

Ethanol Tolerance in *Aspergillus niger* and *Escherichia coli* Phytase

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ABSTRACT

Despite yeast having its own native phytase, the high levels of phytate found in DDGS, a byproduct of ethanol (ETOH) fermentation, suggest that its activity is diminished in the presence of ETOH. Ethanol, a product of grain fermentation, is known to inactivate several hydrolytic enzymes but its effect on phytases is relatively unknown. In this study, two phytases, *Aspergillus niger* (PhyA) and *Escherichia coli* (AppA2), were tested for ETOH tolerance. The *E. coli* phytase displayed greater ethanol tolerance over fungal phytase in the 5 to 10% range. However, ETOH inactivation was found to be reversible for both the enzymes. These differences in ETOH tolerance do suggest that there is a potential to achieve higher ETOH tolerance in phytases by “structure-function” studies to lower phytic acid levels in DDGS and for other applications.

Keywords: Phytase; Ethanol; DDGS; Fermentation; *Aspergillus niger*

1. Introduction

Phytate is the principal storage compound for phosphorus in plants. With the increased utilization of high phytate containing plant meals over the last several decades, extensive research has focused on the deployment of phytases as an animal feed additive. This is to allow monogastric animals (swine, poultry, etc.) which lack a digestive phytase to obtain the phytate’s ortho-phosphate groups, which otherwise will be unavailable.

More recently, demands for additional bio-based fuels have spurred enhanced fermentation of corn and other grains such as sweet sorghum to produce ethanol in response to the growing demand. This has resulted in increased amounts of dried distillers grains with solubles (DDGS) emanating from this process. The DDGS is rich in nutrients and has much potential as an animal feed. However, recent studies have reported that DDGS contains high levels of phytate [1]. Moreover, ethanol is known to inhibit the activity of several hydrolytic enzymes [2,3] and these results suggest that the phytase produced by the native yeast, *Saccharomyces cerevisiae*, [4] would be inhibited by increased concentration of ethanol. In this study, the ethanol tolerances of two phytases that are marketed as animal feed additive are determined. While both are histidine acid phosphatases (HAPs) and share the same active site geometry and catalytic mechanism, one is from *Aspergillus niger* and

the other is produced by *Escherichia coli*, each has its own unique catalytic properties. While no information exist on ethanol tolerance in phytases, any differences in ethanol tolerance in the two enzymes can enhance our understanding of how ethanol interacts with this class of enzymes and this may contribute to the designing of phytases that retains more activity during fermentation and thus lower the phytic acid content of DDGS. In addition, the achievement of a molecular modification to enhance ethanol tolerance in phytase may also have further applications in enhancing the ethanol tolerance of other hydrolytic enzymes.

2. Materials and Methods

2.1. Source of Phytase

Fungal phytase was obtained from the cloned *Aspergillus niger phyA* gene that was overexpressed in *Pichia pastoris*. The recombinant phytase was purified using sequential ion-exchange column chromatographies.

E. coli phytase, the *AppA2* gene product, was a gift from Phytex LLC, Portland, Maine, which was overexpressed in *Pichia pastoris*. The crude culture filtrate was dialyzed against 25 mM glycine, pH 2.8 buffer and loaded onto a MacroPrep™ S column and eluted as a single activity component in a linear salt (0 - 0.5 M sodium chloride) gradient. The final specific activity of the phytase was about 15,000 nkat/mg of protein at 55°C.

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2.2. Phytase Assay

Phytase assays were carried out in 1.0 mL 50 mM acetate buffer, pH 5.0°C at 55°C similar to *A. niger* phytase assay [5]. Liberated inorganic ortho-phosphates were quantitated spectrophotometrically using a freshly prepared AMA reagent consisting of acetone, 10 mM ammonium molybdate, and 5.0 N sulfuric acid, (2:1:1, v/v) [6]. Adding 2.0 mL AMA solution per assay tube terminated phytase assay. After 30 seconds, 0.1 mL of 1.0 M citric acid was added to each tube to fix the color generated by AAM reagent. Absorbance was read at 355 nm after blanking the spectrophotometer with appropriate control. Values were expressed as nkat/mL, where kat is defined as moles of substrate converted per second.

2.3. Ethanol in the Inorganic Phosphate Measurement

A 50 mM acetate buffer containing 0% - 10% ethanol and 200 μ M aliquot of potassium phosphate (K_2HPO_4) in 1 ml volume was mixed with AMA reagent followed by citrate as in the phytase assay to measure the inorganic ortho-phosphate. This is to rule out any inference of ethanol in the detection and quantification of inorganic orthophosphates.

