

Does pH of tyrode solution modify glucose and electrolyte jejunal absorption in rats?

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ABSTRACT

Studies have shown that glucose is able to decrease the pH of the surface epithelium jejunal preparations when added *in vitro* and the existence of a high concentration of protons in the immediate area of the mucosa could be of considerable significance for absorption of electrolytes. The aim of this study is to assess whether the change in pH of Tyrode (solution used for perfusion of the jejunum) interferes with the absorption of glucose and electrolytes. Male Wistar rats weighing 200 to 220 g (n = 6) were utilized. Jejunal absorption of glucose and electrolytes was investigated in rats. A Tyrode solution containing twice glucose, sodium and potassium concentration (pH 7.0, 7.4, 8.0 and 8.5) was infused through the jejunal loops during 40 minutes. The glucose absorption was not significantly affected by Tyrode. However, there was significantly decrease in sodium absorption at pH 7.0 and 8.5 (41.13 ± 2.79 and 41.37 ± 1.71 , respectively, $P < 0.05$) when compared with the uptake at pH 7.4 and 8.0 (61.06 ± 6.50 and 56.28 ± 7.03 , respectively, $P < 0.05$). Moreover, potassium absorption increased at pH 8.0 (1.04 ± 0.07) when compared with the uptake at pH 7.0 (0.59 ± 0.04), 7.4 (0.78 ± 0.08) and 8.5 (0.54 ± 0.05) ($P < 0.05$). These data indicate that the pH of Tyrode has no influence on glucose absorption. However, the major potassium uptake occurs at pH 8.0, while the absorption of sodium is impaired at pH 7.0 and 8.5.

Keywords: Absorption; pH; Glucose; Sodium; Potassium; Jejunum

1. INTRODUCTION

Studies have shown that glucose is able to decrease the pH of the surface epithelium of duodenal and jejunal preparations when added *in vitro* [1,2], although the aci-

dification is not caused by glucose transport alone [3]. This effect is reproduced when the substrate is administered *in vivo* [1]. It is plausible that the generation of protons, which are secreted into this region, is closely related to the intracellular metabolism of glucose and lactate production. The existence of a high concentration of protons in the immediate surface of the mucosa could be of considerable significance for the absorption of electrolytes [1].

Recent studies suggest that changes in extracellular pH affect the ionization state of membrane proteins including transporters and ion channels with specific functions in transporting epithelia [4]. The surface pH of rat intestine had been measured *in vivo*, confirming the phenomenon of acidification in the jejunum followed by an alkalization process in the ileum [5]. However, no difference in the pH of the intestinal surface was detected when the bicarbonate buffer was used or when glucose was included in the buffer. Several factors can affect the intestinal absorption of glucose, electrolytes and water, such as nutritional deficiency [6] or absorption modulation by hormones [7] and peptides [8-12] via the autonomic nervous system. However, the different experimental models used in absorption studies have employed different physiological solutions for infusion, such as Ringer's solution [13], Krebs' solution [14], Tyrode solution [8-11,14,15] and phosphate buffer [16,17], with a broad range of variation in the pH of these solutions (pH 7 to pH 8).

The purpose of the present study was to determine whether a change in the pH of Tyrode solution (used for jejunal perfusion) affects the absorption of glucose and electrolytes. These significant new data concerning the dependence of glucose or electrolytes absorption on the pH of Tyrode solution may help to improve methodological approach in future studies of intestinal transport.

2. MATERIAL AND METHODS

2.1. Animals

Adult male Wistar rats weighing 200 to 220 g were

housed under standard laboratory conditions of a 12:12-h light/dark cycle and controlled temperature ($23^{\circ}\text{C} \pm 3^{\circ}\text{C}$). The animals were fasted for 12 h prior to the experimental procedures, but water was offered ad libitum. All experiments complied with the International Principles of Animal Care and the study received approval from the local Ethics Committee on Animal Experimentation (CETEA/UFMG process n^o 230/2010).

