

# Bioinspired-Interpenetrating Network (IPNs) Hydrogel (BIOF-INPs) and TMD *in Vitro*: Bioadhesion, Drug Release and Build in Free Radical Detection and Defense

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## Abstract

In this work, Bioactive-functionalized interpenetrating network (IPNs) hydrogel (BIOF-INPs) were prepared and investigated *in vitro* for the free radical detection/defense, therapeutic release as well as shear bond strength to dentine, ability to re-mineralize surface of the dentin after application of these bio-inspired materials using a biologically inspired mineralization process *in vitro* as well as investigating antimicrobial properties of the BIOF-INPs against *S. aureus*. The aim of this investigation was to evaluate the suitability and flexibility of the designer materials to act as an “*in vitro*” probe to gain insights into molecular origin of TMD and associated disorders.

## Keywords

TMD, Functionalized IPNs, Bio-Adhesion, Drug Release, Build in Free Radical Detection and Defense

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## 1. Introduction

In the last few decades, tissue engineering has emerged as a promising multidisciplinary approach for the repair

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and regeneration of damaged bone tissue [1] [2].

The molecular events that underline the degenerative temporomandibular joint diseases are poorly understood [3].

Mechanical stresses are generated during functional or para-functional movements of the jaw; adaptive mechanism of the TMJ may be exceeded by free radical accumulation leading to a dysfunctional state (*i.e.* disease state) [4]. Biologically relevant free radicals are very reactive and unstable molecular entities that have an unpaired electron and they can produce participate in the propagation chain reaction to form a new radical. Although oxygen free radicals participate in many physiological processes, they can be harmful to tissue when either their action or their generation have been left uncontrolled. The most common source of free radicals in biological systems is oxygen. The elevation of reactive oxygen species (ROS) leads to oxidative stress that causes molecular damage to the vital structures and functions [5]-[7].

The unsaturated fatty acids of cell membrane lipid are susceptible to peroxidative reaction. Lipid peroxidation of cell membranes has been implicated in the wide range of tissue injuries and diseases. Accumulation of lipid hydroperoxides in a membrane disrupts its function and causes it to collapse and have range of cytotoxic radicals. The most serious one is byproducts aldehydes. They may also react with transition metals like iron or copper to form stable aldehyde such as malodialdehyde for example, which will damage the cell membrane. Because hemoglobin constitutes the largest iron store in the body, it is speculated to be a potential source of redox activity iron, which can catalyze the formation of free radicals that might be damaging to the joint [8].

Antioxidant defense mechanisms involve both enzymatic and non-enzymatic strategies [9].

The aetiological factors of TMJ disorders are as follows: systemic diseases (rheumatoid arthritis, psoriasis, pseudogout, ankylosing spondylitis, etc.), secondary inflammatory component from the neighbouring regions (otitis, maxillary sinusitis, tonsillitis), trauma (chronical), prevalence of dental arch defects e.g. missing of molar teeth [10], malocclusion, endocrinological disturbances, odontogenic infections (impacted third molars) [11].

Presences of specific bacterial organisms such as *Staph. aureus*, *Strep. mitis*, *M. fermentas*, *Actinobacillus actinomycetemcomitans* in the synovial fluid have been found [12]. Serum antibodies against Chlamydia species in patients with mono arthritis of the TMJ have also been reported [13].

To realize this aim, the preparation of bio-active scaffolds which mimic the complexity of bone structure from the macro- to the molecular scale and which enable an activation of cell adhesion and proliferation under the stimulus of natural bio-adhesion *in vitro* through investigation of dentin bond strength as a model system is done.

Chitosan (CS), obtained through the alkaline deacetylation of chitin, is a copolymer of N-acetyl-D-glucosamine and D-glucosamine [14]. CS stands out by a unique combination of favorable biological properties such as nontoxicity, biocompatibility and biodegradability, along with mucoadhesive, bacteriostatic, and wound-healing properties [15]-[18].

