



Magnetic, Electrical and Magnetotransport Properties of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ Obtained by Pyrohydrolytic Method

A. V. Trukhanov^{1*}, S. V. Trukhanov¹, O. P. Smirnov², A. A. Kulikov², H. Szymczak³

¹Laboratory of Magnetic Films Physics, Scientific-Practical Materials Research Centre, Minsk, Belarus

²Petersburg Nuclear Physics Institute, Gatchina, Russia

³Institute of Physics of PAS, Warsaw, Poland

Email: truhanov86@mail.ru

Received 20 January 2015; accepted 5 February 2015; published 9 February 2015

Copyright © 2015 by authors and OALib.

This work is licensed under the Creative Commons Attribution International License (CC BY).

<http://creativecommons.org/licenses/by/4.0/>



Open Access

Abstract

The magnetic, electrical and magnetotransport properties of spinel-type $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ ceramic ferrites have been investigated. The results of the investigations temperature and field dependences of magnetization and resistivity are presented and analyzed. The data of neutron scattering were found to be in good agreement with the Neel model of ferrimagnetism. This compound is magnetomonophase ferrimagnetic in the wide range of temperatures, and shows transition in paramagnet at high temperature (~490 K). $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ is characterized by maximal value of magnetization (28 emu/g) in the range of $\text{Mg}(\text{Fe}_{1-x}\text{Ga}_x)_2\text{O}_{4+\delta}$ solid solutions. A negative magnetoresistance (MR) effect has been observed in $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ ferrite spinel. It has been determined that coefficient of the negative MR increased from 0.23% (at room temperature) to 7.28% (at 220 K). It can be explained with the help of the spin-dependent scattering on magnetic inhomogeneity of granular boundary.

Keywords

Ferrimagnetic Spinel Oxide, Magnetic Structure, Vibration Magnetometry, Neutron Powder Diffraction, Magnetotransport

Subject Areas: Inorganic Chemistry, Inorganic Nonmetallic Materials

1. Introduction

One of the main problems of microelectronics is integration of magnetic materials in microcircuit. Microelec-

*Corresponding author.

tronics is based on semiconducting materials (Si, Ge, GaAs, GaN, etc.). To solve this problem is to search semiconductor materials with ferromagnetic properties. There are a number of requirements for these materials: 1) homogeneity (single-phase); 2) Curie temperature of materials must be higher than 150°C (operating temperature range of microelectronic devices); 3) possibility of forming thin films on semiconducting substrates; 4) magnetoresistive effect. Material searching has some problems because most of homogeneous ferromagnetic (or ferrimagnetic) semiconductors (europium chalcogenides and oxides, cadmium-chromium spinels) are characterized of low Curie temperature [1]. Also these materials are mismatch with semiconducting materials. But it should be noted that there are many active studies in this area [2] [3].

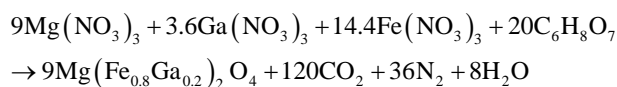
Ferrites constitute a technologically important class of metal oxides. Their high saturation moments and electrical resistance are exploited in high-frequency circuits, where they greatly reduce power losses from eddy currents. Other relevant uses of these materials are found in magnetic recording, microwave technology and heterogeneous catalysis. A key factor in all of these applications is the control of the relevant properties which can be exerted by means of intelligent design of chemical composition and crystal structure. Ferrites with general formula MFe_2O_4 , where M is a divalent cation, have the spinel-type structure, space group $Fd\bar{3}m$. Cation distribution among (tetrahedral) A and (octahedral) B sites is of paramount relevance for many applications of spinel ferrites, since many of their properties (e.g. magnetic, semiconducting, catalytic) depend on such an ionic partition. It is in this context where ternary spinels (which may be regarded as solid solutions between two binary spinels) are particularly amenable. Introduction of the third cation allows control of the ionic partition, and hence of the relevant properties, by means of appropriate choice of substituent cation and chemical composition. However, the extent to which cation distribution can be modified in this way is difficult to foresee, since site occupancy in ternary spinels cannot be predicted (in general) by the concept of site preference energy as applied to binary spinel. Despite the considerable amount of experimental and theoretical work already done on the crystal chemistry of spinel solid solutions, a careful investigation of new systems can still be very valuable for a better understanding of these complex systems.

