

Particle Physics Can Be Investigated from a Thermodynamic Point of View

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Abstract

Based on the idea of that particle decay represents nothing but a kind of thermodynamic process due to its spontaneity, we here explore a new kind of heat engine: particle Carnot engine (PCE), which satisfies Carnot's theorem. The result shows that any single particle carries its quantized intrinsic entropy, and the total entropy never decreases for any decay process. Particle thermodynamic laws analogous to the usual ones are proposed, among which the momentum conservation principle is specially introduced that will determine the irreversibility of particle decay. Moreover, we also develop the operational definitions of particle state functions, including Boltzmann relationship, which can be used to discuss the thermodynamic properties of particle objects. Thus, our study can provide a new theoretical framework to investigate particle physics.

Keywords

Particle Decay, Particle Carnot Engine, Particle Thermodynamic Laws

Subject Areas: Particle Physics, Thermodynamics

1. Introduction

Through generalizing in some interdisciplinary areas [1] [2], thermodynamics has successfully led to a series of new concepts, such as quantum heat engine (employing as working agents multi-level systems instead of gas-filled cylinders) [3] [4], information entropy (an equivalence between information and entropy is postulated) [5] and quantum entanglement (in some way analogous to thermodynamic energy) [6] [7]. These concepts imposing the thermodynamic aspects of microscopic objects of interest, inspire us to study particle physics from a thermodynamic standpoint, and make us believe that a single particle could also exhibit its thermodynamic properties since it carries an intrinsic spin, spreads like a wave and acts as a complicated system. In this paper, we will give the basic elements of our analogy between particle physics and thermodynamics. To provide background for the details of the analogy, we propose the particle thermodynamic laws, and further develop the op-

erational definitions of particle state functions to discuss the thermodynamic properties of particle objects.

2. Analogy between Particle Object and Thermodynamic System

To develop the full analogy between particle object and thermodynamic system, we need to recall that particle decay is exactly representing a kind of spontaneous process, just as heat passing from a higher to a lower temperature (the situation is shown as **Table 1**). So, we take notice of the thermodynamic meaning of particle object with moving velocity v, and treat its relativistic energy $E(m, p) = \sqrt{m^2 + p^2}$ (nature units: $c = \hbar = k_B = 1$

adopted) as the enthalpy function

$$h(s,p) = \sqrt{\overline{m}^2 s^2 + p^2}, \quad s = m/\overline{m} \tag{1}$$

of which the differential reads

$$dh(s, p) = \tau ds + v dp, \ \tau = \overline{m}(1 - v^2)$$
⁽²⁾

here \overline{m} denotes the mean value of mass m with a distribution width Γ , τ is the intrinsic temperature of particle object, and s is the intrinsic entropy with a quantized mean value of $\overline{s} = 1$ (here called entron, *i.e.* quantum of entropy). Therefore, following from the above form, there also exist other functions in which one or more variables is replaced by its slope, the functions with the same information content can be obtained by applying the following transform. For example, a function defined by

$$u(s,v) = h(s,p) - pv = \overline{m}s\sqrt{1 - v^2}$$
(3)

it yields the transition from independent variables (s, p) to (s, v), and then gives

$$du(s,v) = \tau ds - p dv = dq - dw$$
⁽⁴⁾

The differential above suggests that u(s,v) may be treated as the internal energy of moving particle with the natural variables *s* (identified with entropy) and *v* (analogous to usual volume), d*q* is the bound or disorder energy (analogous to heat), and d*w* is the useful energy (analogous to work). Importantly, such analogy can help us to find an interesting connection with Carnot theory, and then explore PCE that works in speed space only when decay occurs. In working process, PCE exchanges energy with its surrounding fields (a special kind of heat reservoirs [8] [9]) through field theory mechanism [10] [11].

Now, let us imagine there is a particle X of mass m_X decaying into Y of $m_Y: X \to Y$, of which the X field can be treated as the working substance to be taken around the cycle. The surroundings consist of constant temperature heat reservoirs, one (the field used to excite X) at τ_X and the other (the Y field) at $\tau_Y \left(=\overline{m}_Y \sqrt{1-v_Y^2}\right) < \tau_X$. The X field system and surroundings make up a PCE. It operates reversibly between

the two heat reservoirs, with, in each cycle (called particle Carnot cycle), heat q_X entering at τ_X , q_Y leaving at τ_Y and useful energy w being delivered. The Carnot cycle operation can be schemed by four steps (see Figure 1):

(i) Isothermal expansion of X field from 1 to 2 at $\tau_x = \overline{m}_x$, with heat $q_x = m_x$ absorbed and a particle X excited.