2.2. General Procedures

The rats were anesthetized with thiopental (Cristália, Brazil) (40 mg/kg i.p.). Following the procedures of median xypho-pubic laparotomy, a 20-cm segment of jejunum after the duodenojejunal ligament was isolated, preserving the nerves and the vascular pedicle. Two cannulae were then inserted into the extremities of the jejunal loop—one for perfusion and the other for fluid drainage. The abdominal wall was then closed in order to prevent tissue dehydration. Both cannulae were exteriorized through the extremities of the abdominal suture. Tyrode solution (137 mM NaCl, 2.7 mM KCl, 1.36 mM CaCl_2 , 0.49 mM MgCl_2 , 11.9 mM NaHCO_3 , and 5 mM D-glucose) was maintained in a bottle connected to the catheter infusion pump at 37°C in a bath. Tyrode solution pH 8.0 (buffered by HCO_3^-) was perfused at a rate of $0.5 \text{ ml} \cdot \text{min}^{-1}$ for 15 min in order to equilibrate the fluids to reach steady state within the jejunal lumen [8].

The rats were divided into four groups: Group 1 (n = 6) received Tyrode solution pH 7.0; Group 2 (n = 6) received Tyrode solution pH 7.4; Group 3 (n = 6) received Tyrode solution pH 8.0; Group 4 (n = 6) received Tyrode solution pH 8.5. The rats in all four groups were submitted to the infusion of a Tyrode solution containing twice the usual concentrations of glucose, sodium and potassium during the 40-min experiment, under the same conditions described above for Tyrode solution. The effluents were collected separately in test tubes at 10-min intervals and maintained in ice, then stored in a freezer at -20°C until the biochemical analysis.

2.3. Biochemical Determinations

The glucose concentration of the effluent was determined by an enzymatic method based on glucose oxidase (Glucose PAP Liquiform, Labtest, Brazil), estimated from a standard glucose concentration curve. Effluent potassium and sodium ion concentrations were measured by flame photometry. The results were expressed by the difference between influx and efflux.

2.4. Statistical Analysis

Split-plot analysis of variance (ANOVA), followed by the Student-Newman-Keuls method, was used for the statistical analysis, with the level of significance set at

5% ($P < 0.05$).

3. RESULTS AND DISCUSSION

Figure 1 shows the effect of Tyrode pH on jejunal glucose absorption. There were no differences in glucose absorption among the four groups.

Figure 2 displays the effect of Tyrode pH on jejunal sodium absorption. There was a significant decrease in sodium absorption at pH 7.0 and 8.5 when compared with uptake at pH 7.4 and 8.0 ($P < 0.05$). Tyrode pH 7.4 increased sodium absorption. Moreover, no statistically significant difference was found between Tyrode pH 8.0 and Tyrode pH 7.4.

A significant decrease in sodium absorption at pH 7.0 and 8.5 as compared with uptake at pH 7.4 and 8.0 probably is a result of adverse conditions of extreme acidosis or alkalosis. However, the physiological pH favored sodium absorption that was not different from pH 8.0.

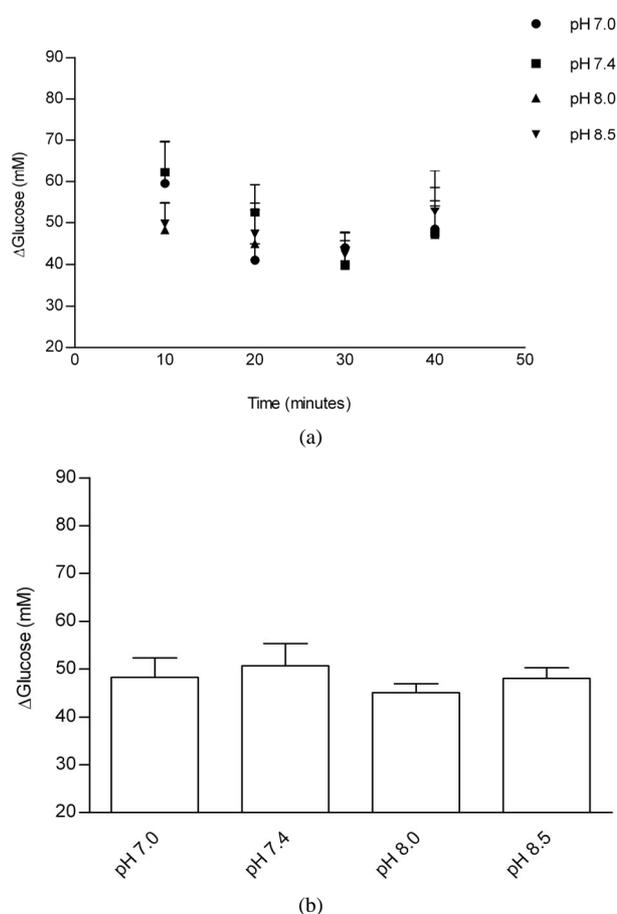


Figure 1. Effects of Tyrode pH on jejunal glucose absorption; (a) represents the time course of the effect of Tyrode pH; and (b) represents mean \pm SEM for all time points of each pH. The results were expressed by the difference between influx and efflux. There was no difference in glucose absorption among the groups perfused with Tyrode solution at different pH values.

