

Preparation of Polyfunctionally Substituted Pyridine-2(1H)-Thione Derivatives as Precursors to Bicycles and Polycycles

Fathi A. Abu-Shanab^{1,2*}, Sayed A. S. Mousa¹, Sherif M. Sherif³, Mohamed I. Hassan¹

¹Department of Chemistry, Faculty of Science, Al-Azhar University, Assiut, Egypt

²Department of Chemistry, Faculty of Science, Gazan University, Gazan, KSA

³Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt

Email: *fathiabushanab@yahoo.com

Received 3 October 2014; revised 19 November 2014; accepted 6 December 2014

Copyright © 2014 by authors and Scientific Research Publishing Inc.

This work is licensed under the Creative Commons Attribution International License (CC BY).

<http://creativecommons.org/licenses/by/4.0/>



Open Access

Abstract

Reaction of acetylacetone with 1 mole of dimethylformamide dimethyl acetal (DMFDMA) affords enamine 2a which reacts with cyanothioacetamide to give pyridinethione 3a. Pyridinethione 3a reacts with methyl iodide, halogenated compounds, aromatic aldehyde and malononitrile/elemental sulfur to yield compounds 7-10 respectively. Reactions of thioether 7 in ethanolic K₂CO₃, 1 mole DMFDMA and 4-(dimethylamino)benzaldehyde give compounds 11, 13, 14 respectively. Enaminone 12 can be prepared by reaction of compound 11 with DMFDMA. We have demonstrated some reactions in order to show the potential usefulness of the prepared compounds for the preparation of new bipyridyl compounds 15, 16, 18, bicyclic compounds 17 and uncommon tricyclic compounds 20, 21, 22 and 23 respectively using DMFDMA.

Keywords

Acetyl Acetone, DMFDMA, Malononitrile Dimmer, Bipyridyl, 5-Acetylpyridinethione

1. Introduction

Formamide acetals are useful reagents in organic synthesis; [1] [2] their main application has been used for functional group transformations [3], but they may also be regarded as one-carbon synthons in the construction of carbon skeletons. One type of reaction, which is potentially valuable for the future purpose, is the reaction of *N,N'*-dimethylformamide dimethyl acetal (DMFDMA) with 1,3-dicarbonyl compounds **1** to give enamines **2** [4] (Figure 1).

*Corresponding author.

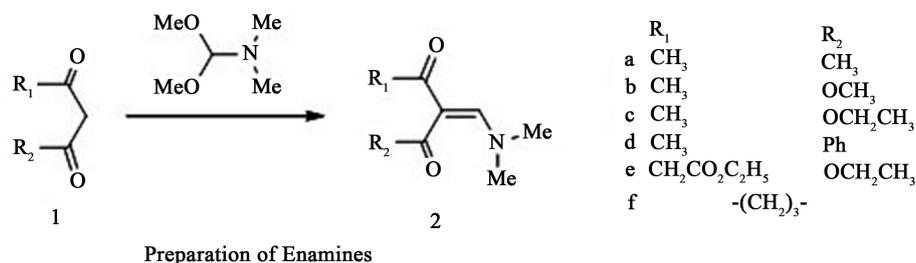


Figure 1. Preparation of Enamines.

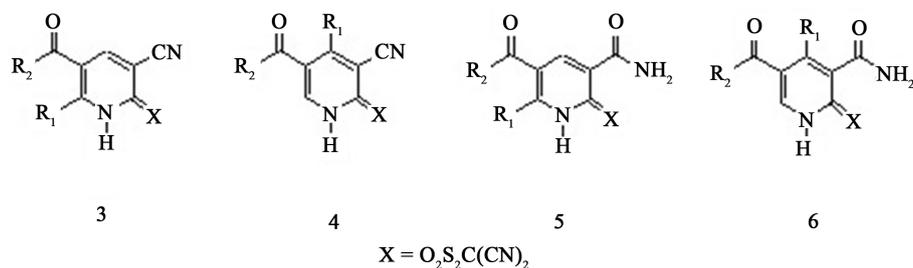
We have reported that enamines **2** were used as precursors in the synthesis of pentasubstituted pyridines **3-6** [5]-[8] (**Figure 2**).

The treatment of acetylacetone (**1a**) with dimethyl formamide dimethylacetal (DMFDMA) in dry DMF under nitrogen and stirring over night afforded the corresponding enamine **2a** which on treatment with cyanothioacetamide and sodium hydride in dry DMF (*in situ*) afforded pyridine-2(1*H*)-thione (**3a**) [6], when the emamine **2a** was treated with cyanothioacetamide in ethanol and piperidine as a base afforded the pyridine-2(1*H*)-thione (**5a**) [7] [12] (**Figure 3**).

2. Results and Discussion

In conjunction of this work, we report here the reaction of acetylacetone **1a** with one mole of *N,N*-dimethylformamide dimethyl acetal (DMFDMA) in dry dioxane gave the corresponding enamine **2a**. The treatment of this enamine (*in situ*) with cyanothioacetamide in ethanol in the presence of sodium ethoxide under reflux gave 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** with a very good yield [7], **Scheme 1**.

We have found that the prepared compound **3a** included three functional groups which are thioamido group, cyano group and acetyl group. These functional groups can be used for the preparation of bicyclic or polycyclic compounds of biological interest. Thus, some illustrative reactions designed to demonstrate the potential usefulness of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** for further heterocyclic synthesis. Therefore, the reaction of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** with methyl iodide in alcoholic sodium hydroxide afforded the corresponding thioether derivative **7**, which in turn is a good intermediate for the preparation of further heterocyclic compounds of biological interest. The structure of the isolated compound **7** is confirmed by spectral analysis. The IR spectrum shows the disappearance of (NH) group. Also, the ¹H NMR spectrum shows the disappearance of the thioamide proton and the appearance of a singlet signal corresponding to (SCH₃) at δ_H = 2.63 ppm. Also, the mass spectrum shows the molecular ion peak at m/e 206 which corresponding to the molecular formula (C₁₀H₁₀N₂OS). The reaction of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** with ethyl chloroacetate or chloroacetamides in ethanolic sodium ethoxide afforded the corresponding 5-acetyl-3-amino-6-methylthieno[2,3-*b*]pyridine derivatives **8a-c** in a good yield. The structure of the isolated compounds is confirmed by elemental and spectral analysis. The IR spectrum shows the disappearance of cyano group and appearance of amino group at ν_{max} = 3427, 3328 cm⁻¹ in compound **8a** as example beside the other functional groups. Also, the mass spectra show the molecular ion peaks fit to all compounds **8a-c**. Also, the ¹H NMR spectra show signals fit to the structure of all compounds **8a-c**. The presence of acetyl group in 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** is useful for the preparation of fused heterocyclic compounds. So that the reaction of compound **3a** with aldehydes like 4-(dimethylamino) benzaldehyde and 4-methylbenzaldehyde in ethanolic sodium hydroxide afforded the corresponding chalcones **9a,b**. The structure of the isolated chalcones is confirmed by elemental analysis as well as spectral analysis. The mass spectra show the molecular ion peak fit to all compounds **9a,b**. As an example compound **9a** shows the molecular ion peak at m/e 323 which corresponding to the molecular formula (C₁₈H₁₇N₃OS). Also, the ¹H NMR spectra of these compounds **9a,b** show the disappearance of the signal corresponding to the methyl of acetyl group and the appearance of two doublets signals corresponding to the two protons of double bond of chalcone. Finally, 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** was treated with malononitrile and sulfur element (Gewald's reaction) in ethanol in the presence of triethylamine as a base to afford 5-(5-amino-4-cyanothiophen-3-yl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **10** in a good yield, **Scheme 1**. The IR spectrum of compound **10** shows the appearance of amino group at ν_{max} = 3435, 3350 cm⁻¹ beside the other

