

The Bowing Parameters of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ Ternary Alloys

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Abstract

On the basis of first principles calculations using density functional theory, we explore the structural and electronic properties of two binaries: CaO and MgO in rock salt structures. Structural properties of the semiconductor $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys are derived from total-energy minimization within the General Gradient Approximation. The band gap bowing parameters dependence is very powerful Calcium composition. The results offer that an average bowing parameter of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys is $b = \sim 0.583 \text{ eV}$. We analyzed the volume deformation, charge transfer and structural relaxation effects of the $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys.

Keywords

Density Functional Theory, Ternary Alloys, Band-Gap Bowing Parameter

1. Introduction

Recently, the binary compounds such as TIAs, AlAs, SeN, GaN and their mixtures such as TlAlAs, ScGaN have been studied theoretically [1]-[3], because the wide range of the band gap is important for microelectronics devices. Generally, monoxide compounds are known as rock salt structures (B1) at room temperature and under pressure. Y. Duan *et al.* [4] have been handled the electronic properties of XO (X = Be, Mg, Ca, Sr, Ba, Zn and Cd) for wurtzite, zincblende and rock salt structure. Ponce *et al.* [5] have analyzed theoretically and experimentally electronic and optic properties of CaS nad CaO. Albuquerque and Vasconcelos [6] are reported structural, electronic and optical properties of CaO. Karki *et al.* [7] [8] have inversigated structural, dynamic and electronic properties of liquid MgO via density functional theory. Makaremi and Nourbakhsh have declerated structural,

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electronic and magnetic properties of MgO nanolayers. Nishi *et al.* [9] have investigated experimentally metastable $\text{Mg}_{1-x}\text{Ca}_x\text{O}$ solid solutions film ZnO layers. Stolbov and Cohen [10] have studied the electronic structure for equilibrium MgO-CaO. Miloua *et al.* [11] have calculated the electronic properties of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$, theoretically. Besides, A. Srivastava *et al.* [12] calculated phase translations in $\text{Mg}_{1-x}\text{Ca}_x\text{O}$ alloys.

In this paper, we represent the bowing parameter of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys by first principles density functional theory. To the best of our knowledge, no theoretical as well as experimental work has been performed thus far for the bowing parameter of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$. The aim of this paper is to understand the attitude of the bowing parameters and contribution to the gap bowing parameters. The paper is methodized as follows: computational methodology is given in Section 2. Results and discussion are represented in Section 3. The study is concluded in Section 4.

2. Computational Details

The calculations for CaO, MgO and $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ in the rock salt structure were investigated within the generalized gradient approximation (GGA) of density functional theory (DFT) using the PWSCF code [13]. In Quantum Espresso, the examining is performed by utilizing the Kohn-Sham [14] formation established on the DFT. Total energies have been calculated by using ultrasoft pseudopotentials and plane-wave basis sets. The exchange-correlation potentials in the GGA [15] is separately used in the calculations. The electronic configurations used for the pseudo potentials were Ca($3p^64s$), Mg($2p^63s$) and O($2s^22p^4$). The Kohn-Sham [14] orbitals were described using a plane wave basis set. The highest kinetic energy of a plane wave in the chosen fundamental set is known the cutoff energy. Specialize assignation of the cutoff energy is important for achieving accurate results with available computational process. The values of cutoff energies used in our calculations are summarized in Table 1.

The plane wave energy cut off is selected 90 Ry. Accurate Brillouin zone investigations are carried out using the standard special k-points technique of Monkhorst and Pack [16]. The Brillouin zone investigation was performed over a $12 \times 12 \times 12$ mesh points. Our calculations involve an 16 atom for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys in a supercell. We start at MgO cluster and finish at CaO cluster.

3. Results and Discussion

3.1. Structural Properties of Binary Compounds

The ternary compounds $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ are bordered by two binary compounds of CaO and MgO. In order to be able to analyze the energy band gaps and bowing parameters of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys, it is wholesome to study the CaO and MgO binary compounds in terms of their structural and electronic properties. By lessening the total energy with regards to the atomic positions and lattice parameters we carried out the structural optimization. Equilibrium lattice parameters are obtained by fitting the total energy with the different volumes according to the Birch equation of states.

The Birch equation of states [17] can be seen in the Equation (1):

$$E = E_0 + \frac{9}{8} B_0 V_0 \left(\left(\frac{V_0}{V} \right)^{2/3} - 1 \right)^2 + \frac{9}{16} B_0 V_0 (B'_0 - 4) \left(\left(\frac{V_0}{V} \right)^{2/3} - 1 \right)^3 \quad (1)$$

where B_0 and B'_0 are the bulk modulus and its pressure derivative at the equilibrium volume V_0 . The calcu-

Table 1. Ground state energies for equilibrium MgO and CaO with various cutoff energies.

Cutoff Energy (RYD)	Ground State Energy, MgO	Ground State Energy, CaO
40	-111.699	-107.228
60	-111.705	-107.234
90	-111.707	-107.237
100	-111.707	-107.237
150	-111.707	-107.237

lated values for the equilibrium of CaO and MgO are 4.805 Å and 4.263 Å, respectively. We represented and compared the equilibrium lattice parameter, bulk modulus B and bulk modulus derivation B' in **Table 2**.

