

Heterostructure Solar Cells Based on Sol-Gel Deposited SnO₂ and Electrochemically Deposited Cu₂O

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ABSTRACT

To fabricate a heterostructure solar cell using environmentally friendly materials and low cost techniques, tin oxide (SnO₂) and cuprous oxide (Cu₂O) were deposited by the sol-gel method and the electrochemical deposition, respectively. The SnO₂ films were deposited from a SnCl₂ solution containing ethanol and acetic acid. The Cu₂O films were deposited using a galvanostatic method from an aqueous bath containing CuSO₄ and lactic acid at a temperature of 40°C. The Cu₂O/SnO₂ heterostructure solar cells showed rectification and photovoltaic properties, and the best cell showed a conversion efficiency of 6.6×10^{-2} % with an open-circuit voltage of 0.29 V, a short-circuit current of 0.58 mA/cm², and a fill factor of 0.39.

Keywords: Sol-Gel Deposition; Electrochemical Deposition; Cu₂O; SnO₂; Solar Cell

1. Introduction

The production of silicon solar cells is expanding, but not sufficiently fast, because of high cost and large energy consumption for the material purification and cell manufacturing. Thus extensive researches are going on, aiming the development of solar cells using cost-effective and environmentally benign materials prepared by simple low-cost techniques. In this work, we attempt to use tin oxide (SnO₂) deposited by a sol-gel technique for the n-type semiconductor of a solar cell. SnO₂ thin films have been attracting interest since they have many important applications such as gas sensors [1,2] and transparent electrodes [3,4]. SnO₂ films have been fabricated by a number of techniques, including spray pyrolysis [5-6], sputtering [7-9], chemical vapor deposition (CVD) [10-12], and sol-gel deposition [3,13,14]. The sol-gel technique has several advantages, such as easy control of film thickness, ability to coat large area, and cost-effectiveness. In this work, we fabricate heterostructure solar cell based on sol-gel deposited SnO₂. It should be noted here that there are very few reports of application of SnO₂ for n-type layer of a heterostructure solar cell [15-18] although SnO₂ is often used as an electrode material.

On the other hand, cuprous oxide (Cu₂O) has gathered much attention as a p-type absorption layer of solar cells

owing to its suitable band gap around 2 eV, material abundance, and non-toxicity. The solar cells based on electrochemically deposited Cu₂O have been fabricated with pn heterojunction with ZnO [19-21], TiO₂ [22,23], and indium-tin-oxide (ITO) [24]. The highest solar conversion efficiency of 1.28% was obtained from ZnO/Cu₂O solar cell [19]. In this work, we attempt fabrication of heterostructure solar cells based on electrochemically deposited Cu₂O and sol-gel deposited SnO₂. One paper has been published on Cu₂O/SnO₂ heterostructure solar cells based on sputtered SnO₂ [16], but to our best knowledge, there are no reports on fabrication of heterostructure cells by a combination of sol-gel deposition of SnO₂ and electrochemical deposition (ECD) of Cu₂O.

2. Experimentals

2.1. SnO₂ Deposition and Characterization

SnO₂ films were deposited by the sol-gel method on ITO coated glass substrates. (For the XRD measurement, we used a glass substrate.) 0.7 M SnCl₂·2H₂O, 10 mL ethanol and 0.25 mL acetic acid were used as a starting source material, solvent and catalyst, respectively. The coating solution was stirred at 70°C for 30 min and aged at 40°C for 3 hours, and then dropped on the ITO substrate, which was rotated at 3000 rpm for 20 sec. Then,

the films were dried at 150°C for 5 min. The procedures from the spin coating to the drying were repeated 2 - 5 times, and then the film was annealed in air at 300°C for 1 hour.

The compositional analysis was carried out by Auger electron spectroscopy (AES) using the model JEOL JAMP 7800. Profile meter Accretch Surfcom-1400D was used to measure the thickness of the film. The X-ray diffraction (XRD) measurement was carried out by the Rigaku SmartLab X-ray diffractometer using CuK α_1 radiation. The optical transmission measurement was performed using a JASCO-570 spectrometer with the ITO substrate as the reference. In addition, the conduction type and photosensitivity of the films were examined by means of the photoelectrochemical (PEC) measurements. The PEC measurement was carried out using the three-electrode cell with saturated calomel electrode (SCE) as the reference electrode. The deposited film was used as the working electrode, and 100 mM of Na₂SO₃ was used for the electrolyte. The backside of the sample was illuminated by pulsed light coming from a Xe lamp (about 100 mW/cm²). The incident light was turned off and on mechanically every five seconds. A ramp voltage was applied between the working and the reference electrodes, and the released current was monitored.