(ii) Emerging of X and subsequent adiabatic evolution from 2 to 3.

(iii) Isothermal compression of X field from 3 to 4 through thermal contact with the entropy sink of Y field at $\tau_{\rm Y} = \overline{m}_{\rm Y} \sqrt{1 - v_{\rm Y}^2}$, releasing $q_{\rm Y} = m_{\rm Y} \sqrt{1 - v_{\rm Y}^2}$ and delivering $w = p_{\rm Y} v_{\rm Y}$. It requires $q_{\rm X} = q_{\rm Y} + p_{\rm Y} v_{\rm Y}$,

i.e.
$$m_{\rm Y} = m_{\rm X} \sqrt{1 - v_{\rm Y}^2}$$
.

(iv) Breaking of thermal contact and continuance of adiabatic compression of X field at vacuum state from 4 to 1.

In the way, PCE is made up, and all PCEs operating between the same two reservoirs have the same efficiency of converting bound energy to useful form, that is

$$\eta_{c} = \frac{q_{\rm X} - q_{\rm Y}}{q_{\rm X}} = 1 - \frac{q_{\rm Y}}{q_{\rm X}} = 1 - \frac{\tau_{\rm Y}}{\tau_{\rm X}}$$
(5)



Figure 1. Temperature-entropy diagram for particle Carnot cycle. In working process of the imagined PCE, the distribution of useful energy is completely determined by Carnot's theorem.

 Table 1. Natural symmetry suggests that a particle object should exhibit thermodynamic behaviors like a complicated system.

Complicated system		Particle object
Ų	⇔	Ų
Angular momentum	⇔	Intrinsic spin
Machinery wave	⇔	de Broglie wave
Thermodynamic behaviors	⇔	Thermodynamic exhibition

here the adiabatic case of $s_X = s_Y$ is used. In turn, when observing on Y, the particle would be at rest, and X in motion with speed v_Y . So that, the temperatures of the two read respectively $\tau'_Y = \overline{m}_Y$, $\tau'_X = \overline{m}_X \sqrt{1 - v_Y^2}$, which give $q'_Y = m_Y = m_X \sqrt{1 - v_Y^2} = q'_X$. It clearly tells us that, because of no useful energy delivered, the efficiency of presented PCE should be zero, namely

$$\eta_c' = \frac{q_x' - q_y'}{q_x'} = 1 - \frac{\tau_y'}{\tau_x'} = 0$$
(6)

Moreover, considering no engine more efficient than PCE, we have $\eta = 1 - q_y/q_x \le \eta_c$, and further

$$\sum_{i} \frac{q_i}{\tau_i} \le 0, \quad i = 3, 4, \cdots$$
(7)

This is referred to as Clausius' inequality, in which, "=" is allowed only for reversible decay process (*i.e.* $\sum_{i} s_i = 0$), and in fact $i \ge 3$ required by the momentum conservation principle (the situation is shown as **Figure 2**). Importantly, inequality (7) can lead directly to the principle of increasing entropy for particle physics: in any decay process, the total entropy of relevant particles increases or remains constant, but cannot decrease. In particular, for π^{\pm} meson decay

$$\pi^{\pm} \rightarrow \mu^{\pm} + \nu_{\mu^{\pm}}$$

$$L \rightarrow e^{\pm} + \nu_{e^{\pm}} + \nu_{\mu^{\mp}}$$
(8)

the entropy increase trend can be shown as

$$\overline{s}_{\pi} \left(=1\right) < \overline{s}_{\mu} + \overline{s}_{\nu_{\mu}} \left(=2\right) < \overline{s}_{e} + \overline{s}_{\nu_{e}} + 2\overline{s}_{\nu_{\mu}} \left(=4\right)$$

$$\tag{9}$$

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Figure 2. Any reversible particle decay process of $X_1 \rightarrow X_2 + X_3 + \dots + X_i$ can be always viewed as a combination of several sub Carnot cycles, which indicates that energies are exchanged between the field bath at high temperature τ_1 and the ones at low temperatures $\tau_2, \tau_3, \dots, \tau_i$.

Thus we say it is the principle that determines the irreversibility of particle decay.

3. Particle Thermodynamic Laws

So far, based on the obtained above, we present the following particle thermodynamic laws.

