

Synthesis and antimicrobial activity of Ibuprofen derivatives

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ABSTRACT

In order to reduce the ulcerogenic effect of ibuprofen, its carboxylic group has been converted into 5-membered and 6-membered heterocyclic rings. Various 1,3,4-thiadaizo (y), imidazo [2,1-b] [1,3,4] thiazoles (1a-d), 1,3-oxazin-2-yl (2), quinazolin-2-yl (3, 4 and 5), hydrazine, pyrazole (6), pyrazone (7, 8) and oxadiazole (9, 12). Derivatives of ibuprofen were prepared by cyclization under various reaction conditions in a very good yield. The microbial inhibitory effect of the new agents has been assessed *in vitro* against Gram-positive and Gram-negative bacteria activity. Compounds (4, 5, and 8) showed the highest antibacterial activities compared to other compounds and standard drugs. However, compound (1b) showed the lowest antibacterial activity compared to other compounds. All compounds have been characterized by IR, ¹H NMR and C.H.N analysis.

KEYWORDS

Ibuprofen; Biological Active Compounds

1. INTRODUCTION

Ibuprofen is (2RS)-1[4-(2-methylpropyl) phenyl] propionic acid [1]. Ibuprofen was the first member of propionic acid derivatives to be introduced in 1969 as a better alternative to aspirin. Gastric discomfort, nausea and vomiting are still the most common side effects though less than aspirin or indomethacin [2,3].

Ibuprofen is effective in reducing high body temperature, and an anti-inflammatory which inhibits normal platelet function. Ibuprofen is reported to be better for joint and muscle pain than other pain killer and has been used

by people for arthritis for years. However, it can cause gastrointestinal upset and bleeding [4].

Ibuprofen is usually given as the free acid, also various salts, esters, and other complexes are also used. These include lysing and sodium salts, guaiacol and pyridoxine esters, iso-butanol ammonium and meglumine derivatives [5-7].

In order to reduce the ulcerogenic effect of ibuprofen, its carboxylic group has been converted into 5-membered heterocyclic rings. The cyclized derivatives were screened for their anti-inflammatory activity by the carrageenan induced rat paw edema method and showed 50% to 86% inhibition, whereas the started drug ibuprofen showed 92% inhibition at the same oral dose [8]. A recent literature survey revealed that different heterocyclic moieties have been widely used by the medicinal chemist in the past to explore its biological activities [9,10].

In the present study, we have aimed to achieve a better antimycobacterial profile at lower concentrations, by preparing ibuprofen derivatives. This report deals with the synthesis of ibuprofen derivatives and antimicrobial and antimycobacterial activities.

2. EXPERIMENTAL WORK

2.1. Instrumentation

Unless otherwise noted, materials were obtained from commercial suppliers and used without further purification, melting points were recorded on electrothermal melting point apparatus. Thin layer chromatography (TLC) controls were carried out on precoated silica gel plates (F254 Merck). The IR spectra were recorded on Shimadzu 8400 FT-IR infrared spectrophotometer using KBr pellets. ¹H NMR spectra were recorded on ¹H NMR Bruken ultra shield 300 MHz. Elemental analysis were carried out using Euro vector EA 3000 A Hally.

2.2. Synthesis of 2-Amino-5-(2-(4-Isobutylphenyl) Ethyl-1,3,4-Thiadiazole (y)

Compound (y) was obtained according to the procedure described by Al-Omar *et al.* [11] and Mazzone *et al.* [12] with a modification. A mixture of equimolar quantities of ibuprofen (0.1 mol), thiosemicarbazide (0.1 mol), and phosphorus oxychloride (30 ml) was refluxed gently for half an hour, after cooling water was added (90 ml) and the mixture was refluxed for 4 h and filtered, the precipitate was filtered and recrystallized from ethanol.