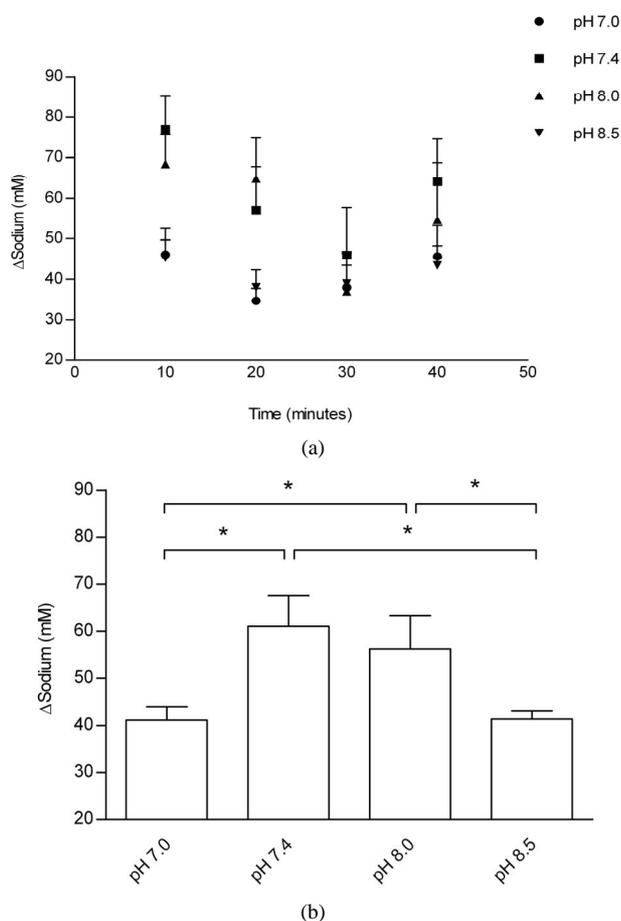


Figure 2. Effects of Tyrode pH on jejunal sodium absorption; (a) represents the time course of the effect of Tyrode pH; (b) represents mean \pm SEM for all time points of each pH. The results were expressed by the difference between influx and efflux. Sodium absorption in the group perfused with Tyrode solution pH 7.4 was higher than that of the other groups ($^*P < 0.05$). Sodium absorption increased at pH 8.0 when compared with Tyrode pH 8.5 and pH 7.0 ($^*P < 0.05$). * Denotes significant ($P < 0.05$) difference in relation to the other groups.

Figure 3 shows the effect of Tyrode pH on jejunal potassium absorption. Absorption in the group that received Tyrode solution pH 8.0 was greater than that in the groups that received solutions with pH 8.5 and 7.4 ($P < 0.05$). It seems that the absorption of potassium is strongly dependent on pH. Moreover, there was a decrease in potassium absorption at pH 7.0, indicating that a high concentration of protons in the Tyrode solution could be of considerable significance for impair potassium jejunal absorption. This impairment maybe due to conformational changes in potassium channels since the activation of these channels occurs in extracellular alkalization [4,18-20].

The novel findings of the present study are that the pH of Tyrode solution in jejunal perfusion did not affect glucose absorption, whereas greater potassium absorption occurred with Tyrode pH 8.0 and greater sodium absorption occurred with Tyrode pH 7.4.

