

Effect of Retrogression and Reaging on Stress Corrosion Cracking of Spray Formed Al Alloy^{*}

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Received 19 November 2015; accepted 25 January 2016; published 28 January 2016

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Abstract

Stress corrosion cracking (SCC) resistance of a spray formed Al-Zn-Mg-Cu alloy underwent retrogression and reaging (RRA) was studied by slow strain rate tests in dry air and 3.5 wt% NaCl solution. The results showed that after RRA treatment, interrupted η phases at grain boundaries and slightly wide precipitate free zones could decrease SCC susceptibility of the alloy. Lots of reticular dislocations appeared in deformation process could prevent hydrogen induced cracking, and then SCC. Abundance transgranular dispersive η' phases separated out again promoted tensile strength to 759.4 MPa. The fracture ways of the specimens were dimple fracture in dry air and sub-cleavage fracture in 3.5% NaCl solution.

Keywords

Spray Forming, Al-Zn-Mg-Cu Alloy, Retrogression and Reaging, Stress Corrosion Cracking

1. Introduction

Al-Zn-Mg-Cu alloys are precipitation hardening alloys, with a vast number of applications in the aerospace industry [1] [2]. Because of limitation on traditional casting process, the strength of the alloys hardly surpasses 700 MPa [3] [4]. Until spray forming has been used on the alloys, the strengths are elevated to over 750 MPa [5] [6].

Such as structural material, it is not only considered the strength and toughness, but also focused on stress corrosion cracking (SCC) resistance [7]-[10]. In regard to the SCC mechanisms on Al-Zn-Mg-Cu alloys, there

How to cite this paper: Su, R.M., Qu, Y.D., Li, X., You, J.H. and Li, R.D. (2016) Effect of Retrogression and Reaging on Stress Corrosion Cracking of Spray Formed AI Alloy. *Materials Sciences and Applications*, **7**, 1-7. http://dx.doi.org/10.4236/msa.2016.71001

^{*}This research was financially supported by the National Natural Science Foundation of China (51574167) and Program for Innovative Research Team in University of Liaoning Province (LT2015020).

are anodic dissolution theory [11], hydrogen induced cracking (HIC) theory [12] and "Mg-H" theory [13] [14]. For these studies and theories, there are some reports of heat treatment on SCC resistance of Al-Zn-Mg-Cu alloy [15]-[17]. Especially, the effects of retrogression and reaging (RRA) treatment on SCC resistance are complete-ly remarkable [18] [19]. But the alloys are prepared with traditional casting process in above-mentioned reports. Some papers [20]-[23] report the effect of aging treatment on spray formed alloy, but the SCC resistance is not involved.

Some differences on microstructure of alloy (e.g. solid solubility or grain size) between spray forming and traditional casting have been proved by results [24]-[26]. Those differences also influence SCC resistance after RRA treatment and the regularity need to study.

To offer a data for optimizing aging treatments on spray forms Al-Zn-Mg-Cu alloy and reference for next step research, this paper studies of RRA treatment on microstructure and SCC resistance of spray form 7075 alloy with scanning electron microscope (SEM), transmission electron microscope (TEM) and slow strain rate tests, compared with T6 and T73 treatments, and discuss results.

2. Experimental

The 7075 aluminum alloy with alloying elements of 5.48 wt% Zn, 2.21 wt% Mg, 1.48 wt% Cu, 0.189 wt% Cr, 0.371 wt% Fe and 0.121 wt.% Si was sprayed with atomization gas of nitrogen (N₂), spray distance of 370 ~ 380 mm, substrate eccentricity of 60 ~ 65 mm, conduit bore of 3.6 mm, incidence angle of $37^{\circ} \sim 39^{\circ}$, spray temperature of 770°C ~ 780°C, crucible temperature of 735°C ~ 745°C, horizontal velocity of 0.15 mm/s, and vertical velocity of 0.18 mm/s. Then resulting bar was extruded at temperature of 400°C, ratios of 30:1 and feeding rate of 1.5 mm/s.