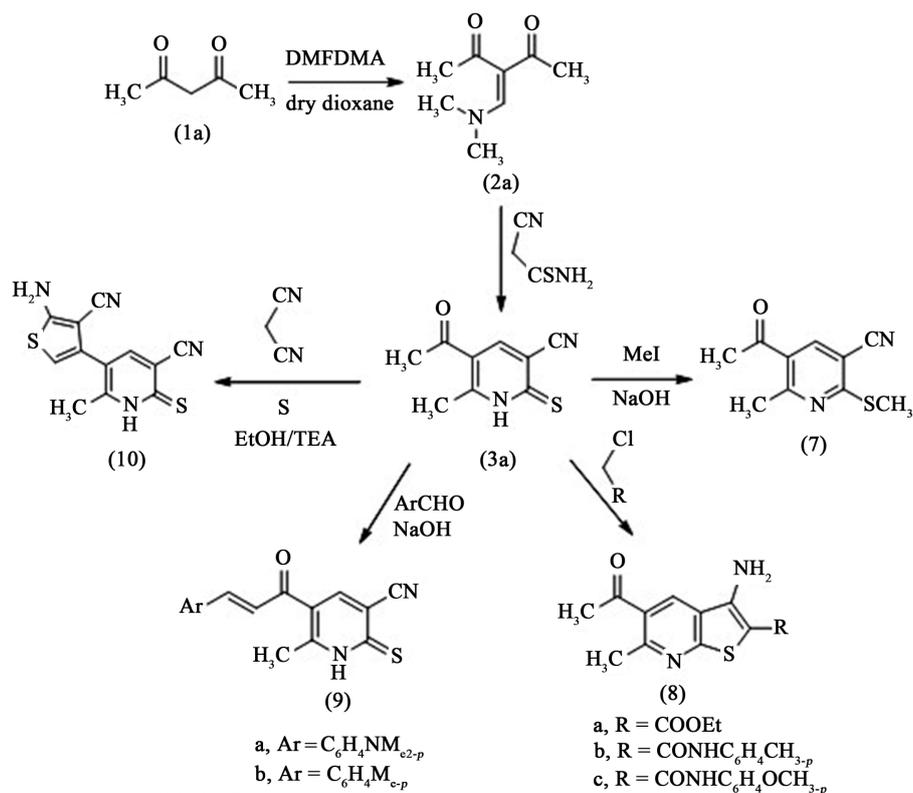


Defferent polysubstituted pyridines have been prepared

Figure 2. Different polysubstituted pridines have been prepared.



Figure 3. Tetrasubstituted pyridinethione have been prepared.



Scheme 1. Synthesis of pyridine (1*H*)-thione derivative 3a and its reactions with MeI, α -chloroketones, aldehydes and malononitrile.

functional groups. Also, 1H NMR spectrum of compound **10** shows singlet signal at $\delta_H = 2.45$ ppm corresponding to methyl group and singlet signal at $\delta_H = 6.95$ ppm corresponding to amino group and singlet signal at $\delta_H = 7.07$ ppm corresponding to CH thiophene ring and singlet signal at $\delta_H = 7.2$ ppm corresponding to CH pyridine ring.

5-Acetyl-6-methyl-2-(methylthio)nicotinonitrile **7** can be used as intermediate for further preparation of hete-

rocylic compounds. So that compound **7** was treated with potassium carbonate in ethanol to afford 5-acetyl-2-ethoxy-6-methylnicotinonitrile **11**. This compound was formed by nucleophilic substitution of SMe by OEt group. The structure of the isolated compound is confirmed by elemental and spectral analyses. The mass spectrum shows the molecular ion peak at m/e 204 corresponding to the molecular formula ($C_{11}H_{12}N_2O_2$). Also, the 1H NMR spectrum shows the disappearance of SMe signal and appearance of two signals; a triplet at $\delta_H = 1.43$ ppm and a quartet at $\delta_H = 4.54$ ppm corresponding to the OEt moiety, in addition to the rest of signals corresponding to the other protons in the molecule. Compound **11** was reacted with *N,N'*-dimethylformamide dimethyl acetal (DMFDMA) in dry xylene to give the corresponding enamine **12** in a good yield. The mass spectrum of compound **12** shows the molecular ion peak at m/e 259 which corresponding to the molecular formula ($C_{14}H_{17}N_3O_2$). Also, the 1H NMR spectrum of compound **12** shows the disappearance of the singlet signal which is related to the methyl of acetyl group and the appearance of two singlet signals at $\delta_H = 2.68$ and 3.04 ppm corresponding to the two methyl groups of NMe_2 moiety. Consequently the 1H NMR spectrum shows the appearance of two doublets at $\delta_H = 6.25$ ppm and 7.87 ppm corresponding to the two protons of the enamine double bond.

Enamine **13** can be prepared in a good yield by reaction of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** with two moles of *N,N'*-dimethylformamide dimethylacetal (DMFDMA) in dry xylene or by the reaction of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile **7** with one mole of *N,N'*-dimethylformamide dimethylacetal (DMFDMA) in dry xylene. The structure of the isolated compound is confirmed by elemental and spectral analysis. Whereas the mass spectrum shows the molecular ion peak at m/e 261 which corresponding to the molecular formula ($C_{13}H_{15}N_3OS$). Also, the 1H NMR spectrum of it shows the disappearance of the singlet signal which is related to the methyl of acetyl group and appearance of two singlet signals at $\delta_H = 2.62$ and 2.64 ppm corresponding to the two methyl groups of NMe_2 moiety. Consequently, the 1H NMR spectrum shows the appearance of two doublets at $\delta_H = 5.28$ ppm and 7.75 ppm corresponding to the two protons of double bond of enamine.

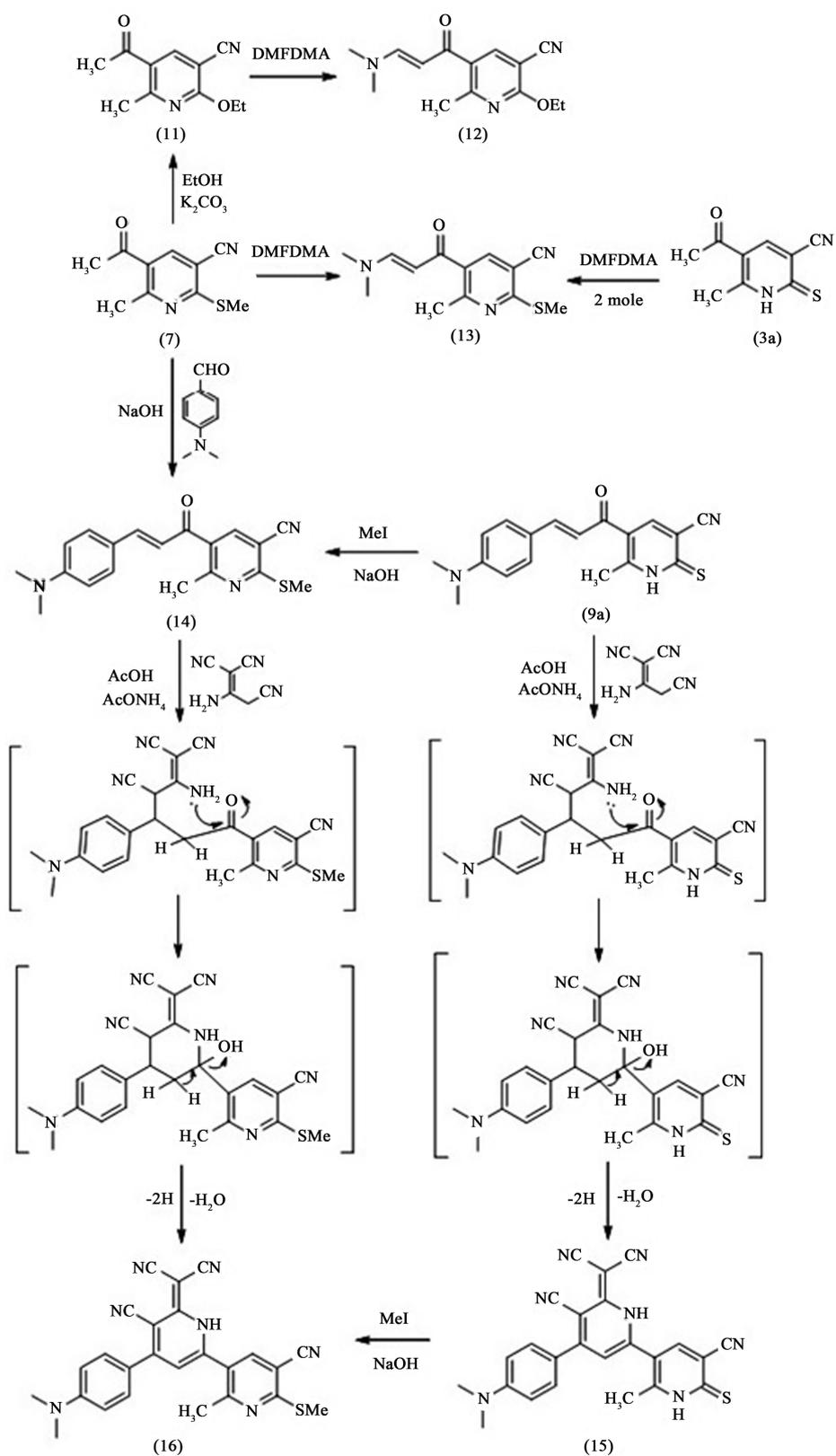
Chalcone **14** can either be prepared by the reaction of compound **7** with (4-(dimethylamino)benzaldehyde) in ethanolic sodium hydroxide or by treatment of compound **9a** with methyl iodide in ethanolic sodium hydroxide. The mass spectrum of compound **14** shows the molecular ion peak at m/e 337 corresponding to the molecular formula ($C_{19}H_{19}N_3OS$). Also, the 1H NMR spectrum of compound **14** shows singlet signal at $\delta_H = 2.62$ ppm corresponding to methyl group and singlet signal at $\delta_H = 2.66$ ppm corresponding to SCH_3 and two singlet signal at $\delta_H = 2.9$, 3.04 ppm corresponding to NMe_2 moiety and appearance of some signals of other protons in molecule.