It's observed that ferrite spinel properties are strongly depend on a number of requirements: method of producing, concentration substitution, size factor, anion nonstoichiometry, and defects. It is known that ferrite spinels as a rule are soft magnetic. They don't need large magnetic fields for their magnetization. It was reported that diamagnetic substitution could induce changes in ferrosinell magnetic structure [4] [5]. In one case, frustrated magnetic structure consists of separate spontaneously magnetized areas formed due to long-range and short-range magnetic order. In such magnetic structures, substitution induces a decrease (in absolute value) of the average AB interaction with respect to AA and BB sites. As a consequence, the competition between negative intrasublattice and intersublattice interactions becomes strong, which leads to the formation of a cluster spin-glass state and frustrated magnetic structure [6]. For example in [7], the authors noticed that composition $CuGa_{0.4}Al_{0.8}Fe_{0.8}O_4$ was characterized by isotropic magnetoresistive effect which increased with temperature decreasing. In the other case, substitution can induce of deviation in directions of magnetic moment vectors. The ferrites with canted B sites cations are called the YK-type ferrites. With specified concentration, the B site cations are no longer collinear. The spins of the B site cations are canted and make a certain angle [8]. In previously works [9] [10], the authors noticed that more perspective compound in the solid solutions range $Mg(Fe_{1-x}Ga_x)_2O_{4+\delta}$ ($0 < x < 1$) for investigation magnetoresistive effect is $Mg(Fe_{0.8}Ga_{0.2})_2O_{4+\delta}$. This compound is characterized by maximal values of magnetization ($M_S = 28$ emu/g), and conductivity ($\sim 10^{-8}$ Sm/m).

2. Experimental

Polycrystalline $Mg(Fe_{0.8}Ga_{0.2})_2O_{4+\delta}$ sample was prepared by pyrohydrolytic method [11] at 1300 K. To carry out the pyrohydrolytic reactions, we prepared solution containing citric acid and a mixture of magnesium, gallium, and iron nitrates. The composition of the initial compound was determined on the basis of thermogravimetry (an ULVAC SINKU-RICO TGD 7000 thermal analyzer). The initial metal molar ratio in solutions was Mg:Ga:Fe = 1:0.4:1.6. The solution prepared was evaporated under continuous stirring in a porcelain bowl on a hot plate until a dense, viscous mass was formed. Then, pyrolysis began from the bottom of the bowl and rapidly (in 1 - 2 min) propagated over the entire volume. The reaction mixture was periodically stirred to complete the pyrolysis. The process was finished only when the reaction ceased (as monitored visually). The precursors obtained in this way were ground and annealed at 1250 - 1300 K.

The pyrohydrolytic reaction of combustion of the nitrates + citric acid initial mixture is as follows:



During the process of production time and temperature of synthesis were decreased in comparison with method of solid-state reactions. Crystal structure and phase analysis of polycrystalline $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ sample was carried out on a DRON-3M diffractometer using CuK_α radiation in the angular interval $25^\circ < 2\theta < 70^\circ$. Phases were identified with the use of the JCPDS PDF2 database. The Powder-2 program package was used for indexing X-ray diffraction patterns and calculating unit cell parameters. The magnetic structure of the $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ was determined by neutron powder diffraction ($\lambda = 1.38 \text{ \AA}$, scanning step $\Delta\theta \sim 0.1^\circ$, angle range $10^\circ \leq 2\theta \leq 120^\circ$) on an reactor VVR-M in Gatchina. Magnetic properties (the temperature dependences of magnetization were measured in a field of 100 Oe in heating mode after zero field cooling (ZFC) and field cooling (FC), and Curie temperature were carried out a Helium Free Liquid system (by vibration magnetometry) in the 4.2 - 650 K temperature range temperature and field dependences of resistivity were investigated in the 200 - 300 K temperature range and magnetic field ranging 0 - 14 T. Energy of activation was calculated using the expression:

