

Optimum Calcium Carbonate Filler Concentration for Flexible Polyurethane Foam Composite

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ABSTRACT

The production cost of flexible polyurethane foam is significantly dependent on the cost of polyol, which constitute the largest percentage of materials used in foam production with the characteristics to induce superior mechanical properties. Suitable fillers that are relatively cheap can be introduced in the foam matrix as replacement for polyol. However, certain compositions of filler have deleterious effect on some relevant mechanical properties of the foam. This paper investigates the effect of CaCO₃ filler in flexible polyurethane foam matrix for the dual purposes of achieving sustained mechanical properties and reduction in production cost. The optimum CaCO₃ composition was found to be 20 wt % representing a concomitant 18.54% reduction in cost of production.

Key words: polyurethane, CaCO₃, tensile strength, elongation at break, indentation hardness.

1. INTRODUCTION

Flexible polyurethane foams are one of the most important classes of cellular plastic used in the manufacture of such materials as foam mattresses, pillows, furniture, cushioning materials for automobiles, packing, recreation, shoes, etc [1]. The global consumption of flexible polyurethane foam was estimated to be above 7 million metric tons in 2007 and the average annual growth rate is about 5% [2]. In general, industries that produce flexible polyurethane foams use fillers to modify the material's properties in some way to achieve dimensional stability ease of retraction from the mold and service density [3-4]. When adding a filler to a polymer to form a conjugated biphasic material, the tension applied to the polymeric matrix will be transferred in part to the

disperse filler phase since it presents properties superior to the pure polymer [5]. The use of several fillers to achieve improved properties in foam has been widely studied [6]. Some notable ones include inorganic materials such as calcium carbonate, dolomite, aluminum silica, titanium dioxide, and talc [6] while some of the organic materials used as filler are carbon black and natural fibers [7-8].

In flexible polyurethane foams, the fillers promote an increase in density and resistance to compression. However, they reduce the resilience and contribute to the increase in permanent deformation. In addition, properties such as tear strength are significantly affected by the introduction of filler [9]. Accordingly, it is necessary to determine the correct concentration of the filler in the polymer matrix, so as to obtain a product of reliable quality [10].

Usage of flexible polyurethane foam in Sub- Sahara Africa has been mainly for mattresses and furniture and recently, there has been a growing demand for durable and high hardness characteristic (i.e. high compression resistant) foam at low cost [11]. Foams with these qualities are of interest to many sectors of the economy, and therefore their preparation, characteristics, and applications are subject interest [12]. However, the cost of petro-chemical based polyol, which constitute the largest percentage of materials used in foam production and which possess the characteristics to induce superior mechanical properties in foam currently commands a high value due to the rising costs of the petro-chemical feed stocks [13].

Several studies have been conducted on the suitability of calcium carbonate as filler in flexible polyurethane foam composite. Notable ones are the works of Sabina et. al.[14] and Latinwo et.al.[12,15]. Sabina et. al.[14] investigated the physico-chemical properties of flexible polyurethane foam containing commercial CaCO_3 and observed that the excess of commercial CaCO_3 utilized in industry causes the increase of hysteresis, possibly causing permanent deformations and damaging the quality of the final product. Latinwo et. al. [12] studied the effect of CaCO_3 of different compositions and particle size distributions on the mechanical properties of flexible polyurethane foam. The study reported that finely divided filler material increased the hardness characteristics of the foam to compositions of up to 35wt%, while coarse filler did not show any appreciable improvement in the property. For all particle sizes, the tensile strength and elongation at break decreased while increasing filler content.

It is evident from previous works that foam properties are optimized at specific filler composition and particle size distribution. However, cost implication was not given prime consideration in these works.

For example, the cost benefit of nano-size filler particle is doubtful considering the cost of nanotechnology. Therefore, this paper presents the results of the effect of various composition of

CaCO₃ as replacement for polyol in the range 0-30 wt % on the mechanical properties of flexible polyurethane foam with the view to determining the optimum properties and cost implication.

2. EXPERIMENTAL METHODOLOGY

The flexible foam production method employed is a continuous one known as the **Slab Stock Foaming** process. In this process, the raw materials are weighed, measured and/or metered continuously from separate lines into an in-line mixer while the mixed reactants still in liquid form are poured into a continuous mould with paper covering on a mould conveyor. In the mould, the foaming proper takes place to form a solid continuous block of foam.