For preparation of bipyridyl derivatives, we have carried out the reaction of chalcones 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **9a** and 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile **14** with malononitrile dimmer [9] in acetic acid and ammonium acetate afforded the corresponding bipyridyl derivatives 6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1,1',6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile **15** and 6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-(methylthio)-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile **16** respectively. The reaction proceeds by Michael addition followed by cyclization through condensation as shown in [Scheme 2](#).

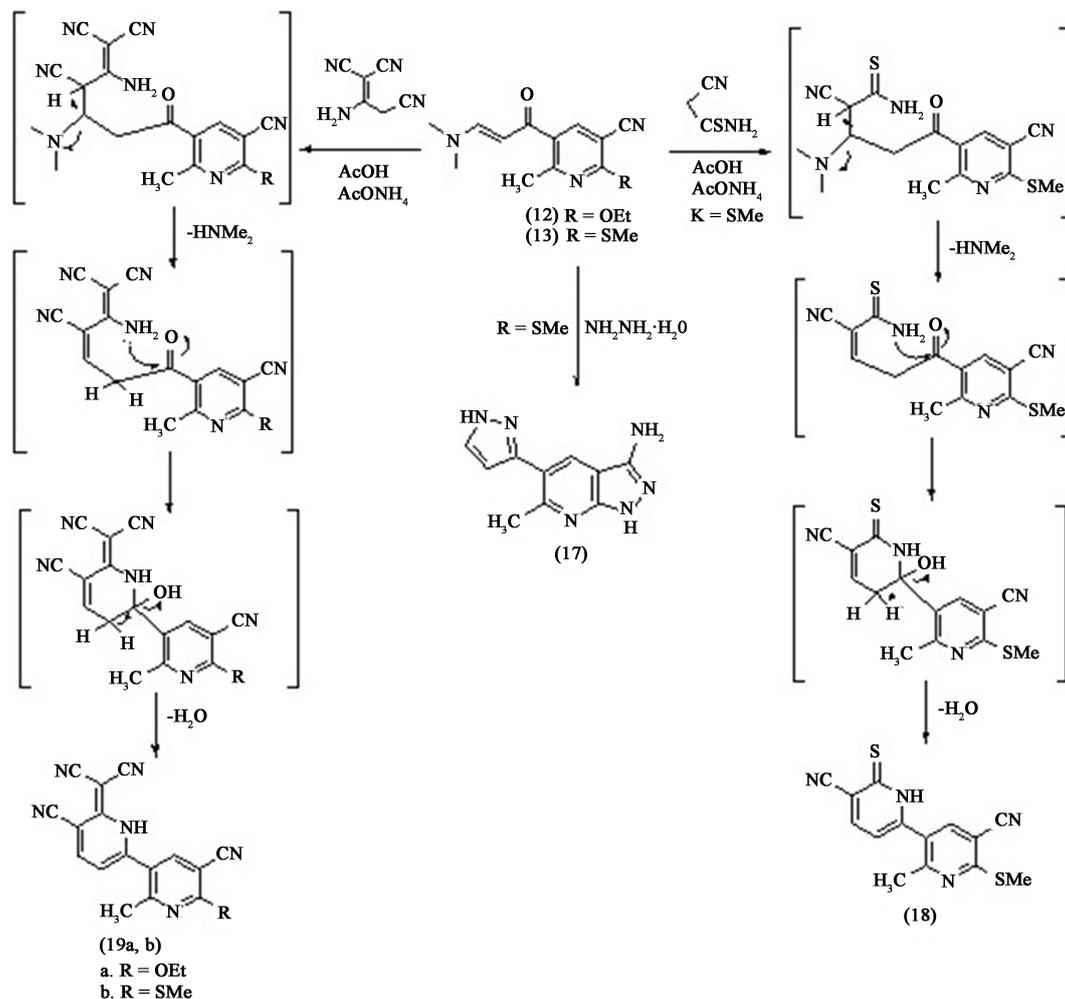
The compound **16** can also be obtained by the reaction of 6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1,1',6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile **15** with methyl iodide in alcoholic sodium hydroxide [Scheme 2](#).

The structure of the isolated compounds **15** and **16** is established by elemental and spectral analysis. Whereas the mass spectra of these compounds show the molecular ion peaks at m/e 435 corresponding to the molecular formula ($C_{24}H_{17}N_7S$), and at m/e 449 corresponding to the molecular formula ($C_{25}H_{19}N_7S$) for **15** and **16** respectively. The IR spectra of both compounds **15** and **16** show the disappearance of the carbonyl group and the appearance of NH group. Also, the 1H NMR spectra of these compounds show signals fit to structures **15** and **16**.

For further preparation of heterocyclic compounds [10], we carried out the following reactions. The reaction of enamine **13** with excess hydrazine hydrate in ethanol afforded 6-methyl-5-(1H-pyrazol-3-yl)-1H-pyrazolo[3,4-b]pyridin-3-amine **17** in a good yield as shown in [Scheme 3](#). The IR spectrum of compound **17** shows the disappearance of the cyano group and the appearance of NH_2 and NH groups at ν_{max} at 3405 cm^{-1} , 3329 cm^{-1} and 3136 cm^{-1} respectively. Also, the mass spectrum of compound **17** shows the molecular ion peak at m/e 214 corresponding to the molecular formula ($C_{10}H_{10}N_6$). Also, the 1H NMR spectrum of compound **17** shows signals fit to the structure.



Scheme 2. Reactions of tetrasubstituted pyridine **7** with DMFDMA, alcoholic K_2CO_3 and p-N,N-dimethylaminobenzaldehyde.



Scheme 3. Reactions of tetrasubstituted pyridine (12,13) with cyanothioacetamide, hydrazine hydrate and malononitrile dimer.

Also, the enamine **13** is treated with cyanothioacetamide in acetic acid and ammonium acetate afforded 2'-methyl-6'-(methylthio)-6-thioxo-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile **18**. The reaction is started by Michael addition of cyanothioacetamide on the double bond followed by elimination of dimethylamine (HNMe_2) and cyclization with the carbonyl group. The structure of the isolated compound **18** is confirmed by elemental and spectral analysis. The IR spectrum of compound **18** shows the disappearance of carbonyl group and appearance of NH group at ν_{max} at 3428 cm^{-1} . The mass spectrum of compound **18** shows the molecular ion peak at m/e 298 corresponding to the molecular formula ($\text{C}_{14}\text{H}_{10}\text{N}_4\text{S}_2$). Also, the ^1H NMR spectrum of compound **18** shows the disappearance of protons of NMe_2 moiety and appearance of NH proton beside the other protons.