$$E_{act} = 2K_B \frac{\partial(\ln \rho)}{\partial(1/T)}$$

where ρ —resistivity; E_{act} —energy of activation; K_B —Boltzmann's const, T —temperature.

Coefficient of negative magnetoresistive effect was calculated using the relation:

$$M_R = \frac{R_B - R_0}{R_B} \times 100\%$$

R_B —value of resistivity in magnetic field (14 T), R_0 —value of resistivity without magnetic field.

3. Results and Discussions

3.1. Crystal Structure

The X-ray powder diffraction patterns of the $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ sample recorded at 25°C and 300°C [11]. The unit cell parameter a is $0.835(3) \text{ nm}$ at 25°C . At 300°C , this value is $0.837(2) \text{ nm}$ [11]. These data show that an increase in temperature considerably increases the unit cell parameter a of the spinel lattice. Apparently, temperature elevation leads to complex disordering processes in the crystal lattice of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ [10]. Data of X-ray diffraction show that this compound is single-phase and thermostable in temperature range until 590 K, which makes this compound perspective for semiconductor electronics (the temperature range of stability for the materials used for semiconducting electronics is from -50°C till 150°C [11]. The analysis of the X-ray diffraction patterns indicates that the sample is described well ($\chi^2 = 2.4$) by an orthorhombic ($\text{SG} = Fd\bar{3}m$) spinel-like unit cell.

3.2. Magnetic Structure and Magnetic Properties

Temperatures dependences (ZFC and FC at 100 Oe) of magnetization are practically coinciding at interval 4.2 - 300 K (see **Figure 1**). This proves of sample magnetic homogeneity. Temperatures dependences data of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ magnetization is evidence of thermal stability and absence of phase separation till 600 K.

From the temperature variation of the mass magnetization of the $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ ceramic (**Figure 1**) and its square (inset **Figure 1**), in the temperature range 300 - 650 K, the Curie temperature T_C of the ceramic was determined to be $\sim 490 \text{ K}$ (as curve bend of magnetization). The theoretical effective magnetic moment per Fe^{3+} ion ($3d^5$ electronic configuration) is $5 \mu_B/\text{f.u.}$, but the experimental data for the $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ ceramic give $1.4 \mu_B/\text{f.u.}$ (summarized magnetic moment) for Fe^{3+} (28% of the theoretical value). The possible causes of the discrepancy are as follows: 1) coexistence of two magnetic spinel phases (or probably frustrated magnetic structure as in case of $\text{CuGa}_{0.4}\text{Al}_{0.8}\text{Fe}_{0.8}\text{O}_4$), 2) noncollinear magnetic vectors within a single magnetic phase (sperimagnetism) as in case of ferrite Yaffet-Kittel type, and 3) compensation of the magnetic moments of the A- and B-sublattices of the spinel due to gallium substitution for Fe^{3+} ions on the octahedral site. We concluded that reduced magnetic moment per Fe^{3+} ion attest to the third possibility. It must be noticed that frustrated