Yield: 80%, M.p. 205°C - 208°C, IR (KBr, cm⁻¹): 3593, 3120, 1630. ¹H NMR (δ, ppm, DMSO-d₆): 0.89 (d, 6H, (CH₃)₂), 1.61 (d, 3H, CH₃CH), 1.76 - 1.84 (m, 1H, CHCH₃), 2.55 (d, 2H, CH₂Ar), 3.56 (q, 1H, CH₃CH), 7 - 10 (m, 4H, ArH), 8.85 (s, 2H, NH₂).

2.3. Synthesis 2-(4-Isobutylphenyl) Ethyl-6-(4-Substitutoryl) Imidazo [2,1-b] [1,3,4] Thiadiazoles (1a-d) [13]

Compound (y) (0.01 mol) and an appropriate α-bromoketones (0.01 mol) was refluxed in dry ethanol (300 ml) for 10 h, excess of solvent was removed under reduced pressure the solid hydrobromide salts suspended in water, and neutralized by aqueous sodium carbonate solution to get free base, it was filtered, washed with water, dried and recrystallized from suitable solvent. 2-(4-Isobutylphenyl) ethyl-6-phenylimidazo [2,1-b] [1,3,4] thiadiazoles (1a) yield: 70%, M.p. 215°C - 217°C, IR (KBr, cm⁻¹): 3095, 3033, 2943, 1576, 1490, 1339, 1130, 843. ¹H NMR (δ, ppm, DMSO-d₆): 0.88 (d, 6H, (CH₃)₂), 1.61 (d, 3H, CH₃CH), 1.95 - 2.08 (m, 1H, CHCH₃), 2.59 (d, 2H, CH₂Ar), 3.56 (q, 1H, CH₃CH), 7.28 - 7.40 (m, 9H, ArH), 8.75 (s, 1H, H-5 imidazole).

2-(4-Isobutylphenyl)ethyl-6-(4-methylphenyl)imidazo [2,1-b] [1,3,4] thiadiazoles (1b) yield: 73%, M.p. 196°C - 198°C, IR (KBr, cm⁻¹): 3085, 3043, 2973, 1596, 1481, 1343, 1150, 843. ¹H NMR (400 MHz, δ, ppm, DMSO-d₆): 0.83 (d, 6H, (CH₃)₂), 1.46 (d, 3H, CH₃CH), 1.12 - 1.51 (m, 1H, CHCH₂), 2.49 (d, 2H, CH₂Ar), 3.52 (q, 1H, CH₃CH), 7.08 - 7.18 (m, 4H, ArH), 8.15 (d, 2H, Ar), 8.30 (d, 2H, Ar), 8.74 (s, 1H, H-5 imidazole).

2-(4-Isobutylphenyl) ethyl-6-(4-methoxyphenyl) imidazo [2,1-b] [1,3,4] thiadiazoles (1c) yield: 70%, M.p. 216°C - 218°C, IR (KBr, cm⁻¹): 3085, 3043, 2973, 1596, 1481, 1343, 1150, 843. ¹H NMR (400 MHz, δ, ppm, DMSO-d₆): 0.88 (d, 6H, (CH₃)₂), 1.29 (d, 3H, CH₃CH), 2.36 - 2.39 (m, 1H, CHCH₃), 2.43 (d, 2H, CH₂Ar), 3.76 (q, 1H, CH₃CH), 7.11 - 7.26 (m, 4H, ArH), 7.40 (m, 5H, ArH), 8.75 (s, 1H, H-5 imidazole).