Oth law: Particle objects have well-defined values of some set of state variables, being of two types: to some extensive variables (such as s, p), there correspond the conjugate intensive variables (such as τ , ν). When two particles unite to form a heavier one (nuclear fusion), they undergo no whole changes if and only if the intensive variables of the interacting sub-objects, have the same values.

1st law: Energy is conserved for all particle processes. The law claims that the particle perpetual mobile of the first kind is absolutely impossible.

2nd law: Exactly as the usual one [12], this law has two equivalent statements:

(i) Particle at lower temperature (corresponding to small mass) cannot decay spontaneously into the one at higher temperature (corresponding to a large mass); while the constraints on the system and the state of the rest of the world are left unchanged (analogous to Clausius').

(ii) It is impossible to find a single particle that produces no effect other than the transformation of its bound energy into an equivalent amount of useful form (analogous to Kelvin's).

In other word, the particle perpetual mobile of the second kind never occurs. The important point is here that, if allow a particle transform its bound energy directly into the useful form, namely $m \rightarrow pv$, an added momentum p is yielded, violating momentum conservation. It is proved that, the deep physics behind entropy increasing must be related to momentum transference, whereas in usual thermodynamics, this is overlooked.

2.5th law: In any particle process, the total momentum cannot increase, nor decrease, but remains constant. As a supplementary item, such law puts restrictions on which processes may occur, or more particularly which processes can never occur even though they are allowed by the 2nd. For example, the decay process of $\mu^{\pm} \rightarrow e^{\pm}$ is forbidden.

3rd law: It is impossible to design a procedure that can control particle temperature to zero, namely, the absolute zero of particle is never attainable. This law disallows any decay process with 100% efficiency, even if for photon (with nonzero mass) [13] [14], no exceptions are made.

4. State Functions

To set up the state functions of a decay particle of mass width Γ , we are allowed to treat it as an open thermodynamic system [15] [16], and substitute complex mass $m' = m - i\Gamma/2$ for m. This complex mass with a module of

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$$|m'| = \sqrt{m^2 + \frac{\Gamma^2}{4}} \approx m + \mu_0, \ \mu_0 = \frac{\Gamma^2}{8m}, \ \Gamma \ll m$$
 (10)

can determine a mass distribution in the following form

$$\rho(m) = \frac{\Gamma}{2\pi \left[\left(m - \overline{m} \right)^2 + \Gamma^2 / 4 \right]}$$
(11)

Combing with the decay function of $\rho_t(t) = e^{-\Gamma t}$, it gives the composite probability distribution of unstable particle, that is

$$\rho_{c}(m,t) = \rho(m)\rho_{t}(t) = \frac{\Gamma e^{-tt}}{2\pi \left[\left(m - \overline{m}\right)^{2} + \Gamma^{2}/4 \right]}$$
(12)

Such the result can help us to write the adjusted enthalpy function in the form of

$$h(s, p, \rho, t) = \sqrt{\left(\overline{m}s + \mu_0 \rho \rho_t(t)\right)^2 + p^2}, \quad s = \rho_c m/\overline{m}$$
(13)

followed by the internal energy

$$u(s,v,\rho,t) = h(s,p,\rho,t) - pv = (\overline{m}s + \mu_0\rho\rho_t(t))\sqrt{1 - v^2}$$
(14)

So that, the total differential of the function reads

$$du(s, v, \rho, t) = \tau ds - p dv + \mu d\rho + \varepsilon dt$$
(15)

with the identifications:

$$\begin{cases} \tau = \left(\frac{\partial u}{\partial s}\right)_{v,\rho,t} = \overline{m}\sqrt{1-v^2} \\ p = -\left(\frac{\partial u}{\partial v}\right)_{s,\rho,t} = \frac{\rho_c |m'| v}{\sqrt{1-v^2}} \\ \mu = \left(\frac{\partial u}{\partial \rho}\right)_{s,v,t} = \mu_0 \rho_t \sqrt{1-v^2} \\ \varepsilon = \left(\frac{\partial u}{\partial t}\right)_{s,v,\rho} = \mu_0 \Gamma \rho_c \sqrt{1-v^2} \end{cases}$$
(16)

Obviously, μ should have the meaning of chemical potential, describing the interaction between particle object and its surroundings [17] [18], ε the decay efficiency of unstable particle, which in the case of $\Gamma \rightarrow 0$ or $\nu \rightarrow 1$, tends to zero, *i.e.* $\varepsilon \rightarrow 0$. This actually tells us that, only the particle stable or moving with unit speed, would have a decay efficiency of zero.

