 (m, 4 H, Ar-H). Anal. Calcd for C₁₅H₁₅N₃ (237.3): C, 75.92; H, 6.37; N, 17.71. Found: C, 75.78; H, 6.22; N, 17.62.

4,6-Dimethyl-2-(4-methylphenylamino)-pyridine-3-carbonitrile (3d)

Yield 91%, m.p. 160°C - 162°C, IR (KBr) ν (cm⁻¹): 3340 (NH), 2230 (C≡N). MS (EI) m/z : 237 (M⁺, 75%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.31 (s, 3 H, CH₃), 2.60 (s, 3 H, CH₃), 2.72 (s, 3 H, CH₃), 4.4 (s, 1 H, NH, D₂O exchangeable), 6.34 - 6.74 (dd, 4 H, Ar-H), 6.8 (s, 1 H, C5-H). Anal. Calcd for C₁₅H₁₅N₃ (237.3): C, 75.92; H, 6.37; N, 17.71. Found: C, 75.80; H, 6.54; N, 17.95.

2-(4-Methoxyphenylamino)-4,6-dimethylpyridine-3-carbonitrile (3e)

Yield 84%, m.p. 105°C - 107°C, IR (KBr) ν (cm⁻¹): 3160 (NH), 2211 (C≡N). MS (EI) m/z : 253 (M⁺, 86%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.40 (s, 3 H, CH₃), 2.72 (s, 3 H, CH₃), 3.75 (s, 3 H, O-CH₃), 4.5 (s, 1 H, NH, D₂O exchangeable), 6.7 (s, 1 H, C5-H), 7.35 - 7.58 (dd, 4 H, Ar-H). Anal. Calcd for C₁₅H₁₅N₃O (253.3): C, 71.13; H, 5.97; N, 16.59; O, 6.32. Found: C, 71.32; H, 6.21; N, 16.72; O, 6.58.

2-(Benzylamino)-4,6-dimethylpyridine-3-carbonitrile (3f)

Yield 69%, m.p. 140°C - 142°C, IR (KBr) ν (cm⁻¹): 3260 (NH), 2230 (C≡N). MS (EI) m/z : 237 (M⁺, 88.0%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.3 (s, 3 H, CH₃), 2.55 (s, 3 H, CH₃), 4.7 (s, 2 H, CH₂), 6.2 (s, 1 H, NH, D₂O exchangeable), 6.8 (s, 1 H, C5-H), 7.15 - 7.0 (m, 5 H, Ar-H). Anal. Calcd for C₁₅H₁₅N₃ (237.3): C, 75.92; H, 6.37; N, 17.71. Found: C, 75.70; H, 6.59; N, 17.48.

2-(2-Chlorophenylamino)-4,6-dimethylpyridine-3-carbonitrile (3g)

Yield 71%, m.p. 100°C - 102°C, IR (KBr) ν (cm⁻¹): 3330 (NH), 2225 (C≡N). MS (EI) m/z : 257 (M⁺, ³⁵Cl, 56%), 259 (M⁺ + 2, ³⁷Cl, 17.6%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.29 (s, 3 H, CH₃), 2.53 (s, 3 H, CH₃), 4.62 (s, 1 H, NH, D₂O exchangeable), 6.8 (s, 1 H, C5-H), 7.5-7.0 (m, 4 H, Ar-H). Anal. Calcd for C₁₄H₁₂ClN₃ (257.72): C, 65.25; H, 4.69; Cl, 13.76; N, 16.30. Found: C, 65.48; H, 4.89; Cl, 13.91; N, 16.54.

2-(4-Chlorophenylamino)-4,6-dimethylpyridine-3-carbonitrile (3h)

Yield 69%, m.p. 190°C - 192°C, IR (KBr) ν (cm⁻¹): 3260.1 (NH), 2218 (C≡N). MS (EI) m/z : 257 (M⁺, ³⁵Cl, 78.2%), 259 (M⁺ + 2, ³⁷Cl, 25.5%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.36 (s, 3 H, CH₃), 2.72 (s, 3 H, CH₃), 5.2 (s, 1 H, NH, D₂O exchangeable), 6.8 (s, 1 H, C5-H), 6.4, 7.2 (dd, 4 H, Ar-H). Anal. Calcd for C₁₄H₁₂ClN₃ (257.72): C, 65.25; H, 4.69; Cl, 13.76; N, 16.30. Found: C, 65.48; H, 4.82; Cl, 13.51; N, 16.48.