In this work, Bioactive-functionalized interpenetrating network (IPNs) hydrogel (BIOF-INPs) were prepared and investigated *in vitro* for the free radical detection/defense, therapeutic release, ability to re-mineralize surface of the dentin after application of these bio-inspired materials using a biologically inspired mineralization process *in vitro* as well as investigating antimicrobial properties of the BIOF-INPs against *S. aureus*. The aim of this investigation was to evaluate the suitability and flexibility of the designer materials to act as an “*in vitro*” probe to gain insights into molecular origin of TMD and associated disorders.

## 2. Preparation of FeEDTA-BIOF-INPs (BIOF-INPs)

The novel prepared by dispersion of therapeutic agent such as naproxen, ibuprofen or aspirin (powder 0.4 gm) in glycerol (5% w/w) using a mortar and a pestle. FeEDTA (5% w/w) was added to the mixture and mortar and pestle used to achieve the uniform mixture. Ten milliliters of glacial acetic acid (1% w/w) was then added with continuous mixing and finally chitosan polymer was spread on the surface of the dispersion and mixed well to form the required gel and left for 24 hours. The strength of the prepared gel (10 gm) is 0.4 g therapeutic agent in each gram of the base. The summary of the newly prepared materials was presented in **Table 1**.

### 2.1. *In Vitro* Study of Therapeutic Agents Release Profile

The *in vitro* release of the therapeutic agents were evaluated using previously reported protocol [19]-[25]. Samples (2 ml) were collected at a regular time intervals and were analyzed spectrophotometrically by U.V. Spectrophotometer (Cintra 5, GBC Scientific equipment, Australia) at  $\lambda_{\max}$  275 nm,  $\lambda_{\max}$  (ibuprofen) 264 nm and  $\lambda_{\max}$

**Table 1.** Gel formulation prepared in the study.

| Gel formulation     |       | Chitosan: FeEDTA<br>VitC 5:1 (w/w%) | Aspirin<br>(w/w%) | Ibuprofen<br>(w/w%) | Naproxen<br>(w/w%) | Tetracycline<br>(w/w%) | pH   |
|---------------------|-------|-------------------------------------|-------------------|---------------------|--------------------|------------------------|------|
| Chitosan-VitC (5:1) | Gel-1 | 5                                   | 0                 | 0                   | 0                  | 0                      | 6.84 |
| Chitosan-VitC-A1    | Gel-2 | 5                                   | 1                 | 0                   | 0                  | 0                      | 6.92 |
| Chitosan-VitC-N1    | Gel-3 | 5                                   | 0                 | 0                   | 1                  | 0                      | 6.75 |
| Chitosan-VitC-II    | Gel-4 | 5                                   | 0                 | 1                   | 0                  | 0                      | 6.56 |
| Chitosan-VitC-T1    | Gel-5 | 5                                   | 0                 | 0                   | 0                  | 1                      | 6.26 |

VitC is vitamin C additive, FeEDTA as  $\text{Fe}^{2+}/\text{Fe}^{3+}$  additive, A is aspirin additive, N is the naproxen, I-ibuprofen. Hydrogels containing chitosan:Vitamin C (5:1) (5%) are synthesized and characterized.

(naproxen) 260 nm [19]-[25]. Each sample was replaced by the same volume of phosphate buffer pH 6.8 to maintain its constant volume and sink condition [19]-[25].

## 2.2. Bioadhesive Investigation

Bioadhesion studies were done using Chatillon apparatus for force measurement using previously reported protocol that has been published in detail [19]-[25]. This method determines the maximum force and work needed to separate two surfaces in intimate contact [19]-[25]. The strength was recorded as a function of the displacement, which allowed to determine the maximal detachment force,  $F_{\max}$ , and the work of adhesion,  $W$ , which was calculated from the area under the strength-displacement curve.

