

# Development of Different Strategies for the Clean-Up of Polychlorinated Biphenyls (PCBs) Congeners Using Pressurized Liquid Extraction

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## ABSTRACT

The extracted fish samples (w/w) using pressurized liquid extraction (PLE) were cleaned up by applying different adsorbent materials such as: combination of acidic/basic silica and alumina, florisil, and silica gel either alone or in combination with another sort of sorbent. Different solvents such as hexane (H), dichloromethane (DCM), with different compositions were used to elute the PCBs target compounds from the spiked fish samples. Overall the mean percentage recoveries for all PCBs congeners using PLE were ranging from 78.6% to 98.7%; from 79.4% to 91.8%; from 65.8% to 104.5% and from 82.4% to 100.9% for cases A, B, C, D and E, respectively. However, the overall mean percentage recoveries for the 13C-PCBs surrogate including: 101, 138, 153, 180 and 209 were ranging from 88.2% to 97.6%; from 79.4% to 90.2%, from 88.3% to 96.4%, from 79.8% to 90.4% and from 83.9% to 95.9%, for cases A, B, C, D and E, respectively. The limits of detection of the proposed method were varied from 0.01 to 0.28 ng/g (w/w) for different PCBs congeners. The proposed method has linear dynamic range from 21.5 ng/g to 109 ng/g. This method was applied to fish samples for the determination of the target PCBs components.

**Keywords:** Polychlorinated Biphenyls (PCBs); Pressurized Liquid Extraction (PLE); Adsorbent Materials; Gel Permeation Chromatography (GPC); Negative Chemical Ionization (NCI); Marine Sample

## 1. Introduction

Persistent Organic Pollutants (POPs) consist of heterogeneous groups including, polychlorinated biphenyls (PCBs), organo-chlorinated pesticides (OCPs), and other organic pollutants [1]. POPs are characterized by high chemical, biological stability and lipophilicity, which makes POPs persist in the environment and bio-accumulate within the food chain [2]. POPs are toxic chemicals that have the capability to stay stable in the environment over long period of time. In the year 2001 Stockholm Convention, the polychlorinated biphenyls (PCBs) organochlorinated pesticides (OCPs) were given special attention and thus were included in the so called “dirty dozen” [3].

It is to be noted that accuracy in generating data is critical in decision making process. Monitoring of PCBs plays a vital role in determining environmental contamination. Therefore, obtaining information for regulatory purpose is needed. Clean-up of the extracted organic compounds looks to be interesting by applying different sorts of adsorbents which leads to strongly binding of the analytes onto adsorbent portion or by their interaction with

matrix in order to recover the maximum amount of analytes from substrate. Sample treatment consists of two steps: one step is to destroy the lipids by using SX3-300 Bio-beads (GPC) and the second step is the clean-up step that is performed by using different adsorbents.

Pressurized liquid extraction (PLE) found to fulfill the least evolve parts of the analytical extraction procedures since it has the advantage of short extraction times combined with automation and low solvent consumption. Furthermore, the technique is simple and adopted to a large number of applications. Extraction procedures in PLEs were effectively used to extract number of POPs such as PCBs [4-6]. Using pressurized liquid extraction (PLE) is widely used for food, feed extraction and in environmental POP investigations since the mid 1990s [2,3,7]. However, the United States Environmental Agency (US EPA) adopted the PLE technique in 1995 (US EPA Method 3545) [8]. PLE has been used for the extraction of PCBs in whole ground carp [9], muscle tissue [9], and oyster tissue [10]. Numerous applications of PLE for the extraction of POPs from biota samples have been reported in the past few years [11-14]. Suchan *et al.* [14]

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extracted OCPs and PCBs from fish fillets applying PLE using two different solvent mixtures (H:DCM; 1:1, v/v) and (H:acetone; 4:1, v/v) and they compared PLE with conventional soxhlet extraction with both solvents tested and they found that PLE was comparable to soxhlet. Recently, Saito *et al.* [15] applied PLE to extract POPs (PCBs, PBDEs, OCPs, etc.) from biological tissues. Jiang *et al.* [16] reported the use of PLE to extract several POPs (PCBs, PCDDs/DFs and PCNs) from freeze-dried seafood samples such as: fish, crab, shrimp, bivalves and cephalopods.