2-(4-Isobutylphenyl) ethyl-6-(4-nitrophenyl) imidazo [2,1-b] [1,3,4] thiadiazoles (1d) yield: 70%, M.p. 205°C - 210°C, IR (KBr, cm⁻¹): 3060, 3021, 2959, 1587, 1494,

1346, 1128. ¹H NMR (400 MHz, δ, ppm, DMSO-d₆): 0.86 (d, 6H, (CH₃)₂), 1.80 (d, 3H, CH₃CH), 1.95 - 2.08 (m, 1H, CHCH₃), 2.47 (d, 2H, CH₂Ar), 3.76 (q, 1H, CH₃CH), 7.16 - 7.37 (m, 4H, ArH), 7.14 - 7.31 (m, 4H, ArH), 8.40 (d, 2H, ArH), 8.16 (d, 2H, ArH), 8.85 (s, 1H, H-5 imidazole)

2.4. Synthesis 1-(4-Oxo-4H-Benzo [d] [1,3] Oxazin-2-yl) Ibuprofen (2)

A solution of acid chloride (5 mmol) in dry pyridine (30 ml), was refluxed for 3h the reaction mixture was cooled and poured into cold diluted HCl, the solid that separated filtered off and recrystallized from benzene to give (2), yield: 77%, M.p. 202°C - 204°C, IR (KBr, cm⁻¹): 3044, 2920, 2850, 1745, 1600. ¹H NMR (δ, ppm, DMSO-d₆): 0.89 (d, 6H, (CH₃)₂), 1.31 (d, 3H, CH₃CH), 1.76 - 1.82 (m, 1H, CHCH₃), 2.43 (d, 2H, CH₂Ar), 3.71 (q, 1H, CH₃CH), 7.15 - 7.66 (m, 4H, ArH) for ibuprofen, 7.0 - 7.3 (m, 4H, ArH) for anthranilic acid.

2.5. Synthesis 1-(4-Oxo-4H-Benzo [d] [1,3] Oxazin-2-yl) Ibuprofen (2)

A solution of acid chloride (5 mmol) in dry pyridine (30 ml), was refluxed for 3 h the reaction mixture was cooled and poured into diluted HCl, the solid that separated filtered off and recrystallized from benzene to give (2), yield: 77%, M.p. 202-204°C, IR (KBr, cm⁻¹): 3044, 2920, 2850, 1745, 1600. ¹H NMR (δ, ppm, DMSO-d₆): 0.89 (d, 6H, (CH₃)₂), 1.31 (d, 3H, CH₃CH), 1.76 - 1.82 (m, 1H, CHCH₃), 2.43 (d, 2H, CH₂Ar), 3.71 (q, 1H, CH₃CH), 7.15 - 7.66 (m, 4H, ArH), for ibuprofen, 7.0-7.3 (m, 4H, ArH) for anthranilic acid.

2.5.1. Synthesis of 1-(3-Oxo-3,4-Dihydroquinazolin-2-yl) Ibuprofen (3)

A mixture of (3) (6 mmol) and ammonium acetate (6 mmol) was fused in an oil bath for 2 h, then poured into water, the separation solid after concentration and cooling was filtered off and recrystallized from ethanol to give (4), yield 66%, M.p. 217°C - 219°C, IR (KBr, cm⁻¹): 3325, 2910, 2850, 1589, 1349.

2.5.2. Synthesis of 1-(3-Thioxo-3,4-Dihydroquinazolin-2-yl) Ibuprofen (4)

A mixture of (4) (5 mmol) and phosphorus pentasulfide (5 mmol) in dry xylene (50 ml) was heated under reflux for 1h, then concentration the solution, the product was obtained by filtration and recrystallized from xylene to give (5), yield 68%, M.p. 222°C - 224°C, IR (KBr, cm⁻¹): 3260, 2920, 2850, 1587, 1325. ¹H NMR (δ, ppm, DMSO-d₆): 0.89 (d, 6H, (CH₃)₂), 1.48 (d, 3H, CH₃CH), 1.12 - 1.51 (m, 1H, CHCH₃), 2.55 (d, 2H, CH₂Ar), 3.56 (q, 1H, CH₃CH), 7.10 - 7.18 (m, 4H, ArH), 6.5 - 7.2 (m, 4H, ArH), 9.1 (s, 1H, NH, exchangeable).