The slab stock foaming process usually commences with the preparation, treatment and conditioning of the chemicals. The polyol and toluene di-isocyanate to be used must be cooled to a temperature between 22 – 25°C and 20 – 22°C respectively. This is important as the density, viscosity and chemical reactivity of the two chemicals vary with temperature. This *chilling* operation is accomplished using Chiller units with a Shell and Tube heat exchanger connected to the storage tanks containing the chemicals. Also, the stannous octoate to be used is pre mixed with polyol in the ratio 1:11. Likewise, amine is also pre mixed with polyol in the ratio 2:3.

Thereafter the Foam **Formulation Sheet** is prepared. This *polyol – based* formulation sheet contain a list of the chemicals needed for the production and their various calculated proportions usually based on parts per 100 parts of polyol. This formulation is then used to calibrate the various meters, gauges and flow lines in the plant. The raw materials are pumped to the mixer head in their correct proportions.

Metering accuracy and continuous uniform flow is maintained as any error will lead to the production of poor quality foam. In other to ensure the metering accuracy, each production line is calibrated to determine the efficiency of the machine before production commenced.

Before the production process proper is carried out a **Laboratory Mix (*box foam*)** or laboratory analysis is done. Experience has shown that efficiency of production is greatly enhanced when the laboratory analysis is integrated into the production programme. Laboratory analysis is a process by which a little quantity of foam is made in the laboratory under controlled conditions. The essence of this is to:

- Evaluate the suitability of the raw materials in a proposed formulation;
- Determine the actual levels of activators and silicone required for efficient production;
- Check for the presence of contaminants in any of the chemicals; and
- Determine appropriate *cream* and *rise* times of a proposed formulation.

Samples from foam produced are cut into various sizes suitable for their respective test equipment. For this work, varying concentration of CaCO_3 ranging from 0 to 30 wt % were introduced into the formulation as shown in Table 1 and the following properties measured: density, tensile strength, elongation, compression set, indentation and hardness characteristics.

Table 1. Formulation Table for Experiment.

CHEMICALS	SAMPLES AND CONCENTRATIONS						
	A	B	C	D	E	F	G
Polyol (g)	1000	950	900	850	800	750	700
CaCO₃ (g)	0.0	50.0	100.0	150.0	200.0	250.0	300.0
TDI (g)	516.00	490.20	464.40	438.60	392.80	387.00	361.20
Water (g)	42.00	39.90	37.80	35.70	33.60	31.50	29.40
Amine (g)	0.80	0.76	0.72	0.68	0.64	0.60	0.56
Stannous (g)	2.00	1.90	1.80	1.70	1.60	1.50	1.40
Silicone (g)	10.00	9.50	9.00	8.50	8.00	7.50	7.00

3. RESULTS AND DISCUSSION

The percentage (%) composition of calcium carbonate (CaCO_3) in each sample varies from 0% in Sample A to 30% in sample G at 5% interval. The foam properties studied and their corresponding results are as shown in Figures 1 – 6.

Figure 1 illustrates the rise time variation and it shows a uniform decrement in rise time as calcium carbonate (CaCO_3) concentration increases. This implies that the foam rises faster when the filler composition increases indicating that the blowing / gas production reaction between toluene-di-isocyanate and water occurs faster. It could therefore be inferred that the introduction of fillers in foam formulation greatly influences the reaction time which is a positive development.

The density of the foam is improved as CaCO_3 content increases to 30% as shown in Figure 2. There is a gradual increase in density value until 20% filler concentration before a sharp rise in density occurred afterwards. This clearly indicate that the filler concentration could be as high as 20% and still produce the same density foam (22.6kg/m^3) as when not introduced at all (21.1kg/m^3).

It is however worthy of note that it is possible to still maintain the density while increasing filler concentration further, if so desired. This could be done by making adjustments in the other chemical compositions. The 20% filler concentration obtained as optimum in this study is just the most convenient to determine the effect of fillers on foam manufacture.