The extracts of bald eagle tissue for PCBs, OCPs and polychlorinated dibenzo-p-dioxins and dibenzofurans (PCDD/DFs) were subjected to silica gel, alumina, acidic silica and activated carbon column for clean-up and fractionation [17]. Activated silica was used for the fractionation containing PCBs, OCPs, and Poly Aromatic Hydrocarbons (PAHs) from the mussel extract [18,19].

The analytical methods used for the determination of PCBs in environmental samples (fish) consist of several steps for sampling, sample treatment, fractionation and detection of targeted compounds. The sample treatment is multi-steps procedure that its basic concept is to convert a complex matrix into a sample that is suitable for the analysis.

Regardless, the technique that it is used for extraction with different types of components such as, lipids, pigments are mostly present in the extract and must be removed in order to identify and to quantify lower levels of analytes and to reduce the deterioration of chromatograms. Several methods of clean-up have been developed in order to remove the co-extracted matrix and minimize their negative effects. The necessity of the clean-up step is to completely remove the bulk of the co-extracted material, as well as those compounds that behave the same as to the analytes that could potentially interfere in the final quantification.

Therefore, the main target of this study is to achieve the followings: 1) to extract the PCBs congeners from the matrix; 2) to remove the un-wanted organic compounds, which may interfere with the compounds of interest; 3) to provide a suitable adsorbent material free from interferences (clean-up); 4) finally to convert the extracted analytes from the matrix into a more suitable concentration level. A comparative study using different types of adsorbents was conducted in order to improve the sample preparations without sacrificing the quality of measurements (sensitivity, accuracy, precision, repeatability, reproducibility and recovery) in fish samples for the analyses of polychlorinated biphenyls (PCBs). Moreover, a GC-MS (NCI) method was applied for the analysis of PCBs.

## 2. Experimental

### 2.1. Standards, Chemicals and Samples

The PCBs (EC-4133) contain the following congener numbers: 2,4,4',5-tetraCB (74); 2,3',4',5-tetraCB (70); 2,2',3,5',6-pentaCB (95); 2,2',4,5,5'-pentaCB (101); 2,2',4,4',5-pentaCB (99); 2,2',3,4,5'-pentaCB (87); 2,3,3',4',6-pentaCB (110); 2,2',3,5,5',6-hexaCB (151); 2,2',3,4',5',6-hexaCB (149); 2,3',4,4',5-pentaCB (118); 2,2',4,4',5,5'-hexaCB (153); 2,3,3',4,4'-pentaCB (105); 2,2',3,4,4',5'-hexaCB (138); 2,2',3,3',4,6'-hexaCB (132); 2,3,3',4,4',6-hexaCB (158); 2,2',3,4',5,5',6-heptaCB (187); 2,2',3,4,4',5',6-heptaCB (183); 2,2',3,3',4,4'-hexaCB (128); 2,2',3,3',4',5,6-heptaCB (177); 2,2',3,3',4,4',6-heptaCB (171); 2,3,3',4,4',5-hexaCB (156); 2,2',3,4,4',5,5'-heptaCB (180); 2,3,3',4,4',5',6-heptaCB (191); 3,3',4,4',5,5'-hexaCB (169); 2,2',3,3',4,4',5-heptaCB (170); 2,2',3,3',4,5,5',6,6'-nonaCB (208); 2,2',3,3',4,4',5,6'-octaCB (195); 2,2',3,3',4,4',5,5'-octaCB (194); 2,2',3,3',4,4',5,5',6-nonaCB (206) and decaCB (209). A  $^{13}\text{C}$ -labeled PCB mixture (EC-4058) was obtained from Cambridge Isotope Laboratories, and contained the following congener numbers: 2,2',3,4,4',5'-hexaCB ( $^{13}\text{C}_{12}$ , 99%) (138); 2,2',4,4',5,5'-hexaCB ( $^{13}\text{C}_{12}$ , 99%) (153); 2,2',3,4,4',5,5'-pentaCB ( $^{13}\text{C}_{12}$ , 99%) (180); and decaCB ( $^{13}\text{C}_{12}$ , 99%) (209). A standard stock solution of 128.8 ng/ml was used to prepare the standard working calibration solutions for most of the PCB compounds.

All solvents were pesticide-grade. Hexane (H) and dichloromethane (DCM) were supplied by Merck (Darmstadt, Germany). Nitrogen gas was used to concentrate the extract. The evaporator (Heidolph-Verwenden, Germany) and SX-3 Bio-Beads (200 - 400 mesh) were purchased from Bio-Rad Laboratories GmbH, in Munich, Germany. Anhydrous sodium sulfate (EMD-Chemical, Darmstadt, Germany) was purchased from Sigma Aldrich Chemie GmbH in Steinheim, Germany. Silica gel (100 - 200 mesh) was obtained from Aldrich (Steinheim, Germany). Aluminum oxide (70 - 230) and florisol were obtained from VWR-Baker. Fish samples were collected from the local market in Kuwait to validate the method on real samples.