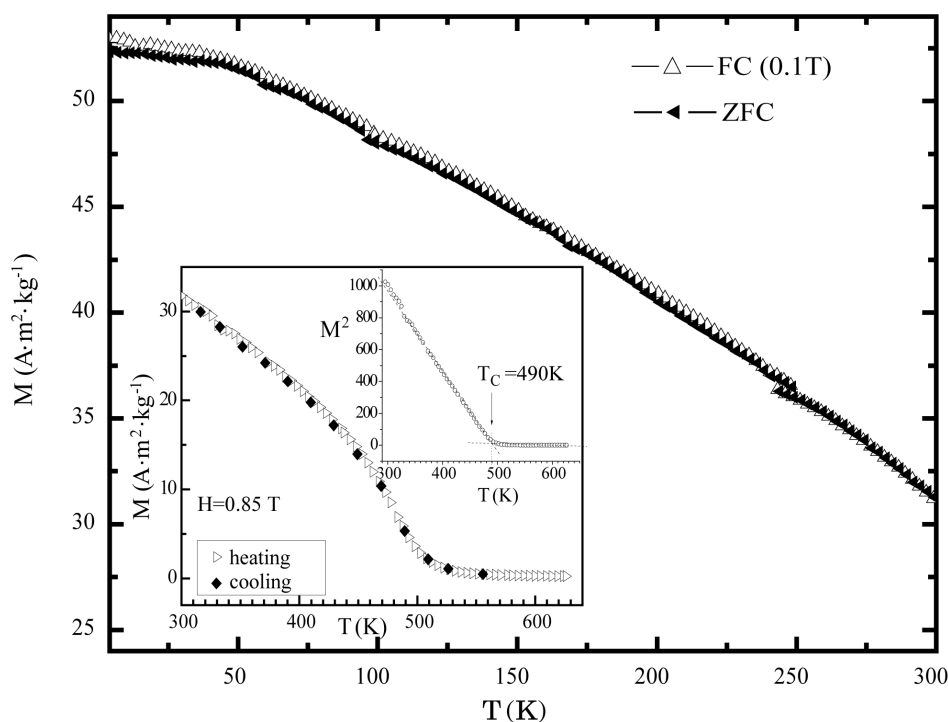


Figure 1. Temperature dependences of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ magnetization (insert-temperature interval 300 - 630 K).

magnetic structure or noncollinear magnetic vectors in spinel structure can be cause of magnetoresistance in ferrite like spinel.

Neutron diffraction data were collected on a 48-counter powder neutron diffractometer [12] using monochromatic neutrons with the wavelength $\lambda = 1.38 \text{ \AA}$ at the VVR-M reactor in Gatchina (Konstantinov Petersburg Nuclear Physics Institute, RAS). The volume of a powder ceramic sample was $\sim 0.3 \text{ cm}^3$. Measurements were carried out in a vanadium container at 290 K. The resulting neutron diffraction pattern for $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ showed diffraction reflections for interplanar spacings in the $d\text{hkl}$ range 0.8 - 7.0 \AA is shown in Figure 2. The unit cell parameters, oxygen atoms coordinates, magnetic moments per formula unit and the occupation numbers for Mg^{2+} and Fe^{3+} ions among tetrahedral and octahedral sites [13] were refined by the full-profile analysis method with the FULLPROF program [14] for space group $Fd\bar{3}m$ and are given in insert of Figure 2. Oxygen parameter is $u = 0.2556(2)$ as part of the lattice constant (distortion of the oxygen arrangement from cubic close-packing).

The distribution of chemical elements at A- and B-sites in this sample is $[\text{Mg}_{0.20(3)}\text{Fe}_{0.72(3)}](\text{Fe}_{0.44(3)}\text{Mg}_{0.40(3)}\text{Ga}_{0.2})_2\text{O}_4$. The orientation and magnitude of magnetic moments were determined simultaneously with the FULLPROF program. Effective magnetic moment of Fe^{3+} in tetrahedral coordination is $1.2 \mu_B$ and effective magnetic moment of Fe^{3+} in octahedral coordination is $1.85 \mu_B$. The average ferrimagnetic moment of Fe^{3+} (summarized magnetic moment of f.u.) is $1.53 \mu_B/\text{f.u.}$ that close to the value calculated according to vibration magnetometry data.