Note that, the Maxwell relations for particle object are a consequence of the fact that the order of the cross derivatives of $u(s, v, \rho, t)$ does not matter, namely

$$\begin{cases} \left(\frac{\partial\tau}{\partial\nu}\right)_{s,\rho,t} = -\left(\frac{\partial p}{\partial s}\right)_{\nu,\rho,t}, \quad \left(\frac{\partial\tau}{\partial\rho}\right)_{s,\nu,t} = \left(\frac{\partial\mu}{\partial s}\right)_{\nu,\rho,t}, \quad -\left(\frac{\partial p}{\partial\rho}\right)_{s,\nu,t} = \left(\frac{\partial\mu}{\partial\nu}\right)_{s,\rho,t} \\ \left(\frac{\partial\tau}{\partial t}\right)_{s,\nu,\rho} = \left(\frac{\partial\varepsilon}{\partial s}\right)_{\nu,\rho,t}, \quad -\left(\frac{\partial p}{\partial t}\right)_{s,\nu,\rho} = \left(\frac{\partial\varepsilon}{\partial\nu}\right)_{s,\rho,t}, \quad \left(\frac{\partial\mu}{\partial t}\right)_{s,\nu,\rho} = \left(\frac{\partial\varepsilon}{\partial\rho}\right)_{s,\nu,t} \end{cases}$$
(17)

accompanied with the chemical potential μ satisfies

$$\left(\frac{\partial\mu}{\partial\tau}\right)_{v,t} = -\frac{s}{\rho}, \ \left(\frac{\partial\mu}{\partial v}\right)_{\tau,t} = -\frac{p}{\rho}$$
(18)

These two are corresponding to the usual forms respectively $(\partial \mu / \partial T)_p = -S/n$, $(\partial \mu / \partial V)_T = -P/n$ in thermodynamics. Figure 3 does make apparent the general trend of increasing chemical potential with increasing mass [19]. This feature can be explained as that, the particle chemical potential is determined by the field interaction with surroundings, the more massive the unstable particle, the higher temperature and the stronger interaction.

If further, apply Legendre transform to $u(s, v, \rho, t)$, we can also get the free energy and enthalpy of particle object, those are

$$f(\tau, \nu, \rho, t) = u(s, \nu, \rho, t) - \tau s = \mu_0 \rho \rho_t(t) \sqrt{1 - \nu^2}$$
(19)

$$\phi(\tau, p, \rho, t) = u(s, \nu, \rho, t) - \tau s + p\nu = \sqrt{\left(\overline{ms} + \mu_0 \rho \rho_t(t)\right)^2 + p^2 - \tau s}$$
(20)

including the variation of $s(u, v, \rho, t)$

$$\delta s(u, v, \rho, t) = \frac{1}{\tau} \delta u + \frac{p}{\tau} \delta v - \frac{\mu}{\tau} \delta \rho - \frac{\varepsilon}{\tau} \delta t$$
(21)

The relationship enables us to explore an equilibrium criterion for an individual particle keeping contact with its surrounding bath *b*. It is that, in equilibrium, the total entropy of the particle object and of the bath should be maximized at fixed temperature, specifically $\delta s = 0$, $\delta^2 s \le 0$, requiring $\tau = \tau_b$, $v = v_b$, $\mu = \mu_b$ and $\varepsilon = \varepsilon_b$. For example, the equilibrium between electron and radiation field can be reached at a temperature about $\sim 10^{11}$ K. Now, we summarize the thermodynamic and particle analogies in Table 2.

Now, consider a particle state with free energy $f(\tau, v, \rho, t)$, and by statistical mechanics [21] we define the partition function



Figure 3. Log-log plot of chemical potential (μ_0/GeV) versus mass (\overline{m}/GeV). Choice of 139 unstable particles from the Monte Carlo file [20] is plotted, the graph can be broken into two parts: particles with $\mu_0 < 10^{-23}$ GeV (hollow triangles) and $\mu_0 > 10^{-16}$ GeV (hollow squares), they are exhibiting approximately identical slope coefficient.

