



Assessment on Water Quality of Waste Water in Sugar Industry and Its Impact on Environment

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How to cite this paper: Sultana, S., Ahmed, A.N., Sultana, S., Biswas, P.K., Saha, B. and Alam, S. (2017) Assessment on Water Quality of Waste Water in Sugar Industry and Its Impact on Environment. *Open Access Library Journal*, 4: e3455. <https://doi.org/10.4236/oalib.1103455>

Received: February 13, 2017

Accepted: March 6, 2017

Published: March 9, 2017

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Abstract

This present study was conducted to determine the physico-chemical properties of sugarmill waste water. Samples were collected from Joypurhat sugarmill area. Different parameters such as pH, EC, TDS, DO, BOD, COD and Na⁺, K⁺, Ca²⁺, Mg²⁺, Cl⁻, SO₄²⁻, HCO₃⁻ etc. of these water samples were analyzed. There was a wide variation in waste water quality of before production and after production of sugarmill. This discharged waste water did not maintain standard values of inland surface water quality (according to Department of Environment, Bangladesh). Waste water from different sugarmills was also studied here. The waste water of Karu & Kong was better than other six sugarmills. This study was done to measure quality of sugarmill waste water and to know the effect of it on surrounding environment.

Subject Areas

Hydrology

Keywords

Sugarmill, Waste Water, BOD, COD, DO, EC, ECR'97

1. Introduction

Sugar industries rank the second among the agro-based industries in Bangladesh. Sugar industry is seasonal in nature and operates only for 120 to 200 days in a year. The effluents released produce a high degree of organic pollution in

both aquatic and terrestrial ecosystems. They alter the physico-chemical characteristics of receiving aquatic bodies and affect aquatic flora & fauna [1]. Waste water from sugarmills with its high Biological Oxygen Demand (BOD) and Chemical Oxygen Demand (COD) rapidly depletes available oxygen supply when discharged into water bodies endangering fish's and other aquatic's life and also creates septic conditions, generating foul-smelling hydrogen sulfide, which in turn can precipitate iron and any dissolved salts, turning the water into black and highly toxic for aquatic life. Suspended solids reduce light penetration capability and as a result, plant production in the receiving water body is diminishing through increasing turbidity that also clogs fish gills. Discharge of water with a high Total Dissolved Solids (TDS) level would have an adverse impact on aquatic life, render the receiving water unfit for drinking and domestic purposes, reduce crop yields if used for irrigation, and exacerbate corrosion in water systems and pipe [2].

Every industry in our country should maintain the Environment Conservation rules of Bangladesh (ECR, 1997) [3]. Firstly we have to know the waste water quality of this sugarmill. After characterization we can compare it with the discharge water quality parameter of ECR'97. Then decision can be made about whether this water discharges into the water sources or not. After this irrigation suitability of it can be investigated. The main objectives of this study include the following points:

- Characterization of Joypurhat sugarmill waste water (JSW) and to compare with the irrigation water of Bangladesh.
- Comparative study on different sugarmill waste water.
- Effect of sugarmill waste water on environment.

2. Method & Materials

2.1. Study Area

This study was done in area of Joypurhat sugarmill Ltd. which is situated at Joypurhat sador, in Joypurhat district at the time of 2014-2015. Location map of sampling site (Joypurhat sugarmill and different sugarmills) is shown in **Figure 1**.

2.2. Sample Collection

Samples were collected in two turns. First, sampling was done before starting production activity of mill in the month of December, 2014. Second, sampling was done after production activity of mill become finished in the month of February, 2015. To get the actual condition of waste water mixture, two samples were collected from each station like light & heavy. Samples were numbered as St-1, St-2, St-3 & St-4 and nearby three pond's samples were numbered as P-1, P-2, P-3. Samples were collected along the flow path of the effluent i.e., the drain. Here pond-1 is directly connected to the discharge path of sugarmill. One sample from Tulshi-gonga river (about 30 km north west away from JSW) as R-1 was collected. A locally used irrigation water (shallow pump) was also collected as a control sample which will be considered as background value. 500 ml PVC