On the ground of neutron powder diffraction data (calculated with Fullprof) we concluded that diamagnetic substitution in $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ doesn't induce formation of frustrated magnetic structure as in case of $\text{Cu-Ga}_{0.4}\text{Al}_{0.8}\text{Fe}_{0.8}\text{O}_4$ and doesn't induce deviation from collinearity of magnetic vector's directions in $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ as in case of ferrite Yaffet-Kittel type.

3.3. Electrical and Magnetotransport Properties

Due to the high resistivity of the sample (at 300 K is approximately $5.9 \text{ M}\Omega\cdot\text{cm}$) and the limitation of current source of the measuring system, resistivity was measured between 300 and 220 K as shown in Figure 3. It is found that the resistivity increases (approximately on the 3 order of magnitude in temperature interval 300 - 220

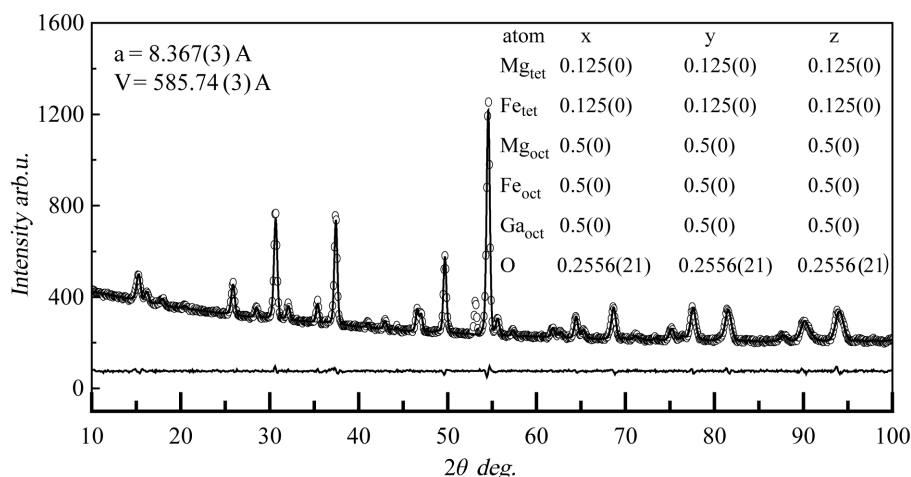


Figure 2. Neutron powder diffraction pattern of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ at room temperature.

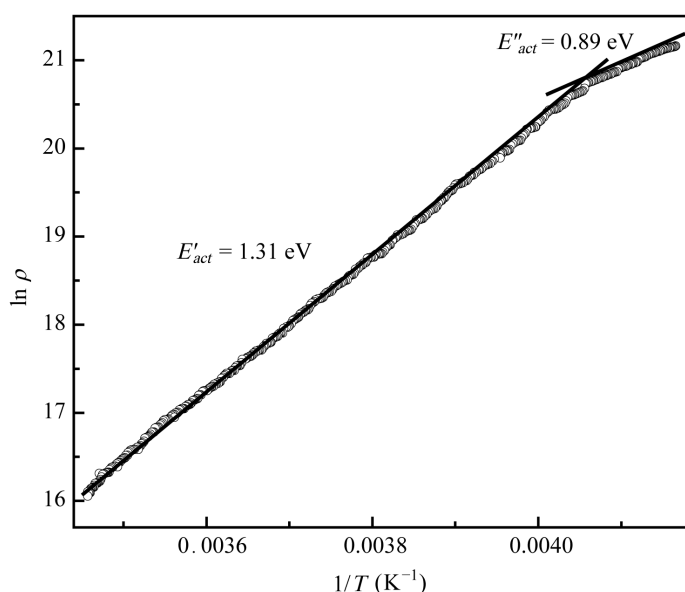


Figure 3. Temperature dependence of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ resistivity.