## 2.3. Microbiological Investigations

A type strain of *Staphylococcus aureus* (ATCC 12600), obtained from the American Type Culture Collection (Manassas, USA) was used as test bacterium for estimating the antibacterial activity of the hydrogels. The antibacterial activity of the prepared tetracycline/antioxidant chitosan hydrogels were tested using the standard Kirby-Bauer agar disc diffusion method (Bauer *et al.*). Five to 6 mm deep Muller-Hinton agar (Oxoid, Basingstoke, UK) plates were inoculated by streaking a standardized inoculum suspension that match a 0.5 McFarland standard and containing  $10^7$  -  $10^8$  colony forming units/ml with a throat cotton swab. For each test sample 500  $\mu\text{g}$  of hydrogel was applied to a 6 mm diameter paper disc. The paper discs were placed on the inoculated Muller-Hinton agar medium and incubated at 37°C for 24 hours. The diameter of the zones of growth inhibition was measured with a caliper. Each measurement was done in triplicate and the testing of each sample was repeated 3 times. The antibacterial efficacy of the prepared gels were compared to antibiotic sensitivity discs (Mast Laboratories, Merseyside UL) containing 30  $\mu\text{g}$  of tetracycline per disc.

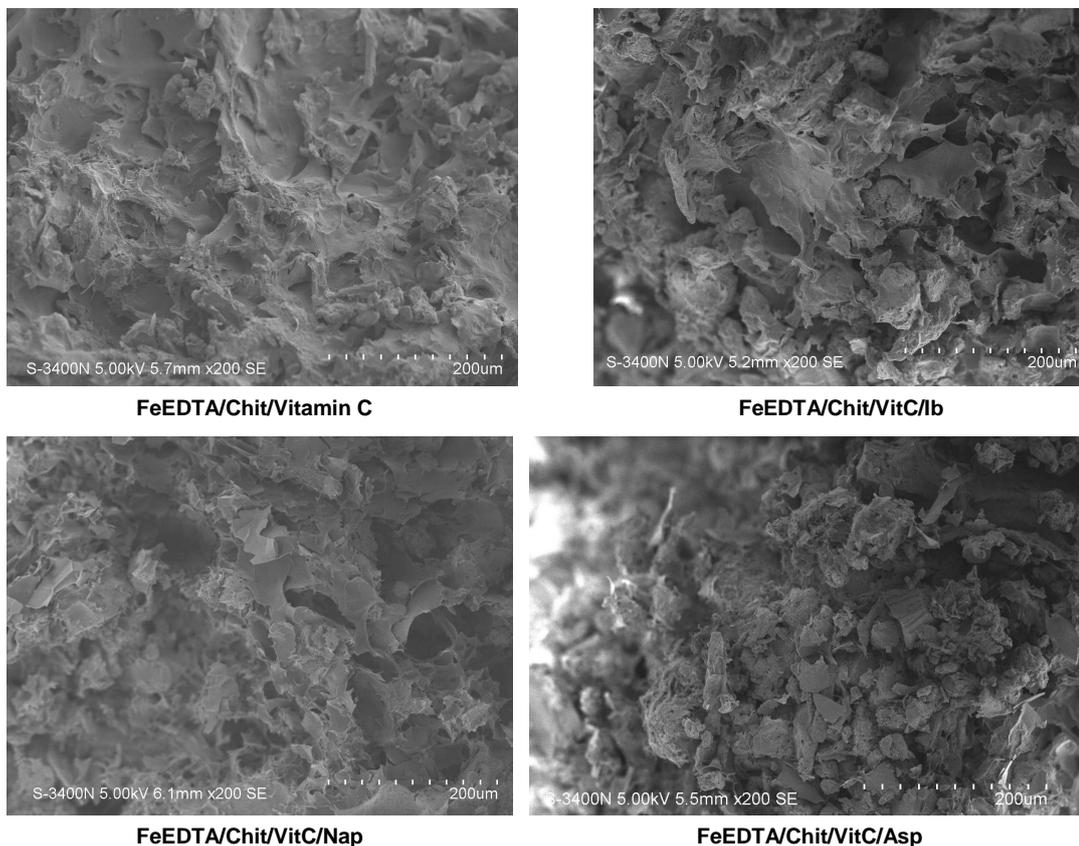
## 3. Results and Discussion

### 3.1. The Characterization of FeEDTABIOF-INPs (Gel 1-4)

The SEM images were obtained to characterize the microstructure of the freeze-dried naproxen composite gels and are presented in **Figure 1**. It could be seen that the hydrogels have a channel-like surface with an extremely porous structure. While the hydrogels have a similar surface appearance, the distribution and the size of the pores are different. The detailed investigation into the correlation of the pore sizes versus the concentration of the cross-linking moieties incorporated into the gel are currently under investigation in our laboratory and will be reported in the due course. The “skin” of the gels can be seen, and the collapse of the surface pores may be due to freeze-drying process.

