

Thermoelectric Power of Cu-Zn Ferrites

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ABSTRACT

A series of Cu-Zn mixed ferrites with composition formula $Cu_{1-x}Zn_xFe_2O_4$ is prepared by the double sintering ceramic technique. Thermoelectric power studies are performed over a temperature range of 300 to 800 K by a differential method. The results showed a negative value for the Seebeck coefficient S for all samples, and all compositions exhibited an n -type semiconductors behavior in the measured range of temperature. The values of charge carrier concentration n and the Fermi energy were determined. The values of n were found to decrease as temperature increased, while Fermi energy directed to more negative values when Zn content is increased. On the basis of these results a mechanism for the conduction in Cu-Zn ferrites is suggested and the properties of the mentioned compounds were determined.

Keywords: Ferrites, Thermoelectric Power, Fermi Energy, Seebeck Coefficient

1. Introduction

Spinel ferrites are commercially important materials because of their excellent magnetic and electrical properties. Interesting physical and chemical properties of the magnetically diluted ferrites arise from the ability of these compounds to distribute the cations amongst the available tetrahedral A- and octahedral B-sites. Ferrites are able to fulfill a wide range of applications from micro wave to radio frequencies, are of great importance from both fundamental and applied research points of view [1,2]. Because Zinc is non-magnetic divalent ions that occupy essentially tetrahedral A sites, when substituted in ferrites [3]. Gonchar and Andreev [4] have reported that Ni-Cu-Zn ferrites with less content of Zn could obtain high Curie temperature, but the initial permeability of Ni-Cu-Zn ferrites reached. The substitution of Cu brings about a structural phase transition accompanied by the reduction in the crystal symmetry due to co-operative Jahn-Teller effect [5,6], which ultimately results in some interesting electrical and magnetic properties. Various investigations [7,8] studied the electrical and thermal power of the spinel ferrites and have found that they have semi-conducting properties of n or p -type. In this article, the electric and thermoelectric power of Cu-Zn ferrites have been studied experimentally as a function of composition and temperature in order to understand the conduction mechanism in these samples.

2. Experimental Technique

The polycrystalline Zinc-substituted Copper ferrites having compositional formula $Cu_{1-x}Zn_xFe_2O_4$ where x stepped $0 \leq x \leq 1$, were prepared by the double sintering ceramic technique [9], from pure oxides powders Fe_2O_3 , CuO and ZnO. Oxides were mixed in stoichiometric proportions and pre-fired at 750°C for 5 h. The pre-fired powders were grounded and compressed at constant pressure (3×10^8 Pa) in the form of disk shape of 13 mm diameter. The samples were finally sintered at 1150°C for 5 hours and then slowly cooled to room temperature with cooling rate of 2°C/min.

The electrical measurements were performed by two-probe method; silver past was applied on both sides of the samples to make good ohmic contact.

Thermoelectric power studies were carried out as a function of composition and temperature by differential method [10]. The sample holder for measuring the Seebeck coefficient consists of two non-magnetic copper electrodes between which the sample is fixed. An auxiliary heating coil is fixed to upper electrode for additional heating to maintain a temperature difference of about 15 degree between the two faces of the sample. A temperature of both surfaces of the sample was measured by the same type of two thermocouples. The Seebeck coefficient S was calculated using the relation

$$S = \Delta E / \Delta T$$

where ΔE is the thermoelectric potential difference pro-

duced across the sample due to the temperature difference ΔT .

3. Results and Discussion

The variation values of Seebeck coefficient with temperature for each sample of Copper zinc ferrite having a compositional formula $\text{Cu}_{1-x}\text{Zn}_x\text{Fe}_2\text{O}_4$ where x stopped $0 \leq 0.2 \leq 1$, have been calculated from the experimental

values of thermoelectric motive force, and depicted against the temperature as shown in **Figures 1(a)-(f)**. It is noticed that the sign of the Seebeck coefficient for all samples is negative, indicating that the Cu-Zn ferrite behave as n-type semiconductor. The conduction mechanism in these ferrites is due to electrons [11], and many authors obtained similar results [12,13].