The test samples were cut from the as-extruded bar for the two-stage solid solution, *i.e.*, 450°C for 1 h and 475°C for 2 hrs, respectively, after which the samples were water quenched to room temperature. Samples were aged at 120°C for 24 hrs as T6 treatment, retrogressed at 200°C for 10min and re-aged at 120°C for 24 hrs (RRA). In addition, others samples were aged as T73 treatment (at 120°C for 8 hrs and 160°C for 16 hrs) for comparison.

Slow strain rate tested by SCC-1 stress corrosion experimental system refer to international standard ISO 7539 - 7:2005, specimen sizes are shown as **Figure 1**, strain rate was 10^{-6} s⁻¹ in dry air or 3.5 wt% NaCl solution at $35^{\circ}C \pm 1^{\circ}C$ until cracking. Appearances of fracture were observed with S-3400 N scanning electron microscope, operating at 15 kV. The 3 mm diameter disks for TEM observation were punched out directly from samples which were mechanically ground down to 60 µm thickness after aging. These disks were electropolished using a DJ-2000 twin-jet electropolisher with a 30% nitric acid solution in methanol at $-30^{\circ}C$. TEM examinations were performed using a JEM-2100 transmission electron microscope.

3. Results

To judge the SCC resistance, stress corrosion index I_{SSRT} was defined by processing various mechanical properties with slow strain rate test.

$$I_{\text{SRRT}} = 1 - \frac{\sigma_{\text{NaCl}} \times (1 + \delta_{\text{NaCl}})}{\sigma_{\text{air}} \times (1 + \delta_{\text{air}})}.$$
(1)

where, δ_{NaCl} is elongation in 3.5 wt% NaCl solution (%), δ_{air} is elongation in dry air (%), σ_{NaCl} is tensile strength in 3.5 wt% NaCl solution (MPa), σ_{air} is tensile strength in dry air (MPa). The SCC resistance increases with the I_{SSRT} close to 0. The slow strain rate test properties of spray formed 7075 alloy after various aging treatments are listed in **Table 1**.

From **Table 1**, it can be seen that there are some differences on three aging treatments in dry air. After T6 treatment, strength maintains at a high level, the tensile strength of the alloy is 777.2 MPa, but the elongation is



Figure 1. Sizes of the specimens for slow strain rate tests (mm).

Table 1. Properties of the alloy after various aging treatments with slow strain rate tests.					
Aging treatment —	σ_b/MPa		$\delta / \%$		I
	air	NaCl	air	NaCl	ISSRT
Τ6	777.2	642.1	6.24	4.22	0.404
T73	676.5	660.8	8.48	7.95	0.078
RRA	759.4	729.5	8.36	7.51	0.125

only 6.24%. After T73 treatment, the elongation rises to 8.48%, but the tensile strength of the alloy is only 676.5 MPa which reduced 13% than T6. The contradiction on tensile strength and elongation between T6 and T73 treatments was remitted by RRA treatment. The tensile strength of the alloy is 759.4 MPa which is closed to T6, the elongation is 8.36% which is near to T73.

The slow strain rate test results in 3.5% NaCl solution are also different. The elongation of the specimen after T6 treatment is 4.22% which is the least among three aging treatments and the tensile strength lost seriously remains 642.1 MPa. After T73 treatment, the tensile strength and elongation of specimens have few lost. The limitation of T73 treatment by which the tensile strength is lower in dry air, so the tensile strength is only 660.8 MPa which is a little more than T6 in corrosive medium (NaCl solution). The tensile strength and elongation of alloy after RRA treatment also lose rarely. Because the tensile strength is higher in dry air, the tensile strength is 729.5 MPa which outclass T6 and T73 treatments in corrosive medium. And the elongation is 7.51% closed T73 level.

In stress corrosion index, the I_{SSRT} of the specimen after T6 treatment is biggest one. The index expresses that the SCC resistance maintain at a low level. Comparing with in dry air, both the tensile strength and elongation of specimens are significantly reduced in 3.5% NaCl solution. The I_{SSRT} with T73 treatment reflect the wonderful SCC resistance. Comparing with in dry air, the losses of tensile strength and elongation are only 2.32% and 6.25% respectively in 3.5% NaCl solution. After RRA treatment, the spray formed 7075 alloy has outstanding tensile properties in dry air by slow strain rate test, the tensile strength is closed to T6 and elongation is near to T73. It can be found that the SCC resistance is little inferior to T73 in 3.5% NaCl solution by slow strain rate test. And the tensile strength and elongation of the alloy are 729.5 MPa and 7.51% respectively. Combination properties of the spray formed 7075 alloy after RRA treatment are excellent.

