

Low-Loss Co₂-Y Ferrites with Added CuO Sintered in Air for High Frequency Application

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Abstract

The sintering characteristics of hexagonal Co₂-Y-type ferrite, Ba₂Co₂Fe₁₂O₂₂, with the addition of 0.6 wt% CuO, were studied in order to allow for preparation in air, as opposed to the conventionally recommended O₂, for industrial production. The dependence of the resistivity, ρ magnetic loss, tan δ , and the permeability, μ , at 1 GHz on the sintering temperature was investigated. A low tan δ of 0.05 with a μ of 2.7 at a frequency of 1 GHz, along with a high ρ (up to 7 × 10⁴ µm), were attained under sintering at 1170°C in air, which were the same features as those of samples sintered at 1200°C in O₂. The dependence of tan δ on grain diameter was also examined, and it was determined that a small grain size (less than 2 µm) is preferable for low tan δ .

Keywords

Ferrite, Hexagonal, Y-Type, Sintering, Low Loss

1. Introduction

Ferrites and other magnetic materials have been widely used as the key elements in microwave devices [1]-[3] such as isolators, phase shifters [4] [5], and circulators [6]. Traditional ferrites, known as spinel types, such as Ni-Zn ferrites have been known to exhibit high permeability in the frequency range up to a few hundred MHz because of restriction by Snoek's law [7] [8]. Therefore, hexagonal ferrites are expected to be promising candidates for expansion of the device frequency to the GHz range [9]. Z- and Y-type ferrites show soft magnetic characteristics with moderate relative permeability, μ , up to 1 GHz [10]-[12]. The former is denoted as Ba₃Me₂Fe₂₄O₄₁, while the latter is labeled Ba₂Me₂Fe₁₂O₂₂, where Me represents a divalent metal ion from the first transition series or, alternatively, it may represent Zn or Mg. In particular, the Y-type ferrite has a high Curie temperature

[11] and higher magnetic resonance than the Z-type, despite a low μ [13] [14]. Therefore, it is more applicable to high frequency devices in the GHz range.

Recently, telecommunication devices applied to mobile phones have broadened their market extensively. For these applications, antennas are essential, and miniaturization of these devices is therefore necessary. Ferrites possess permeability as well as permittivity, and are considered as a candidate material for chip antennas [15]-[19], because the wavelength is reduced proportionally according to $1\sqrt{\mu\epsilon}$ where ϵ is the relative permittivity. The Y-type ferrite, Ba₂Co₂Fe₁₂O₂₂Y (Co₂-Y), with a high Curie temperature of ~330°C [11], is a promising material for this application.

We have demonstrate that Co₂-Y modified by the addition of 0.6 wt% CuO exhibited a moderate μ of ~2.7 and low magnetic loss, tan δ , of 0.05, even at 1 GHz [20]. This material has been prepared by means of a conventional powder metallurgical process, and sintering has been conducted under the conventionally recommended oxygen atmosphere. However, sintering of the Co₂-Y in air would be preferable for industrial production because of cost effectiveness. To date, some studies have been conducted on Y-type Ba₂Zn_{2-2x-2y}Co_{2x}Cu_{2y}Fe₁₂O₂₂ ($0 \leq x \leq 0.1$) sintering in air [14] [21]. In addition, Co₂-Y (with no added CuO) sintering in air has also been reported [22], which showed a high resistivity of 5 × 10⁴ Ω m but did not show values of tan δ . The industrially favorable sintering conditions (in air) of 0.6 wt% CuO added to Co₂-Y will be presented in this study in order to attain the same characteristics as the samples sintered in O₂ in our early study mentioned above. The effective factor of tan δ will also be discussed.

2. Experimental

Samples of Co₂-Y ferrite were prepared as a stoichiometric composition by means of conventional powder metallurgy. Raw material powders of Fe₂O₃, BaCO₃, and Co₃O₄ were well-mixed using ball-milling and calcinated at 1000°C for 2 h in air. The calcinated powders were ground, with the addition of 0.6 wt% CuO powders and of a 1 wt% PVA binder, followed by compacting into predetermined shapes at a pressure of 20 MPa and then sintered at 1200°C for 3 h under atmospheres with varying oxygen content (balance: nitrogen).

The resistivity, ρ , was measured for samples whose dimensions were 13 mm in diameter and 3 mm in thickness. An electrode was then printed on both sides of the samples as a silver paste. The permeability and permittivity frequency response were characterized by means of a network analyzer (Agilent E8364A) and a coaxial airline fixture (KANTOH E.A.D. Co. Model: CSH2-APC-7) up to 18 GHz, after the Nicolson-Ross method [23]. Ring shaped samples (ID: 3.0 mm, OD: 7.0 mm, thickness: 3.5 mm) were used in this characterization.

The sample densities were determined by means of Archimedes' method, while their morphologies were investigated using a scanning electron microscope (SEM) (Hitachi S-800). The grain diameters were defined as the average diagonal length of approximately 30 grains, orienting the hexagonal shape towards the top, in the SEM images.