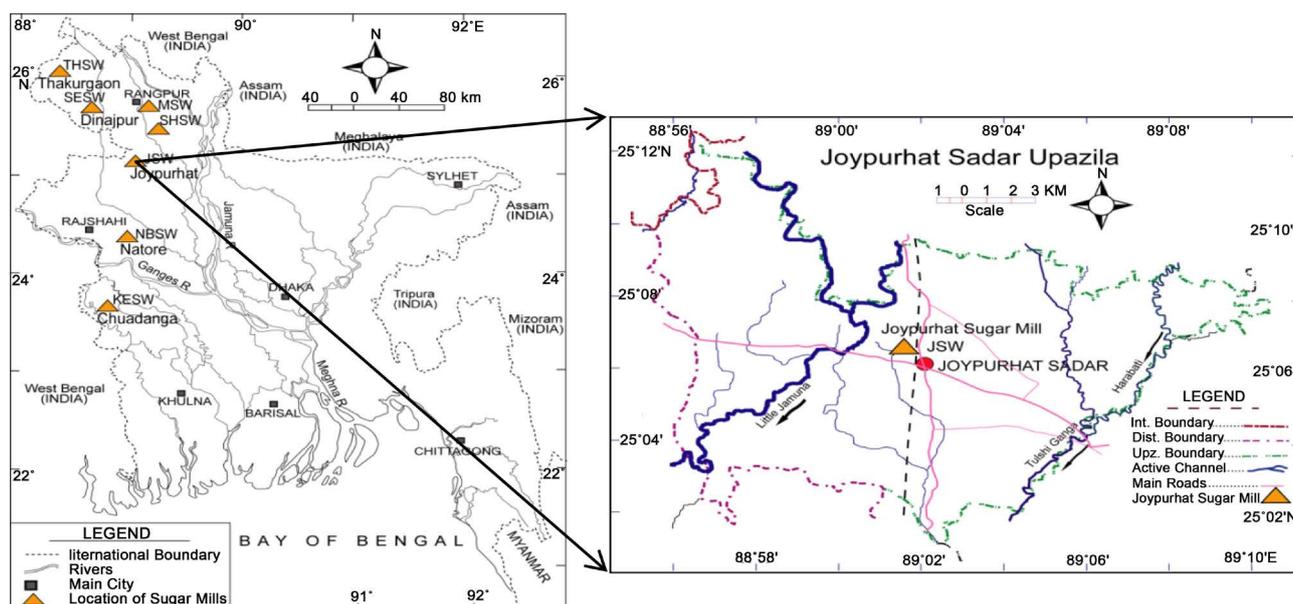


Figure 1. Location map of sampling site (different sugarmills and joypurhat sugarmill).

bottles were used for sample collection. Waste water from six sugarmills was collected at the time of after production. Seven sugarmills studied in this project were situated in north-western part of Bangladesh.

2.3. Sample Analysis

The physical parameters (pH, EC, TDS, Turbidity, temperature) of collected water samples were measured in Institute of Mining, Mineralogy and Metallurgy (IMMM) laboratory. Temperature was measured immediately at the sampling site by a mercury thermometer of 0 to 50°C range and with 0.2°C least count [4]. pH was measured by using digital pH meter (model: HANNA HI 96107). Electrical conductivity was measured by EC meter (model: HANNA HI- 2315). Total Dissolved Solids was measured by TDS meter (model: HANNA HI- 2300). Turbidity was measured by using turbidity meter (model: HANNA HI 93703). Raw water samples were used for physical analysis and physico-chemical analysis. Temperature and pH of sample was measured in the field.

The physico-chemical parameters such as DO, COD, BOD were measured by using related meters. BOD measurement was done in Dhaka laboratory, BCSIR using BOD Track Apparatus, HACH, USA. Fresh samples were used for DO and also for COD & BOD measurement. Dissolved Oxygen was measured by digital DO meter (model: DO 5509). COD was measured by chemical titration method in IMMM laboratory. Different chemical parameters i.e., cations, anions were measured in our laboratory. The cations Sodium (Na^+), Potassium (K^+) were measured by Flame photometer (model: FP 902, PG instruments). Raw samples are used for sodium & potassium measurement. Calcium (Ca^{2+}) and magnesium (Mg^{2+}) were measured by titrimetric method [5]. The anions Chloride (Cl^-), Sulphate (SO_4^{2-}) and bicarbonate HCO_3^- were done by chemical titration/ gravimetric method [6].