Several features can be obtained from **Figures 1(a)-(f)**,

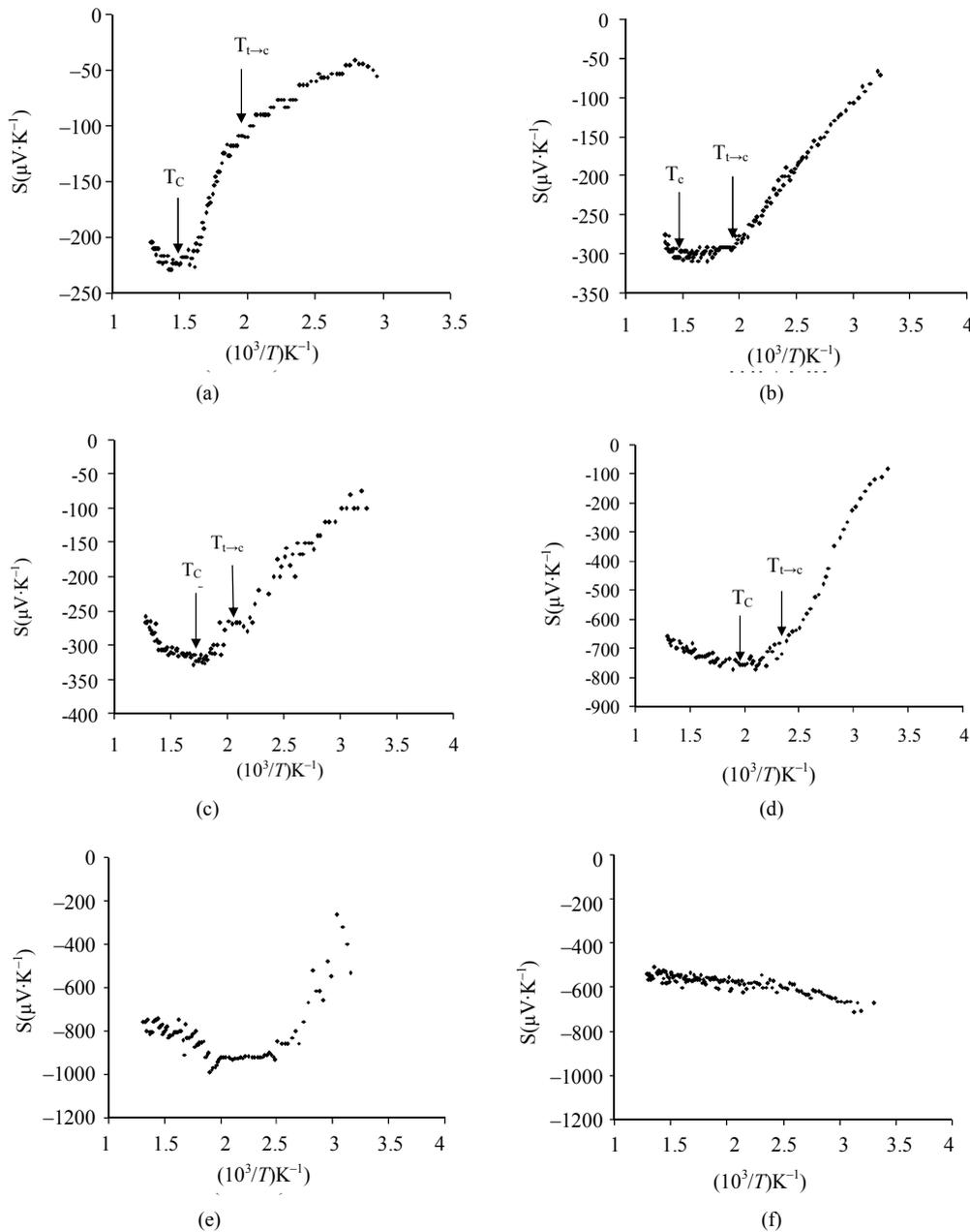


Figure 1. (a) The variation of Seebeck coefficient (S) with $10^3/T$ (for sample of $x = 0.0$); (b) The variation of Seebeck coefficient (S) with $10^3/T$ (for sample of $x = 0.2$); (c) The variation of Seebeck coefficient (S) with $10^3/T$ (for sample of $x = 0.4$); (d) The variation of Seebeck coefficient (S) with $10^3/T$ (for sample of $x = 0.6$); (e) The variation of Seebeck coefficient (S) with $10^3/T$ (for sample of $x = 0.8$); (f) The variation of Seebeck coefficient (S) with $10^3/T$ (for sample of $x = 1.0$).

