

# Preparation of High Ga Content Cu(In,Ga)Se<sub>2</sub> Thin Films by Sequential Evaporation Process Added In<sub>2</sub>S<sub>3</sub>

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## ABSTRACT

High Ga content Cu(In,Ga)Se<sub>2</sub> thin films incorporated sulfur were prepared by sequential evaporation from CuGaSe<sub>2</sub> and CuInSe<sub>2</sub> ternary compounds and subsequently Ga<sub>2</sub>Se<sub>3</sub>, In<sub>2</sub>Se<sub>3</sub> and In<sub>2</sub>S<sub>3</sub> binary compounds. The In<sub>2</sub>S<sub>3</sub>/(Ga<sub>2</sub>Se<sub>3</sub>+ In<sub>2</sub>Se<sub>3</sub>) ratio was varied from 0 to 0.13, and the properties of the thin films were investigated. XRD studies demonstrated that the prepared thin films had a chalcopyrite Cu(In,Ga)Se<sub>2</sub> structure. The S/(Se+S) mole ratio in the thin films was within the range from 0 to 0.04. The band gaps of Cu(In,Ga)Se<sub>2</sub> thin films increased from 1.30 eV to 1.59 eV with increasing the In<sub>2</sub>S<sub>3</sub>/(Ga<sub>2</sub>Se<sub>3</sub>+ In<sub>2</sub>Se<sub>3</sub>) ratio.

**Keywords:** Cu(In,Ga)Se<sub>2</sub> Thin Film; Solar Cell; High Ga Content; Sulfur Incorporation; Sequential Evaporation

## 1. Introduction

Photovoltaic power system has received considerable attention for safety and clean energy resources. It is necessary to fabricate low cost and high efficient solar cells in order to spread the PV system widely. Chalcopyrite Cu(In,Ga)Se<sub>2</sub> is a potential absorber material for high efficiency thin film solar cell because of its favorable band gap and high absorption coefficient for solar radiation. The band gap energy of Cu(In,Ga)Se<sub>2</sub> thin films varies from about 1.0eV to 1.7eV according to the increase in CuGaSe<sub>2</sub> molar fraction which makes it also promising for single-junction and multi-junction solar cell applications [1]. Conversion efficiencies for Cu(In,Ga)Se<sub>2</sub> based solar cells have been significantly improved over recent years and achieved the value of 20% by three-stage process using a multisource vacuum evaporation system equipped with elemental Cu, In, Ga and Se sources [2,3]. The Ga/(In+Ga) ratio of this absorber was around 0.3, which showed a band gap E<sub>g</sub> of about 1.14 eV. It is expected to improve the efficiency by increasing its band gap until 1.4 eV due to a better matching solar spectrum. The conversion efficiencies of Cu(In,Ga)Se<sub>2</sub> thin film solar cells decreased with increasing a Ga/(In+Ga) mole ratio above 0.3 [4]. For example, the efficiencies of Cu(In,Ga)Se<sub>2</sub> thin film solar cells were 12% for Ga/(In+Ga) mole ratio of 0.73 (E<sub>g</sub>=1.5 eV) and 10% for that of 0.91 (E<sub>g</sub>=1.62 eV), respectively [4]. On the other hand, a performance of Cu(In,Ga)Se<sub>2</sub> thin film solar cell with a Ga/(In+Ga) mole ratio of around 0.3 was improved by sulfurization of the film surface such as InS treatment by a wet process [5] and annealing in S vapor atmosphere [6]. We have proposed the process using a vacuum deposition apparatus with three evaporation boats which was the sequential evaporation technology from CuGaSe<sub>2</sub> and CuInSe<sub>2</sub> ternary compounds [7,8]. Our proposed process has advantages to be able to easily control a Ga/(In+Ga) mole ratio in Cu(In,Ga)Se<sub>2</sub> thin films by changing the amount of CuGaSe<sub>2</sub> and CuInSe<sub>2</sub> evaporating materials in the first step and to use inexpensive

equipment for preparation of an absorber layer. In this study, one evaporation source was added in our vacuum deposition apparatus. In<sub>2</sub>S<sub>3</sub> was added as an evaporation material in the third step of our sequential evaporation process and the prepared thin films and solar cells were investigated.