Another type of bipyridyl derivatives **19a,b** can be prepared by the reaction of the enamines **12** and **13** with malononitrile dimer in acetic acid and ammonium acetate. This reaction proceeds by Michael addition of malononitrile dimer, followed by elimination of dimethylamine (HNMe_2) and cyclization through condensation of amino group with carbonyl group as shown in **Scheme 3**. The mass spectrum of compound **19a** shows the molecular ion peak at m/e 328 corresponding to the molecular formula ($\text{C}_{18}\text{H}_{12}\text{N}_6\text{O}$), and compound **19b** shows the molecular ion peak at m/e 330 corresponding to the molecular formula ($\text{C}_{17}\text{H}_{10}\text{N}_6\text{S}$). The IR spectra of the compounds **19a,b** show the disappearance of the carbonyl group and the appearance of NH group beside the other groups. Also, the ^1H NMR spectra of compounds **19a,b** show the disappearance of protons of NMe_2 moiety and appearance of NH proton beside the other protons.

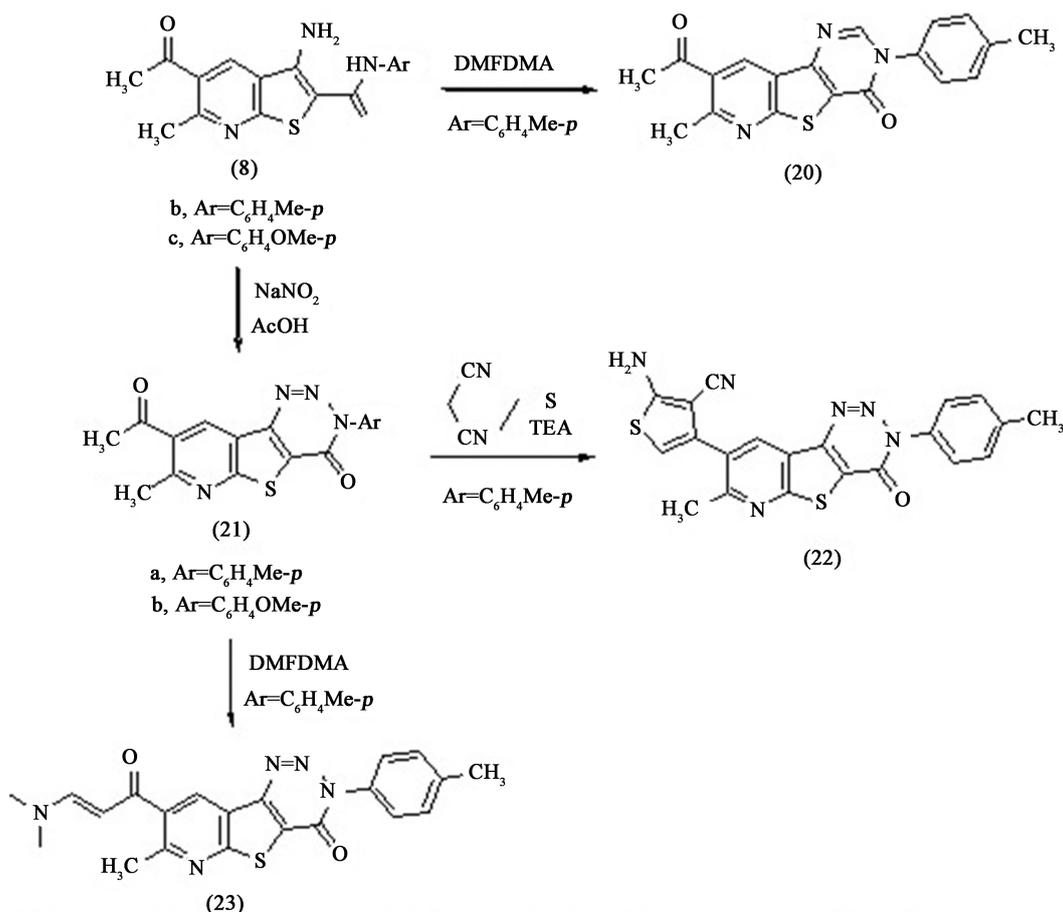
The tricyclic heterocyclic compounds are biologically interest compounds. They are examples of uncommon ring system [11] [12]. Therefore we are interested for the preparation of this type of heterocyclic compound.

Thus 5-acetyl-3-amino-6-methyl-N-(*p*-tolyl)benzo[*b*]thiophene-2-carboxamide **8b** is reacted with *N,N*-dimethylformamide dimethyl acetal (DMFDMA) in dry dioxane afforded 8-acetyl-7-methyl-3-(*p*-tolyl)pyrido [3',2':4,5]thieno[3,2-*d*]pyrimidin-4(3*H*)-one **20**. The IR spectrum of compound **20** shows the disappearance of (NH₂) and (NH) groups. The mass spectrum of compound **20** shows the molecular ion peak at *m/e* 349 which corresponding to the molecular formula (C₁₉H₁₅N₃O₂S). Also, the ¹H NMR spectrum of compound **20** shows the appearance of two singlet signals at δ_H = 8.43 ppm, and 8.52 ppm corresponding to two protons of pyrimidinone and pyridine rings respectively beside other signals for other protons.

For further reaction of 5-acetyl-3-amino-6-methyl-N-substituted[*b*]thiophene-2-carboxamide **8b,c** it reacted with nitrous acid in acetic acid afforded the tricyclic compounds **21a,b** in a good yield as shown in **Scheme 4**. The structures of the compounds **21a,b** are established by elemental and spectral analysis. Whereas the IR spectra of both compounds **21a,b** show the disappearance of the bands corresponding to (NH₂) and (NH) groups. The mass spectrum of the compound **21a** as an example shows the molecular ion peak at *m/e* 350 corresponding to molecular formula (C₁₈H₁₄N₄O₂S).

Also, the ¹H NMR spectra of compounds **21a,b** show the disappearance of the signals which corresponding to (NH₂) and (NH) groups beside the appearance the other signals for other groups.

We have found that the prepared tricyclic compounds **20** and **21a,b** contain acetyl group which is very important for the preparation of new heterocyclic compounds. So that the reaction of **21a** with malononitrile and sulphur element in ethanol and triethylamine (Geweld reaction) afforded 2-amino-4-(7-methyl-4-oxo-3-(*p*-tolyl)-3,4-dihydropyrido[3',2':4,5]thieno[3,2-*d*][1,2,3]triazin-8-yl)thiophene-3-carbonitrile **22**. The IR spectrum of compound **22** shows the disappearance of the carbonyl group of acetyl moiety and the appearance of amino and cyano groups at *v*_{max} at 3427 cm⁻¹ and 2208 cm⁻¹ respectively. Also, the mass spectrum of this compound **22** shows the molecular ion peak at *m/e* 430 which corresponding to the molecular formula (C₂₁H₁₄N₆OS₂).



Scheme 4. Reactions of thienopyridine derivatives (**8b,c**) with DMFDMA and sodium nitrile in acetic.

Also, the compound **21a** is treated with *N,N*-dimethylformamide dimethyl acetal (DMFDMA) in dry xylene afforded the corresponding enamine 8-(3-(dimethylamino)acryloyl)-7-methyl-3-(*p*-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)-one **23** in a good yield, **Scheme 4**. The mass spectrum of compound **23** shows the molecular ion peak at *m/e* 405 corresponding to molecular formula (C₂₁H₁₉N₅O₂S). Also, the ¹H NMR spectrum of compound **23** shows the disappearance of the methyl of acetyl moiety and appearance instead of it two singlet signals at $\delta_{\text{H}} = 3.63$ ppm and 3.67 ppm corresponding to (NMe₂) moiety. Also, it shows the appearance of two doublet signals at $\delta_{\text{H}} = 5.42$ ppm and 7.82 ppm respectively corresponding to the double bond protons of enaminone moiety beside signals for other protons.

