ABB

MD simulations of the p53 oncoprotein structure: the effect of the Arg273→His mutation on the DNA binding domain

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ABSTRACT

A comparative molecular dynamics (MD) simulation study was performed on the p53 oncoprotein to investigate the effect of the Arg273His (R273H) mutation on the p53 \rightarrow DNA Binding Domain (DBD). The two p53 dimer structures of the wild-type and mutant Arg273His (R273H) were simulated with the same thermodynamic and environmental parameters. The obtained results demonstrate that the induced Arg-273His mutation has a considerable effect on the p53→DNA close contact interaction and changes the picture of hydrogen formation. The Arg273His mutation, in some cases, destroys the existing native hydrogen bond, but, in other cases, forms a strong $p53 \rightarrow$ DNA hydrogen bond, which is not proper for the native protein. The MD simulation results illustrate some molecular mechanism of the conformational changes of the Arg273His key amino acid residue in the p53 \rightarrow DNA binding domain, which might be important for the understanding of the physiological functioning of the p53 protein and the origin of cancer.

Keywords: Molecular Dynamics Simulations; p53 Oncoprotein; Effect of the R273H Mutation; DNA Binding Domain

1. INTRODUCTION

The p53 tumor suppressor protein is involved in preventing cancer and plays a central role in conserving genomic stability by preventing a genome mutation [1-5]. p53 (the 53 kilodalton (kDa) protein) is activated either to induce a cell cycle arrest allowing the repair and survival of the cell, or apoptosis to discard the damaged cell. Single amino acid substitutions (mutations) in the p53 structure deactivate the p53 protein, which results in cancer [1-8]. Usually, most of the mutations (95% of all known tumor mutations) occur in the DNA-binding domain (DBD) of the p53 protein. Thus, an oncogenic form of p53 is predominantly a full-length p53 protein with a single amino acid substitution in the DBD. Most of these mutations destroy the ability of the protein to bind to its target DNA sequences, and thus prevent the transcriptional activation of these genes. Tumors with inactive p53 mutants are aggressive and often resistant to ionizing radiation and chemotherapy.

Some reported observations of the effect of the G245S, R248Q, R249S, and R273H mutations suggest that they disturb essentially the stability of the p53 core domain. A detailed analysis of p53 mutations shows that the vast majority of the mutations in p53 cluster in the conserved regions of the DNA-binding core domain (residues 96 -292). Around 20% of all mutations are concentrated in five "hotspot" codons in the core domain: 175, 245, 248, 249, and 273. In [4], examining the Arg273His (R273H) mutation and p53-DNA interactions, it was found that at least three R273H monomers are needed to disable the p53 tetramer. In [6], it is shown that mutations in p53 occur at a rate of approximately 70% in hormone-refractory prostate cancer and that R273H p53 mutation (p53R273H) facilitates androgen-independent growth in castrated nude mice, etc.

In studying the p53 protein structural behavior and its mutation transition, molecular dynamics (MD) simulation is a very efficient technique. Nevertheless, few published papers addressed MD simulation of the p53 protein [9-11]. The two major computational difficulties that block the efficient use of MD for the p53 protein could be outlined (as detailed in [9]) as follows: 1) the large size of p53; 2) the problem of accurately modeling the zinc-binding interface in p53c. The p53 protein functions as a tetramer in a cell, but the p53c monomer is already 50 Å in diameter. Simulation of the p53 tetramer



structure in explicit water is extremely difficult because a large number of water molecules are needed to solvate this protein. The MD simulations performed in [9] were limited to the consideration of the p53c monomer only. It was found that the monomer p53c model is stable in vitro and that such model is still suitable for the protein stability analysis, although the monomer p53c-DNA binding is weaker.

In the present study, we focus on the p53 dimer structure (the A and B chains) to investigate the p53-DNA binding phenomena. The p53 dimer structure is a more adequate model than a monomer one. In the dimer representation, the p53 protein will symmetrically surround the relevant DNA molecule from two sides, which allows the formation of a binding interface in a native manner [12-15]. Next, we perform a comparative MD analysis of the p53 dimer—DNA interaction between the wild-type and mutant Arg273His (R273H) versions of the p53 protein.

2. MATERIALS AND METHODS

In **Figure 1**, the side and top views of the p53 mouse protein are presented (PDB entry file: 3EXJ). For the p53 dimer structure, two chains (A and B) which surround DNA symmetrically are shown along with two catalytic centers [Zn(CYS)3(HIS)1] (the zinc-binding interfaces are displayed in the right pictures; zinc is shown as a gray sphere). On two p53-relevant structures described above (the mouse p53 protein: the wild-type and mutant R273H version (PDB entry file: "3EXJ")), we have performed several model calculations (periodic PME-NPT; periodic PME-NVT; non-periodic cutoff; and nonperiodic no-cutoff).