## 2. Experimental

### 2.1. Preparation of Cu(In,Ga)Se<sub>2</sub> Thin Films Added In<sub>2</sub>S<sub>3</sub>

The evaporating materials of CuGaSe<sub>2</sub> and CuInSe<sub>2</sub> were synthesized by reacting stoichiometric amounts of high-purity elements (Cu, In, Ga, Se) in sealed and evacuated quartz ampoules. The detail procedure was described in Reference [9]. The CuInSe<sub>2</sub> and CuGaSe<sub>2</sub> ingots were removed from the quartz ampoules. In<sub>2</sub>Se<sub>3</sub>, Ga<sub>2</sub>Se<sub>3</sub> and In<sub>2</sub>S<sub>3</sub> compounds available in the market were used as an evaporating material. Mo layer used as a back contact was prepared by rf magnetron sputtering onto soda-lime glass substrate in Ar ambient. Our evaporation process consists of the four steps, which schematic profile was shown in **Figure 1**. Before fabrication of Cu(In,Ga)Se<sub>2</sub> thin films, the Mo/soda-lime glass substrates were heated in vacuum for 5min at 500°C with infrared lamp. After cooling down to 200°C, in the first step, Cu-In-Ga-Se layer was evaporated from CuGaSe<sub>2</sub> and CuInSe<sub>2</sub> compounds onto the Mo/soda-lime glass. The CuGaSe<sub>2</sub>/(CuGaSe<sub>2</sub>+ CuInSe<sub>2</sub>) mole ratio of the evaporating materials kept at constant of 0.8. In the second step, In-Ga-Se layer was deposited from In<sub>2</sub>Se<sub>3</sub> and Ga<sub>2</sub>Se<sub>3</sub> compounds at a substrate temperature of 490°C. The (In<sub>2</sub>Se<sub>3</sub> + Ga<sub>2</sub>Se<sub>3</sub>)/(CuGaSe<sub>2</sub> + CuInSe<sub>2</sub>) mole ratio kept at constant of 0.2. In the third step, S was deposited from In<sub>2</sub>S<sub>3</sub> compound at a substrate temperature of 490°C. The In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio was varied from 0 to 0.13 in this experiment. Finally, only Se was effused at the same substrate temperature.

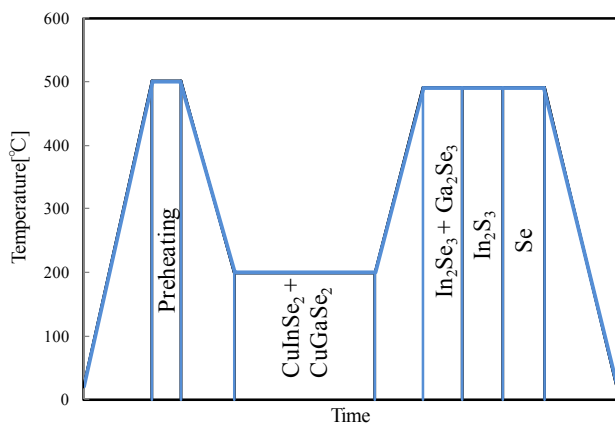


Figure 1. Schematic profile of our sequential evaporation process.

## 2.2. Fabrication of Solar Cells

The solar cells with a configuration of Al/ZnO:Al/i-ZnO/CdS/Cu(In,Ga)Se<sub>2</sub>/Mo/SLG substrate were fabricated. CdS buffer layer with a thickness of 70 nm was deposited by the chemical bath deposition technique using a CdI<sub>2</sub> (2.0×10<sup>-3</sup> M)-thiourea (0.166M)- ammonia (1M) aqueous solution during heating from room temperature to 65°C. i-ZnO buffer layer with a thickness of 100 nm was deposited by rf-magnetron sputtering from non-doped ZnO target in Ar gas at room temperature. Transparent conductive ZnO:Al film with a thickness of 0.4 μm was subsequently deposited by rf-magnetron sputtering from a 2wt%Al<sub>2</sub>O<sub>3</sub> doped ZnO target in Ar gas at room temperature. Al grids for the front electrode were formed by a vacuum evaporation with W boat using a metal mask. No antireflection coating was applied. The size of a solar cell is 5 mm x 5 mm.

## 2.3. Characterization

The surface composition of thin films were determined by an electron probe microanalysis (EPMA). The effective range of electron for production of the characteristic X-rays in EPMA analysis is roughly estimated to be around 0.4 μm for Cu (In,Ga)Se<sub>2</sub> thin films [10]. The growth orientation of thin films was studied by X-ray diffraction (XRD) in the θ-2θ mode using Cu Kα radiation. The surface and cross-section morphology and grain size of the thin films were studied by scanning electron microscopy (SEM). Current-voltage characteristics of solar cells were measured using standard 1-sun (AM1.5, 100mW/cm<sup>2</sup>) illumination. The quantum efficiencies of solar cells were measured using a spectrophotometer with illumination normalized against calibrated photodiode.