3. Experimental

All melting points are uncorrected. IR spectra were recorded on a Perkin-Elmer 17,100 FTIR spectrometer as KBr disks. NMR spectra were recorded on a Varian Gemini (400 MHz) spectrometer for solutions in CDCl₃ or DMSO-*d*₆ with tetramethylsilane (TMS) as an internal standard unless otherwise. Mass spectra were obtained on Finnigan 4500 (low resolution) spectrometers using electron impact (EI) at Micro-analytical Center Cairo University Giza Egypt.

Preparation of 5-acetyl-3-cyano-6-methylpyridine-2(1H)-thione (3a):

A mixture of acetylacetone (**1a**) (1.0 g, 10 mmol), dry dioxane (10 mL) and *N,N*-dimethylformamide dimethyl acetal (1.19 g, 10 mmol) was stirred under dry condition at room for 24 h. In a second flask, a mixture of dry ethanol (15 mL), and sodium metal (0.46 g, 20 mmol) was left under stirring for 10 min. Then cyanothioacetamide (1.0 g, 10 mmol) was added to the mixture. The mixture was left for further 10 min. The contents of the second flask were transferred into the first flask, and the resulting mixture was stirred for further 1 h, followed by converting stirring into reflux for 4 h. After cooling, the mixture was poured onto acidified ice/cold water. The product was recovered by filtration and recrystallised from ethanol as orange crystals (76%), Mp. 232° - 233°, similar to be published before [7] and mixed Mp.

Preparation of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile (7)

Mixture of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** (1.92 g, 10 mmol) in ethanol as solvent and sodium hydroxide (0.4 g, 10 mmol) with stirring for 1 hr., and add methyl iodide (0.62 ml, 10 mmol) with stirring until precipitate formed. The product was recovered by filtration and recrystallised from ethanol as white crystals (1.52 g, 74%), Mp. 140°C - 142°C; ¹H-NMR (CDCl₃): $\delta = 2.54$ (3H, s, CH₃ py.), 2.63 (3H, s, SCH₃), 2.77 (3H, s, CH₃CO), 8.07 (1H, s, CH py.); IR (KBr) ν 2227 (CN), 1685 cm⁻¹ (C=O); MS (EI)⁺: *m/z* 206 M⁺; Anal. Calcd for C₁₀H₁₀N₂OS (206.27): C, 58.23; H, 4.89; N, 13.58. Found: C, 58.03; H, 4.73; N, 13.41.

General procedure for the preparation of compounds 8a-c

In dry flask, a mixture 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** (1.92 g, 10 mmol) and α -chloro compounds (10 mmol) in ethanol and sodium ethoxide (20 mmol) was left under reflux for two hours. The mixture was left for cooling and poured onto ice cold water. The solid product was recovered by filtration and recrystallised from the proper solvent.

Ethyl 5-acetyl-3-amino-6-methylthieno[2,3-*b*]pyridine-2-carboxylate (8a): Obtained using ethyl 2-chloroacetate (1.06ml, 10 mmol). The product was recrystallised from acetic acid as yellow crystals (2.16 g, 77.7%), Mp. 220°C - 222°C; ¹H-NMR (DMSO): $\delta = 1.25$ (3H, t, CH₃ ethyl), 2.6 (3H, s, CH₃ py.), 2.66 (3H, s, CH₃CO), 4.25 (2H, q, CH₂ ethyl), 7.29 (2H, s, NH₂), 8.95 (1H, s, CH py.); IR (KBr) ν 3427, 3328 (NH₂), 1679 cm⁻¹ (C=O); MS (EI)⁺: *m/z* 278 M⁺; Anal. Calcd for C₁₃H₁₄N₂O₃S (278.33): C, 56.10; H, 5.07; N, 10.06, Found: C, 55.96; H, 4.94; N, 9.97.

5-Acetyl-3-amino-6-methyl-N-(*p*-tolyl)thieno[2,3-*b*]pyridine-2-carboxamide (8b): Obtained using 2-chloro-N-(*p*-tolyl)acetamide (1.83 g, 10 mmol). The product was recrystallised from ethanol as yellow crystals (2.7 g, 79%), Mp. 218°C - 220°C; ¹H-NMR (DMSO) $\delta = 2.26$ (3H, s, CH₃ Ar), 2.64 (3H, s, CH₃ py.), 2.73 (3H, s, CH₃CO), 7.12 (2H, d, Ar), 7.47 (2H, s, NH₂), 7.55 (2H, d, Ar), 9.04 (1H, s, CH py.), 9.4 (1H, s, NH); IR (KBr) ν 3428, 3312 (NH₂, NH), 1685 cm⁻¹ (C=O); MS (EI)⁺: *m/z* 339 M⁺; Anal. Calcd for C₁₈H₁₇N₃O₂S (339.42): C, 63.70; H, 5.05; N, 12.38, Found: C, 63.56; H, 4.93; N, 12.15.

5-Acetyl-3-amino-N-(4-methoxyphenyl)-6-methylthieno[2,3-*b*]pyridine-2-carboxamide (8c):

Obtained using 2-chloro-N-(4-methoxyphenyl)acetamide (1.99 g, 10 mmol). The product was recrystallised from ethanol as yellow crystals (2.8 g, 79%), Mp. 240°C - 242°C; ¹H-NMR (DMSO) $\delta = 2.65$ (3H, s, CH₃ py.), 2.73 (3H, s, CH₃CO), 3.76 (3H, s, CH₃O), 6.9 (2H, d, Ar), 7.45 (2H, s, NH₂), 7.56 (2H, d, Ar), 9.04 (1H, s, CH

py.), 9.4 (1H, s, NH); IR (KBr) ν 3428, 3310, 3251 (NH₂, NH), 1680 cm⁻¹ (C=O); MS (EI)⁺: m/z 355 M⁺; Anal. Calcd for C₁₈H₁₇N₃O₃S (355.42): C, 60.83; H, 4.82; N, 11.82, Found: C, 60.76; H, 4.73; N, 11.69.

General procedure for the preparation of compounds 9a,b

A mixture of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** (1.92 g, 10 mmol) in ethanol as solvent in presence of sodium hydroxide (0.4 g, 10 mmol) with aromatic aldehydes (10 mmol) with stirring for 2hr. then poured onto ice, cold water and acidified with conc. hydrochloric acid until the precipitate was formed. The solid product was recovered by filtration and recrystallised from ethanol.

5-(3-(4-(Dimethylamino)phenyl)acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile (9a):

Obtained using 4-(dimethylamino)benzaldehyde (1.49g, 10 mmol). Mp. 140°C - 142°C as yellow crystals (2.45 g, 76%); ¹H-NMR (CDCl₃) δ = 2.65 (3H, s, CH₃), 2.79 (6H, s, NMe₂), 6.70 (2H, d, Ar), 7.06 (1H, d, CH chalcone), 7.74 (2H, d, Ar), 7.85 (1H, d, CH chalcone), 8.07 (1H, s, CH py.), 13.2 (1H, br., NH); IR (KBr) ν 3437 (NH), 2225 (CN), 1685 cm⁻¹ (C=O); MS (EI)⁺: m/z 323 M⁺; Anal. Calcd for C₁₈H₁₇N₃OS (323.42): C, 66.85; H, 5.30; N, 12.99, Found: C, 66.69; H, 5.17; N, 12.86.

6-Methyl-2-thioxo-5-(3-(p-tolyl)acryloyl)-1,2-dihydropyridine-3-carbonitrile (9b): Obtained using 4-methylbenzaldehyde (1.2 g, 10 mmol). Mp. = 240°C - 242°C as yellow crystals (2.2 g, 74.8%); ¹H-NMR (DMSO) δ = 2.31 (3H, s, CH₃ Ar), 2.58 (3H, s, CH₃ py.), 7.24 (2H, d, Ar), 7.49 (1H, d, CH chalcone), 7.6 (1H, d, CH chalcone), 7.69 (2H, d, Ar), 8.58 (1H, s, CH py.), 13 (1H, br., NH); IR (KBr) ν 3434 (NH), 2231 (CN), 1659 cm⁻¹ (C=O); MS (EI)⁺: m/z 294 M⁺; Anal. Calcd for C₁₇H₁₄N₂OS (294.38): C, 69.36; H, 4.79; N, 9.52, Found: C, 69.19; H, 4.68; N, 9.45.