2-(N-Methyl-N-phenylamino)-4,6-dimethylpyridine-3-carbonitrile (3i)

Yield 74%, m.p. 110°C - 112°C, IR (KBr) ν (cm⁻¹): 2232 (C≡N). MS (EI) m/z : 237 (M⁺, 38.8%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.39 (s, 3 H, CH₃), 2.57 (s, 3 H, CH₃), 3.6 (s, 3 H, N-CH₃), 6.7 (s, 1 H, C5-H), 6.9 - 7.4 (m, 5 H, Ar-H). Anal. Calcd for C₁₅H₁₅N₃ (237.3): C, 75.92; H, 6.37; N, 17.71. Found: C, 75.73; H, 6.54; N, 17.92.

4,6-Dimethyl-1H-pyrazolo[3,4-b]pyridine-3-amine (4)

A mixture of 2-chloro pyridine **2** (0.01 mol) and hydrazine hydrate (99%, 0.03 mol) in absolute ethanol (20 mL) was heated under reflux for 6 h. The reaction mixture was concentrated to one third of its volume. The product that formed was collected by filtration, washed with cold ethanol, and then crystallized from ethanol to afford **4**.

Yield 95%, m.p. 280°C - 282°C (Lit. mp. 280°C [18]).

General procedure for the synthesis of compounds (5a-f)

A mixture of 3-amino pyrazolopyridine **4** (0.01 mol) and the appropriate aldehyde, namely: benzaldehyde, 4-chlorobenzaldehyde, 4-dimethylaminobenzaldehyde, *p*-anisaldehyde, furfural and cinnamaldehyde (0.01 mol) in absolute ethanol (30 mL) containing few drops glacial acetic acid, was refluxed for 5 h. The reaction mixture was cooled and poured onto ice. The solid formed was filtered off, washed with water and re-crystallized from the methanol to afford Schiff bases **5a-f**.

N-Benzylidene-4,6-dimethyl-1H-pyrazolo[3,4-b]pyridine-3-amine (5a)

Yield 72%, m.p. 245°C - 247°C, IR (KBr) ν (cm⁻¹): 3265 (NH), 1320 (C=N). MS (EI) m/z : 250 (M⁺, 58.9%). ^1H NMR (DMSO- d_6) (500 MHz) δ (ppm): 2.36 (s, 3 H, CH₃), 2.72 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.3 - 7.6 (m,

5H, Ar-H), 8.1 (s, 1 H, CH), 13.0 (s, 1 H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₅H₁₄N₄ (250.3): C, 71.98; H, 5.64; N, 22.38. Found: C, 71.68; H, 5.41; N, 22.59.

***N*-(4-Chlorobenzylidene)-4,6-dimethyl-1*H*-pyrazolo[3,4-*b*]pyridine-3-amine (5b)**

Yield 79%, m.p. 290°C - 292°C, IR (KBr) ν (cm⁻¹): 3290 (NH), 1350 (C=N). MS (EI) m/z : 284 (M⁺, ³⁵Cl, 49.5%), 286 (M⁺ + 2, ³⁷Cl, 16.5%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.42 (s, 3 H, CH₃), 2.68 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.3, 7.6 (dd, 4 H, Ar-H), 7.9 (s, 1 H, CH), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₅H₁₃ClN₄ (284.74): C, 63.27; H, 4.60; Cl, 12.45; N, 19.68. Found: C, 63.43; H, 4.85; Cl, 12.62; N, 19.39.