2.2. Cu₂O/SnO₂ Heterostructure Fabrication and Characterization

The heterostructure solar cells were fabricated by depositing Cu₂O by ECD on the SnO₂ film. We used a solution containing 0.2 M CuSO₄ and 1.6 M lactic acid. pH of the solution was adjusted to 12.5 by adding KOH. Cu₂O was deposited galvanostatically at 40°C with a current density of -1.0 mA/cm². All samples were deposited for 10 min. Indium metal electrodes were deposited by thermal evaporation on the Cu₂O layer. The electrode size is 1 mm². Photovoltaic properties were characterized using an AM 1.5 solar simulator. The radiation power of light was about 100 mW/cm². The light was illuminated on the ITO side (SnO₂ side) of the sample.

3. Results and Discussion

3.1. SnO₂ and Cu₂O Films

The thickness of the SnO₂ film was about 0.6 μ m when the coating-drying cycle was repeated 3 times. The film is transparent in the visible range, and the band gap obtained from the optical transmission spectrum is 3.9 eV. The differential AES spectrum of the prepared film is shown in **Figure 1**. The O/Sn composition ratio obtained from it is about 2, *i.e.*, the film is almost stoichiometric. A weak chlorine signal was observed near 190 eV. The XRD pattern of the film deposited on the glass substrate

is shown in **Figure 2**. The pattern shows three characteristic XRD peaks of SnO₂ with a broad background signal due to glass. The SnO₂ peaks are broad, which indicates that the SnO₂ film consists of nano crystallites.

The PEC results for the SnO₂ film are shown in **Figure 3**. The current was not changed by illumination for the cathodic scan, while during the anodic scan, the current was changed due to the light chopping. When the junction of the semiconductor-electrolyte is illuminated, photogenerated electrons/holes are separated in the space charge region. The photogenerated minority carriers

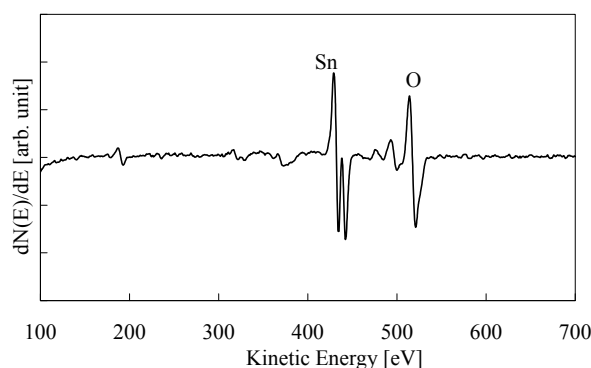


Figure 1. AES spectrum for sol-gel deposited SnO₂.

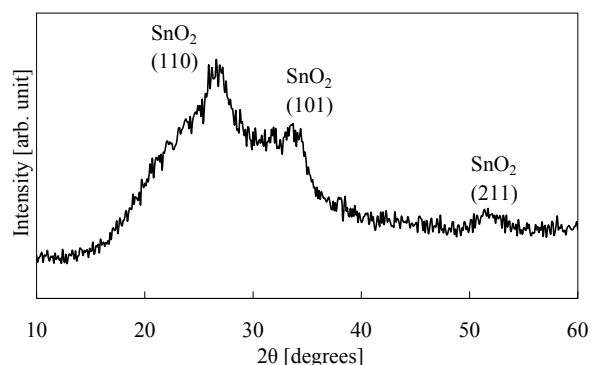


Figure 2. X-ray diffraction pattern of sol-gel deposited SnO₂.

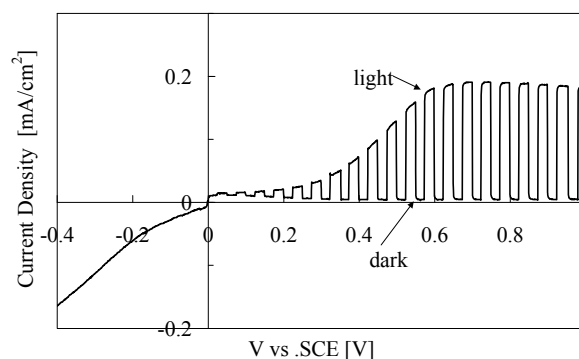


Figure 3. Photoelectrochemical measurement results for SnO₂.

arrive at the interface of the semiconductor-electrolyte to participate in the electrochemical reaction at the film/electrolyte interface. The current becomes more positive under the light illumination during the anodic scan. This implies that the minority carriers generated here are holes. Thus, the prepared film is n-type semiconductor.