2.5.3. Synthesis of 1-[4-(5-Amino [1,3,4] Thiadiazol-2-yl) Quinazolin-2-yl] Ibuprofen (5)

A mixture of (5) (5 mmol) and thiosemicarbazide (5 mmol) and phosphorus oxychloride (5 mmol) was warmed at 60°C for 1 h and the temperature was raised to 90°C for additional 2 h, the contents were poured onto crashed ice, cooled to 10°C, pH adjusted to 8 - 10 M NaOH and the resulting solid was crystallized from DMF to give (6), yield 63%, M.p. 231°C - 233°C, IR (KBr, cm⁻¹): 3200, 1620, 1210, 1075, 980. ¹H NMR (δ, ppm, DMSO-d₆): 0.98 (d, 6H, (CH₃)₂), 1.53 (d, 3H, CH₃CH), 1.74 - 1.78 (m, 1H, CHCH₂), 2.39 (d, 2H, CH₂Ar), 3.56 (q, 1H, CH₃CH), 7.16 - 7.37 (m, 4H, ArH), 6.2 - 7.2 (m, 4H, ArH), 8.81 (s, 1H, NH₂).

2.6. Synthesis of Ibuprofen Hydrazide (z)¹⁴

A mixture of ibuprofen (0.03 mol) and phosphorus pentachloride (0.07 mol) in a hydrous carbon tetrachloride (20 ml) were refluxed for 2 h at 100°C, the solvent was distilled off for further reaction without any purification. To the ibuprofenoyl chloride (0.03 mol) was added hydrazine hydrate (0.1 mol) dropwise below 5°C and the resultant mixture was stirred for 5 h at room temperature. A solid that separated out was washed with aqueous NaHCO₃ (10%) and dried in vacuum, it was recrystallized from methanol to obtain pure crystalline solid z, yield: 76%, M.p. 188°C - 200°C, IR (KBr, cm⁻¹): 3335 (NH, NH₂), 1345 (CN), 1628 (C=N), 1680 (C=O).

2.7. Synthesis of Compounds (6-8, 9) [15]

To a solution of (z) (0.001 mol) in ethanol (10 ml) appropriate diketone (ethylacetoacetate, acetylacetone, ethylcyanoacetate) (0.002 mol) was added and the reaction mixture was refluxed on a water bath for 12 h in presence of catalytic amount of glacial acetic acid (2 - 3 drops), the reaction contents were cooled to room temperature and the obtained product (7, 8, 9) was filtered, dried and purified by recrystallization from ethanol.

2.7.1. Synthesis of 5-[2-(4-Isobutylphenyl) Ethanone]-3, 5-Dimethyl-1H-Pyrazol (6)

Yield 71%, M.p. 265°C - 266°C, IR (KBr, cm⁻¹): 1683 (C=O), 1622 (C=N), 1350 (C-N). ¹H NMR (δ, ppm, DMSO-d₆): 0.87 (d, 6H, (CH₃)₂), 1.29 (d, 3H, CH₃CH), 1.76 - 1.85 (m, 1H, CHCH₃), 2.42 (d, 2H, CH₂Ar), 3.74 (q, 1H, CH₃CH), 7.15 - 7.66 (m, 4H, ArH), 2.76 (s, 3H, CH₃ pyrazole), 5.9 (s, 1H, CH pyrazole).

2.7.2. Synthesis of 5-[2-(4-Isobutylphenyl) Ethanone]-3-Methyl-1H-Pyrazol-5(4H)-One (7)

Yield 76%, M.p. 240°C - 242°C, IR (KBr, cm⁻¹): 1722,

1718 (C=O), 1615 (C=N). ¹H NMR (δ, ppm, DMSO-d₆): 0.88 (d, 6H, (CH₃)₂), 1.26 (d, 3H, CH₃CH), 1.78 - 1.85 (m, 1H, CHCH₂), 2.43 (d, 2H, CH₂Ar), 3.62 (q, 1H, CH₃CH), 7.08 - 7.26 (m, 4H, ArH), 0.9 (s, 3H, CH pyrazole).