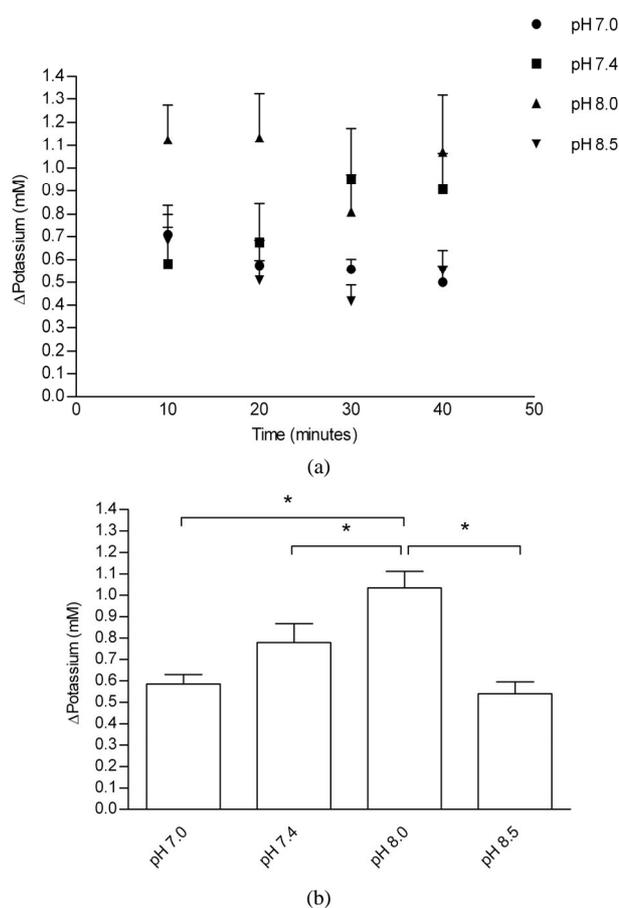


Figure 3. Effects of Tyrode pH on jejunal potassium absorption; (a) represents the time course of the effect of Tyrode pH; (b) represents mean \pm SEM for all time points of each pH. The results were expressed by the difference between influx and efflux. Potassium absorption in the group perfused with Tyrode solution pH 8.0 increased when compared with uptake at pH 7.0, 7.4 and 8.5 ($^*P < 0.05$). * Denotes significant ($P < 0.05$) difference in relation to the other groups.

The transport of glucose is mainly affected by the amount of sodium in the intestinal lumen. A high concentration of sodium on the mucosal surface facilitates the glucose influx into the epithelial cells. Glucose and sodium share the same co-transporter 1 (SGLT1) in the brush border membrane. Thus, glucose transport is an example of secondary active transport, provided indirectly by the active transport of sodium out of the cell, which maintains the concentration gradient across the luminal border of the cell. Glucose then passively passes from the enterocytes to the circulation through glucose transporter 2 (GLUT-2) in the basolateral membrane [21]. The Na^+/H^+ exchange which occurs by NHE1 found in basolateral membrane, NHE2 and NHE3 both at the brush border membrane of the intestinal epithelium [22-25] and Na^+ absorption through the Na^+/Cl^- co-transporter [22] are alternative sodium absorption pathways. It could be suggested that these alternative pathways to sodium uptake were impaired

by Tyrode pH, which would explain the dissociation of glucose and sodium uptakes regarding Tyrode pH.

In addition, acidification (pH 7.0) or alkalization (pH 8.5) produced by changes in the pH of Tyrode solution impaired sodium absorption. Based on the absorption studies, Tyrode pH 7.4 is the best choice for improving sodium absorption, followed by Tyrode pH 8.0.

Acidification produced by Tyrode pH 7.0 impaired potassium absorption. Moreover, the physiological condition (Tyrode pH 7.4) or alkalization (Tyrode pH 8.5) reduced potassium absorption when compared to Tyrode pH 8.0. The mechanism proposed for potassium absorption in the small intestine is passive diffusion through either an intercellular or paracellular pathway, favoring a potential transmembrane gradient due to water absorption secondary to the active transport of sodium or other solutes that concentrate potassium within the intestinal lumen [26]. It has also been suggested that potassium may be absorbed by an active transport mechanism operating through the cellular pathway [26]. Interestingly, potassium is the most abundant ion in intracellular fluid and plays an important role in acid-base homeostasis. Moreover, the results of the present study suggest that potassium absorption mechanisms seem to be dependent on pH. The active transport mechanism is likely to be influenced by pH.

4. CONCLUSION

In conclusion, Tyrode pH did not affect jejunal glucose absorption, but did affect jejunal electrolyte absorption. The optimal value for jejunal potassium absorption was pH 8.0 and the optimal value for sodium absorption was 7.4, which did not differ significantly from the result obtained with pH 8.0. These results suggest that a Tyrode solution with pH 8.0 is the best choice in studies involving jejunal perfusion and glucose and electrolyte absorption.

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