The film thickness of ECD-Cu₂O was about 0.7 μm and the direct band gap obtained from the optical transmission was about 2.5 eV. The PEC results for the Cu₂O film are shown in **Figure 4**. The current was not changed by illumination for the anodic scan, and thus only the results for the cathodic scan are shown there. Negative photo current was observed, which implies that the minority carriers generated here are electrons. Thus, the prepared film is a p-type semiconductor with a good photosensitivity.

3.2. Cu₂O/SnO₂ Heterostructures

The heterostructures were fabricated with SnO₂ as a window layer and Cu₂O as an absorption layer. **Figure 5** shows I-V characteristic of the Cu₂O/SnO₂ heterostructure cell fabricated by repeating the coating-drying cycle 3 times for the SnO₂ deposition. The rectification and photovoltaic behaviors appeared. The heterostructure cells were fabricated with different repetition time of the coating-drying cycle, and the photovoltaic properties of all the samples are shown in **Table 1**. The properties strongly depends on the repetition number of the coat-

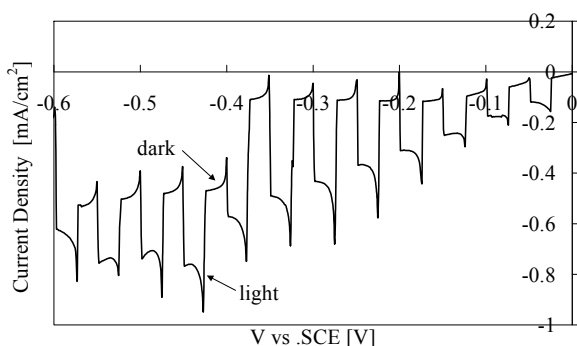


Figure 4. Photoelectrochemical measurement results for ECD-Cu₂O.

Table 1. Photovoltaic properties for the Cu₂O/SnO₂ heterostructure cells.

Number of coating-drying cycles for SnO ₂	Open circuit voltage [mV]	Short circuit current [mA/cm ²]	Fill factor	Efficiency [%]
2	20	3.2×10^{-1}	0.25	1.6×10^{-3}
3	290	5.8×10^{-1}	0.39	6.6×10^{-2}
4	75	4.9×10^{-1}	0.27	1.0×10^{-2}
5	10	7.8×10^{-2}	0.27	2.3×10^{-4}

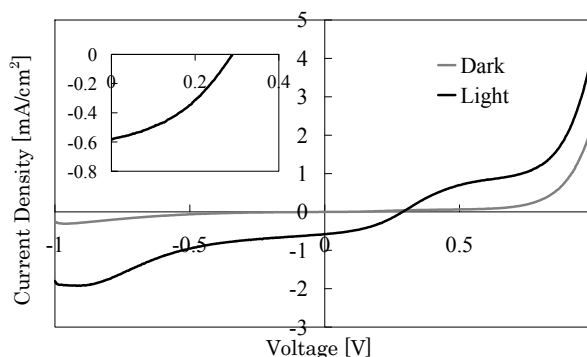


Figure 5. I-V characteristic of the heterostructure cell consisting of sol-gel deposited SnO₂ and ECD-Cu₂O. The coating-drying cycle was repeated 3 times in the sol-gel process. The inset is an expanded figure of photovoltaic properties.

ing-drying cycle, and the highest efficiency of 6.6×10^{-2} % was obtained with three times repetitions. The reason for dependence on the repetition number is not understood. The film thickness increases in proportion to the repetition number, and too large thickness may results in generation of defects such as crack because of strain.

Since each of the layers in the heterostructure showed fairly high photosensitivity as shown in **Figures 3 and 4**, the poor photovoltaic characteristics will be due to some disorder at the interface. Apparently, the SnO₂ layer did not dissolve in the Cu₂O deposition solution, but we cannot exclude possibility that the SnO₂ surface reacted with the Cu₂O deposition solution, to form a thin defective layer. Another possible reason of the poor performance is the Cu₂O/electrode interface. The forward characteristics under the illumination is not a simple diode curve but has a plateau around 0.7 V. This may be due to non-ohmic characteristics of the interface with the electrode.

4. Conclusion

SnO₂ films have been deposited by the sol-gel method. The films showed clear n-type conduction and photosensitivity. We have fabricated Cu₂O/SnO₂ heterostructure cells by depositing Cu₂O by ECD on the sol-gel deposited SnO₂ layer. Rectification property was observed, and the best cell showed an efficiency of 6.6×10^{-2} % under AM1.5 illumination.

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