***N*-[4-(Dimethylamino)benzylidene]-4,6-dimethyl-1*H*-pyrazolo[3,4-*b*]pyridine-3-amine (5c)**

Yield 65%, m.p. 230°C - 232°C, IR (KBr) ν (cm⁻¹): 3280 (NH), 1260 (C=N). MS (EI) m/z : 293 (M⁺, 85%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.42 (s, 3 H, CH₃), 2.68 (s, 3 H, CH₃), 3.2 (s, 6 H, CH₃), 6.9 (s, 1 H, C5-H), 7.3 (d, 2H, Ar-H), 7.6 (d, 2H, Ar-H), 8.34 (s, 1 H, CH), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₇H₁₉N₅ (293.37): C, 69.60; H, 6.53; N, 23.87. Found: C, 69.87; H, 6.78; N, 23.61.

***N*-(4-Methoxybenzylidene)-4,6-dimethyl-1*H*-pyrazolo[3,4-*b*]pyridine-3-amine (5d)**

Yield 84%, m.p. 250°C - 252°C, IR (KBr) ν (cm⁻¹): 3142 (NH), 1310 (C=N), 1250 (C-O). MS (EI) m/z : 280 (M⁺, 41%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.35 (s, 3 H, CH₃), 2.72 (s, 3 H, CH₃), 3.7 (s, 3 H, O-CH₃), 6.8 (s, 1 H, C5-H), 7.2, 7.2 (d, 2H, Ar-H), 7.5 (d, 2H, Ar-H), 8.02 (s, 1 H, CH), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₆H₁₆N₄O (280.32): C, 68.55; H, 5.75; N, 19.99; O, 5.71. Found: C, 68.81; H, 5.94; N, 19.75; O, 5.49.

***N*-(Furan-2-yl)methylene]-4,6-dimethyl-1*H*-pyrazolo[3,4-*b*]pyridine-3-amine (5e)**

Yield 63%, m.p. 235°C - 237°C, IR (KBr) ν (cm⁻¹): 3230 (NH), 1250 (C=N). MS (EI) m/z : 240 (M⁺, 54%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.46 (s, 3 H, CH₃), 2.71 (s, 3 H, CH₃), 6.9 (s, 1 H, C5-H), 7.3 - 7.5 (m, 3 H, furan-H), 8.0 (s, 1 H, CH), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₃H₁₂N₄O (240.26): C, 64.99; H, 5.03; N, 23.32; O, 6.66. Found: C, 64.72; H, 5.25; N, 23.59; O, 6.84.

(*E*)-4,6-Dimethyl-*N*-(3-phenylallylidene)-1*H*-pyrazolo[3,4-*b*]pyridine-3-amine (5f)

Yield 81%, m.p. 135°C - 137°C, IR (KBr) ν (cm⁻¹): 3142 (NH), 1210 (C=N). MS (EI) m/z : 276 (M⁺, 83.5%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.25 (s, 3 H, CH₃), 2.62 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.1 - 7.4 (m, 5 H, Ar-H), 8.3 (d, 1 H, CH=N), 8.8 (t, 1 H, CH=CH), 9.6 (d, 1 H, CH-ph), 13.0 (s, H, NH, pyrazole, D₂O ex-

changeable). Anal. Calcd for C₁₇H₁₆N₄ (276.34): C, 73.89; H, 5.84; N, 20.27. Found: C, 73.62; H, 5.59; N, 20.12.

General procedure for the synthesis of compounds (6a-b)

A mixture of 3-amino pyrazolopyridine 4 (0.01 mol) and phenylisocyanate or phenylisothiocyanate (0.01 mol) in pyridine (20 mL) was refluxed for 5 h. The reaction mixture was poured on to crushed ice containing few drops of HCl. The solid product was filtered off and crystallized from acetic acid to afford 6a-b.

1-(4,6-Dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)-3-phenylurea (6a)

Yield 83%, m.p. 250°C - 252°C, IR (KBr) ν (cm⁻¹): 3321, 3275 and 3123 (NH), 1690 (C=O). MS (EI) m/z : 281 (M⁺, 45%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.32 (s, 3 H, CH₃), 2.49 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.3 - 7.64 (m, 5 H, Ar-H), 8.5, 7.9 (s, 2 H, 2 NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₅H₁₅N₅O (281.13): C, 64.04; H, 5.37; N, 24.90; O, 5.69. Found: C, 64.29; H, 5.13; N, 24.73; O, 5.91.