### 3.2. Bio-adhesion *in Vitro* Model

Higher adhesiveness of the gels is desired to maintain an intimate contact with skin or tooth structure and results are summarized in **Table 2**. Chitosan hydrogels showed the highest adhesive force and the work of adhesion this can be expected because of the well known intrinsic bio-adhesive properties of chitosan [26] [27]. The adequate



**Figure 1.** SEM photographs of interior morphology of the selected gels under investigation for (a) Gel-1, (b) Gel-2, (c) Gel-3, and (d) Gel-4.

**Table 2.** Bioadhesion testing *in vitro*.

| Hydrogel | Adhesive Force (N) ± SD (Dentin) | Work of Adhesion (Ncm) ± SD (Dentin) |
|----------|----------------------------------|--------------------------------------|
| Gel-1    | 1.91 ± 0.35                      | 7.92 ± 0.34                          |
| Gel-2    | 1.79 ± 0.44                      | 8.49 ± 0.42                          |
| Gel-3    | 1.89 ± 0.60                      | 6.94 ± 0.29                          |
| Gel-4    | 1.97 ± 0.35                      | 7.35 ± 0.34                          |

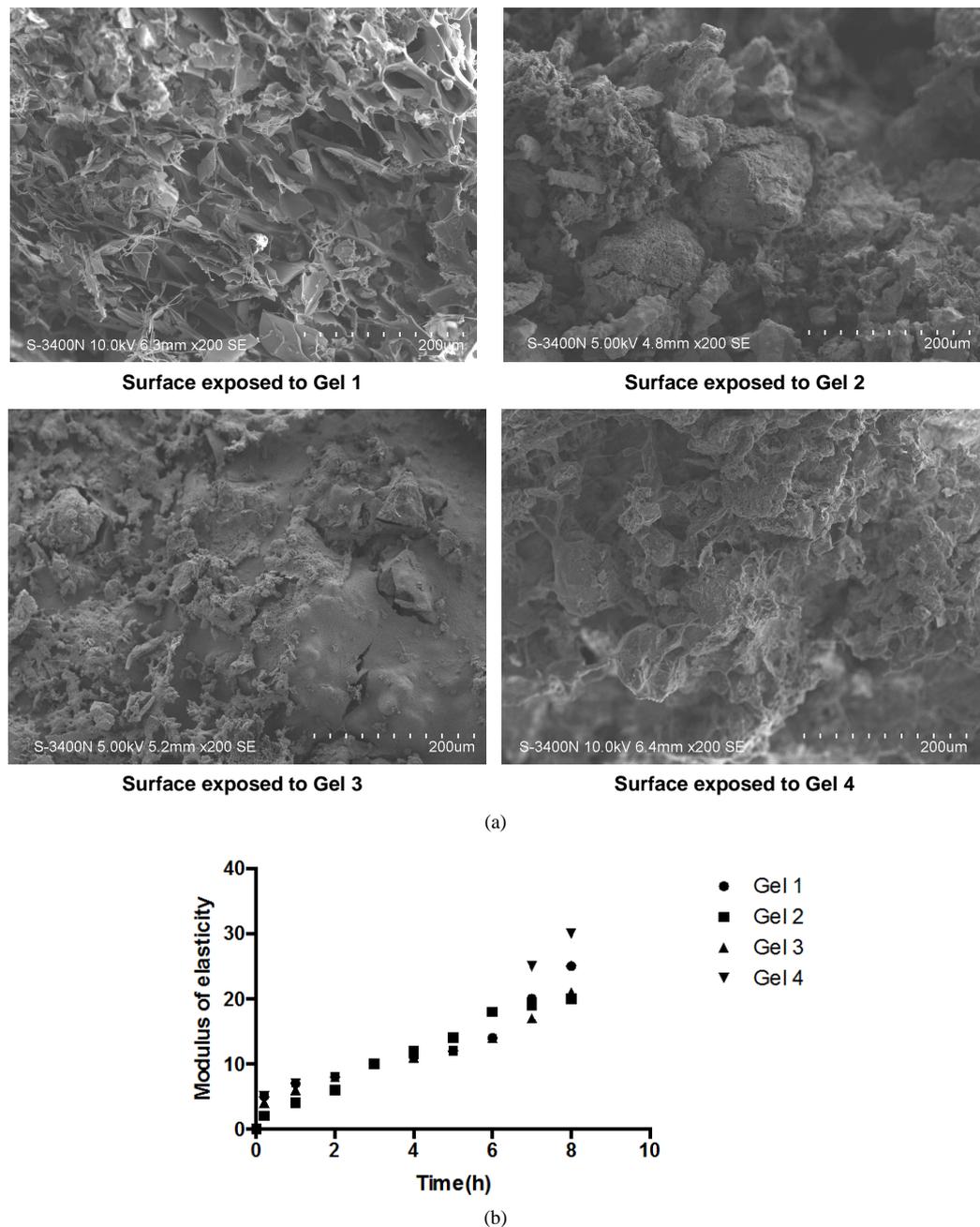
The presented values are an average (n = 5).

