

# **Contribution to Comparative Study of Physical-Chemical Characteristics of Diack Basalt and Bandia Limestone** for Use in Railway Engineering

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## Abstract

This paper presents a comparative study of Physical-Chemical characteristics of Limestone and Basalt (from Senegalese quarries). First, chemical tests show that Basalt is richer in silica 51.59% versus 2.84% for Limestone. Basalt is made up of silica minerals and essentially carbonated minerals with a CaO percentage of 50.05%. Chemical results also show that Basalt is richer in iron 12.71% versus 0.44% for Limestone. Finally, they revealed a fire loss of 40.91% for Limestone and 2.44% for Basalt. Second, physical analysis results show that Diack Basalt has the best characteristics with a flattening coefficient of 5% between 5% and 20%; the percentage of pollutants is 0.36% less than 1%; the Los Angeles coefficient is 12.21% below 15, while Bandia Limestone gives a flattening coefficient of 3%; the Los Angeles coefficient of 40.17% and the percentage of pollutant (2.4%) well above 2%. It is noted that the percentage of Limestone pollutant is too high. These important results show the net advantage of Basalt compared to Limestone in terms of physical-chemical characteristics.

#### **Keywords**

Ballast, Railways, Basalt, Limestone, Physical-Chemical Characteristic

# **1. Introduction**

Rail transport is one of the important factors in the field of land transport. Its development has grown very significantly in recent years mainly with the variety with the appearance of high-speed trains but also with the considerable increase in axle loads [1]. Ballasted railways are made up of a set of elements working in perfect correlation (rails, sleepers, fastening systems, ballast, support platform, etc.). If the ties rested directly on the platform, they would sink more or less into the natural terrain, the strength of which is generally insufficient to support the load transmitted by the ties; track levelling would be compromised. The study of the mechanical behavior of materials used in public works in general has been of interest to the scientific community for a very long time [2]-[11], among others]. The case of materials used as railway ballast is also a branch of public works that attracts a lot of interest. In recent decades, ballast has been increasingly studied in order to better understand its behavior and degradation modes [[12]-[20], among others]. In Senegal, many quarries of sand, limestone, sandstone, basalt, laterites, clays, etc. are exploited mainly in the Dakar and Thiès regions. Diack Basalt is widely used for the surface layer of the bituminous pavements. However, with the increase in traffic volume and the load of axles, Diack Basalt is increasingly used as a base or foundation layer of soft pavements. It is often used as a severe untreated 0/31.5, in the form of Grave Bitumen 0/20 with higher performance. This material is well known and is a good reference material in Senegal that will later be used with other reference materials in the United States. However, the amount of basalt reserves available should be questioned because the only deposits exploited are Diack's. The Dakar deposits are not accessible for environmental reasons [21]. In Bandia, Limestone appears at the outcrop in the form of large blocks and kidneys deeply karstified and lapidated, very hard, rich in shell debris; these Limestones are currently employed for the production of aggregates and for the manufacture of lime. However, the work of [22] [23] shows that basalt remains the reference material for ballast in Senegal.

Limestone that can be used as a stony material in road engineering is found to the west of Bandia. At the outcrop, they appear in the form of large boulders and very hard kidneys, rich in shell debris. The Diack deposit hosts the main quarries for the production of crushed basalt aggregates in Senegal. The outcrops are in the form of two peaks (peaks A and B) separated by a zone without outcrops [24] and consist of fine, medium and coarse-grained facies. The particles sizes collected in the Diack and Bandia quarries are 20/50 mm classes, knowing that the materials studied are intended for Senegalese ballasted railways. Figure 1



Bandia LimestoneDiack BasaltFigure 1. Overview of materials in Bandia and Diack quarries [23].

shows the materials studied. The shape of the aggregates is polyhedral and sharp-edged. The aggregates should therefore not be too long or too flat. They also have good angularity.

This paper presents an experimental work on materials from local quarries in Senegal. The purpose is the comparative study between Bandia Limestone and Diack Basalt for use as ballast in railway structures. They are part of the feasibility study campaign for the use of materials from Senegalese quarries as ballast, such as the study presented in [22] [23].

#### 2. Context

In Senegal, the exploitation of geological formations for aggregate production generally concerns massive rocks from the Senegal-Mauritanian basin [23]. Many quarries are exploited mainly in the Dakar and Thiès regions. The western part of the Senegal-Mauritanian basin was the site of a major could not flight towards the end of the Tertiary over the whole of Cape Verde and the Thiès Plateau, and in the quaternary on the Dakar peninsula. The Diack deposit is located between the longitudes 16' - 43' and 16' - 45' west and latitudes 14' - 40' and 14' - 41' north, at the eastern end of the Thiès plateau and on Ngoundiane hill, located 80 km east of Dakar. This deposit is home to Senegalese main quarries for the production of crushed basalt aggregates. The first finite-tertiary volcanic episode took place between the Oligocene and the Upper Miocene. It manifests itself in basalt effusions in the Cape Verde peninsula (Cap Manuel, Gorée, Fann) and lava intrusions corresponding to dykes in the Thiès region (Diack, Sène Sérère), or tectonic veins (Keur Mamour, Ravin des voleurs, Thiéo, Bellevue, Sandock, Fouloume) [24]. Tertiary volcanism is therefore essentially fractural. The limestone formations of the Senegalese-Mauritanian basin are of upper Cretaceous age in Paleocene and are present in a large extension in the Cape Verde peninsula and the Thiès Plateau. The Paleocene rises east and west of the Horst de Ndiass in the cliffs of Thiès and in Dakar. During the Paleocene, the sedimentation environment became more and more limestone and was characterized by the development of reefs made up of limestone, limestone clay and marl. At the end of the Paleocene the Horst of Ndiass began to emerge and a karst relief of Paleocene limestone (Figure 2).