K) as the temperature decreases, and calculated E_{act} that corresponds energy of impurity activation. Behavior of resistivity $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ is characterized by semiconducting type of conductivity. As shown in **Figure 3** ρ at 300 K is equal $5.9 \text{ M}\Omega\cdot\text{cm}$, and ρ at 220 K $\sim 2.1 \text{ G}\Omega\cdot\text{cm}$.

The strong increase in resistivity with decreasing T can be understood in terms of a thermally activated hopping type transport mechanism of the itinerant electrons on the B-sublattice. It was determined that in this compound predominate impurity type of conductivity. From the hopping amplitude $t \sim 0.1 \text{ eV}$ an electron conduction bandwidth $\sim 1 \text{ eV}$ is expected. However, experimental data cannot be explained within a simple band theory. They can be better ascribed to some thermally activated motion of charge carriers. It have been shown that below 250 K energy of impurity activation is $E_{act}^2 \sim 0.89 \text{ eV}$, and above 250 K till 300 K energy of impurity activation is $E_{act}^1 \sim 1.31 \text{ eV}$. As has been shown by neutron diffraction data there isn't some spin canting between A- and B-sublattices. This hasn't been taken into account by an appropriate spin correlation factor leading to a spin-dependent part of the activation energy.

Behavior of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ magnetoresistance (MR) in the presence of 14 T field and at different temperatures was investigated (see **Figure 4**). In our case $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ we have intergranular contribution in MR. This contribution depends on spin-polarized carrier scattering on magnetic heterogeneity of granular

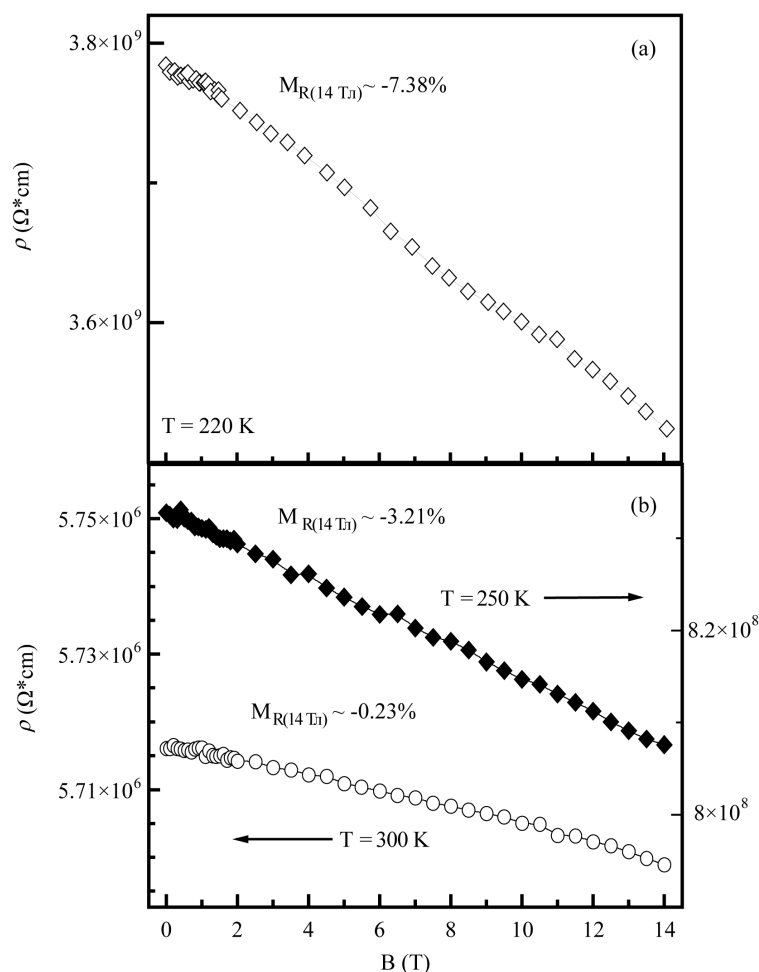


Figure 4. Field dependence of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ resistivity at various temperatures. (a) At 220 K; (b) At 300 K and 250 K.

boundary. This is due to multidirectional of crystallines magnetization vectors. For this sample 7.38% negative MR is observed at 220 K in the presence of 14 T field. Then MR coefficient is monotonic decreases with temperature increasing. At room temperature MR is 0.23%. It can be explained by decreasing of spin polarization coefficient of magnetic heterogeneity local spins in granular boundary.