1-(4,6-Dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)-3-phenylthiourea (6b)

Yield 89%, m.p. 200°C - 202°C, IR (KBr) ν (cm⁻¹): 3360, 3290 and 3163 (NH), 1610 (C=S). MS (EI) m/z : 297 (M⁺, 60.5%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.34 (s, 3 H, CH₃), 2.52 (s, 3 H, CH₃), 6.7 (s, 1 H, C5-H), 7.3 - 7.82 (m, 5 H, Ar-H), 8.5, 9.2 (s, 2 H, 2 NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₅H₁₅N₅S (297.38): C, 60.58; H, 5.08; N, 23.55; S, 10.78. Found: C, 60.84; H, 5.25; N, 23.39; S, 10.92.

***N*-(4,6-Dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)acetamide (7a)**

A solution of 3-amino pyrazolopyridine 4 (0.01 mol) and acetic anhydride (30 mL) was heated under reflux for 4 h and then cooled. The solid product formed was filtered off and crystallized from ethanol to afford compound 7a.

Yield 89%, m.p. 290°C - 292°C, IR (KBr) ν (cm⁻¹): 3145 (NH), 1680 (C=O). MS (EI) m/z : 204 (M⁺, 80.5%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.02 (s, 3 H, CH₃), 2.34 (s, 3 H, CH₃), 2.61 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 8.2 (s, 1 H, NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₀H₁₂N₄O (204.23): C, 58.81; H, 5.92; N, 27.43; O, 7.83. Found: C, 58.59; H, 5.65; N, 27.67; O, 7.53.

***N*-(4,6-Dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)benzamide (7b)**

To a stirred iced solution of 3-amino pyrazolopyridine 4 (0.01 mol) in CH₂Cl₂ (8 mL) containing few drops (5 mL) of triethylamine benzoyl chloride (0.01 mol) was added drop wise and stirring continuous for 1 h at r.t.. The

solid product formed was filtered off and re-crystallized from ethanol to afford compound 7b.

Yield 83%, m.p. 243°C - 245°C, IR (KBr) ν (cm⁻¹): 3200 (NH), 1660 (C=O). MS (EI) m/z : 266 (M⁺, 75%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.4 (s, 3 H, CH₃), 2.7 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.4 - 7.7 (m, 5 H, Ar-H), 8.1 (s, 1 H, NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₅H₁₄N₄O (266.3): C, 67.65; H, 5.30; N, 21.04; O, 6.01. Found: C, 67.39; H, 5.12; N, 21.29; O, 6.32.

General procedure for the synthesis of compounds (7c-h)

Oxalyl chloride (0.01 mol) and a drop of DMF were added to a stirring solution of appropriate substituted carboxylic acid (0.01 mol), namely: 2-chlorobenzoic acid, 2-bromobenzoic acid, 2-methoxybenzoic acid, 2-fouric acid, propionic acid and cinnamic acid in dry CH₂Cl₂ (8 mL). The reaction mixture was allowed to stir at room temperature for 12 h, and then 3-aminopyrazolopyridine 4 (0.01 mol) was added and the reaction mixture was continued to stir at room temperature for further 3 h. Excess of CH₂Cl₂ was removed under reduced pressure. The residue was purified by shaking with diluted HCl in separating funnel, separate the aqueous layer and then dried with MgSO₄ to afford 7c-h.

2-Chloro-*N*-(4,6-dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)benzamide (7c)

Yield 61%, m.p. 138°C - 140°C, IR (KBr) ν (cm⁻¹): 3325 (NH), 1685 (C=O). MS (EI) m/z : 300 (M⁺, ³⁵Cl, 69.5%), 302 (M⁺ + 2, ³⁷Cl, 23.2%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.29 (s, 3 H, CH₃), 2.52 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.4 - 7.75 (m, 4 H, Ar-H), 7.9 (s, 1 H, NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₅H₁₃ClN₄O (300.7): C, 59.91; H, 4.36; Cl, 11.79; N, 18.63; O, 5.32. Found: C, 59.63; H, 4.58; Cl, 11.92; N, 18.82; O, 5.64.