Fracture surfaces of slow strain rate test in dry air and 3.5% NaCl solution after RRA treatments are shown as Figure 2. From Figure 2(a), it can be seen that tensile fracture of specimen after RRA treatments in dry air is dimple fracture, dimples are small and uniformity, and dimple diameter is almost 2 µm. In 3.5% NaCl solution, Figure 2(b), the fracture is similar to cleavage fracture but on typical river or ligule patterns, and there are some tear ridges on surface of fracture, so this fracture can be classified as quasi-cleavage fracture.

Figure 3 shows TEM images of spray formed 7075 alloy after T6, T73 and RRA treatments. From Figure **3(a)**, it can be seen that abundance transgranular η' phases are thin, isolated and dispersed after T6 treatment. Sizes of the η' phases are 1 ~ 2 nm. At grain boundaries, η phases are continuous and precipitate free zones (PFZ) are about 5 nm. From Figure 3(b), it can be found that the transgranular η' phases after T73 treatment are bigger and more againate than them after T6 treatment. Volume fraction of η' phases depressed and sizes of the η' phases are 3 ~ 5 nm. At grain boundaries, η phases interrupt and PFZ widen to 25 nm. From TEM image after RRA, as Figure 3(c), it can be found that the tiny dispersive homogeneous η' phases were separated out again in grains. At this moment, the main strengthening phases were η' and η . And precipitated phases on grain boundaries were thick and discontinuous η phases. Those interrupted η phases at grain boundaries are analogous to them after T73 treatment.

4. Discussion

The usual precipitation sequence of 7xxx series aluminum alloys can be summarized as [27]: SSSS (Super-saturated solid solution) \rightarrow GP zones \rightarrow metastable $\eta' \rightarrow$ stable η . GP zones are metastable, coherent solute clusters of Zn, Mg and Cu. The metastable η' phases, Al, Cu and Mg components base on a solid solution of $MgZn_2$, $Mg(ZnCuAl)_2$ or $Mg(Zn_2, AlMg)$ appear as discrete platelet particles that are semi-coherent with the matrix, which is known to populate within the grains, and η is pseudostable, non-coherent of the same phase



Figure 2. Fracture surfaces of slow strain rate test in different environment after RRA treatments: (a) in dry air (b) in 3.5% NaCl solution.



Figure 3. TEM images of alloy after various aging treatments: (a) T6, (b) T73, (c) RRA.

appearing as rods or plates, which is known to populate the grain boundary. Strengths of the alloy are immediately influenced by sizes and volume fraction of strengthening phases.

The strength of the alloy mainly relies on matrix precipitates (MPt) during the whole aging process, the strength of the alloy changes with characteristics of GP zone, η' and η . The best strength depends on fine homogeneously dispersive MPt. Plasticity, toughness and corrosion resistance of the alloy are remarkably influenced by structure and chemical property of grain boundary precipitates (GBP). There is a popular belief that continuous GBP is harmful to properties of the alloy. Because relative movement of crystalline grains in deformation process has been impeded by continuous GBP, plasticity and toughness of the alloy are completely injured. On the other hand, the continuous GBP is preferentially dissolved as anodes in anodic dissolution. Because the potential of GBP, PFZ and matrix is -1.05 V, -0.85 V and -0.75 V respectively, the potential difference (PD) between GBP and precipitate free zones (PFZ) is less than the PD between GBP and matrix. With regard to corrosion resistance of the alloy, widening PFZ can remit the corrosion sensibility and improve corrosion resistance of alloy.

And slow strain rate test was an accelerated SCC process which is stable and placid. This method can verify whether material have SCC susceptibility or not in a short time. According to the slow strain rate tests, crystalline grains slip deformation and some dislocations left.