## 3. Results and Discussion

### 3.1. Film Composition

From EPMA analysis, the thin films prepared at various In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio had almost a stoichiometry composition in I-III-VI<sub>2</sub> compound. **Figure 2** shows the compositional ratio of Ga/(In+Ga) and S/(Se+S) in the thin films. In this experiment, the CuGaSe<sub>2</sub>/(CuGaSe<sub>2</sub>+CuInSe<sub>2</sub>) mole ratio in the evaporating materials was kept at constant of 0.8. The Ga/(In + Ga) mole ratio in the thin films prepared in the range of In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio from 0 to 0.13 was within the range from 0.855 to 0.747. The Ga/(In+Ga) mole ratio slightly decreased with increasing the In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio due to the presence of In in the third step. These values are considered to be a high Ga content which is the purpose of this study. On the other hand, the S/(Se+S) mole ratio in the tin films increased from 0 to 0.04 with increasing the In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio. A slightly S incorporation into the thin films was confirmed from EPMA analysis.

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### 3.2. Crystal Structure

**Figure 3** shows XRD patterns for the thin films prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio of 0 to 0.13. XRD spectrum exhibited several peaks corresponding to diffraction lines of the chalcopyrite phase in Cu(In,Ga)Se<sub>2</sub>, in particular split of 220/204 and 312/116 diffraction lines. The 112 diffraction line was the strongest. The position of X-ray diffraction peaks for Cu(In,Ga)Se<sub>2</sub> thin films prepared at various In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio was almost same although the Ga/(In+Ga) and S/(Se+S) mole ratio in the thin films was slightly different.

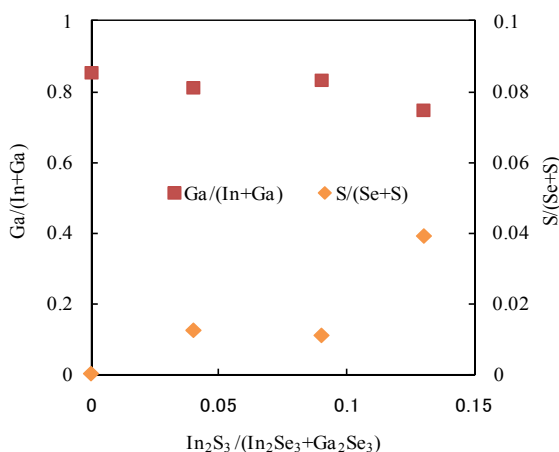


Figure 2. Compositional ratio of the prepared thin films determined by EPMA.

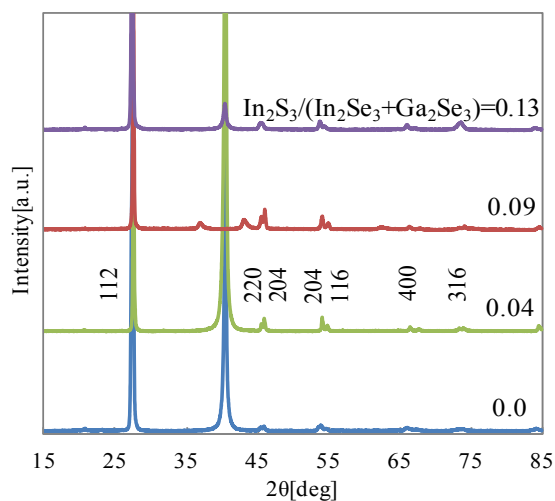


Figure 3. XRD patterns of the thin film prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>)=0-0.13.