In the p53 protein structure, two chains (A and B)

symmetrically surround (yellow arrows) the related DNA sequence located in the central DNA binding domain (DBD). In the p53-DBD interaction, three arginines (R248, R273, and R280), one serine (S241), and one alanine (A276) are responsible for DNA binding. In **Figure 1**, a positional snapshot of the p53 chain A {Arg248, Arg273, Arg280, Ser241, Ala276} with respect to DNA is separately displayed.

Considering the above noted difficulty of zinc binding in p53c, experiments point to the importance of zinc coordination for achieving the correct folding and correct binding of p53 to a specific DNA in intact cells. In our simulation, however, both non-bonded and bonded approaches are used to describe the zinc-p53c binding interface. It should be stressed that it is quite challenging to maintain interface stability at room temperature during long-time simulations [9]. We have followed these criteria and have not fixed zinc locally in its binding interface, even though it is suitable to simulate proteins in which zinc is required for the catalytic function [9]. In this study, we focus on the peculiarities of the global structural changes of the p53 protein dimer. We have performed a comparative analysis of the wild-type and mutant versions between the mouse (mouW, mouM) p53 proteins. For the mouse p53 structures, we have examined the effect of the Arg273His (R273H) mutation on the p53-DNA binding domain.

All simulations were performed with the AMBER (versions 7 to 11) MD software package for studying biomolecules [16,17]. In our periodic MD simulations, the electrostatic interactions were treated with the Particle Mesh Ewald (PME) algorithm [16-18]. Non-periodic MD involved both cutoff and no-cutoff simulations. To perform no-cutoff calculations, the AMBER



Figure 1. The side and top views of the p53 mouse protein (PDB entry file: 3EXJ). For the p53 protein structure, two DNA chains (A and B) are shown along with two catalytic centers $[Zn(CYS)_3(HIS)_1]$ of the zinc-binding interface (zinc is shown as gray spheres).

versions were used that had been adapted to the MDGRAPE-2 and 3 hardware. For MDGRAPE-2 and 3, all the particle interactions are calculated [19]. The Cornell *et al.* all-atom force field was used in the MD simulations [20,21]. A system was solvated with TIP3P molecules [22] generated in a rectangular and spherical (non-periodic) water baths. The temperature was kept constant by using the Berendsen algorithm [23]. Only bond lengths involving hydrogen atoms were constrained using the SHAKE method [24]. The result of simulations and images of the simulated proteins were analyzed using the RasMol [25], MOLMOL [26], and Visual Molecular Dynamics (VMD) [27] software. For some details of our MD simulations, see [28-30]).

MD Protocol

Below follows an example of the MD protocol of periodic PME-NPT simulations.

Potential function:

ntf = 1, ntb = 2, igb = 0, nsnb = 1ipol = 0, gbsa = 0, iesp = 0dielc = 1.0, cut = 10.0, intdiel = 1.0scnb = 2.0, scee = 1.2Frozen or restrained atoms: ibelly = 0, ntr = 0Molecular dynamics: nstlim = 10000, nscm = 2, nrespa = 1t = 0.0, dt = 0.001, vlimit = 20.0 Berendsen temperature regulation: temp0 = 300.0, tempi = 0.0, tautp = 1.0**Pressure regulation**: ntp = 1pres0 = 1.0, comp = 44.6, taup = 1.0SHAKE: ntc = 2, jfastw = 0tol = 0.00001**Ewald parameters**: verbose = 0, ew_type = 0, nbflag = 1, use_pme = 1 vdwmeth = 1, eedmeth = 1, netfrc = 1 Box X = 100.202, Box Y = 68.001, Box Z = 57.148 Alpha = 90.0, Beta = 90.0, Gamma = 90.0 NFFT1 = 100, NFFT2 = 72, NFFT3 = 60 Cutoff = 10.0, Tol = 0.100E-04Ewald Coefficient = 0.27511 Interpolation order = 4

3. RESULTS AND DISCUSSION

To compare the conformational changes between the relaxed and original 3D states in the native and mutant structures in **Figures 2-5**, the side and top views of the p53 mouse protein are shown (PDB entry file: 3EXJ).

We have performed a comparative analysis of the wildtype and mutant versions between the mouse (mouW, mouM) p53 proteins. For the mouse p53 structures, we have examined the effect of the Arg-273His (R273H) mutation on the p53-DNA binding domain.

Arginine (Arg) is an alpha-amino acid with pKa = 12.48; its side chain consists of a 3-carbon aliphatic straight chain and is positively charged in neutral, acidic, and even most basic environments. Because of conjugation between the double bond and the nitrogen lone pairs, the positive charge is delocalized, enabling the formation of multiple H-bonds. Histidine (His) is an aromatic amino acid with pKa = 6.5; its side chain consists of a positively charged imidazole ring which is aromatic at all pH values. This means that at physiologically relevant pH values, relatively small shifts in pH will change its average charge. Below a pH of 6, the imidazole ring is mostly protonated [15]. The differences in the chemical structure and properties between arginine and histidine during nanosecond dynamical changes could influence the final (relaxed) states of all amino acids. In Figures 2 and 4, the position of Arg273His \rightarrow DNA for the p53 mouse protein (chain A) are shown at final state t = 3 ns.