compositions of $x \leq 0.6$ have two change of slope in S against temperature, the first change at lower temperature can be referred to structural change from tetragonal to cubic system. The transition from tetragonal to cubic $T_{t \rightarrow c}$ was found to decrease with increasing Zn content as shown in **Figure 2**. In other words the tetragonality is in agreement with the results of [14].

At higher temperature Seebeck coefficient S decrease in negativity with increasing temperature, up to a particular temperature, hereafter referred to as Seebeck coefficient transition temperature T_c . However, by further increase of temperature, the values are found to decrease in negativity. The minimum values observed in Seebeck coefficient may be attributed to a magnetic transition, where the ferromagnetic material becomes paramagnetic. On the other hand, in the case of samples of $x = \geq 0.8$, the Seebeck coefficient is found to be increased slightly as the temperature increased unlike other samples, and there is no transition T_c was found.

This is due to the fact that these samples having higher Zinc concentration with weak tetrahedral A-site and octahedral B-site super exchange interaction. This behavior is obvious because both samples have higher concentration of Zinc with paramagnetic structure at room temperature [14]. Similarly, they behave like normal solid with slightly increase in Seebeck coefficient values with temperature. However, the change of slope could not be observed in the case of the two samples *i.e.* of $x = \geq 0.8$ since they are diamagnetic at room temperature [15].

Generally, the change of slope for samples of $x \leq 0.6$ is attributed to change in conductivity mechanism. The conduction at lower temperature (below Curie temperature) is due to hopping of electrons [12] between Fe^{3+} and Fe^{2+} ions, whereas at higher temperature (above Curie temperature) due to polaron hopping [15].

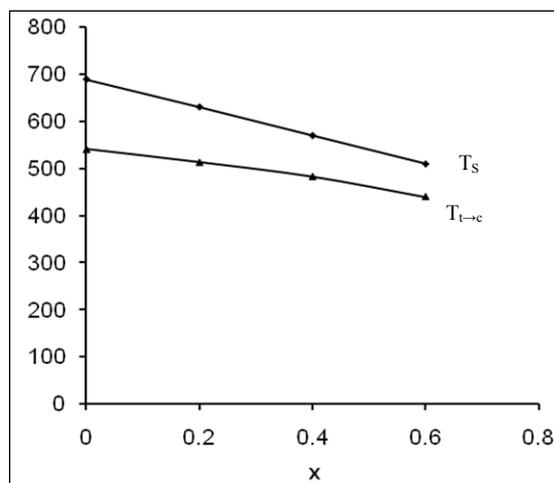


Figure 2. The variation of $T_{t \rightarrow c}$ and T_s with composition x .

Following according to Wu *et al.* [16] the Seebeck coefficient S is expressed in terms of Fe^{3+} and Fe^{2+} ions, considering the small polaron hopping conduction mechanism as follows:

$$S = -(k/e) \ln \left\{ \beta \frac{[Fe^{3+}]_B}{[Fe^{2+}]_B} \right\}$$

where $[Fe^{3+}]_B$ and $[Fe^{2+}]_B$ are the concentration of the Fe^{3+} and Fe^{2+} ions in the octahedral sites, respectively and $\beta = 1$. Dawoud *et al.* [17] reported that the increase of Zn substitution tends to increase the Fe^{2+} ions as demonstrated in above equation.

In addition, the increase of Zn content is associated with a decrease in copper content. In turn the possibility of Cu^+ formation and $Cu^{2+} \leftrightarrow Cu^{1+}$ hopping process and the number of holes involved will be reduced. As a result S becomes more negative with the increase of x (Zinc content).