3. Result and Discussion

3.1. Physico-Chemical Analysis of Joypurhat Sugarmill Waste Water

The physico-chemical, chemical properties of Joypurhat sugarmill waste water (JSW) were determined. Physico-chemical properties of waste water at before production and after production time were presented in **Table 1**, **Table 2** respectively.

Temperature: Temperature is basically important for its effect on certain chemical and biological radiations taking place in water for organism and inhibiting aquatic media. Salequzzaman *et al.* [7] studied the temperature of five different locations and observed that temperature is between 30°C and 40°C. In this experiment average temperature after production time is in the range of 23°C to 25°C. Here in open drain the temperature of waste water decreases with increase of time and distance.

pH: pH is the most important parameter of water for aquatic life and microbial activity. Before starting the mill the pH at the discharge point ranges 7.4 to 7.7 and after starting the mill the pH ranges 4.0 to 6.1. It was seen that after production time, pH of effluent water did not maintain the standard (6-9). Here

Table 1. Results of physico-chemical analysis of sugarmill waste water at before production time.

	Temp (°C)	pH	EC (µ/cm)	TDS ppm	DO (mg/l)	BOD mg/l	COD mg/l	Turbidity (FTU)
St-1	24.1	7.7	498	236	5.1	22	412.78	930.2
St-2	24.5	7.5	469	222	7.9	27	385.42	651.1
St-3	25.1	7.4	689	326	9.1	45	370.79	454.5
St-4	25.0	7.5	745	345	8.6	6.45	355.08	103.3
P-1	24.5	7.5	474	223	8.2	28	460.89	60.0
P-2	23.2	7.4	407	192	9.1	41	454.53	409.0
P-3	23.0	7.4	375	178	9.0	20	282.42	771.1
R-1	24.0	7.8	172	85	5.4	8.1	124.26	20.91

Table 2. Results of physico-chemical analysis of sugarmill waste water at after production time.

Sample ID	Temp (°C)	pH	EC (µ/cm)	TDS ppm	DO (mg/l)	BOD mg/l	COD mg/l	Turbidity (FTU)	Colour
St-1	23.9	4.0	4675	2572	2.6	69	673	863	deep yellowish
St-2	24.0	4.4	2190	1205	2.1	55	506	589	dark brown
St-3	24.0	4.8	2245	1235	0.3	128	521	>1000	brown to black
St-4	24.0	5.0	2450	1348	0.4	105	519	>1000	Blackish
P-1	24.0	6.1	3090	1699	2.8	98	612	>1000	Yellowish
P-2	24.0	8.3	573	315	4.4	18	109	40.81	Colourless
P-3	24.1	7.3	570	314	6.6	20	114	0.00	light greenish
R-1	24.2	7.1	365	245	6.2	8.9	68	5.6	Turbid
Control	24.1	6.9	193.4	106.37	4.7	8.3	48	2.1	Colorless

pH of control sample was 7.2.

Electrical conductivity: The EC of water is an indicator of salinity and hazard that gives the total salt concentration in water (Brady and Well, 2002) [8]. There is a tremendous increase in concentration of EC during production time. Before production the EC ranges 474 - 745 $\mu\text{s}/\text{cm}$ and after production the EC ranges 2190 - 3090 $\mu\text{s}/\text{cm}$. According to DoE the EC of effluent should be below 1200 $\mu\text{s}/\text{cm}$. After production the discharge water at all stations did not maintain the standard, the values are above the standard. Four stations have higher EC than river and pond water. But pond-1 is different from other two ponds because this pond is directly connected to mill drain.

Turbidity: Turbidity ranges from 109 - 930 FTU at before production time. After production time turbidity was found over 1000 FTU. Too much turbid water is not suitable for aquatic flora and fauna because sunlight cannot pass through the turbid water.

Total Dissolved Solid: TDS also increase tremendously after production. Before starting TDS of effluent ranges 222 to 345 ppm and after production TDS ranges 1205 to 1699 ppm. According to DoE values of after production time are below the inland surface water standard (2100 ppm). TDS value of control sample (106 ppm) was lower than JSW.

Dissolved Oxygen: According to DoE standard, effluent should be within the range of 4.5 to 8 mg/l. After production time DO is almost absent that means at all stations the DO values are <1 , only st-1 contains 2.6 mg/l. Before production time DO was good (range 5.1 - 9.1 mg/l). Control sample of that area has DO value 4.7 mg/l.