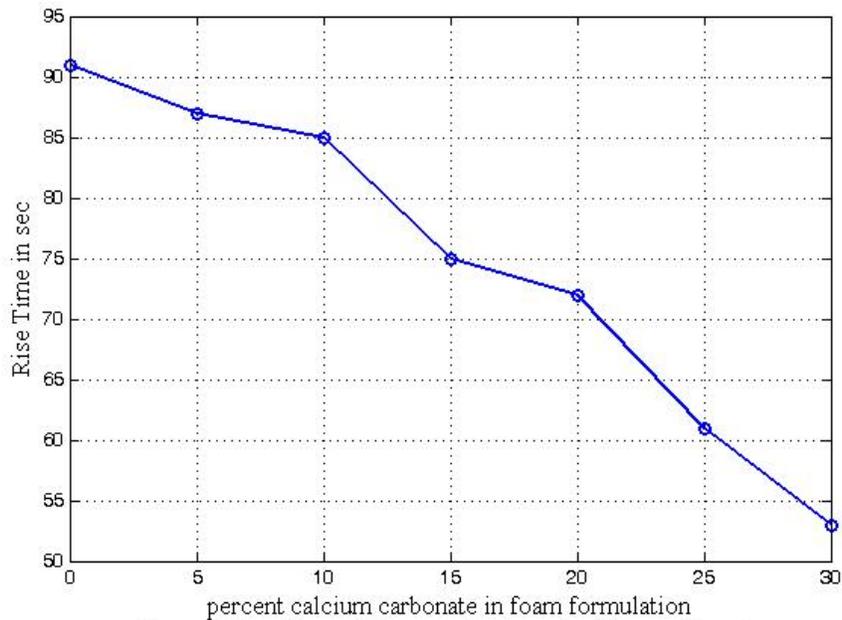


Figure 1 Rise Time with concentration of calcium carbonate

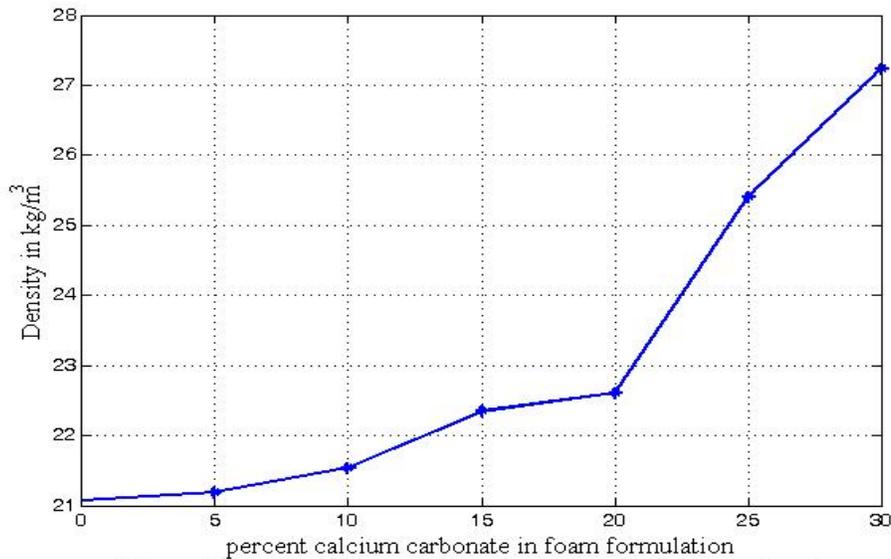


Figure 2 Density of foam with concentration of calcium carbonate

Also, one of the properties of foam that distinguishes good foam from a bad one is its percentage elongation which describes its elastic properties. Figure 3 shows that the elastic property of the foam was maintained between the normal ranges of 90–110 kN/m² up till the point where the filler concentration is about 20%. Thereafter, a sharp drop in the elastic property was experienced. This depicts also that the elastic property of foam can be maintained while introducing fillers into the concentration.