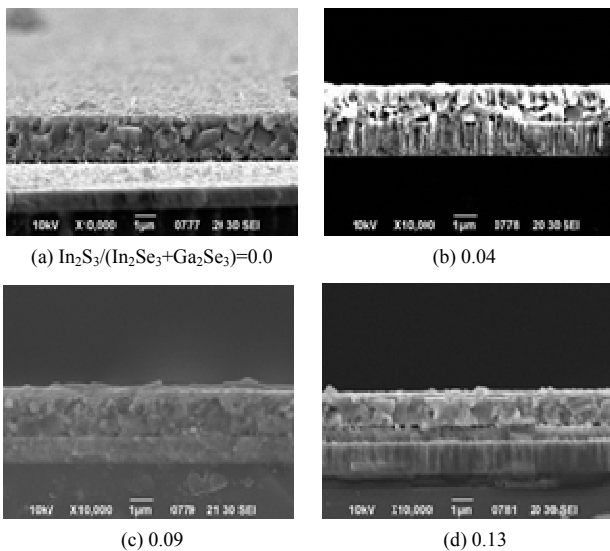
### 3.3. Grain Size

SEM micrographs of the cross section of Cu(In,Ga)Se<sub>2</sub> thin films prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio of 0 to 0.13 are shown in **Figure 4**. This Cu(In,Ga)Se<sub>2</sub> thin films had a high Ga content such as Ga/(In+Ga) mole ratio of the range from 0.855 to 0.747. The grain size in Cu(In,Ga)Se<sub>2</sub> thin film was estimated to be smaller than 1.0 μm. It is well known in general that efficiencies of polycrystalline solar cells increase with increasing grain sizes in the absorber materials. Therefore, the large grain growth in Cu(In,Ga)Se<sub>2</sub> thin films is required for the fabrication of high-performance photovoltaic devices. In comparison with **Figures 4(a)** and **(b)**, the grain size in **Figure 4(b)** was seemed to be larger than that in **Figure 4(a)**, suggesting the promotion of the grain growth by slightly In<sub>2</sub>S<sub>3</sub> supplying.

### 3.4. Band gap Engineering

The solar cells with a configuration of ZnO: Al/i-ZnO/CdS/Cu(In,Ga)Se<sub>2</sub>/Mo/soda-lime glass substrate were fabricated by using Cu(In,Ga)Se<sub>2</sub> thin films prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub> + Ga<sub>2</sub>Se<sub>3</sub>) = 0-0.13. The best solar cell demonstrated V<sub>oc</sub>=500mV, I<sub>sc</sub>= 19.07mA/cm<sup>2</sup>, FF = 0.39 and η = 4.1% without AR-coating, which used Cu(In,Ga)Se<sub>2</sub> thin film prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub> + Ga<sub>2</sub>Se<sub>3</sub>)=0.04. The efficiencies for Cu(In,Ga)Se<sub>2</sub> thin film solar cells were not so good. However, the remarkable change was observed in the quantum efficiency of Cu(In,Ga)Se<sub>2</sub> thin film solar cells, which was shown in **Figure 5**. The quantum efficiency from 400 nm to 600nm for Cu(In,Ga)Se<sub>2</sub> thin film solar cells prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) = 0.04 and 0.09 increased rather than that at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) = 0. Moreover, the absorption band edge in the long wavelength region shifted to the short wavelength range with increasing In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio. For a direct transition, the dependence of the absorption coefficient α on the photon energy hν is given by

$$\alpha h\nu = A(h\nu - E_g)^{1/2} \quad (1)$$



**Figure 4.** SEM micrographs of the cross-section of Cu(In,Ga)Se<sub>2</sub> thin films prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) =0-0.13.

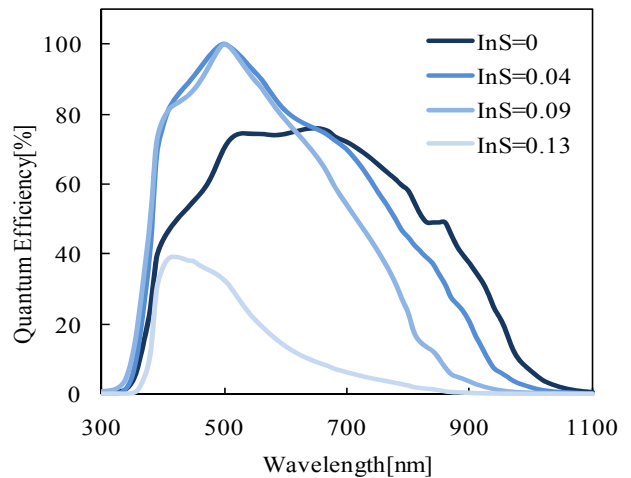
Assuming a very short minority carrier diffusion length L<sub>n</sub>, the quantum efficiency QE can be approximated by