Biological Oxygen Demand: One of the most important parameter for aquatic life is BOD. After production time BOD ranges from 55 to 128 mg/l, which was 22 - 45 mg/l in before production time. Control sample of that area has BOD 8.3 mg/l. JSW should be <50 mg/l to discharge into the water body.

Chemical Oxygen Demand: It is another important parameter for aquatic life. During production time, this water contains higher COD values ranges from 506 to 673 mg/l. COD values decreases with the distance of flow path and st-1 contains highest COD values (673 mg/l). COD values of four stations were much higher than the control value (48 mg/l).

3.2. Chemical Analysis of Joypurhat Sugarmill Waste Water

The chemical properties (cations & anions) of waste water at before and after production time were presented in **Table 3**, **Table 4**.

3.2.1. Analysis of Cations

Sodium: There is a tremendous change in concentration of sodium at the time of before (16 - 38 ppm) and after production time (352 - 890 ppm). According to surface water standard the Na^+ in fresh water should be 6.3 ppm [9]. After production, sodium concentration was higher than the background or control value (14.8).

Potassium: According to Garrels and Mackenzie the surface water standard

Table 3. Analysis of cations of sugarmill waste water.

Sample ID	Before production				After production			
	Na ²⁺ ppm	K ⁺ ppm	Ca ²⁺ ppm	Mg ²⁺ ppm	Na ⁺ ppm	K ⁺ ppm	Ca ²⁺ ppm	Mg ²⁺ ppm
St-1	17.5	18.5	BDL	BDL	475	242	460	141.6
St-2	16	12	BDL	BDL	352	206	400	206.4
St-3	18	17	BDL	BDL	890	210	188	93.6
St-4	38	20	BDL	BDL	508	212	240	7.2
P-1	25	14	BDL	BDL	653	167	200	0.00
P- 2	19	24	BDL	BDL	37	26	186	0.00
P-3	26	20	BDL	BDL	43	24	160	0.00
R-1	15	23	BDL	BDL	25	63	100	26.4
	Control				14.8	7.1	14.5	13.2

*BDL = Below detection limit.

Table 4. Analysis of anions of sugarmill waste water.

Sample ID	Before production			After production		
	SO ₄ ²⁻ ppm	Cl ⁻ ppm	HCO ₃ ⁻ ppm	SO ₄ ²⁻ ppm	Cl ⁻ ppm	HCO ₃ ⁻ ppm
St-1	1.33	22.15	175.3	5.5	3500.69	BDL
St-2	0.44	15.51	198.25	2.0	700.14	BDL
St-3	0.51	64.25	259.25	1.63	443.12	BDL
St-4	0.18	70.90	207.4	1.88	664.69	BDL
P- 1	0.58	22.16	221.1	1.40	797.63	BDL
P- 2	0.00	44.31	230.2	0.00	221.56	BDL
P-3	0.00	60.90		0.00	230.02	BDL
R-1	0.00		0.00	0.00	35.45	BDL
Control					44.32	41.18

the K⁺ in fresh water should be 2.3 ppm [9]. The same condition was found for potassium. It ranges from 12 to 20 ppm at before production and 167 - 242 ppm at after production time. Potassium content of after production is much higher than the control value (7.1 ppm). Higher concentration of sodium and potassium are not harmful because these two cations will increase fertility status of agricultural soil.

Calcium: According to surface water standard the Ca²⁺ in fresh water should be 15 ppm [9]. In this experiment, calcium concentration could not be determined by gravimetric method. But after production time it was found in range of 240 - 460 ppm. Background or control value of that area was 14.5 ppm.

Magnesium: According to surface water standard the Mg²⁺ in fresh water should be 4.1 ppm [9]. Mg²⁺ concentration could not be determined in before production time by gravimetric method. After production time wide variation

(7.2 - 141.6 ppm) was found in four stations. It was also found that ponds and river water was changed during production time. Control value of that area was 13.2 ppm.

3.2.2. Analysis of Anions

Chlorides: According to surface water standard the Cl^- in fresh water should be 7.8 ppm (Garrels and Mackenzie, 1971) [9]. Before production Cl^- ranges 22.1 - 70.90 ppm and after production time it ranges from 664.69 to 350.690 ppm. Cl^- concentration was much higher than the Control value (44.31 ppm).