2.4. Effect of Ethanol on Phytase Activity

Phytases (8 μ L PhyA and 15 μ L AppA2) were incubated with 0 to 10% ethanol in 1 mL volume at room temperature for 10 min. Then they were transferred to a 55°C water bath for 2 min before phytase assay.

2.5. Stability of Phytases after Exposure to Ethanol

A 100 μ L aliquot of PhyA and 4 μ L of AppA2 were incubated at room temperature for 30 min in 0% - 10% ethanol in 1 mL 50 mM acetate buffer, pH 5.0. After incubation, an aliquot of PhyA (10 μ L) and AppA2 (15 μ L) were incubated with 75 μ L of 10 mM phytate for 1 min at 55°C. The liberated inorganic *ortho*-phosphates were measured as above.

3. Results

3.1. Ethanol Tolerance of Microbial Phytase

Samples of both PhyA and AppA2 phytase were incubated for 10 minutes in 0% to 10% ethanol to determine their respective ethanol tolerance. The results are shown in **Figure 1**. The PhyA phytase had activity (566 nkat/mL) at 0% ethanol to no activity at 10% denaturant concentration. However, *E. coli* phytase, AppA2, which had a comparably lower activity at 0% ethanol (275 nkat/mL), still had some activity (55 nkat/mL) at 10% dena-

urant concentration. This means a loss of 80% activity for AppA2 due to 10% ethanol.

3.2. Phytase Activity at Varying Concentration of Ethanol

The activity of the two phytases as a function of various concentration of ethanol is shown in **Figure 2**. The loss of activity in both the phytases were very similar in 0.5 to 3% range of ethanol concentration. While AppA2 phytase retained nearly 20% of its activity at 10% ethanol, the *A. niger* phytase was completely inactivated at 10% ethanol.

3.3. Ethanol Does Not Effect Phytase Assay

The possibility of ethanol interference with the phytase assay was shown not to be a significant factor (**Figure 3**).

3.4. Ethanol Effect on Phytase Is Reversible

Since inactivation of the enzyme by ethanol does not result in permanent denaturation of either of the enzyme (**Figure 4**), it indicates that the inhibition of phytases by

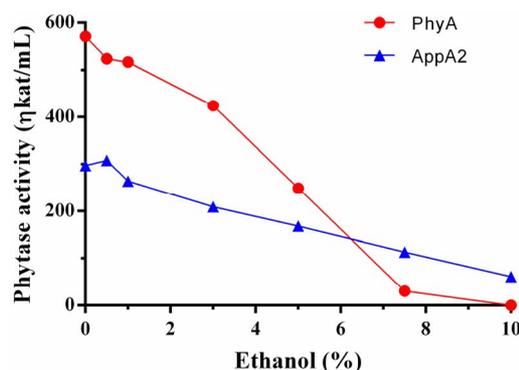


Figure 1. Effect of ethanol on phytase enzyme activity. Phytases (8 μ L phytase A and 15 μ L AppA2) were incubated with 0 to 10% ethanol in 1 mL volume at room temperature for 10 min. Then they were transferred to a 55°C water bath for 2 min before phytase assay.

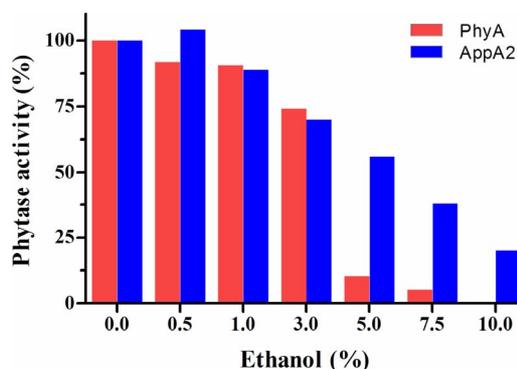


Figure 2. Effects of various ethanol concentration on phytase activity over the 0% - 10% range.