### 2.2. Instrumentation

PCBs were quantified on an Agilent 5973 inert mass selective detector, and on an Agilent Technology 6890 network gas chromatography (GC) system coupled with mass spectrometry (MS) with a negative chemical ionization (NCI) ion source. The system was operated in selective ion monitoring (SIM) mode, and 1  $\mu\text{l}$  of sample solution was injected into the GC in the auto-sampler's splitless mode. The capillary column used was a DB-5MS (30 m

× 0.25 mm I.D., 0.25 µm film thickness). The initial oven temperature was 100°C, which was held constant for 0.5 min. It was then increased to 140°C at a rate of 20°C/min, with no holding time, followed by increase in temperature to 200°C at a rate of 4°C/min with 13 min holding time, and finally increase in temperature to 270°C at a rate of 4°C/min, where it was maintained for 4 min. The helium carrier gas flow rate was maintained at 1.1 ml/min. The transfer line temperature of the GC-MS interface and the ion source temperature were held at 300°C and 280°C, respectively. The MS was conducted in the NCI mode with methane as the reagent gas (40 ml/min).

### 2.3. Sample Preparation and Purification

The edible portions of the fish sample were homogenized. The wet fish sample (5 g) was mixed with anhydrous sodium sulfate to reduce the amount of water and then it was extracted using PLE system. The wet fish samples were grinded to small sizes, which might facilitate the analyte transport to the solvent particle surface. The grinded samples were filled into the cell and the dead volume of the cell was filled with intermatrices, such as hydromatrix. The extraction was performed with 10% (H:DCM). The extract was concentrated by a rotary evaporator to 10 ml. The samples were cleaned with gel permeation chromatography (GPC), GPC column was packed with 12 g SX-3 bio-beads and filled with 1:1 (hexane:dichloromethane), when it is not used. 100 ml [H:DCM (1:1)] was used to elute the extract, and GPC column was used to remove less than 1.0% fat. The first 45 ml was discarded and the remaining solvent was collected. The extract was concentrated to appropriate volume (1 ml) using rotary-vapor concentrator and finally the 1 ml goes under gentle stream of clean dry nitrogen. The extract then was subjected to different layer of adsorbents (eight adsorbents) as described in **Table 1**. The column was plugged with glass wool. A slurry method was used to fill the column. The multilayer column (glass column with

15 mm inner diameter and 30 cm long) was packed from top to bottom. Each column was prepared freshly before its use as part of quality control procedures. Elution of samples was carried out at a rate of about 1 ml/min, using different solvent mixtures as shown in **Table 1**.

### 2.4. Pressurized Liquid Extraction (PLE)

Automated PLE extraction was used (FMS, Waltham, Massachusetts, USA). A stainless-steel extraction cell was supported with Teflon end-caps and filters. The PLE system was controlled by means of a PC using DMS 6000 software that shows the real time, temperature and pressure. The pump, flow rate, solvent time, valves status and cooling were adjusted during the extraction run by the software. Extraction was carried out at a temperature above the solvent's boiling point and under its pressure to maintain the liquid state of the organic solvent, which keeps the solvent below its critical condition, as well as maintaining its viscosity and its salvation power. Under the selected conditions, the extraction efficiency was enhanced, and the amount of solvent required was minimized. Following is the PLE program that was utilized in the effective extraction of the PCBs:

- 1) Filling cells with solvent (input: open; output: open); time = 2 min;
- 2) Pressurizing cell (input: open; output: close); time = 1.0 min;
- 3) Heating and maintaining pressure (input: close; output: close); temp.= 120°C; time = 30 min;
- 4) Cooling (pump: off; input: close; output: close); time = 15 min; fan = on;
- 5) Depressurizing (input: close; output: open); time= 0.02 min;
- 6) Rinsing sample (input: open; output: open); time= 2.5 min;
- 7) Purging with N<sub>2</sub> (N<sub>2</sub>: 35 psi) (input: close; output: open); time = 1.0 min;
- 8) Opening all valves; Time = 0.02 min.