2-Bromo-*N*-(4,6-dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)benzamide (7d)

Yield 69%, m.p. 150°C - 152°C, IR (KBr) ν (cm⁻¹): 3330 (NH), 1670 (C=O). MS (EI) m/z : 344 (M⁺, 70%), 346 (M⁺ + 1, 69%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.32 (s, 3 H, CH₃), 2.60 (s, 3 H, CH₃), 6.9 (s, 1 H, C5-H), 7.1 - 7.4 (m, 4 H, Ar-H), 8.1 (s, 1 H, NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₅H₁₃BrN₄O (345.19): C, 52.19; H, 3.80; Br, 23.15; N, 16.23; O, 4.63. Found: C, 52.42; H, 3.53; Br, 23.39; N, 16.58; O, 4.82.

2-Methoxy-*N*-(4,6-dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)benzamide (7e)

Yield 71%, m.p. 118°C - 120°C, IR (KBr) ν (cm⁻¹): 3370 (NH), 1640 (C=O), 1270 (C-O). MS (EI) m/z : 296 (M⁺, 65%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.32 (s, 3 H, CH₃), 2.60 (s, 3 H, CH₃), 3.7 (s, 3 H, O-CH₃), 6.8 (s, 1 H, C5-H), 7.1-7.4 (m, 4 H, Ar-H), 8.5

(s, 1 H, NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₆H₁₆N₄O₂ (296.32): C, 64.85; H, 5.44; N, 18.91; O, 10.80. Found: C, 64.59; H, 5.21; N, 18.71; O, 10.62.

N-(4,6-Dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)furan-2-carboxamide (7f)

Yield 65%, m.p. 130°C - 132°C, IR (KBr) ν (cm⁻¹): 3400 (NH), 1650 (C=O). MS (EI) m/z : 256 (M⁺, 85%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.40 (s, 3 H, CH₃), 2.72 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.4 - 7.6 (m, 3 H, Ar-H), 8.3 (s, 1 H, NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₃H₁₂N₄O₂ (256.26): C, 60.93; H, 4.72; N, 21.86; O, 12.49. Found: C, 60.68; H, 4.49; N, 21.52; O, 12.13.

(*E*)-*N*-(4,6-Dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)cinnamamide (7g)

Yield 66%, m.p. 110°C - 112°C, IR (KBr) ν (cm⁻¹): 3260 (NH), 1680 (C=O), 1580 (C=C). MS (EI) m/z : 292 (M⁺, 55%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.20 (s, 3 H, CH₃), 2.51 (s, 3 H, CH₃), 6.8 (s, 1 H, C5-H), 7.3 - 7.6 (m, 5 H, Ar-H), 8.2 (s, 1 H, NH, D₂O exchangeable), 8.3 (d, 1 H, CH), 8.6 (d, 1 H, CH), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₇H₁₆N₄O (292.34): C, 69.85; H, 5.52; N, 19.17; O, 5.47. Found: C, 69.64; H, 5.21; N, 19.42; O, 5.21.

N-(4,6-Dimethyl-1*H*-pyrazolo[3,4-*b*]pyridin-3-yl)-3-phenylpropanamide (7h)

Yield 64%, m.p. 128°C - 130°C, IR (KBr) ν (cm⁻¹): 3275 (NH), 1690 (C=O). MS (EI) m/z : 296 (M⁺, 56%), 297 (M⁺ + 1, 55%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.35 (s, 3 H, CH₃), 2.4 (q, 2 H, CH₂), 2.55 (s, 3 H, CH₃), 6.9 (s, 1 H, C5-H), 7.8 (t, 2 H, CH₂), 8.0 (t, 2 H, CH₂), 8.9 (s, 1 H, NH, D₂O exchangeable), 13.0 (s, H, NH, pyrazole, D₂O exchangeable). Anal. Calcd for C₁₁H₁₃BrN₄O (297.15): C, 44.46; H, 4.41; Br, 26.89; N, 18.85; O, 5.38. Found: C, 44.69; H, 4.71; Br, 26.63; N, 18.71; O, 5.52.

General procedure for the synthesis of compounds (8a-b)

A mixture of 3-amino pyrazolopyridine 4 (0.01 mol) and appropriate active methylene reagents, namely: acetylacetone, ethyl acetoacetate (0.01 mole of each) in DMF (30 mL), containing few drops of piperidine was refluxed the 3 h. After cooling, the solid product formed was collected by filtration and re-crystallized from DMF/EtOH (1:3) to afford 8a-b.