5-(5-Amino-4-cyanothiophen-3-yl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile (10)

In dry flask, a mixture 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** (1.92 g, 10 mmol), malononitrile (0.66 g, 10 mmol) and sulfur (0.32 g, 10 mmol) in ethanol and few drops of triethylamine as base was left under reflux for three hours. The mixture was left for cooling then poured onto ice cold water. The product obtained was recrystallised from a mixture of ethanol/DMF (3:1) as brown crystals (1.9 g, 69.8%), Mp. > 300°C; ¹H-NMR (DMSO) δ = 2.45 (3H, s, CH₃), 6.95 (2H, s, NH₂), 7.07 (1H, s, CH thiophene), 7.2 (1H, s, CH py.); IR (KBr) ν 3435, 3350 (NH₂), 3250 (NH), 2210 cm⁻¹ (CN); Anal. Calcd for C₁₂H₈N₄S₂ (272.35): C, 52.92; H, 2.96; N, 20.57, Found: C, 52.85; H, 2.92; N, 20.40.

5-Acetyl-2-ethoxy-6-methylnicotinonitrile (11)

In dry flask, a mixture of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile **7** (2.06 g, 10 mmol) in ethanol and potassium carbonate was left under reflux for 3hr. after cooling the mixture was poured onto ice cold water. The product was recovered and recrystallised from EtOH/H₂O (1:1) as yellowish crystals (1.6 g, 78%), Mp. 78°C - 80°C; ¹H-NMR (CDCl₃) δ = 1.43 (3H, t, CH₃ ethyl), 2.66 (3H, s, CH₃), 3.04 (3H, s, CH₃CO), 4.54 (2H, q, CH₂ ethyl), 7.85 (1H, s, CH py.); IR (KBr) ν 2228 (CN), 1688 cm⁻¹ (C=O); MS (EI)⁺: m/z 204 M⁺; Anal. Calcd for C₁₁H₁₂N₂O₂ (204.23): C, 64.69; H, 5.92; N, 13.72, Found: C, 64.51; H, 5.83; N, 13.54.

5-(3-(Dimethylamino)acryloyl)-2-ethoxy-6-methylnicotinonitrile (12)

In dry flask a mixture of 5-acetyl-2-ethoxy-6-methylnicotinonitrile **11** (2.04 g, 10 mmol) in dry xylene as solvent and *N,N'*-dimethylformamide dimethyl acetal (DMFDMA) (1.32 ml, 10 mmol) was left under reflux for 2 hr., cool and the solvent was evaporated. The product was recovered and recrystallised from EtOH/H₂O (1:1) as yellow crystals (1.9 g, 73.3%), Mp. 68°C - 70°C; ¹H-NMR (CDCl₃) δ = 1.3 (3H, t, CH₃ ethyl), 2.62 (3H, s, CH₃), 2.68, 3.04 (6H, 2s, NMe₂), 4.58 (2H, q, CH₂ ethyl), 6.25 (1H, d, CH), 7.87 (1H, d, CH), 8.2 (1H, s, CH py.); IR (KBr) ν 2230 (CN), 1684 cm⁻¹ (C=O); MS (EI)⁺: m/z 259 M⁺; Anal. Calcd for C₁₄H₁₇N₃O₂ (259.31): C, 64.85; H, 6.61; N, 16.20, Found: C, 64.56; H, 6.47; N, 16.11.

5-(3-(Dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile (13)

(A) In dry flask, a mixture of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile **7** (2.06 g, 10 mmol) in dry xylene as solvent and *N,N'*-dimethylformamide dimethyl acetal (DMFDMA) (1.32 ml, 10 mmol) was left under reflux for 2 hr., cool and poured in dry backer and the solvent was evaporated. The product was recovered and recrystallised from EtOH/H₂O (1:1) as yellow crystals (2 g, 76.6%), Mp. 100°C - 102°C; (B) In dry flask a mixture of 5-acetyl-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **3a** (1.92 g, 10 mmol) in dry xylene as solvent and *N,N'*-dimethylformamide dimethyl acetal (DMFDMA) (2.64 ml, 20 mmol) was left under reflux for 2 hr., cool and poured in dry backer and the solvent was evaporated. The product was recovered and recrystallised from EtOH/H₂O (1:1) as yellow crystals (2.1 g, 80.4%), Mp. and mixed Mp. 100°C - 102°C; ¹H-NMR (CDCl₃) δ = 2.62, 2.64 (6H, 2s, NMe₂), 2.9 (3H, s, CH₃ py.), 3.15 (3H, s, SCH₃), 5.28 (1H, d, trans CH), 6.28 (1H, d, cis CH), 7.75 (1H, d, trans CH), 8.07 (1H, s, CH py.), 10.15 (1H, d, cis CH); IR (KBr) ν 2227 (CN),

1685 cm^{-1} (C=O); MS (EI)⁺: m/z 261 M⁺; Anal. Calcd for C₁₃H₁₅N₃OS (261.35): C, 59.74; H, 5.79; N, 16.08, Found: C, 59.63; H, 5.45; N, 15.8.

5-(3-(4-(Dimethylamino)phenyl)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile (14)

(A) mixture of 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **9a** (3.23 g, 10 mmol) in ethanol as solvent and sodium hydroxide (0.4g, 10mmol) with stirring for 1hr., and add methyl iodide (0.62 ml, 10 mmol) with stirring until precipitate was formed. The product was recovered by filtration and was purified by recrystallised from ethanol as yellow crystals (2.5 g, 74%), Mp. 160°C - 162°C; (B) mixture of 5-acetyl-6-methyl-2-(methylthio)nicotinonitrile **7** (2.06 g, 10 mmol) in ethanol as solvent in presence of sodium hydroxide (0.4 g, 10 mmol) with 4-(dimethylamino)benzaldehyde (1.49 g, 10 mmol) with stirring for 2 hr., until precipitate formed and dilute with water. The product was recovered by filtration and purified by recrystallised from ethanol as yellow crystals (2.4 g, 71%), Mp. and mixed Mp. 160°C - 162°C; ¹H-NMR (CDCl₃) δ = 2.62 (3H, s, CH₃), 2.66 (3H, s, SCH₃), 2.9, 3.04 (6H, 2s, NMe₂), 6.83 (2H, d, Ar), 7.46 (2H, d, Ar), 6.67 (1H, d, CH), 7.38 (1H, d, CH), 7.85 (1H, s, CH py.); IR (KBr) ν 2217 (CN), 1648 cm^{-1} (C=O); MS (EI)⁺: m/z 337 M⁺; Anal. Calcd for C₁₉H₁₉N₃OS (337.45): C, 67.63; H, 5.68; N, 12.45, Found: C, 67.49; H, 5.62; N, 12.48.

6-(Dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1,1',6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (15)

In dry flask, a mixture 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2-thioxo-1,2-dihydropyridine-3-carbonitrile **9a** (3.23 g, 10 mmol) and malononitrile dimmer (1.32 g, 10 mmol) in acetic acid and presence of ammonium acetate was left under reflux for three hours. The mixture was left for cooling and poured onto ice, cold water. The product was recovered by filtration and recrystallisation from ethanol as brown crystals (3.25 g, 74.7%), Mp. 260°C - 262°C; ¹H-NMR (DMSO) δ = 2.38 (3H, s, CH₃), 3.06 (6H, s, NMe₂), 6.83 (2H, d, Ar), 7.5 (1H, s, CH py.), 7.93 (2H, d, Ar), 8.21 (1H, s, CH py.), 11.93 (1H, br, NH), 12.4 (1H, br, NH); IR (KBr) ν 3334, 3207 (2NH), 2206 cm^{-1} (CN); MS (EI)⁺: m/z 435 M⁺; Anal. Calcd for C₂₄H₁₇N₇S (435.51): C, 66.19; H, 3.93; N, 22.51, Found: C, 66.06; H, 3.78; N, 22.35.