$$QE = 1 - \exp(-\alpha W) \quad (2)$$

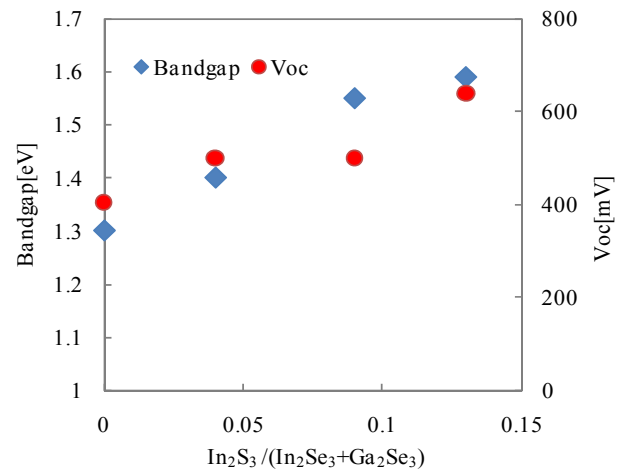
where W is the width of the space charge region. From equations (1) and (2), the following equation is deduced

$$\ln(1 - QE) \times h\nu = -WA(h\nu - E_g)^{1/2} \quad (3)$$

so that a plot of [hν x ln(1-QE)]<sup>2</sup> against hν can be used to extrapolate the band gap E<sub>g</sub> [11]. From this manner, the band gaps estimated from the QE spectra were changed from 1.30 eV to 1.59 eV, which shown in **Figure 6** including the open circuit voltage V<sub>oc</sub> of Cu(In,Ga)Se<sub>2</sub> thin film solar cells. The value of band gap is expected to be 1.56eV for Cu(In,Ga)Se<sub>2</sub> thin film with Ga/III=0.855 from the data reported by Paulson et al [12]. However, the band gap of 1.30eV obtained from Cu(In,Ga)Se<sub>2</sub> thin film prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>)=0 in this experiment was extremely a small value. It has been reported that efficient Cu(In,Ga)Se<sub>2</sub> thin film solar cells with Ga/(In+Ga) mole ratio of 0.3 fabricated by three stage process had a double



**Figure 5.** Quantum efficiency of Cu(In,Ga)Se<sub>2</sub> thin film solar cells prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>)=0-0.13.



**Figure 6.** Dependence of band gap and open circuit voltage on In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>) mole ratio.

graded band gap structure [13]. Therefore, it is presumed that the value of 1.3 eV demonstrates the bottom of the double graded band gap structure. This result is suggestive that Cu(In,Ga)Se<sub>2</sub> thin film solar cells with a high Ga/(In+Ga) mole ratio have a deep valley structure. The deep valley prevents the carrier collection and causes the deterioration of solar cell performance. The similar tendency for Cu(In,Ga)Se<sub>2</sub> thin film solar cells with Ga/(In+Ga) mole ratio of 0.3 fabricated on Mo coated Ti foils by three stage process has been reported [14]. On the other hand, Cu(In,Ga)Se<sub>2</sub> thin film prepared at In<sub>2</sub>S<sub>3</sub>/(In<sub>2</sub>Se<sub>3</sub>+Ga<sub>2</sub>Se<sub>3</sub>)=0.04 demonstrated a band gap of 1.4 eV, which was suitable for a better matching solar spectrum. Thus the cell performance was improved. Therefore, In<sub>2</sub>S<sub>3</sub> slightly supplying is one of the promising methods to improve the performance of Cu(In,Ga)Se<sub>2</sub> thin film solar cells.

#### 4. Conclusion

For photovoltaic device applications, Cu(In,Ga)Se<sub>2</sub> thin films were prepared by sequential evaporation process. The effect of In<sub>2</sub>S<sub>3</sub> supplying in the third step was examined. XRD study showed that Cu(In,Ga)Se<sub>2</sub> thin films had a chalcopyrite structure. EPMA analysis demonstrated that Cu(In,Ga)Se<sub>2</sub> thin films have Ga/(In+Ga) mole ratio of 0.855-0.747 and S/(Se+S) mole ratio of 0-0.04. From SEM micrograph, Cu(In,Ga)Se<sub>2</sub> thin films were formed with small grains. From the quantum efficiency analysis, Cu(In,Ga)Se<sub>2</sub> thin film solar cells with a high Ga/(In+Ga) mole ratio prepared by sequential evaporation process had a deep valley structure, which was the most remarkable point in this study. This result indicates that it is expected to obtain the improvement in Cu(In,Ga)Se<sub>2</sub> thin film solar cells with a high Ga/(In+Ga) mole ratio by controlling an adequate double graded band gap structure. The performance of Cu(In,Ga)Se<sub>2</sub> thin film solar cell was improved by using slightly In<sub>2</sub>S<sub>3</sub> compound in the third step.

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