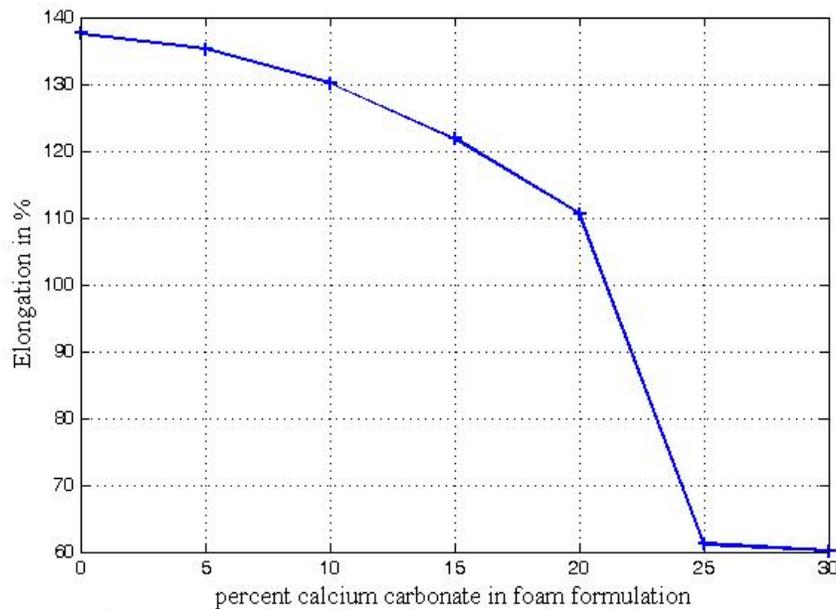


Figure 3 Elongation of foam with concentration of calcium carbonate

Furthermore, in Figure 4 the tensile strength of the foam was maintained until 20% filler content (145-120 kPa) after which there was a sharp and sudden decrease in tensile strength. The compression and load bearing ability of the foam was uniform (3.5%) between 0-10% CaCO_3 despite the introduction of the filler until 20% filler addition with an exception at 15% CaCO_3 (5.5%) (see Figure 5). When there was no filler in the formulation the foam resilience steadily increased to 148 (see Figure 6) as the filler was being introduced.