4. Charge Carrier Concentration (n)

In the case of low mobility semiconductor like ferrites having exceedingly levels, the value of N , the density of state can be taken as 10^{22} cm^{-3} [18].

$$n = N/V \left[N/1 + \exp(-Se/k) \right]$$

The value of charge carriers concentration/unit volume have been calculated for all the samples by using the values of the Seebeck coefficient S , and the relation between n and temperature is depicted in **Figures 3(a)-(c)** for samples of $x = 0.0, 0.2$ and 0.4 respectively. The values of n for all samples are also listed in **Table 1**. The figures illustrated that, the values of n decreased with the increasing the temperature.

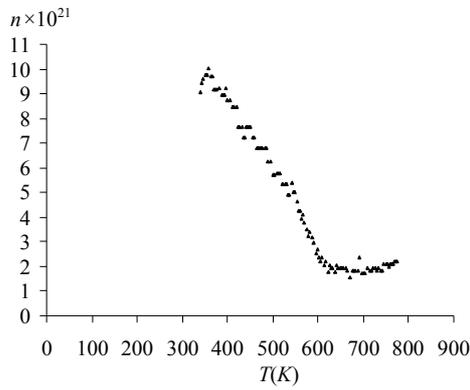
5. Fermi Energy (EF)

The thermoelectric power data were utilized to determine the Fermi energy. The Fermi energy of Cu-Zn ferrite was calculated employing the following expression [10,19].

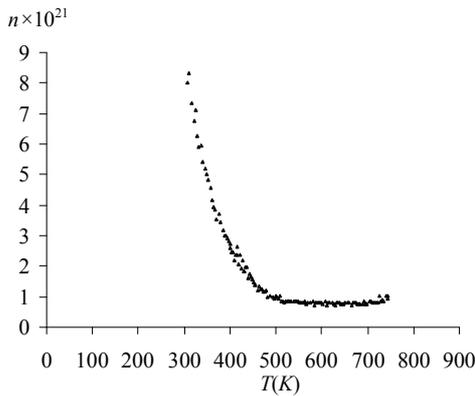
$$E_F = eST - AKT$$

where K , e , S and T are Boltzmann constant, electronic charge, Seebeck coefficient and absolute temperature respectively. A is a dimensionless constant related to the kinetic energy of the charge carrier, and has values of 0 or 2. Two series of values of E_F were obtained, over the temperature range from 300 K up to 800 K, for $A = 0$ and $A = 2$. **Figure 4(a)-(c)** represent the variation of Fermi energy with temperature. The extrapolation of the linear part of the two series of E_F , for $A = 0$ and $A = 2$ intersects at $T = 0$.

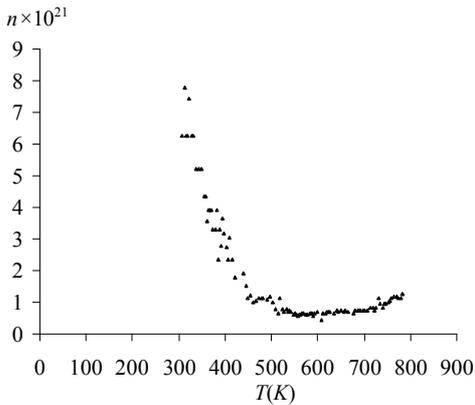
The corresponding value of E_F , the point of interception at ordinate, gives the Fermi energy at zero tempera-



(a)



(b)



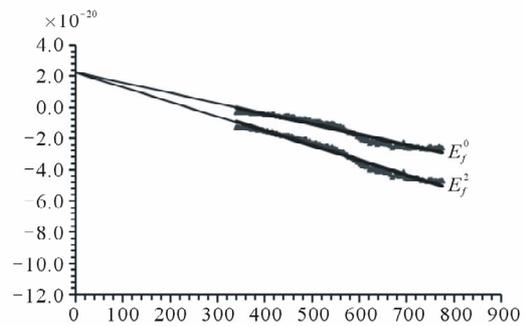
(c)

Figure 3. (a) Temperature dependence of charge carrier n for sample of $x = 0.0$; (b) Temperature dependence of charge carrier n for sample of $x = 0.2$; (c) Temperature dependence of charge carrier n for sample of $x = 0.4$.