Theoretically in case of thermally activated hopping-type conductivity we may estimate that coefficient of spin polarization deeply decreases with increasing semiconductor's band gap. Controlling of electron spin direction in this compound is an intricate problem. In case of $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ magnetic semiconductor low coefficient of magnetoresistance may be associated with low degree of electron spin polarization.

4. Conclusions

This compound $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ which was prepared by pyrohydrolytic method was magnetomonophase ferrite spinel with $T_C \sim 490 \text{ K}$. In the process of production, time and temperature of synthesis were decreased in comparison with method of solid-state reactions. It has been determined that diamagnetic substitution Fe^{3+} ions by Ga^{3+} ions in $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ doesn't induce formation of frustrated magnetic structure and doesn't induce deviation from collinearity of magnetic vector's directions. The distribution of chemical elements is $[\text{Mg}_{0.20(3)}\text{Fe}_{0.72(3)}](\text{Fe}_{0.44(3)}\text{Mg}_{0.40(3)}\text{Ga}_{0.2})_2\text{O}_4$. Effective magnetic moment of Fe^{3+} in tetrahedral coordination is $1.2 \mu_B$ and effective magnetic moment of Fe^{3+} in octahedral coordination is $1.85 \mu_B$. The average ferrimagnetic moment of Fe^{3+} calculated from neutron diffraction data is $1.53 \mu_B/\text{f.u.}$ which is close to the value calculated according to vibration magnetometry data ($1.4 \mu_B/\text{f.u.}$). Coefficients of the negative MR are increasing from

0.23% (at room temperature) to 7.28% (at 220 K). The nature of negative MR in $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ is caused by increasing spin polarization of local spins in the magnetic heterogeneity of granular boundary.

Highest values of magnetization and electroconductivity in the range of $\text{Mg}(\text{Fe}_{1-x}\text{Ga}_x)_2\text{O}_{4+\delta}$ solid solutions with the aggregate MR-phenomenon make this compound promising for practical application. By integrating such thin $\text{Mg}(\text{Fe}_{0.8}\text{Ga}_{0.2})_2\text{O}_{4+\delta}$ layers into spin-dependent tunnelling heterostructures, it can be demonstrated that this material can be useful for magnetic tunnel junctions or as a spin-filtering insulating barrier in the little explored type of tunnel junction called spin-filter.