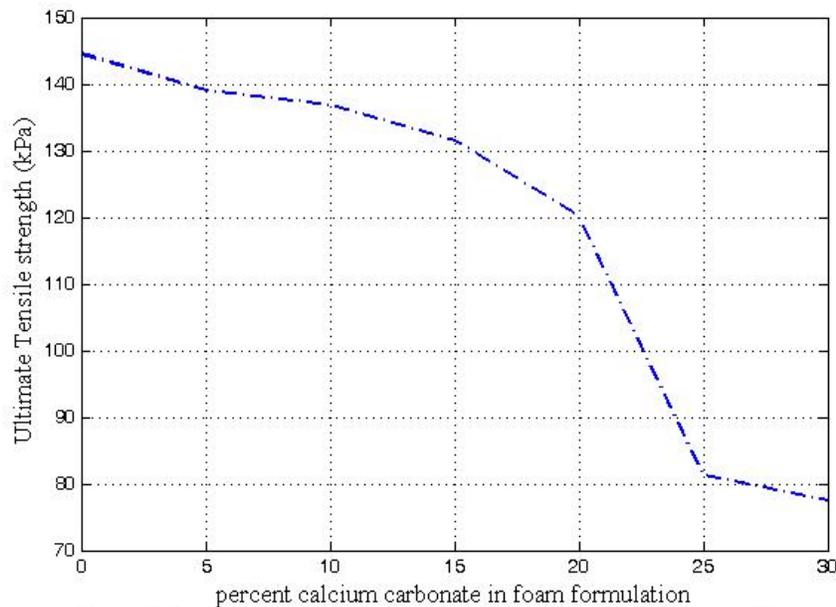


Figure 4 Tensile Strength of foam with concentration of calcium carbonate

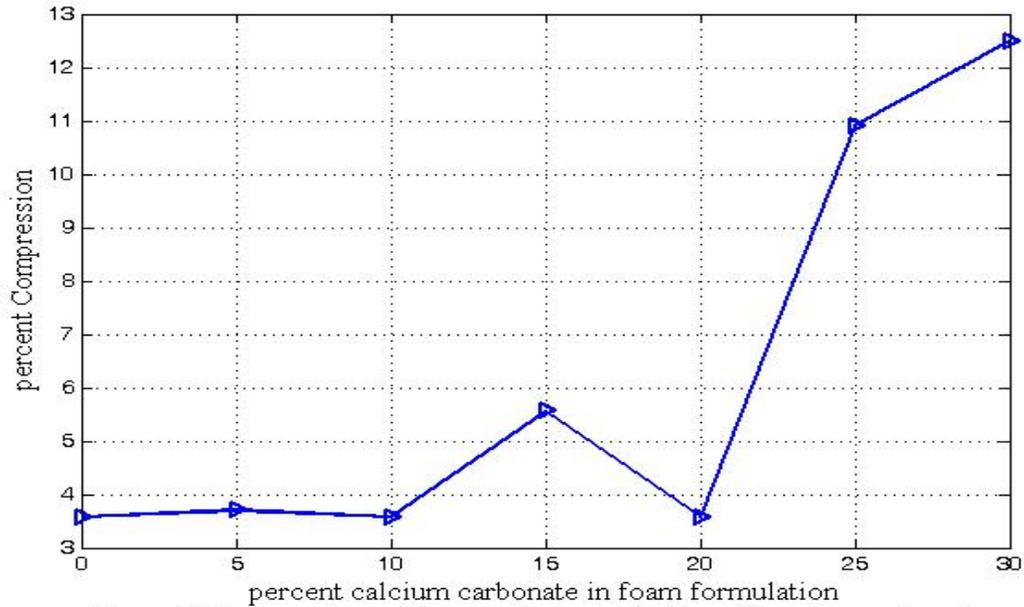


Figure 5 Compression of foam with concentration of calcium carbonate

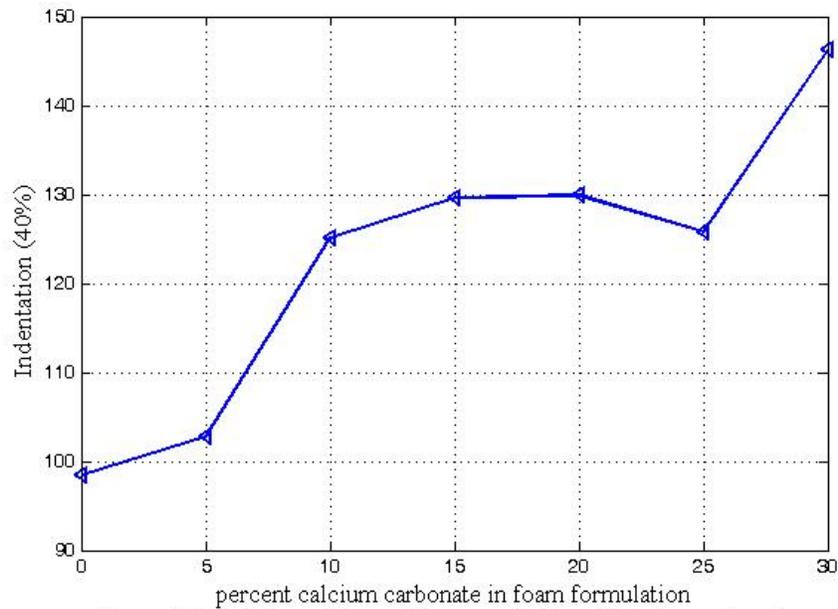


Figure 6 Indentation of foam with concentration of calcium carbonate

Cost effectiveness of this foam formulation is another significant aspect of this study. The result of this research has shown that CaCO_3 filler can be introduced into the foam formulation up to 20% optimum content with enhanced foam qualities as when no filler was used. Cost analysis of the cost of production for formulation without filler (sample A) and formulation with 20% filler

concentration (sample E) is presented below after carrying out a market survey. The current market prices per kilogram of each chemical used and the foam formulations for samples A and E are shown in Tables 2 and 3 respectively based on their requirements for production.

Table 2. Cost Per kg of Each Chemical in Foam Formulation

CHEMICALS	COSTS OF 1 KG (#)
Polyol	273.00
CaCO₃	70.00
TDI	356.00
Water	50.00
Amine	608.00
Stannous	1000.00
Silicone	800.00

Table 3. Total Cost of Production per kg of Each Chemical in Foam Formulation

CHEMICALS	SAMPLES AND COST ESTIMATES			
	Sample A (0%)	COST ESTIMATE (#)	Sample E (20%)	COST ESTIMATE (#)
Polyol (g)	1000	273.00	800	218.40
CaCO₃ (g)	0.0	0.00	200.0	14.00
TDI (g)	516.00	183.70	392.80	139.84
Water (g)	42.00	2.10	33.60	1.68
Amine (g)	0.80	0.49	0.64	0.39
Stannous (g)	2.00	2.00	1.60	1.60
Silicone (g)	10.00	8.00	8.00	6.4
Total Costs		<u>#469.29</u>		<u>#382.31</u>

To estimate the percentage reduction in cost of production given the total costs in Table 3, equation 1

$$\% \text{ Reduction} = \frac{\text{sample A total cost} - \text{sample E total cost}}{\text{sample A total cost}} \times 100 \quad 1$$

⇒

$$\% \text{ Reduction} = \frac{469.29 - 382.31}{469.29} \times 100 = 18.54\%$$

Hence, the cost of production was reduced by about 18.54% while the foam qualities and properties were either maintained or improved upon as the CaCO_3 filler was introduced into the foam formulation to an optimum 20% content. Thus CaCO_3 can effectively replace polyol in flexible foam manufacture. Therefore the introduction of CaCO_3 into foam formulation will not only help reduce the cost of production but also help to maintain/or improve the foam qualities and its physical properties thereby making this an attractive venture.

4. CONCLUSION

This study has shown that calcium carbonate addition in foam formulation increase the rate of the blowing / gas production reaction between toluene-di-isocyanate and water with reduction in the rise time. However the elastic and tensile strength characteristics of foam are not affected by the presence of the filler. If calcium carbonate filler is properly introduced into the foam formulation, the desired tensile strength and elastic qualities of the foam can be achieved, and in effect significantly improves the foam resilience ability as the desired load bearing ability is enhanced.

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