The statistics in **Figure 3** (mouW) and **Figure 5** (mouM) present the MD calculation results on the positional changes of five amino acid residues related to the direct $p53 \rightarrow DNA$ contact:

(mouW)

---top(a) Res{Arg273} \rightarrow DNA{P(DG395)}

---bottom(b) Res{Arg273} \rightarrow DNA{P(DT394)} (mouM)

---top(a) Res{His273} \rightarrow DNA{P(DG395)}

---bottom(b) Res{His273} \rightarrow DNA{P(DT394)}

While constructing the $p53 \rightarrow DNA$ distance diagrams in **Figure 3** (mouW) and **5** (mouM), we estimated the distance distributions between the corresponding residue heavy atoms (from the N- to CO- ends) and two DNA phosphorus atoms P(DG395) and P (DT394):

d[Res(N)-DNA(P)], d[Res(CA)-DNA(P)],

d[Res(CB)-DNA(P)], ..., d[Res(C)-DNA(P)],

d[Res(O)-DNA(P)]. (The order of the interatomic distances is $d\sim Å$).

In Figure 3(a) (the top figure), the 3-ns dynamics of the R273 \rightarrow DNA (d[R273-P(DG395)]) interaction distance is shown for the native mouse p53 protein (mouW, chain A). For the wild-type protein, the minimal Arg-273 \rightarrow DNA distance is around 3.5 Å, which is necessary for p53-DNA bonding. So, p53 forms in this region a close contact with DNA. In Figure 3(b) (the bottom figure), the 3-ns dynamics of the Arg273 \rightarrow DNA (d[R273-P(DT394)]) interaction distance is shown for the mouse p53 protein (mouW, chain A). The minimal distance d[R273-P(DT394)]) increases from 3 - 3.5 Å to 6 - 7 Å. It is clear that in contrast to Figure 3(a), the



Figure 2. Snapshots show the amino acid Arg273 \rightarrow DNA contact for the p53 mouse protein (left: chain A; right: chain B). The positions of two closest to Arg273 phosphorus atoms of DNA, P(DG395) and P(DT394), are shown according to the notation of the 3EJX entry file of PDB.



Figure 3. Distance diagrams of the 3-ns dynamics of the Arg273→DNA contact of the mouse p53 protein (chain A): (a) Arg273-P(DG395); (b) Arg273-P(DT394).



Figure 4. Snapshots show the amino acid His273 \rightarrow DNA contact for the p53 mouse protein (left: chain A; right: chain B). The positions of two closest to His273 phosphorus atoms of DNA, P(DG395) and P(DT394), are shown according to the notation of the 3EJX entry file of PDB.

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Figure 5. Distance diagrams of the 3-ns dynamics of the His273 \rightarrow DNA contact of the mouse p53 protein (chain A). (a) His273-P(DG395); (b) His273-P(DT394).

Arg273 is going apart from the P(DT394) atom of the DNA.

In Figure 5(a) (the top figure), the 3-ns dynamics of the His273 \rightarrow DNA (d[H273-P(DG395)]) interaction distance is shown for the mutant p53 protein (mouM, chain A). For the mutant p53 protein, the minimal His 273 \rightarrow DNA distance has to be around 6.0 - 6.5 Å. Thus, in comparison with Figure 3(a), the induced Arg273His mutation will destroy the p53 \rightarrow DNA close contact. In Figure 5(b) (the bottom figure), the 3-ns dynamics of the His273 \rightarrow DNA (d[H273-P(DT394)]) interaction distance is shown for the mutant p53 protein (mouM, chain A). In this case, for the mutant p53 protein, the minimal His 273 \rightarrow DNA distance has to be around 3 - 3.5 Å. A comparison between Figures 3(b) and 5(b) is straight-forward.

4. CONCLUSIONS

Our MD simulation results on the structure of the p53 oncoprotein with Arg273His (R273H) mutation suggest that the induced mutation disturbs essentially the stability of the p53 core domain. By comparing the relaxed configurations and distance distribution diagrams between the native (Figures 2 and 3) and mutant (Figures 4 and 5) p53 proteins, we clearly see that the Arg273His mutation has a considerable effect on the p53 \rightarrow DNA contact. The induced Arg273His mutation essentially changes the picture of hydrogen formation: in some cases, it destroys the existing native hydrogen bond; but, in other cases, it forms a strong $p53 \rightarrow DNA$ hydrogen bond, which is not proper for the native protein. Our MD simulation results illustrate the molecular mechanism of the conformational changes of the Arg273His and key amino acid residue in the p53 \rightarrow DNA binding domain. In conclusion, the obtained results might be important for the understanding

of the physiological functioning of the p53 protein and the origin of cancer [1-8].

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