After T6 treatment, abundance tiny transgranular η' phases keep the tensile strength at 777.2 MPa. And because the η phases at grain boundaries are preferentially dissolved as anodes, the galvanic corrosions formed by aluminum matrix and continuous η phases lead to bad SCC susceptibility of the alloy after T6 treatment.

After T73 treatment, the transgranular η' phases grow up and volume fraction of η' phases depressed, so the tensile strengths of the alloy is only 676.5 MPa. But, after T73 treatment, interrupted η phases can constitute an obstacle to forming the galvanic corrosions and improve SCC resistance of the alloy.

From TEM of alloy after RRA, as **Figure 3(c)**, it can be found that the tiny dispersive homogeneous η' phases were separated out again in grains. At this moment, the main strengthening phases were η' and η . After RRA, it is the reason of high strength that η' phases were separated out by alloying elements into which lots of GP zones and η' smaller than critical dimension were dissolved at retrogression treatment. And precipitated phases on grain boundaries were thick and discontinuous η phases. Those interrupted η phases at grain boundaries are analogous to them after T73 treatment and can constitute an obstacle to forming the galvanic corrosions and improve SCC resistance of the alloy. After RRA treatment, the alloy both has high strength and favorable SCC resistance.

The effects of PFZ still have different opinions in academia. But, as the result of this experiment, wide PFZ can improve plasticity and SCC resistance of alloy. This result fits Jiang's conclusion [28] [29].

Apart from the above, Wei [30] affirms that hydrogen atoms released by hydration surface which has been formed with hydrones (H_2O) and fresh aluminum alloys infiltrate plastic zone of crack tip via diffusion after studying on fatigue crack growth of aluminum alloys. Those hydrogen atoms can speed crack growth up. For the Al-Zn-Mg-Cu alloys, hydrogen atoms can be released at cathode when anodic dissolution. Such hydrogen atoms are even few, but they influence crack formation and growth, even crack growth by anodic dissolution, to a certain extent.

SCC resistance is also influenced by dislocations in material. Hydrogen atoms form atmospheres around dislocations, and those hydrogen atmospheres move with dislocations together. Chu [31] reported that the cleavage crack nuclei could be founded by dislocation reaction of moving dislocations with hydrogen atmospheres, by which hydrogen was stayed in crack nuclei. All the phenomena above mentioned cause hydrogen embrittlement of material. For the alloy after T6 treatment, relationship between dislocation and SCC resistance can be explained by above views. But there are interlaced tangled reticular dislocations in the matrix of spray formed 7075 aluminum alloy after RRA treatment during slow strain rate test, as **Figure 4**. Reticular dislocations which have bad mobility are unusual in tradition casting Al-Zn-Mg-Cu alloys. Strengthening phases (the black blocks) separated out in these reticular dislocations can pin them, so the possibility of hydrogen embrittlement caused by above-mentioned moving dislocations sharply fall down. In macroscopic view, it is shown that fractures of specimens after RRA treatment by slow strain rate test are dimple fracture in dry air and quasi-cleavage fracture in 3.5% NaCl solution, as in **Figure 2**. So, it can be consider that reticular dislocations are the traps for hydrogen in a qualified sence, whether in HIC or "Mg-H" theory, conducive to improving SCC resistance of alloys.

5. Conclusions

After retrogression and reaging (RRA) treatment, abundance tiny transgranular η' phases separated out again maintain the tensile strength of the spray formed 7075 alloy at 759.4 MPa in dry air and 729.5 MPa in 3.5 wt% NaCl solution by slow strain rate test.

According RRA treatment, interrupted η phases and wide PFZ can improve SCC resistance of the alloy. And reticular dislocations in the matrix during slow strain rate test can decrease the possibility of hydrogen crack and are conducive to improving SCC resistance of alloys.

Fractures of specimens after RRA treatment by slow strain rate test are dimple fracture in dry air and quasicleavage fracture in 3.5% NaCl solution.



Figure 4. TEM images of deformation zone after RRA and slow strain rate test (reticular dislocations).

Acknowledgements

This research was financially supported by the National Natural Science Foundation of China (51574167) and Program for Innovative Research Team in University of Liaoning Province (LT2015020).

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