6-(Dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-(methylthio)-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (16)

(A) In dry flask a mixture of 5-(3-(4-(dimethylamino)phenyl)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile **14** (3.37 g, 10 mmol) and malononitrile dimmer (1.32 g, 10 mmol) in acetic acid acid and ammonium acetate was left under reflux for four hours, cool. The solid product was recovered by filtration and recrystallised from acetic acid as brown crystals (3.4 g, 76%), Mp. 220°C - 222°C; (B) mixture of 6-(dicyanomethylene)-4-(4-(dimethylamino)phenyl)-2'-methyl-6'-thioxo-1,1',6,6'-tetrahydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (**15**) (4.35 g, 10 mmol) in ethanol as solvent in presence of sodium hydroxide (0.4 g, 10mmol) and methyl iodide (0.62 ml, 10 mmol) with stirring until precipitate formed. The product was recovered by filtration and recrystallised from acetic acid as brown crystals (3.2 g, 71.5%), Mp. and mixed Mp. 220°C - 222°C; ¹H-NMR (DMSO) δ = 2.61 (3H, s, CH₃), 2.65 (3H, s, SCH₃), 2.99, 3.01 (6H, 2s, NMe₂), 6.82 (2H, d, Ar), 7.09 (1H, s, CH py.), 7.73 (2H, d, Ar), 8.66 (1H, s, CH py.), 10.3 (1H, br, NH); IR (KBr) ν 3345 (NH), 2213 cm^{-1} (CN); MS (EI)⁺: m/z 449 M⁺; Anal. Calcd for C₂₅H₁₉N₇S (449.54): C, 66.80; H, 4.26; N, 21.81, Found: C, 66.69; H, 4.18; N, 21.66.

6-Methyl-5-(1H-pyrazol-3-yl)-1H-pyrazolo[3,4-b]pyridin-3-amine (17)

In flask a mixture of 5-(3-(dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile **13** (2.61 g, 10 mmol) and excess of hydrazine hydrate was left under reflux for four hours, cool. The solid product was recovered by filtration and recrystallised from ethanol as yellowish crystals (1.6 g, 75%), Mp. 260°C - 262°C; ¹H-NMR (DMSO) δ = 2.49 (3H, s, CH₃), 5.54 (2H, s, NH₂), 6.47 (1H, d, CH pyrazole), 7.8 (1H, s, CH py.), 8.3 (1H, d, CH pyrazole), 11.75 (1H, s, NH), 12.91 (1H, s, NH); IR (KBr) ν at 3405, 3329, 3136 cm^{-1} (NH₂, NH); MS (EI)⁺: m/z 214 M⁺; Anal. Calcd for C₁₀H₁₀N₆ (214.23): C, 56.07; H, 4.71; N, 39.23, Found: C, 55.85; H, 4.56; N, 39.16.

2'-Methyl-6'-(methylthio)-6-thioxo-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (18)

In dry flask a mixture of 5-(3-(dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile **13** (2.61 g, 10 mmol) and cyanothioacetamide (1 g, 10 mmol) in acetic acid and ammonium acetate was left under reflux for four hours. Cool and poured the mixture onto ice cold water. The product was recovered by filtration and recrystallised from ethanol as brown crystals (2.3 g, 77.1%), Mp. 170°C - 172°C; ¹H-NMR (DMSO) δ = 2.63 (3H, s, CH₃), 2.65 (3H, s, SCH₃), 7.7 (1H, d, CH py.), 8 (1H, d, CH py.), 8.14 (1H, s, CH py.), 12.25 (1H, br, NH); IR (KBr) ν = 3428 (NH), 2221 cm^{-1} (CN); MS (EI)⁺: m/z 298 M⁺; Anal. Calcd for C₁₄H₁₀N₄S₂ (298.39): C, 56.35; H, 3.38; N, 18.78, Found: C, 56.12; H, 3.24; N, 18.58.

General procedure for the preparation of compounds 19a,b

In dry flask, a mixture of 5-(3-(dimethylamino)acryloyl)-2-ethoxy-6-methylnicotinonitrile **12** (2.59 g, 10 mmol) or 5-(3-(dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile **13** (2.61 g, 10 mmol) and malononitrile dimmer (1.32 g, 10 mmol) in acetic acid and ammonium acetate was heated under reflux for four hours, cool. The solid product was recovered by filtration and recrystallised from ethanol

6-(Dicyanomethylene)-6'-ethoxy-2'-methyl-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile (19a):

Obtained using 5-(3-(dimethylamino)acryloyl)-2-ethoxy-6-methylnicotinonitrile **12**. Mp. 200°C - 202°C as brown crystals (2.4 g, 73.1%); ¹H-NMR (DMSO) δ = 1.39 (3H, t, CH₃), 2.62 (3H, s, CH₃), 4.50 (2H, q, CH₂), 7.58 (1H, d, CH py.), 8.48 (1H, d, CH py.), 8.7 (1H, s, CH py.), 11.3 (1H, br, NH); IR (KBr) ν 3330 (NH), 2218 cm⁻¹ (CN); MS (EI)⁺: m/z 328 M⁺; Anal. Calcd for C₁₈H₁₂N₆O (328.34): C, 65.85; H, 3.68; N, 25.60, Found: C, 65.71; H, 3.52; N, 25.43.

6-(Dicyanomethylene)-2'-methyl-6'-(methylthio)-1,6-dihydro-[2,3'-bipyridine]-5,5'-dicarbonitrile

(19b): Obtained using 5-(3-(dimethylamino)acryloyl)-6-methyl-2-(methylthio)nicotinonitrile **13**. Mp. 190°C - 192°C as brown crystals (2.3 g, 69.7%); ¹H-NMR (DMSO) δ = 2.58 (3H, s, CH₃), 2.64 (3H, s, SCH₃), 6.5 (1H, d, CH py.), 8.2 (1H, d, CH py.), 8.69 (1H, s, CH py.), 11.31 (1H, br, NH); IR (KBr) ν 3340 (NH), 2212 cm⁻¹ (CN); MS (EI)⁺: m/z 330 M⁺; Anal. Calcd for C₁₇H₁₀N₆S (330.37): C, 61.80; H, 3.05; N, 25.44, Found: C, 61.63; H, 2.89; N, 25.27.

8-Acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d]pyrimidin-4(3H)-one (20)

A mixture of 5-acetyl-3-amino-6-methyl-N-(p-tolyl)thieno[2,3-b]pyridine-2-carboxamide **8b** (3.39 g, 10 mmol) in dry dioxane and DMFDMA (1.32 ml, 10 mmol) with stirring for 12 hrs. The product was recovered by filtration and recrystallised from acetic acid as gray crystals (2.6 g, 74.5%), Mp. 200°C - 202°C; ¹H-NMR (DMSO) δ 2.26 (3H, s, CH₃ Ar), 2.68 (3H, s, CH₃ py.), 2.69 (3H, s, CH₃CO), 7.16 (2H, d, Ar), 7.52 (2H, d, Ar), 8.43 (1H, s, CH pyrimidinone), 8.52 (1H, s, CH py.); IR (KBr) ν at 1691, 1649 cm⁻¹ (2C=O); MS (EI)⁺: m/z 349 M⁺; Anal. Calcd for C₁₉H₁₅N₃O₂S (349.41): C, 65.31; H, 4.33; N, 12.03, Found: C, 65.19; H, 4.26; N, 11.95.

General procedure for the preparation of compounds 21a,b

A mixture of N-substituted-5-acetyl-3-amino-6-methylthieno[2,3-b]pyridine-2-carboxamide **8b,c** (10 mmol) in acetic acid and sodium nitrite (1.38 g, 20mmol) with stirring for 1 hr. the precipitate was formed and dilute with water. The product was recovered by filtration and recrystallised from ethanol.

8-Acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)-one (21a): Obtained using 5-acetyl-3-amino-6-methyl-N-(p-tolyl)thieno[2,3-b]pyridine-2-carboxamide **8b** (3.39g, 10 mmol). Mp. 170°C - 172°C as gray crystals (3 g, 85.7%); ¹H-NMR (DMSO) δ = 2.4 (3H, s, CH₃ Ar), 2.74 (3H, s, CH₃ py.), 2.77 (3H, s, CH₃CO), 7.4 (2H, d, Ar), 7.54 (2H, d, Ar), 9.17 (1H, s, CH py.); IR (KBr) ν 1700, 1687 cm⁻¹ (2C=O); MS (EI)⁺: m/z 350 M⁺; Anal. Calcd for C₁₈H₁₄N₄O₂S (350.40): C, 61.70; H, 4.03; N, 15.99, Found: C, 61.56; H, 3.94; N, 15.78.