Sulphates: According to surface water standard the SO_4^- in fresh water should be 11.2 ppm (Garrels and Mackenzie, 1971) [9] Sulphate concentrations in after production time ranges 1.88 to 5.5. SO_4^- ppm conc. of control sample was null.

Bicarbonates: Before production time conc. of bicarbonate was in range of 175.3 to 259.25 ppm. After production time HCO_3^- could not be detected by gravimetric method. According to surface water standard HCO_3^- in fresh water should be 58.4 ppm (Garrels and Mackenzie, 1971) [9]. Average value (217.25 ppm) of after production time was much higher than the control value (41.18 ppm).

3.3. Variation of Different Parameters with Stations

Samples were collected from different stations such as along discharge path (st-1, 2, 3 & 4) and three from ponds (beside drain) and one from river (30 km away from discharge path). A wide variation was found in different parameters of JSW with stations. Graphical presentation of some important parameters of JSW is shown in **Figure 2**. Pond-1 was different than other two ponds because it was situated between st-1 & st-2 and directly connected to sugarmill drain.

3.4. Comparison of Joypurhat Sugarmill Waste Water with Different Standards of Water

The water quality parameters of JSW were compared with DoE (Department of Environment) standard and inland surface water standard and comparison is shown in **Table 5**. The BOD and EC results are shown in **Figure 3** where comparison of samples from different stations is made with standards. ECR'97 was followed here but this guide had no COD standard value for inland surface and irrigation water. So, COD value could not be shown in this table and JSW value was compared with local irrigation water only. Average temperature of JSW (23.5°C) was between the range of inland surface and irrigation water. In this case temperature is not a problem because it depends on the thermal condition of environment. The main problem is very low or negligible dissolved oxygen (DO) value of waste water. Lower pH of JSW is also a problem. TDS value of JSW was lower than inland surface water value. Though TDS was lower the JSW could not be discharged into open environment because of very low DO, lower pH and higher BOD.

3.5. Comparative Study on Different Sugarmills Waste Water

Physico-chemical properties of seven sugarmills waste water were presented in

Table 6. These seven sugarmills were located in the north-western part of Bangladesh. The waste water quality of sugarmill depends on some issues such as operation procedure of mill, maintenance of machines, open or covered drainage system, underground drainage system etc.

Each sugarmill studied in this project has waste water of very low pH or acidic in nature. Considering BOD it was seen that SESW > SHSW > NBSW > MSW > KASW > THSW > JSW. Among seven, waste water of Shetabgonj sugarmill was