3. Results and Discussion

The Co₂-Y samples with the added 0.6 wt% CuO were prepared in various O₂ volume configurations at a sintering temperature of 1200°C. The O₂ volume fraction varied from 15% - 100%. The dependence of μ , tan δ at 1 GHz, the density, and ρ , on the O₂ volume fraction are shown in Figure 1(a) and Figure 1(b), respectively. Both μ and tan δ decrease with increasing O₂ volume fraction. The lowest tan δ (at 0.05) with a μ of 2.7 was achieved at a full O₂ atmosphere (100%), while the atmospheric case, *i.e.*, sintering in air (20% O₂), exhibited a high tan δ of 0.15 along with a high μ of ~4. It can be seen that the density increases while ρ decreases with decreasing O₂ volume fraction. A high density of 5.24 × 10³ kg/m³ and a low ρ of 1.2 × 10⁴ Ωm were achieved in air. Low tan δ is required for energy-conversion devices such as inductors and antennas. High ρ is also required, because winding coils or printed electrodes make contact with ferrites directly. These characteristics were not attained when sintering at 1200°C conducted in air, as shown in Figure 1. Therefore, the sintering temperature in air must be considered.

The dependence of ρ on the sintering temperature in O₂ and in air is shown in Figure 2, where open and black circles denote sintering in O₂ and in air, respectively. ρ decreases with sintering temperature for both atmospheres, although the change in air is steeper than that in O₂. A resistivity greater than $7 \times 10^4 \Omega$ m in the case of sintering in O₂ is reached below 1180°C in air.

The change in μ and tan δ at 1 GHz due to sintering temperature in O₂ and in air are shown in Figure 3(a) and



Figure 1. Dependence of (a) μ and tan δ , and (b) density and ρ , on O₂ volume fraction at sintering.



Figure 2. Dependence of ρ on sintering temperature in O₂ and in air.



Figure 3. Change in (a) μ and (b) tan δ at 1 GHz due to sintering temperature in O₂ and in air

Figure 3(b), respectively. μ and tan δ increase with sintering temperature for both atmospheres, and the behavior in the air case are shifted to a lower temperature than that of the O₂ case. The target characteristics, namely, μ of 2.8 with a low tan δ (less than 0.05 at 1 GHz), are attained at 1170°C in the case of sintering in air, which is almost identical to the sample sintered at 1200°C in O₂. It was proven that the crystal structure of this sample was that of a Co₂-Y ferrite using X-ray diffraction as shown in **Figure 4**, where no spinel phase was detected as in [21].

The fractured surfaces of the samples with identical characteristics and sintered in different atmospheres are shown in **Figure 5**. Here, **Figure 5(a)** is an image of a sample sintered at 1200°C in O₂, while **Figure 5(b)** shows a sample sintered at 1170°C in air. The density of the former is 4.82×10^3 kg/m³ and that of the latter is 4.90×10^3 kg/m³, respectively. Both samples have the same morphological aspect with small grains isolated by fine pores. In addition, the majority of the grains have a pseudo-hexagonal platelet shape with thin thickness. The grain sizes, defined as the diagonal length of the hexagonal faces, are estimated as being 2.0 µm in the O₂ case and 1.8 µm for the air case from these images. It is assumed that the same magnetic characteristics are attributed to the same morphological nature in both samples.

The relationship between grain diameter and $\tan \delta$ is shown in **Figure 6**. All samples fabricated in this study are plotted on this figure regardless of sintering conditions. The value of $\tan \delta$ increases with grain diameter, and rises abruptly from 0.05 to 0.15 around a grain diameter of ~2 µm. It has been reported previously that energy dissipation in inductors composed of NiZn ferrites is affected by grain size, and that the dissipation was minimized at a single domain size of 2 - 3 µm [24]. The magnetization process is classified into a spin rotational mode or a magnetic domain wall motion depending on magnetic domain sizes. The rotational mode dominates in the case of small grain diameters equal to magnetic single domain sizes, while larger grain sizes lead to twoor multi-domain structures.

The abrupt increase in tan δ by more than 0.1 beyond a diameter of 2 µm could be attributed to the switching of the magnetization mode from the spin rotational mode to the magnetic domain wall motion. The critical size of a magnetic single domain could be estimated as being approximately 2 µm in the Co₂-Y ferrite. Therefore, controlling the grain size has an effect on reducing tan δ .







Figure 5. Fractured surface of the samples sintered at (a) 1200°C in O_2 and (b) 1700°C in air. Grain sizes were estimated as being ~2.0 µm and ~1.8 µm, respectively.



Figure 6. The relationship between grain diameter and $\tan \delta$.

It is apparent that the Co_2 -Y ferrite with low tan δ even at 1 GHz fabricated in air is favorable for industrial production, and is a promising magnetic material for application to high frequency devices.

4. Conclusion

The sintering characteristics of the hexagonal Co₂-Y-type ferrite, Ba₂Co₂Fe₁₂O₂₂, with the addition of 0.6 wt% CuO in air were examined in order to apply the ferrite to industrial production. It was found that sintering at 1170°C in air resulted in a low tan δ of 0.05, with a high μ of 2.7 at 1 GHz and a high ρ of 7 × 10⁴ µm, identical to a sample sintered at 1200°C in the conventionally recommended oxygen atmosphere. The relationship between grain size and tan δ was also examined. It was found that tan δ was dependent on grain size, and that a size of less than 2 µm is preferable for reducing tan δ , which suggests the ferrite can be applied to microwave devises with low energy dissipation.

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