2.7.3. Synthesis of 5-[2-(4-Isobutylphenyl) Ethanone]-3-Amino-1H-Pyrazol-5(4H)-One (8)

Yield 66%, M.p. 214°C - 216°C, IR (KBr, cm⁻¹): 3425 (NH), 1710 (C=O), 1646 (C=N), 1330 (C-N). ¹H NMR (δ, ppm, DMSO-d₆): 0.83 (d, 6H, (CH₃)₂), 1.28 (d, 3H, CH₃CH), 1.78 - 2.35 (m, 1H, CHCH₂), 2.99 (d, 2H, CH₂Ar), 3.70 (q, 1H, CH₃CH), 7.10 - 7.18 (m, 4H, ArH), 2.06 (s, 2H, NH₂), 2.4 (s, 2H, CH₂).

2.8. Synthesis of 5-[2-4-Isobutylphenyl) Ethyl]-5- Substituted-1, 3, 4-Oxadiazole (9-12)

A mixture of x (0.001 mol), substituted aromatic acids (0.002 mol) and phosphorus oxychloride (15 ml) was refluxed on oil bath at 100°C - 110°C for 6 h, the excess of phosphorus oxychloride was distilled off and cooled residue was poured into ice cold. The contents were neutralized with ammonia to offered crude products (15 - 18), which were filtered, dried and purified by recrystallization from 1,4-dioxane.

(9) Yield 76%, M.p. 260°C - 262°C, IR (KBr, cm⁻¹): 1605, 1551 (C=N). ¹H NMR (δ, ppm, DMSO-d₆): 0.89 (d, 6H, (CH₃)₂), 1.46 (d, 3H, CH₃CH), 2.35 - 2.38 (m, 1H, CHCH₂), 2.59 (d, 2H, CH₂Ar), 3.77 (q, 1H, CH₃CH), 7.12 - 8.82 (m, 9H, ArH).

(10) Yield 56%, M.p. 263°C - 265°C, IR (KBr, cm⁻¹): 1635, 1578 (C=N), 1148 (C-O-C). ¹H NMR (δ, ppm, DMSO-d₆): 0.86 (d, 6H, (CH₃)₂), 1.48 (d, 3H, CH₃CH), 2.31 - 2.36 (m, 1H, CHCH₂), 2.43 (d, 2H, CH₂Ar), 3.40 (q, 1H, CH₃CH), 7.11 - 8.26 (m, 9H, ArH), 3.76 (s, 2H, CH₂).

(11) Yield 65%, M.p. 281°C - 283°C, IR (KBr, cm⁻¹): 1686-1599 (C=N), 1150 (C-O-C), 714 (C-Br). ¹H NMR (δ, ppm, DMSO-d₆): 0.91 (d, 6H, (CH₃)₂), 1.46 (d, 3H, CH₃CH), 1.95 - 2.08 (m, 1H, CHCH₂), 2.51 (d, 2H, CH₂Ar), 3.59 (q, 1H, CH₃CH), 7.16 - 8.32 (m, 8H, ArH).

(12) Yield 63%, M.p. 223°C - 225°C, IR (KBr, cm⁻¹): 1665-1585 (C=N), 1155 (C-O-C). ¹H NMR (δ, ppm, DMSO-d₆): 0.92 (d, 6H, (CH₃)₂), 0.77 (d, 3H, CH₃CH), 1.76 - 1.84 (m, 1H, CHCH₂), 2.57 (d, 2H, CH₂Ar), 3.70 (q, 1H, CH₃CH), 7.44 - 8.83 (m, 8H, ArH).