8-Acetyl-3-(4-methoxyphenyl)-7-methylpyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)-one (21b): Obtained using 5-acetyl-3-amino-N-(4-methoxyphenyl)-6-methylthieno[2,3-b]pyridine-2-carboxamide **8c** (3.55 g, 10 mmol). Mp. = 220°C - 222°C as gray crystals (2.9 g, 79.4%); ¹H-NMR (DMSO) δ = 2.74 (3H, s, CH₃ py.), 2.81 (3H, s, CH₃CO), 3.85 (3H, s, CH₃O), 7.15 (2H, d, Ar), 7.61 (2H, d, Ar), 9.26 (1H, s, CH py.); IR (KBr) ν 1687 cm⁻¹ (2C=O); Anal. Calcd for C₁₈H₁₄N₄O₃S (366.40): C, 59.01; H, 3.85; N, 15.29, Found: C, 58.90; H, 3.76; N, 15.17.

2-Amino-4-(7-methyl-4-oxo-3-(p-tolyl)-3,4-dihydropyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-8-yl)thiophene-3-carbonitrile (22):

In dry flask a mixture 8-acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)-one **21a** (3.5g, 10 mmol), malononitrile (0.66 g, 10 mmol) and elemental sulfur (0.32 g, 10mmol) in ethanol and few drops of triethylamine as base was heated under reflux for three hours. The mixture was left for cooling and poured onto ice cold water. The product was recovered by filtration and recrystallised from a mixture of ethanol/DMF (3:1) as brown crystals (3 g, 69.7%), M.p 260°C - 262°C; IR (KBr) ν 3427 (NH₂), 2208 (CN), 1683 cm⁻¹ (C=O); MS (EI)⁺: m/z 430 M⁺; Anal. Calcd for C₂₁H₁₄N₆OS₂ (430.51): C, 58.59; H, 3.28; N, 19.52, Found: C, 58.43; H, 3.14; N, 19.36.

8-(3-(Dimethylamino)acryloyl)-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)-one (23):

In dry flask a mixture 8-acetyl-7-methyl-3-(p-tolyl)pyrido[3',2':4,5]thieno[3,2-d][1,2,3]triazin-4(3H)-one **21a** (3.5 g, 10 mmol) and DMFDMA (1.32 ml, 10 mmol) in dry dioxane was left under reflux for two hours. The

mixture was left for cooling and evaporates the solvent. The product was recovered by filtration and recrystallised from ethanol as brown crystals (2.9 g, 71.6%), Mp. 210°C - 212°C; $^1\text{H-NMR}$ (DMSO) δ = 2.39 (3H, s, CH₃ Ar), 2.66 (3H, s, CH₃ py.), 3.63, 3.67 (6H, 2s, NMe₂), 5.42 (1H, d, CH), 7.41 (2H, d, Ar), 7.54 (2H, d, Ar), 7.82 (1H, d, CH), 9.12 (1H, s, CH py.); IR (KBr) ν 1693, 16.44 cm⁻¹ (2C=O); MS (EI)⁺: m/z 405 M⁺; Anal. Calcd for C₂₁H₁₉N₅O₂S (405.48): C, 62.21; H, 4.72; N, 17.27, Found: C, 62.08; H, 4.59; N, 17.11.

4. Conclusion

From the biological importance of pyridine-2(1H)-thione derivatives, we have used it in order for the preparation of biologically important bipyridyles, bi- and uncommon tricyclic compounds.

References

- [1] Granik, V.G., Zhidkova, A.M. and Glushkov, R.G. (1977) Advances in the Chemistry of the Acetals of Acid Amides and Lactams. *Russian Chemical Reviews*, **46**, 361. <http://dx.doi.org/10.1070/RC1977v046n04ABEH002137>
- [2] Abdulla, R.F. and Brinkmeyer, R.S. (1979) The Chemistry of Formamide Acetals. *Tetrahedron*, **35**, 1675-1735.
- [3] Anelli, P.L., Brocchetta, M., Palano, D. and Visigalli, M. (1997) Mild Conversion of Primary Carboxamides into Carboxylic Esters. *Tetrahedron Letters*, **38**, 2367-2368. [http://dx.doi.org/10.1016/S0040-4039\(97\)00350-X](http://dx.doi.org/10.1016/S0040-4039(97)00350-X)
- [4] Malesic, M., Krbavcic, A., Golobic, A., Golic, L. and Stanovenik, B. (1997) The Synthesis and Transformation of Ethyl 2-(2-acetyl-2-benzoyl-1-ethenyl)amino-3-dimethylaminopropenoate. A New Synthesis of 2,3,4-Trisubstituted Pyrroles. *Journal of Heterocyclic Chemistry*, **34**, 1757-1762.
- [5] Abu-Shanab, F.A., Elnagdi, M.H., Aly, F.M. and Wakefield, B.J. (1994) α,α -Dioxoketene Dithioacetals as Starting Materials for the Synthesis of Polysubstituted Pyridines. *Journal of the Chemical Society, Perkin*, **1**, 1449-1452. <http://dx.doi.org/10.1039/p19940001449>
- [6] Abu-Shanab, F.A., Redhouse, A.D., Thompson, J.R. and Wakefield, B.J. (1995) Synthesis of 2,3,5,6-Tetrasubstituted Pyridines from Enamines Derived from N,N-Dimethylformamide Dimethyl Acetal. *Synthesis*, **5**, 557-560. <http://dx.doi.org/10.1055/s-1995-3954>
- [7] Abu-Shanab, F.A., Aly, F.M. and Wakefield, B.J. (1995) Synthesis of Substituted Nicotinamides from Enamines Derived from N,N-Dimethylformamide Dimethyl Acetal. *Synthesis*, **8**, 923-925.
- [8] Abu-Shanab, F.A., Hessen, A.M. and Mousa, S.A.S. (2007) Dimethylformamide Dimethyl Acetal in Heterocyclic Synthesis: Synthesis of Polyfunctionally Substituted Pyridine Derivatives as Precursors to Bicycles and Polycycles. *Journal of Heterocyclic Chemistry*, **44**, 787-791. <http://dx.doi.org/10.1002/jhet.5570440406>
- [9] Carboni, R.A., Conffman, D.D. and Howard, E.G. (1958) Cyanocarbon Chemistry. XI.¹ Malononitrile Dimer. *Journal of the American Chemical Society*, **80**, 2838-2840. <http://dx.doi.org/10.1021/ja01544a061>
- [10] Abu-Shanab, F.A., Sherif, S.M. and Mousa, S.A.S. (2009) Dimethylformamide Dimethyl Acetal as a Building Block in Heterocyclic Synthesis. *Journal of Heterocyclic Chemistry*, **46**, 801-827. <http://dx.doi.org/10.1002/jhet.69>
- [11] Melani, F., Cecchi, L., Colotta, V., Filacchini, G., Martini, C., Giannicini, G. and Lucacchini, A. (1989) Dipyrzolo[5,4-b:3',4'-d]pyridines. Synthesis, Inhibition of Benzodiazepine Receptor Binding and Structure-Activity Relationships. *Farmaco*, **44**, 585-594.
- [12] Abu-Shanab, F.A.M., Mousa, S.A.S., Eshak, E.A., Sayed, A.Z. and Al-Harrasi, A. (2011) Dimethylformamide Dimethyl Acetal (DMFDMA) in Heterocyclic Synthesis: Synthesis of Polysubstituted Pyridines, Pyrimidines, Pyridazine and Their Fused Derivatives. *International Journal of Chemistry*, **1**, 207-214.

Scientific Research Publishing (SCIRP) is one of the largest Open Access journal publishers. It is currently publishing more than 200 open access, online, peer-reviewed journals covering a wide range of academic disciplines. SCIRP serves the worldwide academic communities and contributes to the progress and application of science with its publication.

Other selected journals from SCIRP are listed as below. Submit your manuscript to us via either submit@scirp.org or [Online Submission Portal](#).