water absorption capacity together with the cationic nature of the chitosan containing IPN promotes binding to the surface of dentin structure. Further investigations are currently on the way in our laboratory to establish the correlation between adhesive force and the skin surfaces and results will be reported elsewhere.

The correlation between the force and work of adhesion is noticeable for all. Further experiments are to be conducted on the skin samples to evaluate the bio-adhesive capacity of the designer hydrogels.

Interestingly the surface of the dentin surface has been significantly influenced by the treatment with Gels 1-4, confirming previously reported unique properties of the chitosan to promote formation of the hydroxyapatite crystals as well as prevent demineralization of enamel and dentin which has been reported previously [28] and therefore supports the initial hypothesis of the model system being suitable to investigate molecular mechanism of TMD *in vitro* using dentin as a prototype of fibrous articular disc model (Figure 2).

Once dentine, which represents our model materials, is subjected to the biochemically induced oxidative free radical stress, demineralisation of the dentine matrix initiates mineral loss from the extra-fibrillar as well as intra-fibrillar compartments of type I collagen fibrils [29] [30]. The loss of minerals from the intra-fibrillar com-



**Figure 2.** (a) SEM images of the dentin surface exposed to **BIOF-INPs** 1-4 for 3 weeks; (b) Modulus of elasticity and reaction to the hydrogels.

partments results in lower mechanical properties of dentine [30]. With the dissolution of minerals, the exposed collagen is subjected to various exogenous as well as host-derived collagenolytic enzymes, resulting in enzyme-mediated collagen breakdown [31]. The regeneration treatment of affected dentine is challenging due to the loss of the integrity of the dentine collagen matrix [32] [33]. Hence, modifying the dentine matrix by strengthening the collagen fibrillar matrix appears to be a rational approach for withstanding mechanical forces, resisting degradation by proteases as well to reducing collagen-bound proteoglycans [24] [34]. Cross-linking of the collagen matrix provides an effective solution for preserving and repairing the de-mineralised dentine structure. This study has shown that newly developed hydrogels are capable to improve the modulus of elasticity of de-mineralised dentine, the detailed investigation of the potential mechanism is currently on the way in our laboratory.

### 3.3. *In Vitro* Release of Therapeutic Agents from FeEDTA Bioactive-Functionalized Interpenetrating Network (IPNs) Hydrogel (BIOF-INPs)

The *in vitro* release of therapeutic agents from Fe<sup>2+</sup> (BIOF-INPs) was carried out using USP dissolution apparatus type I. As the regression analysis of the obtained results for two kinetic models including zero order and Higushi's model showed that Higushi's model gave the highest value of  $r^2$  with significant difference ( $p < 0.05$ ) [24]. Higushi's model, where the cumulative amount of the released drug per unit area is proportional to the square root of time, is more suitable model to describe the release kinetics of from the gel preparations examined in the present study. The release of therapeutic agents from the FeEDTA-BIOF-INPs was studied and gels containing corresponding antioxidant, as shown in **Figure 3**.