3. RESULTS AND DISCUSSION

2-amino-5-(4-i-butylphenyl)ethyl-1,3,4-thiadiazole (y) was obtained by direct cyclization of a ibuprofen and thi-

osemicarbazide in the presence of phosphorus oxychloride, the latter reflexed with substituted α -haloarylketones in dry ethanol yielded the imidazothiadiazole (1a-d) in good yield. It is well established that this reaction proceeds via the intermediate iminothiadiazole [4], which undergoes dehydrocyclization to form the desired fused heterocycle under reflux temperature spontaneously. The electronic and steric factors at 5th position of 2-amino-5-substituted-1,3,4-thiadiazole are crucial in determining the course of its reaction with substituted α -haloarylketones the strongly electronegative groups impart less nucleophilic character to the nitrogen at 4th position of the 1,3,4-thiadiazole. Various α -haloarylketones were prepared by the bromination of corresponding ketones.

The formation (y) by the reaction between ibuprofen and thiosemicarbazide was confirmed by IR spectra, which showed the presence of amino ($-\text{NH}_2$) band and absence of carbonyl stretching of carboxylic acid. The structure of imidothiazole derivatives (1a-d) were established by the absence of amino ($-\text{NH}_2$) band in IR spectra and appearance of imidazole proton (H-5) around δ 8.6 in the ^1H NMR spectra, the synthetic route of the compounds (y and 1a-d) is outlined in **Figure 1**.

The required 1-(4-oxo-4H-benzo[d][1,3]oxazin-2-yl) ibuprofen (2) was synthesized by refluxing chloroibuprofen (x) with anthranilic acid in boiling pyridine. The reactivity of nitrogen nucleophiles was investigated aiming for synthesis of novel heterocyclic synthesis of interest application. Thus, the reaction of (z) with ammonium acetate afforded quinazolinone derivatives (3), the interaction of 3 with P_2S_5 in dry xylene afforded quinazolin-3H-4-thione derivative (4). In the present investigation, the consideration of 4 with thiosemicarbazide using phosphorus oxychloride [14] gave (5) (**Figure 2**). For the synthesis of the compounds (6 - 12), the sequence of reactions shown in **Figure 3**. The reaction of ibuprofen hydrazide (z) with appropriate diketone in ethanol in the presence of glacial acetic acid. The structures of compounds (6 - 8) were established by means of IR, ^1H NMR

and elemental analysis. The C.H.N. analysis data for some of the synthesized compounds is given in **Table 1**. The infrared spectra display a $\text{C}=\text{O}$ stretching on at ($1700 - 1775 \text{ cm}^{-1}$) and the presence of bands at ($1551 - 1686 \text{ cm}^{-1}$) for $\text{C}=\text{N}$ and disappearance of (NH_2) bands, furthermore, the ^1H NMR spectra showed disappeared NH_2 and the presence of singlet at δ 2.2 ppm for CH_2 (pyrazole) ring. The oxadiazole derivatives (9) were proved by IR which showed absorption peak at (1560 cm^{-1}) due to $\text{C}=\text{N}$ Stretch vibration, its structure was further supported by its ^1H NMR spectra, which showed the NH_2 disappeared and multiplet was obtained in the aromatic region at δ ppm for sub. ArH protons. The synthesized compounds (6 - 12) were recorded in **Figure 3**.

Biological Activity

Since the prepared ibuprofen derivatives in this work were built from two biologically active components (5 and 6-membered heterocyclic rings) they were expected to possess biological activity, thus studies on the antibacterial activity of synthesized ibuprofen derivatives have been carried out against two pathogenic organisms including *Staphylococcus aureus* (Gram positive) and *Escherichia coli* (Gram negative) using cup-plate method. The results of the antibacterial studies are shown in **Table 2**. It was noticeable that the nature of substituted on heterocyclic molecules affected their biological activities against the studied bacteria 15. Thus, among the tested compounds (1d, 11, 12) which were substituted with (Cl or NO_2) groups showed high biological activity against *Escherichia coli* and slight to moderate activity against *Staphylococcus aureus*. Compound (1c) which was substituted with (OCH_3) group showed high biological activity against *Staphylococcus aureus* while it showed slight activity against the bacteria.

All compounds other showed high activity against Gram positive bacteria and Gram negative bacteria represented by large inhibition zones.