Fable 2. The suggeste	d analogies betwee	n thermodynamic system	(TDS) and	particle object	(PO).
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Variables	Internal energy	Enthalpy	Free energy	Free enthalpy	Temperature
TDS	U	Н	F	Φ	Т
РО	и	h	f	ϕ	τ
Variables	Entropy	Pressure	Volume	Chemical potential	Density
TDS	S	Р	V	μ	n
РО	S	р	ν	μ	ρ

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$$Z = \int e^{f/\tau} d\Omega$$
 (22)

where Ω denotes the occupation number of particle in phase space. Making the usual assumption that the integral is dominated by the most probable states, we obtain $Z \approx e^{f/r} \Omega$, yielding

$$u = \tau \ln Z = \tau \ln \Omega + f \tag{23}$$

Hence, to agree with Eq. (19) identifies

$$s = \ln \Omega, \ \Omega = e^{\rho_c m/\bar{m}}$$
(24)

This is an analogue of Boltzmann relationship for particle physics, which can be made the basis of a theory of fluctuations about particle object in analogy to that of fluctuations about thermodynamic system. Specifically, the fluctuations in s, v and ρ are given by

$$\overline{\left(\delta s\right)^2} = C_v = s, \ \overline{\left(\delta v\right)^2} = pv\kappa_\tau = v, \ \overline{\left(\delta\rho\right)^2} = \rho$$
(25)

where $C_v = \tau (\partial s / \partial \tau)_v$ is the isochoric heat capacity, $\kappa_\tau = (1/v) (\partial v / \partial p)_\tau = 1/p$ the isothermal compressibility. Meanwhile, the relationship can also be used to examine the increasing behavior of particle entropy, and thus the evolution function of total decay entropy of π^{\pm} is given by

$$\overline{s}_{tot}(t) = \overline{s}_{\pi} e^{-\Gamma_{\pi}t} + \left(1 - e^{-\Gamma_{\pi}t}\right) \left[\overline{s}_{\mu} e^{-\Gamma_{\mu}t} + \overline{s}_{\nu_{\mu}} e^{-\Gamma_{\nu_{\mu}}t} + \left(1 - e^{-\Gamma_{\mu}t}\right) \left(\overline{s}_{e} e^{-\Gamma_{e}t} + \overline{s}_{\nu_{e}} e^{-\Gamma_{\nu_{e}}t} + \overline{s}_{\nu_{\mu}} e^{-\Gamma_{\nu_{\mu}}t} \right) \right]$$
(26)

Note the stable particle condition of $\Gamma_e = \Gamma_{v_e} = \Gamma_{v_\mu} = 0$ and $\overline{s_{\pi}} = \overline{s_{\mu}} = \overline{s_{\nu_{\mu}}} = \overline{s_{e}} = \overline{s_{\nu_{e}}} = 1$, we have

$$\overline{s}_{tot}\left(t\right) = e^{-\Gamma_{\pi}t} + 2\left(1 - e^{-\Gamma_{\pi}t}\right)\left(2 - e^{-\Gamma_{\mu}t}\right)\Big|_{t \to \infty} = 4$$
(27)

with a change ratio

$$\dot{\bar{s}}_{tot} = \Gamma_{\pi} e^{-\Gamma_{\pi} t} \left(3 - 2 e^{-\Gamma_{\mu} t} \right) + 2 \Gamma_{\mu} e^{-\Gamma_{\mu} t} \left(1 - e^{-\Gamma_{\pi} t} \right) \ge 0$$
(28)

equal to Γ_{π} as t = 0 (meaning only π^{\pm} decay at the beginning), and "=" allowed only $t \to \infty$. The result shows once again that: the total decay entropy of relevant particles increases or remains constant, but cannot decrease.

Finally, by the definition of particle temperature, we can further get the following relation

$$p\tau = \alpha v \tag{29}$$

with $\alpha \left(=\rho_c \overline{m}^2\right)$ called the probability density of mass, which is analogous to the usual ideal gas equation PV = NT.

5. Summary

Up to now, we have explored a new theory for the description of particle physics. And conceptually different from the conventional standpoint, this theory, by treating particle objects as a kind of thermodynamic system, has led to a number of apparently new results: (i) Particle decay behavior is restricted by Carnot's theorem; (ii) There exist particle thermodynamic laws analogous to the usual ones. Especially, the momentum conservation principle is introduced to supplement the second law; (iii) Some thermodynamic functions, including an equilibrium criterion, are developed to describe particle states; (iv) Boltzmann relationship for particle physics is obtained. We here emphasize any microscopic particle in nature carries its intrinsic entropy with an absolute value near unit, and the total particle entropy never decreases for any decay process. It is important that we find a new route that can be followed to investigate particle physics in the mathematical framework of thermodynamics.

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