**Table 1. Type of adsorbent; eluent solvent mixture used in the elution of PCBs.**

Case No.	Adsorbent	Solvent mixture	Spiked PCBs (ng/g)	Spiked <sup>13</sup> C-PCBs (ng/g) + recoveries (%)
A.	10 g acidic silica (24%) + 10 g basic silica (1N) + 5 g Al <sub>2</sub> O <sub>3</sub> + 2 g Na <sub>2</sub> SO <sub>4</sub>	60 ml H:DCM (1:1; v/v)	21.8	24 (88.2 - 97.6)
B.	6 g silica gel + 2 g Na <sub>2</sub> SO <sub>4</sub>	135 H + 15 ml DCM	21.8	24 (79.4 - 90.2)
C.	17g florisil + 2g Na <sub>2</sub> SO <sub>4</sub>	70 ml H + 50 ml H:DCM; 70:30, v/v+ 40 ml DCM	54.5	24 (88.3 - 96.4)
D.	17g florisil + 2g Na <sub>2</sub> SO <sub>4</sub>	70 ml H + 50 ml H:DCM; (1:1; v/v)	21.8	24 (79.8 - 90.4)
E.	17g florisil + 2g Na <sub>2</sub> SO <sub>4</sub>	70 ml H + 50 ml (DCM: H; 10%)	21.8	24 (83.9 - 95.9)

The PLE system was washed after 2 sets of extraction run using the following program:

- 1) Filling column (input: open; output: open); time = 2.5 min;
- 2) Flushing bypass (input: open; output: close); time = 1.0 min;
- 3) Depressurizing (input: close; output: open); time = 0.02 min;
- 4) Purging with N<sub>2</sub> (input: close; output: open); time = 0.02 min;
- 5) Ending stage (all are closed).

### 3. Results and Discussion

#### 3.1. Extraction Solvent, Adsorbents Column and Elution Solvent for the Clean-Up

Several facts have to be considered when choosing extraction solvent, such as: physical chemical properties (*i.e.* polarity, boiling point, density and toxicity [20]). However, when using different solvent extraction mixture for the fish sample, calculating the recoveries of the targeted compounds with respect to the solvent mixture used for the extraction, there was no significant differences found between the results obtained with the PLE method. The polarity of the used extraction solvent mixture, plays a role in decreasing the turbidity and the content of matrix pigment in the extracted samples [21].

In general, the solvent selection is important for the extraction of contaminants from the matrix of interest. Hexane, dichloromethane and acetone with different combinations were selected usually for the extraction and clean-up of PCBs, due to the fact that they are frequently used to extract PCBs from biotic and abiotic samples. However, with all the solvents combinations used in PLE extraction and clean-up of the extract resulted in comparably good recoveries of all PCBs which were above 70% and close to 100% for most solvents compositions tested. This shows that all the extraction of PCBs was completely recovered under the PLE and clean-up conditions.

Different chromatographic columns with different solvent mixtures were studied, in order to find the best adsorbent for the elution of PCBs congeners. The adsorbent materials, the solvent composition which were used in the experimental test for PCBs were presented in **Table 1**. Using acidic and basic silica column showed good recoveries for all the PCBs congeners (**Figure 1**, case A). The 6 g silica with 135 ml Hexane + 15 ml DCM, showed good recoveries, as shown in **Figure 1** (case B). Using 17 g florisil and 70 ml hexane + 50 ml (H:DCM; 70:30, v/v) + 40 ml DCM, the recoveries of the spiked sample were varied from 97.4% to 91.8% (**Figure 1**, case C). When using florisil (17 g), with different eluant com-

positions: 1) 70 ml hexane and 50 ml [DCM:H (50%)]; 2) 70 ml hexane + 50 ml [10% (H:DCM)], a comparative recoveries were obtained for most of PCBs as shown in (**Figure 1**, case D and E).

Under the PLE extraction and the clean-up parameters in the present study, recoveries of PCBs were typically excellent over the entire conditions investigated. The use of glass column sorbents for the clean-up method that is filled with adsorbent as described in **Table 1**. The results showed that clean-up elution using any of the adsorbents from case A to E, the recoveries were comparable and satisfactory. However, the chromatograms for all of the cleaned samples were well resolved and they were having less background noise. Moreover, the optimal solvent for the clean-up elution was also investigated as described in **Table 1**. The results of the recoveries were also satisfactory for all PCBs congeners as shown from case A to case E.