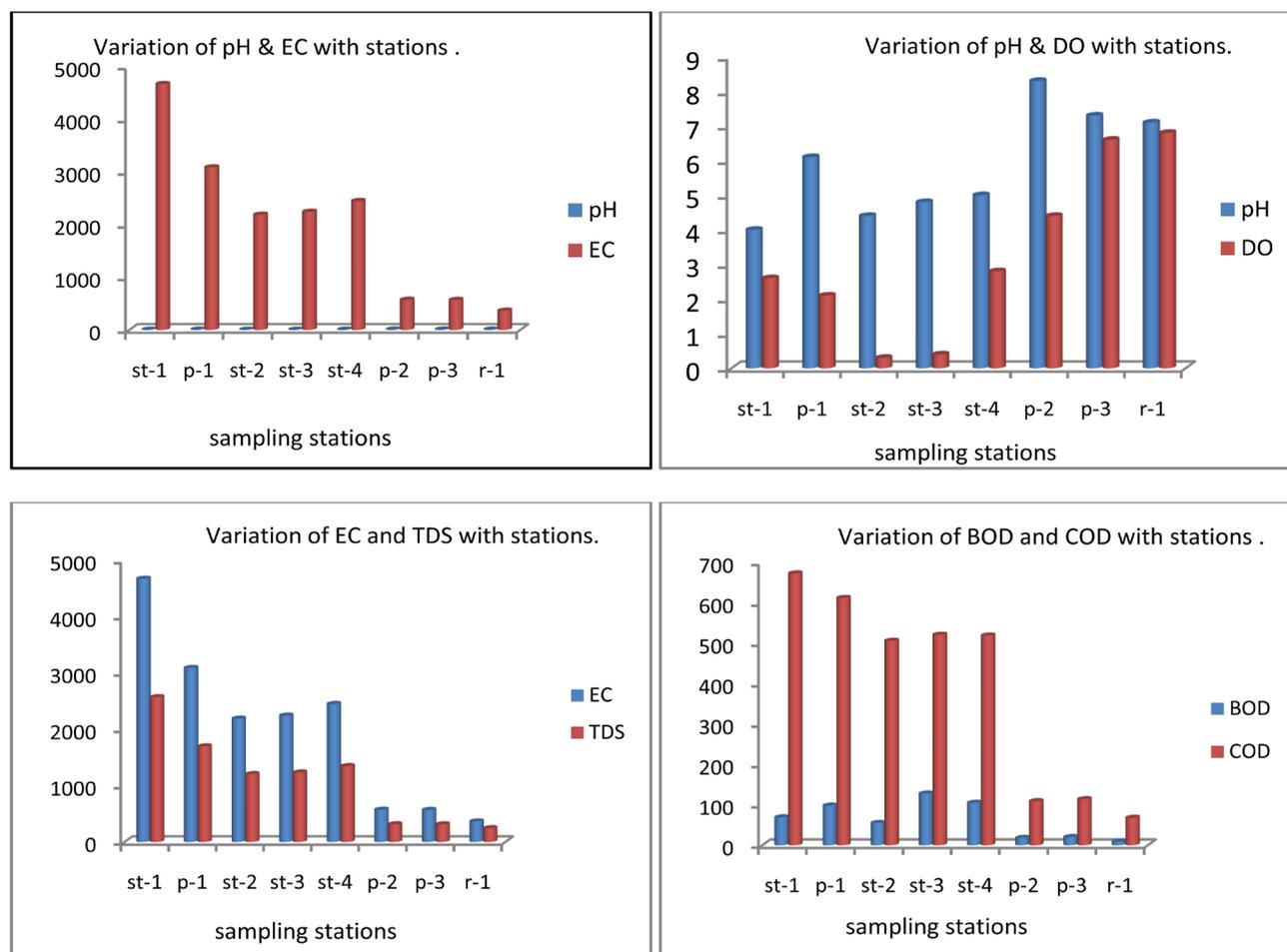


Figure 2. Variation on some parameters of JSW with stations.

Table 5. Comparison of Joypurhat sugarmill waste water with standard values of inland surface water and irrigation water.

Parameter	JSW water (after production)	Inland surface water	Irrigation water	Local irrigation water (control)
Temp (°C)	23 - 24	40	20 - 30	24.1
pH	4.0 - 6.1	6.0 - 9.0	6.5 - 8.5	7.2
DO (mg/l)	0.3 - 2.8	4.5 - 8.0	5 or more	4.7
EC (μ/cm)	2190 - 3090	1200	700 - 3000	193.6
TDS (mg/l)	1205 - 1699	2100	450 - 2000	106
BOD	55 - 128	50 (sugar industry to discharge)	10 or less	8.3
COD	506 - 673			48

(Source: ECR Act 1995 and Rules 1997).