2,4,8,10-Trimethylpyrido[2',3':3,4]pyrazolo[1,5-*a*]pyrimidine (8a)

Yield 82%, m.p. 190°C - 192°C, MS (EI) m/z (%) 226 (M⁺, 75%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.16 (s, 3 H, CH₃), 2.25 (s, 3 H, CH₃), 2.37 (s, 3 H, CH₃), 2.54 (s, 3 H, CH₃), 6.7 (s, 1 H, C9-H), 7.82 (s, 1 H, C3-H). Anal. Calcd for C₁₃H₁₄N₄ (226.28): C, 69.00; H, 6.24; N, 24.76. Found: C, 69.26; H, 6.01; N, 24.53.

4-Hydroxy-2,8,10-trimethylpyrido[2',3':3,4]pyrazolo [1,5-a]pyrimidine (8b)

Yield 85%, mp 290°C - 292°C, IR (KBr) ν (cm⁻¹): 3340.45 (OH). MS (EI) m/z : 228 (M⁺, 67%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.24 (s, 3 H, CH₃), 2.35 (s, 3 H, CH₃), 2.46 (s, 3 H, CH₃), 5.82 (s, 1 H, OH, D₂O exchangeable), 6.7 (s, 1 H, C9-H), 7.62 (s, 1 H, C3-H). Anal. Calcd for C₁₂H₁₂N₄O (228.25): C, 63.15; H, 5.30; N, 24.55; O, 7.01. Found: C, 63.43; H, 5.58; N, 24.73; O, 7.29.

General procedure for the synthesis of compounds (10a-b)

To an ice-cooled solution of the 3-amino pyrazolopyridine **4** (0.01 mol) in AcOH (20 mL) and conc. HCl (3 mL) a cold solution of sodium nitrite (0.23 g in 5 mL of water) was added dropwise. The solution was stirred at room temperature for 3 h at 0°C - 5°C, and then treated with a solution of acetylacetone or ethyl cyanoacetate (0.01 mol) in the presence of sodium acetate (1 g) in ethanol (50 mL). The reaction mixture was stirred in an ice-cold bath for 1 h. The solid product obtained was filtered off, washed with water, and crystallized from ethanol to afford 10a-b.

4-Acetyl-3,8,10-trimethylpyrido[2',3':3,4]pyrazolo [5,1-c]-1,2,4-triazine (10a)

Yield 62%, m.p. 240°C - 242°C, IR (KBr) ν (cm⁻¹): 1760 (C=O). MS (EI) m/z : 256 (M⁺, 68%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 2.12 (s, 3 H, CH₃), 2.32 (s, 3 H, CH₃), 2.75 (s, 3 H, CH₃), 2.91 (s, 3 H, CH₃), 6.7 (s, 1 H, C9-H). Anal. Calcd for C₁₃H₁₃N₅O (255.28): C, 61.17; H, 5.13; N, 27.43; O, 6.27. Found: C, 61.32; H, 5.39; N, 27.75; O, 6.51.

Ethyl 3-amino-8,10-dimethylpyrido[2',3':3,4]pyrazolo [5,1-c]-1,2,4-triazine-4-carboxylate (10b)

Yield 74%, m.p. 160°C - 162°C, IR (KBr) ν (cm⁻¹): 3348, 3250 (NH₂), 1711 (C=O), 1260 (C-O). MS (EI) m/z : 286 (M⁺, 74.5%). ¹H NMR (DMSO-*d*₆) (500 MHz) δ (ppm): 1.5 (t, 3 H, CH₂-CH₃), 2.35 (s, 3 H, CH₃), 2.54 (s, 3 H, CH₃), 4.2 (q, 2 H, CH₂-CH₃), 5.4 (s, 2H, NH₂, D₂O exchangeable), 6.7 (s, 1 H, C9-H). Anal. Calcd for C₁₃H₁₄N₆O₂ (286.29): C, 54.54; H, 4.93; N, 29.35; O, 11.18. Found: C, 54.74; H, 4.75; N, 29.64; O, 11.31.

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