#### 3.2. Effect of PLE Operating Parameters

Temperature is an important factor in PLE to enhance the recoveries. Increasing temperature has positive effects on increasing the extraction efficiency of the analytes, leading to improve the mass transfer of the analytes and consequently cause the analytes to increase their recoveries. Moreover, increasing temperature leads to reduce the surface tension and the viscosity of the solvent, which facilitate and improve the sample wetting and matrix penetration [10]. PLE is a solid-liquid extraction step, taking place in close-vessels at elevated temperature and pressure. The pressure is applied to maintain the organic solvents in its liquid state. The temperature should be above the solvents' atmospheric boiling points. These conditions allow the solvents to be below their critical conditions, but also enhance their salvation power and lower their viscosities, which lead to a higher diffusion rate for the extraction of the analytes. Applying all these conditions, the extraction efficiency is increased and the solvent needed is minimized.

Time is one of the important parameters to be considered and therefore, the time devoted to the optimization and development of the extraction procedure can be reduced. However, PLE is an attractive alternative method compared with conventional soxhlet method, due to the following reasons; its fast, less solvent consumption, and sequentially it allows extracting 6 samples with the use of different sample vessels sizes.

In general, increasing temperature (>140°C) causes serious disruption in the solute-matrix interactions resulting from Van der Waals forces, hydrogen bonding and or dipole attractions. These interactions could affect the recovery percentage obtained [10].