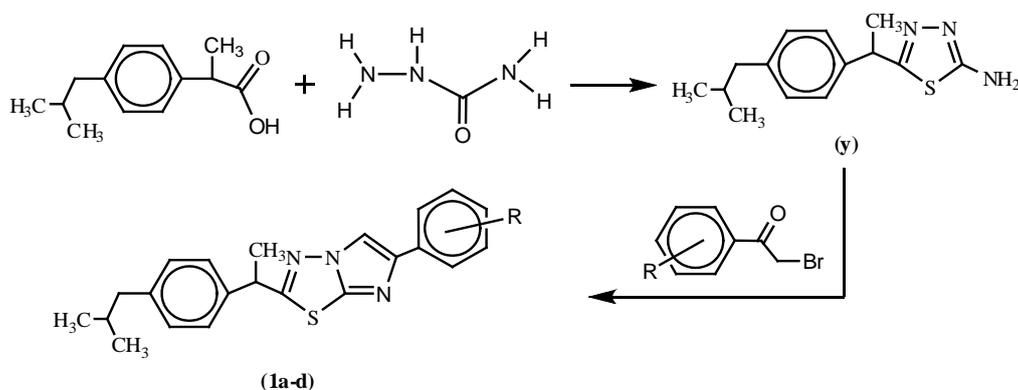


Figure 1. Synthesis compounds (1a-1d).

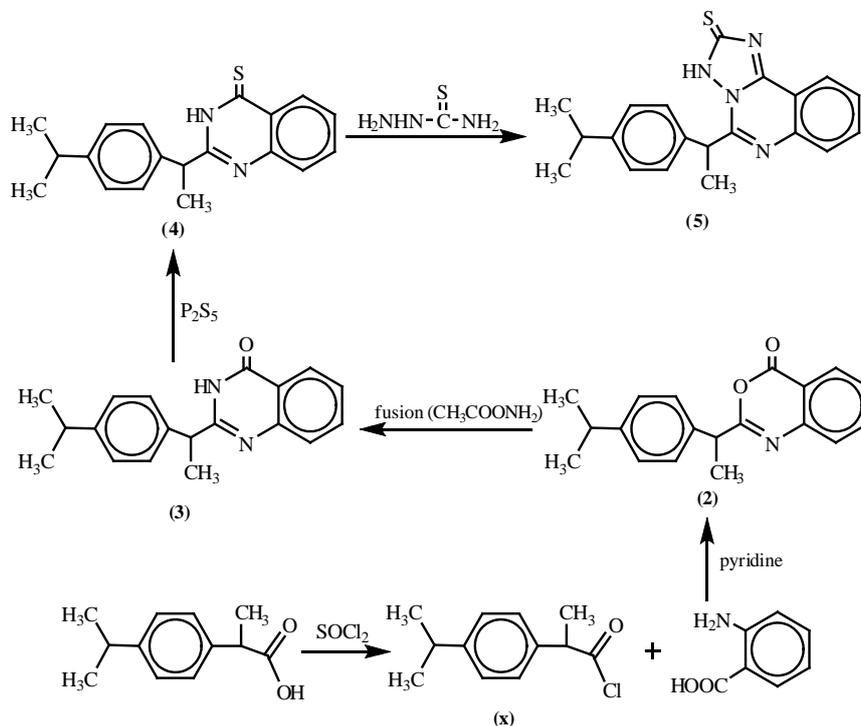


Figure 2. Synthesis compounds (2 - 5).