3.2. Structural Properties of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$

In this paper, we examined the effectiveness of the Vegard's law for rocksalt $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys in its ground states. Ca composition x dependent lattice constant of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ compounds is expressed as:

$$a(x) = xa_{\text{CaO}} + (1-x)a_{\text{MgO}} \quad (2)$$

where $a(x)$, a_{CaO} and a_{MgO} are the lattice constants of the $\text{Ca}_x\text{Mg}_{1-x}\text{O}$, CaO and MgO, respectively. The volume of the deflection from Vegard's law can be calculated by Equation (2). The numerical calculations are fulfilled for the different situations. Seven compositions of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys were checked: 0.125, 0.25, 0.375, 0.5, 0.625, 0.75 and 0.875. The calculated results of lattice parameters with respect to x compositions are itemized in **Table 3** from which we find that the lattice parameter increases with increasing Calcium composition because these results are due to the atomic radius. The atomic radius of Calcium and Magnesium are 194 and 145 picometers, respectively. We represented and compared the equilibrium lattice parameter of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys in **Table 3**.

The results also propose that the composition-dependent lattice parameter of the $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys can be represented by a third-order polynomial equation, $a(x) = 0.80725x^3 - 1.38387x^2 + 2.73844x + 16.09511$ (Å).

The bowing parameter, b , is significant for investigating the band gap energy ternary alloys. The band gap energy of ternary alloys describe by the band gap energy of binary compounds, a quadratic interpolation of

Table 2. Itemized lattice parameter a , bulk modulus B and bulk modulus derivation B' for the binary compounds MgO and CaO in rocksalt structure at equilibrium volume.

	a (Å)	B (GPa)	B'
MgO (Present work)	4.263	148.39	3.760
MgO Ref. [18]	4.15	160.0	4.15
MgO Ref. [12]	4.165	176.01	3.25
MgO Ref. [19]	4.25	159.7	4.26
CaO (Present work)	4.805	148.014	1.824
CaO Ref. [20]	4.81	110	4.26
CaO Ref. [12]	4.953	106.47	3.62
CaO Ref. [11]	4.71	127.0	4.11

Table 3. Itemized lattice parameter a for the ternary alloys $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ in rocksalt structure at equilibrium volume. All values are Å.

	Present Work	Ref. [12]	Ref. [12]	Ref. [11]
$\text{Ca}_{0.125}\text{Mg}_{0.875}\text{O}$	4.3381	-	-	-
$\text{Ca}_{0.25}\text{Mg}_{0.75}\text{O}$	4.4224	4.48	4.58	4.35
$\text{Ca}_{0.375}\text{Mg}_{0.625}\text{O}$	4.4912	-	-	-
$\text{Ca}_{0.5}\text{Mg}_{0.5}\text{O}$	4.5455	4.65	4.7	4.49
$\text{Ca}_{0.625}\text{Mg}_{0.375}\text{O}$	4.6220	-	-	-
$\text{Ca}_{0.75}\text{Mg}_{0.25}\text{O}$	4.6881	4.812	4.82	4.61
$\text{Ca}_{0.875}\text{Mg}_{0.125}\text{O}$	4.7519	-	-	-

composition amount x and the bowing parameter.

The band gap energy of ternary alloys $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ given by

$$E_g(x) = xE_{g,\text{CaO}} + (1-x)E_{g,\text{MgO}} - bx(1-x). \quad (3)$$

Here, $E_g(x)$ is the band gap energy of the ternary $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ compound, $E_{g,\text{CaO}}$ is the band gap energy of the CaO compound, $E_{g,\text{MgO}}$ is the band gap energy of the MgO compound and b is the band gap bowing parameter of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$. Bowing parameter is associated with the band gap energy. Figure 1 shows bowing parameter as a function of x for ternary alloys. We calculated nearly linear variation for b different composition, x , determining powerful. Our calculations show that according to the band gap energy of ternary alloys, bowing parameter decreases. The results are given by Table 4 and our results show that an average bowing parameter of $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ is ~ 0.583 eV.

The combination-dependent bowing parameter function [21] $b(x)$ was described as

$$b(x) = \frac{x E_{g,\text{CaO}} + (1-x) E_{g,\text{MgO}} - E_g(x)}{x(1-x)} \quad (4)$$

The band gap of the ternary alloys are correlated with the band gaps of the binary compounds. The band gap bowing parameters b have three physically contributions [17].