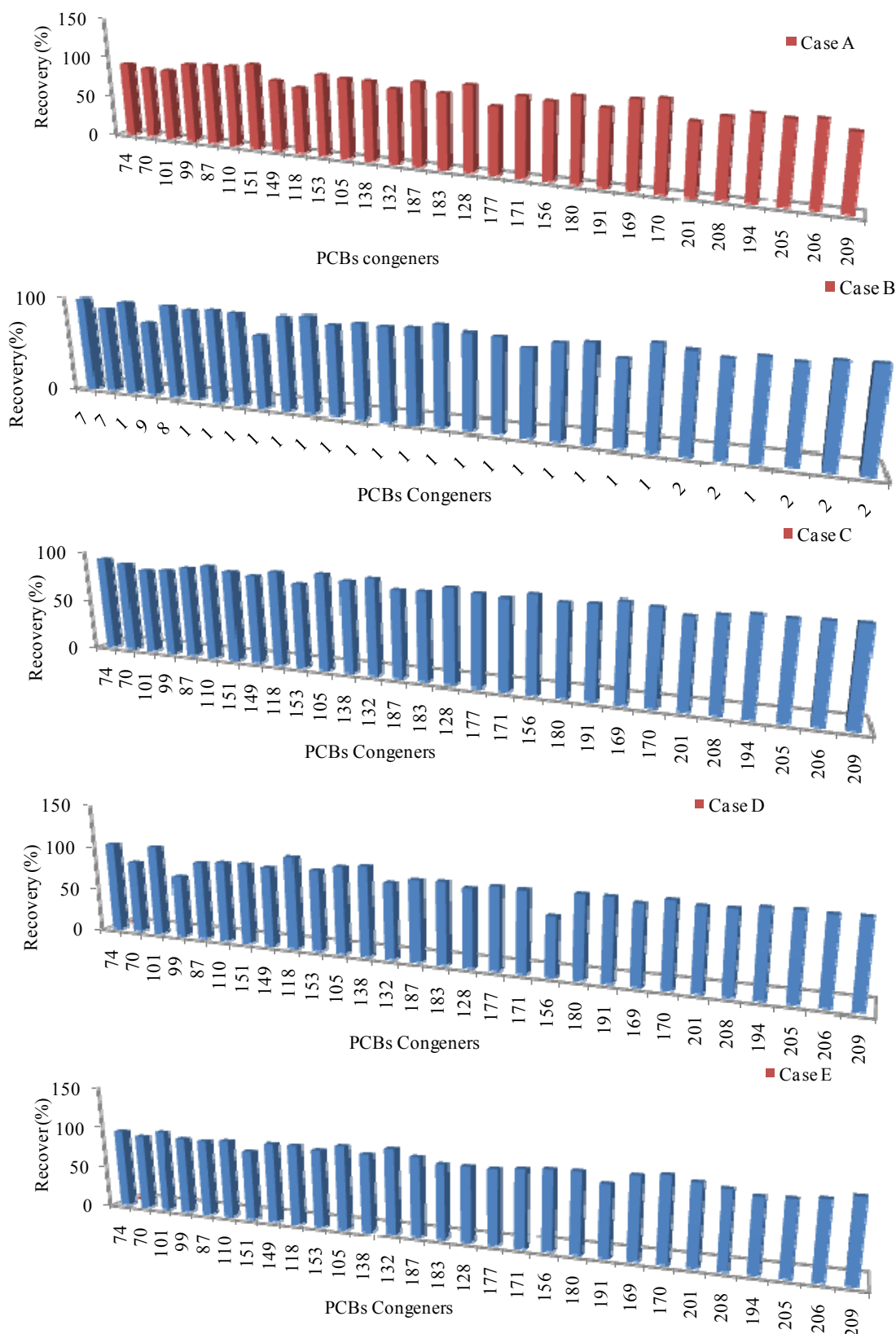


Figure 1. Recoveries percentages of the selected PCBs were obtained by spiking the standards before the extraction, using different adsorbents and solvent mixtures. The detail of each case is described in Table 1.

The extraction recoveries were decreased at temperatures above 140°C for most of the compounds. Temperatures above 140°C could result in the co-extraction of contaminants, which would affect the GC/MS (NCI) analysis. At a higher temperatures (>140°C), the chromatograms showed more background noise that lead to the identification of the peaks was difficult. This was assumed to be due to the presence of co-extracted material at higher temperatures. The highest extraction efficiencies were obtained at temperatures ranging from 100°C to 140°C. Based on these results, 120°C was selected for verification and optimization of the PLE method. Pressure produced no significant effect on the extraction process. A pressure of 1500 psi has been used in several studies to extract analytes from environmental matrices [22], as the higher pressures are generally applied to keep solvents in their liquid states [23,24].

### 3.3. Characterization of the Gel Permeation Chromatography (GPC) Column

The technique is based on molecular size separation and is primarily used to fractionate and to remove lipids (>500 Å), which elutes first from the column. A standard solution of PCBs was transferred into a gel permeation chromatography (GPC) column packed with 12 g of SX-3 Bio-Beads gel (200 - 400 mesh). The column was washed with 25 mL [hexane:DCM (1:1, v/v)] mixture. Then 100 mL [hexane:DCM (1:1, v/v)] solvent mixture was used to elute PCBs. The first 45 mL was discarded, since all of the lipids were eluted out. The next fraction (45 - 100 mL) was collected, since all of the PCBs were completely recovered in this elution. The recoveries of all targeted compounds were in the range of 85.2% to 102.6%. The advantages of GPC over concentrated sulfuric acid or saponification are its nondestructive nature, which allows large amounts of lipids to be handled, and it has greater applicability for unknown contaminants.

### 3.4. Method Validation

#### 3.4.1. Quality Control

Set of experiments were conducted to obtain acceptable and reliable data by using mirex as internal standard (I.S.D). The extracted fresh and blank samples were spiked with I.S.D. (300 ng/g) before extraction. All analytical data were assessed for compliance with acceptable criteria for method validation; the average of recoveries was required to be within 70% - 125%. However recoveries obtained for this experiment was generally over 75% and close to 100% for most of the PCBs congeners. Thus, the recoveries were considered to be satisfactory, and no interference or serious co-elution was encountered during the evaluation process. A surrogate mixture of <sup>13</sup>C-PCBs

containing the following numbers: 101, 138, 153, 180 and 209 were spiked to the extracted samples, the recoveries obtained for all the cases from A to E were calculated and were found to be above 79% for most of the <sup>13</sup>C-PCBs. **Table 1**, shows the total concentration spiked for each case as well as the range of the recoveries.

#### 3.4.2. Calibration Graph

Triplicate injection of 1 µl of a set of four standard solutions containing different concentrations of PCBs (21.5, 52.5, 109 pg/µl) were checked under the optimization conditions of GC/MS (NCI). Internal standard (mirex = 300 pg/µl) was added to the standard solutions. The calibration graphs were obtained in the range 21.5 pg/µl to 109 pg/µl with a correlation coefficients ( $r^2$ ) exceeding 0.996 for most of the PCBs compounds. The correlation coefficients ( $r^2$ ) were satisfactory in the concentration range assayed.

The results obtained for the retention times were acceptable over sets of standards and samples that were studied throughout the experiment. **Table 2**, represents the calibration data, correlation coefficient ( $r^2$ ), regression equations, calibration range, retention times, limit of detections and limit of quantitation for the group of PCBs were analyzed by NCI mode.