The controlled release of the therapeutic agents under investigation depends on pH value of the media, with slower release kinetics at the higher pH conditions, which was attributed to amino groups of chitosan protonation, resulting in a soluble and positively charged polysaccharide leading to faster swelling in the acidic media. The plot reveals that there are 2 stages for the release of the therapeutic agents in all Gels. The first stage of release was initially rapid (burst release), which maybe result from the rapid diffusion of therapeutic agents form the FeEDTABIOF-INPs and the initial swelling. The second stage of release of the all investigated therapeutic agents demonstrates a slow, controlled release. The burst release has an advantage in reaching a biologically relevant concentration much faster, where as slow release will control the sustainable concentration of therapeutic agents such as naproxen, ibuprofen or aspirin in plasma, for example for a prolonged period of time.

### 3.4. Free Radical Defense Capability of the Prepared Hydrogels

The amount of uncontrolled ROS is the main cause of the inability of healing process to continue and therefore it would be ideal to utilize the antioxidant capacity of the “designer hydrogels” to detect and able to “fight the free radical excess” have been assessed using previously described model using well-established that HO radical can be generated from a reaction known as the biologic Fenton reaction and this reaction requires the presence of H<sub>2</sub>O<sub>2</sub>.

Bovine serum albumin (BSA), a completely water-soluble protein, was polymerized by hydroxyl radicals generated by the Fenton reaction system of Fe<sup>2+</sup>/EDTA/ H<sub>2</sub>O<sub>2</sub>/ascorbate [19]-[25]. As a result, the protein loses its water-solubility and the polymerized product precipitates. The decrease in the concentration of the water-soluble protein can easily be detected.

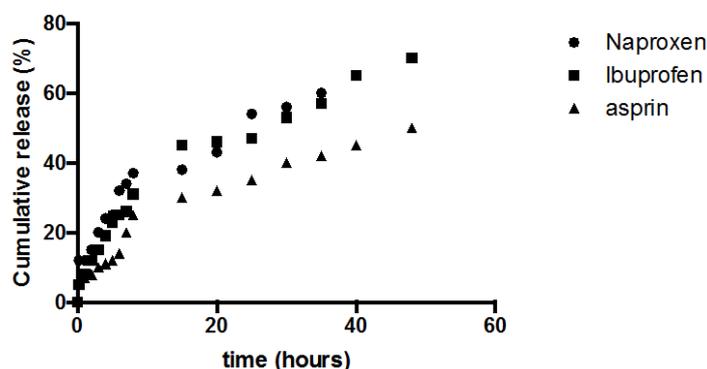
Therefore we adopted the method for recording changes in water solubility of the model protein bovine serum albumin (BSA) exposed to free radicals generated by an inorganic chemical system. As clearly demonstrated by the **Figure 4**, upon exposure to standard H<sub>2</sub>O<sub>2</sub> in the form of FeEDTA BIO-INPs (build in free radical material defense material) and H<sub>2</sub>O<sub>2</sub> as a base line determinate free radical generation under “prototype *in-vitro* free radical damage”, upon incorporation of the chitosan substituted hydrogels, the build in antioxidant capacity and therefore free radical defense of the *in-vitro* model has been activated and are of significant value to take notice. This model represents the practical approach of *in-situ* monitoring and test the amount of free radical production and synergistic antioxidant defense of the system. Further investigations and fine-tuning of the system are currently on the way in our laboratory.

The in-depth investigation of interaction between free radical “*in-situ*” production and functional chitosan biomaterials are currently on the way in our laboratory and will be reported in due course.