Finally, the total band gap bowing parameter can be written by resolving into its components as:

$$b = b_{\text{VD}} + b_{\text{CE}} + b_{\text{SR}}. \quad (5)$$

The effect of volume deformation causes the first term of bowing parameter, b_{VD} , in Equation (5). The no-

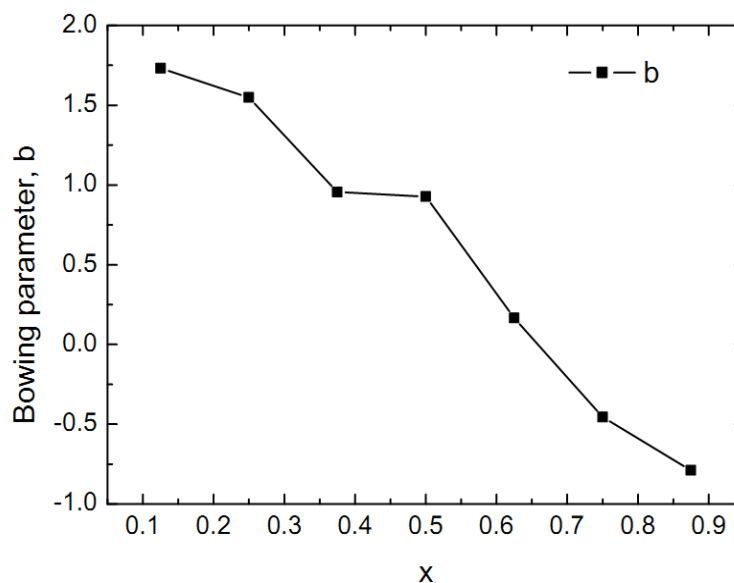


Figure 1. Composition dependence of the bowing parameter for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys.

Table 4. Itemized bowing parameter and contribution of the bowing parameter a for the ternary alloys $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ in rocksalt structure at equilibrium volume.

x	12.5 %	25 %	37.5 %	50 %	62.5 %	75 %	87.5 %
b (eV)	1.731	1.548	0.956	0.928	0.165	-0.455	-0.790
b_{VD}	4.234	3.957	3.741	3.412	3.272	3.083	2.900
b_{CE}	-2.504	-2.410	-2.784	-4.340	-3.436	-3.538	-3.690
b_{SR}	0.001	0.001	0.001	1.856	0.329	0.000	0.000

tional response of MgO(CaO) to hydrostatical pressure states this term via the effect of the contribution of balanced lattice constant $a_{\text{MgO}}(a_{\text{CaO}})$ to the lattice constant of the alloy value $a(x)$.

$$b_{\text{VD}} = \frac{E_{\text{CaO}}(a_{\text{CaO}}) - E_{\text{CaO}}(a)}{1-x} - \frac{E_{\text{MgO}}(a_{\text{MgO}}) - E_{\text{MgO}}(a)}{x} \quad (6)$$

A charge transfer in MgO and CaO at $a = a(x)$ indicates the second term, b_{CE} , of the total band gap bowing parameter,

$$b_{\text{CE}} = \frac{E_{\text{CaO}}(a)}{1-x} - \frac{E_{\text{MgO}}(a)}{x} - \frac{E_{\text{CaMgO}}(a)}{x(1-x)} \quad (7)$$

The third term, b_{SR} refers to the structural relaxation which takes place during the passing from the unrelaxed to the relaxed alloy

$$b_{\text{SR}} = \frac{E_{\text{CaMgO}}(a) - E_{\text{CaMgO}}(a_{\text{eq}})}{x(1-x)} \quad (8)$$

In **Table 4**, the values of bowing parameter, b and its volume deformation component b_{VD} , charge exchange component b_{CE} and structural relaxation component b_{SR} are expressed for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys.

The results also propose that the combination-dependent bowing parameter of the $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys can be represented by a second-order polynomial equation, $b(x) = -1.805x^2 - 1.725x + 2.010$ eV. The results clearly show that b_{CE} and b_{SR} are weak and composition dependent. The prime contribution to the gap bowing is from the b_{VD} . The interval of the calculated band gap bowing coefficient for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys is from 2.900 ($x = 0.875$) to 4.234 ($x = 0.125$). One can note that for all contributions of x the main addition to the gap bowing is owing to the b_{VD} effect. This could be correlated to the strong ionicity dissociable of the corresponding binary compounds (CaO and MgO). The contribution of the volume deformation term to the bowing parameter b_{VD} is gradually decreasing with x concentration. Suddenly, in the case of $x = 0.5$ the addition of the b_{SR} has increasing. The addition of the other structural relaxation b_{SR} is weak. Furthermore, b_{CE} and b_{SR} is nearly no effect to the bowing parameter. Namely, the b_{VD} is the only are component which enables the bowing parameter for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys.

4. Conclusion

We have examined the electronic properties of the rocksalt $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys as a function of Calcium composition x by using the GGA method within DFT. The electronic band structures, which are calculated by using the lattice parameters composed from Vegard's law. In the GGA, a band gap bowing parameters are achieved for rock salt $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys. These results propose the physical condition of a composition dependent band gap energy for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys. Our calculations show that the $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloy's the bowing parameters are very strong Calcium composition. The average bowing parameter is 0.583 eV for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ ternary alloys. The results clearly show that the b_{VD} is the only dominant component of the bowing parameter for $\text{Ca}_x\text{Mg}_{1-x}\text{O}$ alloys. Additionally b_{CE} and b_{SR} are weak.

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