#### 3.4.4. Limit of Detection (LOD) and Limit of Quantitation (LOQ)

The limit of detections (LODs) were obtained as the compound concentration that cause a peak height equal to three times the baseline noise, whereas the limit of quantitations (LOQs) were obtained as the compound concentration that cause a peak height equal to ten times the baseline noise. Thus LODs and LOQs obtained for PCBs using these conditions are presented in **Table 2**, and the values were found to be between 0.01 ng/g (w/w) to 0.23 ng/g (w/w) and 0.03 ng/g (w/w) to 0.93 ng/g (w/w) for LODs and LOQs, respectively.

#### 3.4.5. Precision and Accuracy

The precision of the proposed method is expressed in terms of relative standard deviation (RSD). Wet fish sample was spiked at three different levels of PCBs standards (21.5 ng/g, 52.5 ng/g, 109 ng/g). Repeatability was performed five times ( $n = 5$ ) in the same days under the same conditions. The calculated values showed that the relative standard deviations (RSDs) for most of PCBs compounds were ranging between 0.51% and 17.4%. Reproducibility for PCBs was ranging from 75.2% to 103.2%, with RSDs between 1.8% and 22.6 %.

### 3.5. Concentrations of PCBs in Fish Samples

The fish samples were collected from Kuwait's fish mar-

**Table 2. Calibration data, correlation coefficient ( $r^2$ ), regression equations, calibration range, retention times, LOD and LOQ, for the different PCBs-congeners analyzed by NCI mode.**

PCBs	Correl. Coeff. ( $r^2$ )	Regress. Equation	Calib. range (pg/ $\mu$ l)	Retention time, min $\pm$ S.D. ( $n = 3$ )			LOD (ng/g)	LOQ (ng/g)	m/z
				21.5 (pg/ $\mu$ l)	54.5 (pg/ $\mu$ l)	109 (pg/ $\mu$ l)			
74	1.000	3.89e - 0.001	21.5 - 109	17.53 $\pm$ 0.09	17.46 $\pm$ 0.01	17.56 $\pm$ 0.01	0.02	0.06	290/292
70	1.000	7.63e - 001	21.5 - 109	17.66 $\pm$ 0.01	17.63 $\pm$ 0.01	17.63 $\pm$ 0.01	0.01	0.03	290/292
101	0.998	1.12e + 000	21.5 - 109	18.92 $\pm$ 0.01	18.91 $\pm$ 0.01	18.92 $\pm$ 0.01	0.09	0.32	326/328
99	1.000	5.07e - 000	21.5 - 109	19.19 $\pm$ 0.01	19.18 $\pm$ 0.0	19.18 $\pm$ 0.0	0.01	0.04	326/328
87	0.995	6.57e - 001	21.5 - 109	20.38 $\pm$ 0.0	20.36 $\pm$ 0.01	20.37 $\pm$ 0.01	0.03	0.09	326/328
110	0.997	1.02e + 000	21.5 - 109	20.99 $\pm$ 0.01	20.98 $\pm$ 0.0	20.97 $\pm$ 0.0	0.20	0.68	326/328
151	0.996	8.43e + 000	21.5 - 109	21.77 $\pm$ 0.0	21.76 $\pm$ 0.0	21.76 $\pm$ 0.0	0.23	0.78	326/328
149	0.995	2.14 + e000	21.5 - 109	22.62 $\pm$ 0.02	22.61 $\pm$ 0.01	22.61 $\pm$ 0.0	0.16	0.52	360/362
118	0.995	7.91e + 000	21.5 - 109	22.87 $\pm$ 0.01	22.82 $\pm$ 0.0	22.81 $\pm$ 0.01	0.02	0.05	326/328
153	0.995	1.23e + 001	21.5 - 109	24.63 $\pm$ 0.28	24.61 $\pm$ 0.02	24.61 $\pm$ 0.01	0.10	0.34	360/362
105	1.000	1.43e + 000	21.5 - 109	24.82 $\pm$ 0.02	24.94 $\pm$ 1.74	24.92 $\pm$ 0.02	0.28	0.93	326/328
138	1.000	7.51e + 000	21.5 - 109	27.19 $\pm$ 0.01	26.15 $\pm$ 0.08	27.16 $\pm$ 0.01	0.04	0.15	360/362
132	0.996	3.98e + 000	21.5 - 109	27.41 $\pm$ 0.0	27.34 $\pm$ 0.01	27.38 $\pm$ 0.01	0.03	0.09	360/362
187	0.995	5.61e - 000	21.5 - 109	29.02 $\pm$ 0.01	29.00 $\pm$ 0.01	29.00 $\pm$ 0.01	0.04	0.13	360/362
183	0.996	5.57e + 000	21.5 - 109	29.60 $\pm$ 0.01	29.58 $\pm$ 0.02	29.58 $\pm$ 0.01	0.02	0.06	394/396
128	1.000	6.57e + 000	21.5 - 109	30.17 $\pm$ 0.01	30.12 $\pm$ 0.01	30.12 $\pm$ 0.01	0.02	0.07	360/362
177	0.994	4.36e + 000	21.5 - 109	32.29 $\pm$ 0.01	32.28 $\pm$ 0.01	32.29 $\pm$ 0.01	0.04	0.15	394/396
171	0.993	4.91e + 000	21.5 - 109	32.75 $\pm$ 0.01	32.73 $\pm$ 0.01	32.73 $\pm$ 0.01	0.03	0.10	394/396
156	1.000	1.02e + 000	21.5 - 109	33.01 $\pm$ 0.0	32.96 $\pm$ 0.01	32.95 $\pm$ 0.02	0.21	0.69	360/362
180	0.996	6.02e + 000	21.5 - 109	34.59 $\pm$ 0.01	34.58 $\pm$ 0.01	34.57 $\pm$ 0.0	0.05	0.15	394/396
191	0.994	6.4e + 000	21.5 - 109	35.15 $\pm$ 0.01	35.14 $\pm$ 0.02	35.13 $\pm$ 0.01	0.01	0.04	394/396
169	0.999	7.05e + 000	21.5 - 109	36.49 $\pm$ 0.01	36.43 $\pm$ 0.0	36.41 $\pm$ 0.02	0.01	0.05	360/362
170	0.994	5.13e + 000	21.5 - 109	36.91 $\pm$ 0.01	36.89 $\pm$ 0.0	36.89 $\pm$ 0.01	0.04	0.14	394/396
201	0.993	4.86e + 000	21.5 - 109	37.57 $\pm$ 0.0	37.560.01	37.56 $\pm$ 0.01	0.03	0.10	394/396
208	0.993	4.1e + 000	21.5 - 109	39.87 $\pm$ 0.0	39.86 $\pm$ 0.0	39.86 $\pm$ 0.0	0.02	0.07	462/464
194	0.993	1.02e + 001	21.5 - 109	41.21 $\pm$ 0.01	41.20 $\pm$ 0.01	41.20 $\pm$ 0.0	0.09	0.33	428/430
205	0.994	1.05e + 001	21.5 - 109	41.50 $\pm$ 0.0	41.49 $\pm$ 0.0	41.49 $\pm$ 0.01	0.08	0.25	428/430
206	0.993	8.42e + 000	21.5 - 109	43.28 $\pm$ 0.03	43.27 $\pm$ 0.03	43.27 $\pm$ 0.0	0.01	0.04	462/464
209	0.993	6.08e + 000	21.5 - 109	44.85 $\pm$ 0.01	44.83 $\pm$ 0.01	44.84 $\pm$ 0.01	0.01	0.03	496/498