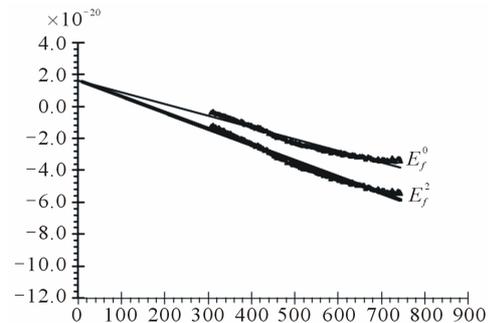
ture $E_F(0)$; and **Figures 4(a)-(c)** indicates the variation of Fermi energy with temperature. Also **Table 1** lists the values of Fermi energy obtained for each composition. It is noticed that $E_F(0)$ is positive for sample of $x = 0$ and sample of $x = 0.2$ then it decreases to the negative value *i.e.* at the left hand side of the gap, as zinc ion increased.

Table 1. The values of Seebeck coefficient, Curie temperature transition temperature from tetragonal to cubic $T_{t \rightarrow c}$ and Fermi energy for Cu-Zn ferrites.

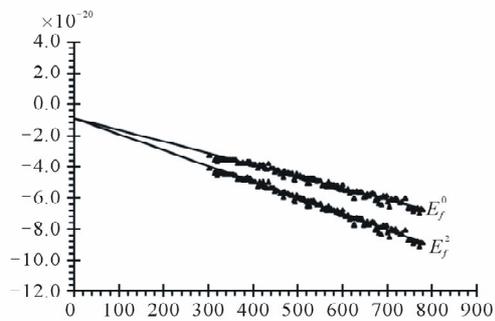
x	S at 374 K	T_S (K)	$T_{t \rightarrow c}$ (K)	E_f (ev)
0.0	-53.8	688	540	2.4×10^{-20}
0.2	-158	630	512	1.6×10^{-20}
0.4	-166.7	569	482	0.4×10^{-20}
0.6	-515	509	439	-0.6×10^{-20}
0.8	-800	-	-	-0.8×10^{-20}
1.0	-636.7	-	-	-1.0×10^{-20}



(a)



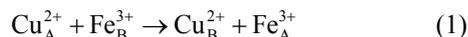
(b)



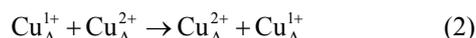
(b)

Figure 4. (a) Plot of E_f against temperature for sample of $x = 0.0$; (b) Plot of E_f against temperature for sample of $x = 0.2$; (c) Plot of E_f against temperature for sample of $x = 1.0$.

Mazen *et al.* [10] reported that for Cu ferrite, $E_F(0)$ is positive at $T = 0$ K. Patil *et al.* [11] reported that when heating CuFe_2O_4 sample, two temperature domains are noted in which the concentration of B-site Cu^{2+} ions is increased and the interstice cation exchange is given as



This reaction gives n-type condition, while at low temperature the conduction process for CuFe_2O_4 at lower temperature is



This gives only p-type conduction depending on the relative concentration of Cu^{1+} of Cu^{2+} ions on A-sites. For the present system Fermi energy is positive for Cu ferrite and reaction (2) is favored. While as the Zinc content is increased, Fermi energy goes to negative and reaction (1) is favored since the increasing of the Zinc content tend to decrease Cu content at A-site this tend to increase the negativity of Fermi energy at absolute zero.

6. Conclusions

- 1) The Cu-Zn ferrite behaves as n-type semiconductor.
- 2) The transition from tetragonal structure to cubic was found to decrease with increasing of Zn content.
- 3) The minimum values observed in Seebeck coefficient may be attributed to a magnetic transition, where the ferromagnetic material becomes paramagnetic.
- 4) The conduction at lower temperature (below Curie temperature) is due to hopping of electrons between Fe^{3+} and Fe^{2+} ions, whereas at higher temperature (above Curie temperature) due to polaron hopping.
- 5) For the present system Fermi energy is positive for Cu ferrite while as the Zinc content is increased, Fermi energy goes to negative value.

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