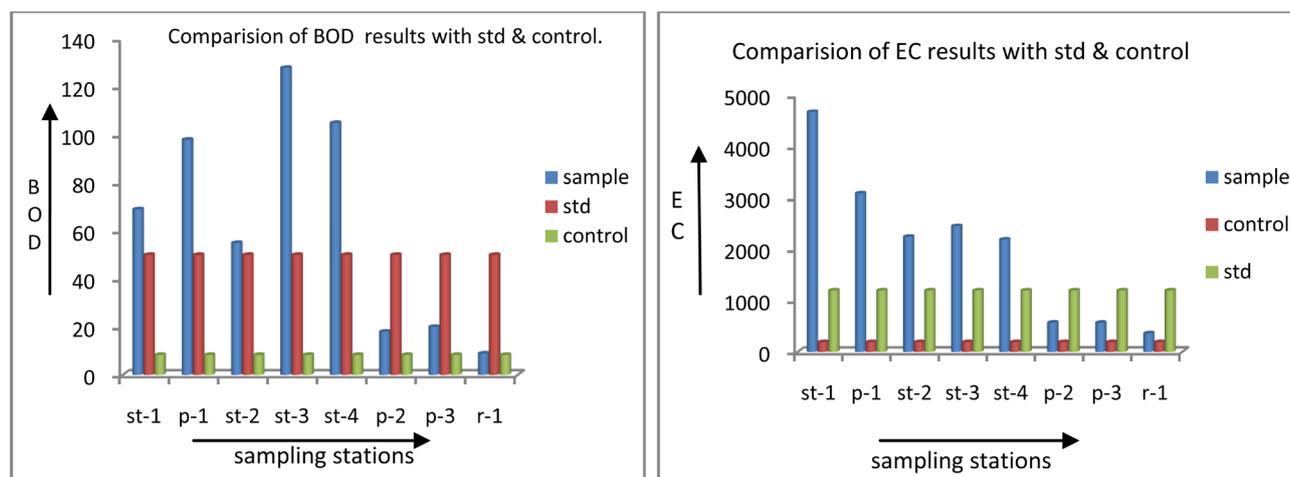


Figure 3. Comparison of JSW with standard and control values.

Table 6. Results of different sugarmills waste water (after production).

Sample id	pH	EC µs/cm	TDS ppm	DO mg/l	BOD mg/l	COD mg/l
JSW	4.6	2890	1590	0.3	89.25	554.75
NBSW	6.7	1072	536	0.00	125	598
MSW	6.6	440	220	0.00	114	560
SHSW	4.4	387	193	0.00	128	498
SESW	4.3	1792	896	0.00	139	603
THSW	4.3	3420	1710	0.00	102	661
KASW	3.9	518	259	0.00	104	98

JSW = Joypurhat sugarmill waste water, NBSW = North Bangle sugarmill waste water, MSW = Mahimagonj sugarmill waste water, SHSW = Shampur sugarmill waste water, SESW = Setabgonj sugarmill waste water, THSW = Thakurgon sugarmill waste water, KASW = Karu & Kong sugarmill waste water.

worser than any other sugarmill in with tremendous odor. DO value of seven sugarmills are nill which is alarming for aquatic life. In case of EC it was found that THSW > JSW > SESW > NBSW > KASW > MSW > SHSW. The waste water of MSW and KASW sugarmills were better than others because these two sugarmill have treatment plant. Six sugarmills of **Table 6** have COD over 400 mg/l except KASW (98 mg/l). The waste water of Karu & Kong sugarmill was better than others.

3.6 Effect of Joypurhat Sugarmill Waste Water on Environment

Though sugarmill performs only 4 - 6 months, it does harm to surrounding environment. During production time, waste water discharges from Joypurhat sugarmill at a rate of 30.28 cubic meters per hour [10]. After closing sugarmill, colour of drain water becomes black. Nearby two ponds which were not directly connected to sugarmill drain that were also changed after production of sugarmill. Excess BOD, COD, low pH and negligible DO value of waste water causes harm to connected river. When the production of joypurhat sugarmill becomes finished then Tulshi Gongga river water turns into black colour. As a result that river (Tulshi Gongga) water becomes unsuitable for aquatic life. Rural people who

use this water for their household work are suffering from skin diseases. People of that area are disturbed by strong fouling odour of sugarmill waste water.

4. Conclusions

1. Due to high BOD, COD, low pH and zero level of DO, Joypurhat sugarmill waste water is not suitable for irrigational use. The waste water quality of this sugarmill is harmful for aquatic life. A suitable treatment method should be developed for this sugarmill.
2. If it is treated, it can be an essential source of irrigation water. Because Joypurhat sugarmill waste water is rich in sodium, potassium, calcium, magnesium which can increase fertility status of soil.
3. Studying on different sugarmill waste water (considering BOD), the order was found as follows: SESW > SHSW > NBSW > MSW > KASW > THSW > JSW. The waste water of Mahimagonj sugarmill and Karu & Kong sugarmill was found better than other sugarmills because they have treatment plant. So, every sugarmill should have treatment plant.

Acknowledgements

We would like to thank Bangladesh Council of Scientific and Industrial Research (BCSIR) authority for giving financial support for this R & D (research and development) project. We also want to thank Director Dr. Nazim Zaman, Institute of Mining, Mineralogy and Metallurgy (IMMM), BCSIR, Joypurhat for his cooperation and inspiration. We are also grateful to the scientists, technicians and lab attendants for their support to complete this project.

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