ket and tested by the presented method in order to evaluate the levels of PCBs compounds. The PCBs levels in fish samples were ranging from 0.81 ng/g to 4.46 ng/g (w/w) that is in agreement with the levels evaluated by the previously published two methods [25,26]. The levels obtained from the two methods were as follows: 0.76 - 5.72 ng/g (w/w) and 0.44 - 2.53 ng/g (w/w), for method [25] and method [26], respectively. Moreover, comparison of the levels obtained in this study with those obtained by Yang from China are equal to 3.77 ng/g (w/w) in fish [27], indicating that the levels obtained in this study are comparative to the levels obtained by other methods. Based on this study, total  $\Sigma$  PCBs concentrations in Kuwait's seafood are deemed to be generally not high.

#### 4. Conclusion

The comprehensive method was established, based on PLE extraction, clean-up of PCBs extracts from fish samples with different adsorbent materials. The detected PCBs congeners showed good accuracy, precision, and linearity range in the method studied. The advantage derived from using NCI source along with GC/MS detection is being capable of obtaining high selective method, resulting in eliminating the interferences of substances from sample matrices. Furthermore, allowing detection of PCBs in fish tissue at low ng/g levels. This method provides better recoveries when using any of the selected adsorbent with the clean-up solvent mixture. The relative standard deviation of the method was acceptable and within the required range. PLE offers the advantage of reducing the amount of organic solvent consumed and it can be easily automated.

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