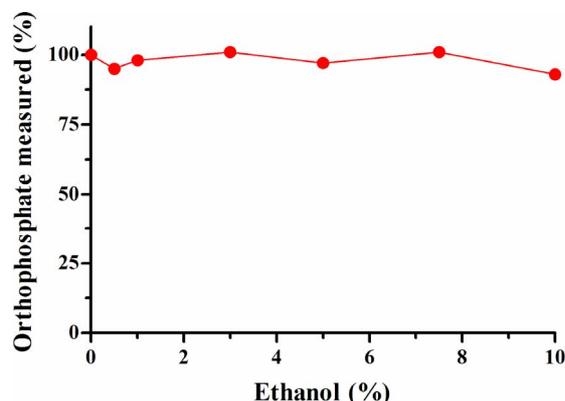


Figure 3. Ethanol in the inorganic phosphate measurement. 200 μM potassium orthophosphate (K_2HPO_4) was mixed with 0 to 10% ethanol in 1 mL 50 mM acetate buffer, pH 5.0 followed by AMA reagent and citrate. This is to rule out any interference of ethanol in the phytase assay.

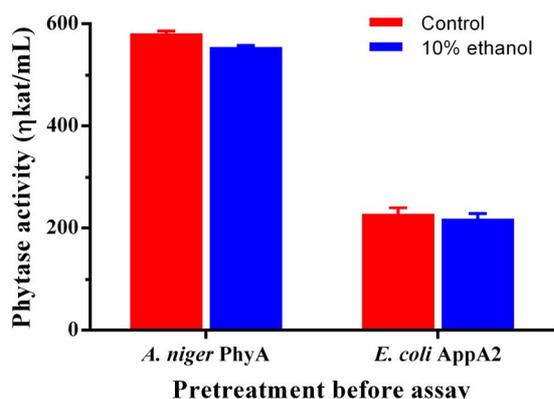


Figure 4. Recovery of phytases after exposure to ethanol. 100 μL phytase PhyA and 4 μL of AppA2 were incubated at room temperature for 30 min in presence and absence (control) of 10% ethanol in 1 mL 50 mM acetate buffer, pH 5.0. After incubation, an aliquot of Phy A (10 μL) and AppA2 (15 μL) were incubated with 75 μL of 10 mM phytate for 1 min at 55°C. The liberated inorganic orthophosphates were measured as above.

ethanol is a reversible process.

4. Discussion

Both *A. niger* and *E. coli* phytase displayed activity inhibition with increasing amounts of ethanol. However, at concentrations of ethanol above 3% the *E. coli* phytase retained significantly more activity than *A. niger* PhyA phytase. Both of these enzymes not only share a common catalytic mechanism, but they also display considerable amino acids divergence in their molecular structure.

Previous studies have shown that a change of just a single amino acid can alter physical properties of these enzymes [7,8]. In addition, researchers have shown that significant differences exist in these two enzymes in their response to sodium chloride [9]. The addition of sodium

chloride increases activity of the fungal phytase in the pH range 1.5 - 6.0. No increase in activity was achieved when AppA2 was tested with the same sodium chloride solution. Ullah and coworkers [9] attributed the differences in the response to sodium chloride to divergence in the electrostatic environment in the active site for the two enzymes.

While no information exist on the ethanol tolerance of any phytase, ethanol inhibition of other hydrolytic enzymes such as, cellulase, has been cited [2,10,11] and this has hindered the development of simultaneous saccharification and fermentation techniques in bio-ethanol production to maximize yields and lower both the cost and energy requirements.

A number of benefits have been proposed for the addition of phytase during fermentation. First, the hydrolysis of phytic acid results in more free minerals e.g., calcium, magnesium, zinc, iron, etc., that are needed for yeast metabolism and whose availability results in a higher fermentation rate [12]. Another is that starch hydrolysis with a α -Amylase has also been shown to be benefitted by the addition of a phytase to relieve phytic acid inhibition of α -Amylase during fermentation [13].

In the present study, a significant difference in ethanol tolerance has been observed in two commercially marketed phytases. Inactivation of both enzymes by ethanol is reversed by the removal of ethanol. The fact that differences in ethanol tolerance do exist supports the thesis that further tolerance can be achieved by structural modification of the enzyme. Such molecular modification may have application in elevating the ethanol tolerance in other hydrolytic enzymes.

5. Conclusion

The results presented in this study clearly demonstrated that ethanol, the main product of starch fermentation, severely inactivates two commercially important phytases' catalytic activity even at a low concentration of 10%. The differences in ethanol tolerance in the two phosphorhydrolases can enhance our understanding of how ethanol interacts with this class of enzymes and this may contribute to the design of phytases that retain more activity during fermentation and thus lower the phytic acid content of dried distillers grains with solubles (DDGS). The achievement of a molecular modification to enhance ethanol tolerance in phytase may find applications in other hydrolytic enzymes.

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