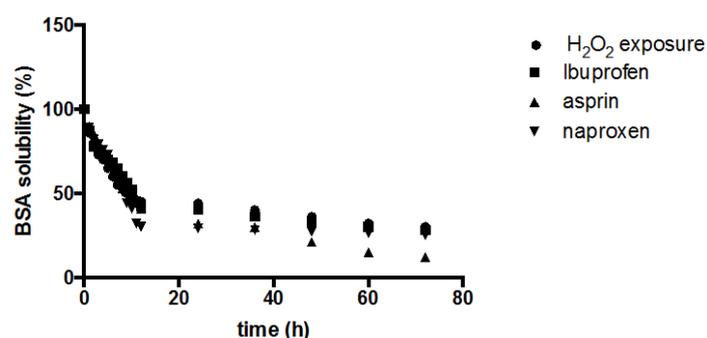
### 3.5. Insight into Microbiological Investigations

Paper discs impregnated with the chitosan hydrogels without tetracycline gave no inhibition zones. However all the test samples containing tetracycline give inhibition zones larger than the clinical breakpoint inhibition zone diameters (European Committee on Antimicrobial Susceptibility Testing, Basel, Switzerland) for *S. aureus* sensitivity for tetracycline. Using the students T test, no statistically significant difference ( $p < 0.05$ ) between the averages of the inhibition zone diameters for all the samples were found (**Table 3**).

In the present study, tetracycline was selected because of its wide application, both locally and systemically, in TMJ potential treatment. A commercially available standard bacterial strain was selected to investigate whether the preliminary goal of preserving the activity of antibiotics, after incorporation into preparations, can be obtained.



**Figure 3.** Cumulative release (%) of naproxen, ibuprofen and aspirin from 5% w/w FeEDTA-BIOF-INPs in phosphate buffer pH 6.8.



**Figure 4.** Plot of changes in water solubility of the model protein bovine serum albumin (BSA) over a time (72 hours) exposed to free radicals detected by a FeEDTABIOF-INPs and produced  $H_2O_2$ /ascorbate system as a source of free radicals.

**Table 3.** (Tetracycline) tested for antibacterial activity against *Staphylococcus aureus* NCTC 12600.

| Sample no. (n = 10)              | Tetracycline inhibition zone diameters |         |                |                  |                     |                  |
|----------------------------------|--|---------|----------------|------------------|---------------------|------------------|
|                                  | Tetracycline disc (30 $\mu$ g)         | Ch/VitC | Ch/VitC/FeEDTA | Ch/VitC/FeEDTA/T | Ch/VitC/FeEDTA/T/HA | Ch/VitC/T/HA     |
| Average $\pm$ standard deviation | 26.44 $\pm$ 0.28                       | 0       | 0              | 23.67 $\pm$ 0.55 | 20.09 $\pm$ 1.63    | 22.80 $\pm$ 0.64 |

Clinical breakpoints for tetracycline: 30  $\mu$ g/disc > 22 mm is sensitive and <19 mm is resistant.

Our results showed that the cross-linked chitosan sponges were able to deliver active antibiotic for up to 5 days. Chitosan hydrogel scaffolds were designed in this study as carriers for antibiotics and showed a steady release of the medication. Three-dimensional chitosan matrices have been shown to be excellent tissue engineering scaffolds for cell attachment and growth. Chitosan has a scalloped structure and has been used in tissue engineering to culture hepatocytes, fibroblasts and cartilage cells because of its ability to promote cell attachment and growth [19]-[25].

In our investigation, chitosan was selected as the carrier for tetracycline, mainly because it can both carry and deliver the medication, but also because it has other useful bioactivities such as antioxidant and anti-inflammatory properties [19]-[25].

#### 4. Conclusion

We developed and tested functional dual-action restorative materials based on FeEDTA-BIOF-INPs “build in free radical defense and detector”, chitosan (antioxidant containing material as well as hydroxyapatite formation stimulant), naproxen, ibuprofen (non-steroidal anti-inflammatory medication), aspirin (pain relief medication

and free radical scavengers), tetracycline (model antibiotic compound) and combinations thereof. We quantified the effects of functional designer biomaterials on the dentin bond strength of a composite and evaluate the capability of newly designed hydrogels to play an integral role of “build in free radical defense mechanism by using BSA solubility as a “molecular prototype of the site of free radical attack *in vitro*”. The newly developed hydrogels also increase modulus of elasticity of the demineralized dentin. Therefore, the model functional multi-dimensional restorative repair materials (**BIOF-INP**) with the build in free radical defense mechanism act as a “proof of concept” for the potential molecular probe towards understanding TMD mechanism *in vitro* and potential development of the efficient treatment *in vivo* in the future.

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