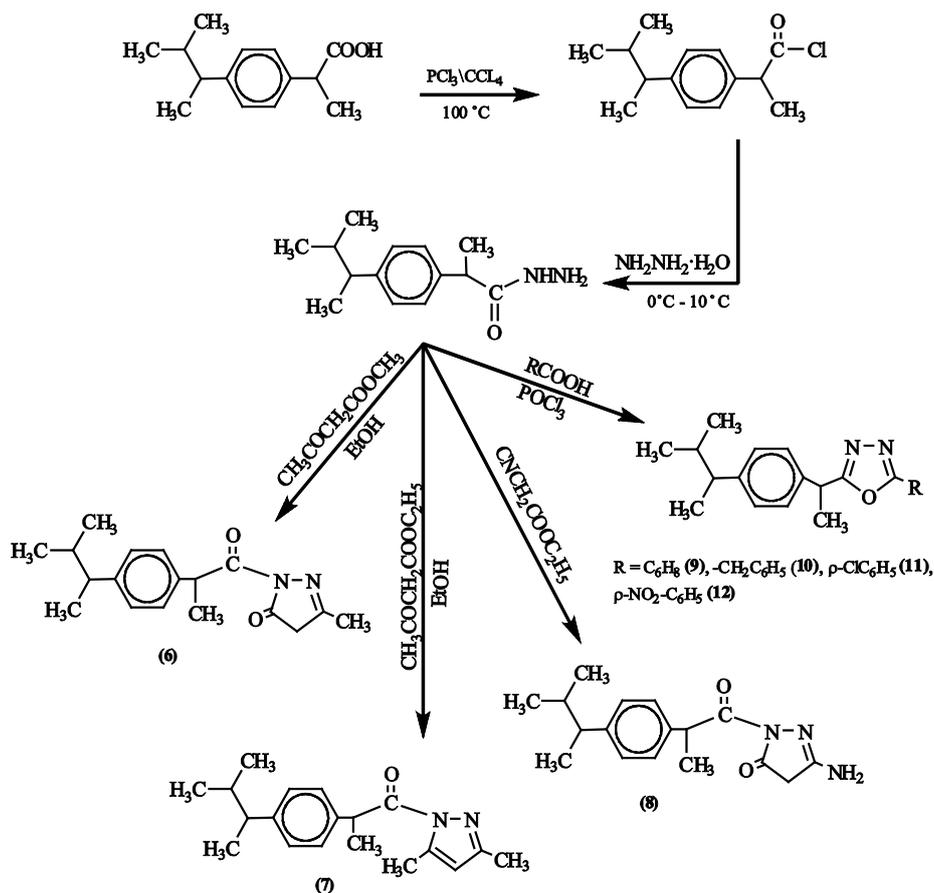


Figure 3. Synthesis compounds (6 - 12).

Table 1. C.H.N analysis for some of the prepared compounds.

Compounds No.	Calculated			Found		
	C %	H %	N %	C %	H %	N %
y	63.560	7.630	16.880	63.601	7.528	16.747
1a	85.496	5.852	10.687	85.328	5.648	10.625
2	78.175	7.840	4.560	78.096	7.839	4.426
4	74.534	5.832	8.695	74.436	5.823	8.657
5	69.613	6.077	15.469	69.626	6.135	15.451
6	72.483	7.382	9.395	72.327	7.355	8.974
8	63.878	7.984	10.969	63.903	7.792	10.958
9	78.931	7.189	9.150	78.377	7.058	9.035

Table 2. Antibacterial activity of compounds (1, 1a-d) and (2-12).

Compounds No.	Gram positive bacteria	Gram negative bacteria
	Staphylococcus aureus	Escherichia Coli
1	++	++
1a	++	+++
1b	+	++
1c	+++	+
1d	+	+++
2	++	++
3	++	++
4	+++	+++
5	+++	+++
6	++	++
7	++	++
8	+++	+++
9	++	++
10	++	++
11	+	+++
12	+	+++

4. CONCLUSION

The present work reports a new investigation of heterocyclic systems (oxadiazole, oxazole, thiadiazol, quinoxaline, pyrazole) derived from Iubprofen. And the systems were prepared with the objective of developing better antibacterial molecules. Among these, compounds (4, 5, and 8) showed maximum activity compared to other compounds. The other prepared compounds also showed maximum activity compared to standard drugs.

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