References

- [1] Nagaev, E.L. (1988) Photoinduced Magnetism and Conduction Electrons in Magnetic Semiconductors. *Physica Status Solidi B*, **145**, 11-64. <http://dx.doi.org/10.1002/pssb.2221450102>
- [2] Ohno, H., Shen, A., Matsukura, F., Oiwa, A., Endo, A., Katsumoto, S. and Iye, Y. (1996) (Ga,Mn)As: A New Diluted Magnetic Semiconductor Based on GaAs. *Applied Physics Letters*, **69**, 363-365. <http://dx.doi.org/10.1063/1.118061>
- [3] Dietl, T., Ohno, H., Matsukura, F., Cibert, J. and Ferrand, D. (2000) Zener Model Description of Ferromagnetism in Zinc-Blende Magnetic Semiconductors. *Science*, **287**, 1019-1022. <http://dx.doi.org/10.1126/science.287.5455.1019>
- [4] Villain, J. (1979) Insulating Spin Glasses. *Journal of Physics B*, **33**, 31-42.
- [5] Poole, C.P. and Farach, H.A. (1982) Monte Carlo Study of the Local-Field Distribution in the Diluted Manganiferromagnetic Ising Model on the Triangular Lattice. *Physica B*, **47**, 55-59.
- [6] Coey, J.M.D. (1978) Amorphous Magnetic Order. *Journal of Applied Physics*, **49**, 1646-1655. <http://dx.doi.org/10.1063/1.324880>
- [7] Antoshina, L.G. (2001) The Behaviour of the Magnetostriction and Magnetoresistance of the Ferrite $\text{CuGa}_{0.4}\text{Al}_{0.8}\text{Fe}_{0.8}\text{O}_4$ with Frustrated Magnetic Structure. *Journal of Physics: Condensed Matter*, **13**, 127-133. <http://dx.doi.org/10.1088/0953-8984/13/5/103>
- [8] Akther Hossain, A.K.M., Tabata, H. and Kawai, T. (2008) Magnetoresistive Properties of $\text{Zn}_{1-x}\text{Co}_x\text{Fe}_2\text{O}_4$ Ferrites. *Journal of Magnetism and Magnetic Materials*, **320**, 1157-1162. <http://dx.doi.org/10.1016/j.jmmm.2007.11.009>
- [9] Nipan, G.D., Ketsko, V.A., Stognij, A.I., Trukhanov, A.V., Kol'tsova, T.N., Kop'eva, M.A., Elesina, L.V. and Kuznetsov, T.N. (2010) Inorganic Materials, Properties Solid of $\text{Mg}(\text{Fe}_{1-x}\text{Ga}_x)_2\text{O}_{4+\delta}$ Solutions in Stable and Metastable State. *Inorganic Materials*, **46**, 429-433. <http://dx.doi.org/10.1134/S0020168510040199>
- [10] Nipan, G.D., Ketsko, V.A., Stognij, A.I., Trukhanov, A.V., Kol'tsova, T.N., Beresnev, E.N., Kop'eva, M.A., Elesina, L.V. and Kuznetsov, T.N. (2010) DMS-Solutions $\text{Mg}(\text{Fe}_{1-x}\text{Ga}_x)_2\text{O}_{4+\delta}$. *Doklady Physical Chemistry*, **430**, 39-42. <http://dx.doi.org/10.1134/S0012501610020077>
- [11] Ketsko, V.A., Beresnev, E.N., Kop'eva, M.A., Rjabkova, L.V., Baranchikov, A.E., Stognij, A.I., Trukhanov, A.V. and Kuznetsov, N.T. (2010) Specifics of the Pyrohydrolytic and Solid-Phase Syntheses of Solid Solutions in the $(\text{MgGa}_2\text{O}_4)_x(\text{MgFe}_2\text{O}_4)_{1-x}$ System. *Russian Journal of Inorganic Chemistry*, **55**, 427-429. <http://dx.doi.org/10.1134/S0036023610030216>
- [12] Golosovskii, I.V., Kharchenkov, V.P., Bulkin, A.P., *et al.* (1988) Preprint LIYaF, No. 2, 1374.
- [13] Corliss, L.M. and Hastings, J.M. (1953) A Neutron Diffraction Study of Magnesium Ferrite. *Physical Review*, **90**, 1013-1018. <http://dx.doi.org/10.1103/PhysRev.90.1013>
- [14] Rodrigues-Carvajal, J. (2007) Program FULLPROF, Version 4.00. LLB CEA/Saclay, France.

Scientific Research Publishing (SCIRP) is one of the largest Open Access journal publishers. It is currently publishing more than 200 open access, online, peer-reviewed journals covering a wide range of academic disciplines. SCIRP serves the worldwide academic communities and contributes to the progress and application of science with its publication.

Other selected journals from SCIRP are listed as below. Submit your manuscript to us via either submit@scirp.org or **Online Submission Portal**.