The limestone reserves are mainly located in the Dakar region (Bargny and Rufisque) and the Thiès region (Bandia, Popéguine Déyane, Mbour, and Panthior). Basalt is a volcanic rock, black and compact, well suited for use as ballast aggregate. In Senegal, Basalt extraction is mainly focused on Diack. There are several operators on Diack Hill, a community in the rural community of Ngoundiane. In 1970, basalt reserves at Diack were estimated at 10 million tons. However, regardless of the type of material used, ballast comes from crushing rock extracted from hard stone quarries.

#### **3. Experimental Protocol**

Methodology adopted consists first of all of an extensive bibliographical research,





to assess the state of knowledge about ballasts. Sampling was carried out in the Diack Basalt and Bandia Limestone quarries. These samples were subjected to physical and mechanical testing in accordance with current standards to assess their suitability in ballasts. Tests carried out are: particle size analysis (NF P 94-056), cleanliness (NF P 18 591), absorption (NF EN 1097-6), density (NF EN 1097-6), Deval (NF 18 577), Los Angeles (NF 18 573) and chemical analyses (NF EN 1744-1 + A1 February 2014, P18-660-1). With each test performed, a comparison is made between the intrinsic characteristics obtained on the two materials.

#### 4. Results and Discussions

The comparative study between limestone and basalt ballasts was carried out through chemical and geotechnical tests. Geotechnical tests are included in granulometric analysis, specific weight, cleanliness, resistance to wear and fragmentation, shape and compaction.

#### 4.1. Chemical Analysis

The chemical composition of the aggregates is determined by analyzing the rock powder. The main techniques used to determine the percentage of the major items listed in the table below met the following standards [26] [27].

Results of the chemical analysis are recorded in **Table 1**. It presents the different percentages of the major elements, minor elements and traces contained in the Bandia Limestone and in the Basalt of Diack.

The results show that basalt is richer in silica 51.59% versus 2.84% for limestone. This difference is due to the petrographic nature of these two materials. Basalt is made up of silica minerals and essentially carbonated minerals with a CaO percentage of 50.05%. These results also show that basalt is richer in iron 12.71% versus 0.44% for limestone. This iron richness is due to the presence of ferromagnesian minerals in basalt. Chemical analysis revealed a fire loss of 40.91% for limestone and 2.44% for basalt. This difference reflects the presence of the higher organic matter content in limestone, resulting from its chemical to biochemical nature.

Chemical	Techniques used	Content (%)		
elements	Techniques used	Limestone	Basalt	
SiO <sub>2</sub>	Insolubilization	2.84	51.59	
$Al_2O_3$	Coloring and reading by Spectrophotometer	0.09	1.84	
Fe <sub>2</sub> O <sub>3</sub>	Coloring and reading by Spectrophotometer	0.44	12.71	
CaO	Complexometry (EDTA)	50.05	19.27	
MgO	Complexometry (EDTA)	2.53	6.29	
Na <sub>2</sub> O	Flame photometry	1.33	2.81	
K <sub>2</sub> O	Flame photometry	0.35	1.05	
TiO <sub>2</sub>	Coloring and reading by Spectrophotometer	0.03	0.8	
MnO	Coloring and reading by Spectrophotometer	0.01	0.02	
$P_2O_5$	Coloring and reading by Spectrophotometer	0.09	0.09	
SO <sub>3</sub>	Precipitation by BaCl <sub>2</sub>	1.26	0.21	
H <sub>2</sub> O	Steaming at 105° for 24 hours	0.12	0.87	
Fire losses	Calcination at 950°	40.91	2.44	

Table 1. Chemical characteristics of Bandia Limestone and Diack Basalt.

EDTA means Ethyl Dimethyl Tetra Acid.

#### 4.2. Particle Size Analysis

The particle size analysis was performed on Bandia Limestone and Diack Basalt under NF P 94-056 (1996). **Figure 3** represents the results of the granulometric analysis.

Examination of **Figure 3** shows that the ballasts have a tight particle size distribution with diameters between 20 and 50 mm. However, there is a high percentage of fine for limestone in the order of 30% compared to 6% for basalt. This difference in the percentage of fine reflects the cohesive structure of limestone that contributes to the generation of fine when the grains come into contact during sieving.

#### 4.3. Physical and Mechanical Characteristics of Materials

The tests carried out on the materials studied yielded the following results recorded in Table 2.

According to UIC code [28], the basalt results have the best intrinsic characteristics. Indeed, the cleanliness test shows that limestone has more impurity 2.96% than basalt 0.32%. The double porosity of limestone explains the absorption rate 2.4% higher than basalt, which is low in porous. The presence of ferromagnesian minerals at the basalt explains the specific density of 3.05 higher than limestone at a value of 2.49. The value of 5 as a flattening coefficient requires that basalt have the best shape than limestone. Depending on the petrographic nature, basalt is a material derived from a magmatic rock. This gives it resistance to wear of 9.77% and fragmentation of 40.17% against 2.31% and 12.21% respectively for limestone.



Figure 3. Grain size curves of aggregates of Bandia Limestone and Diack Basalt.

Table 2. Intrinsic	characteristics	of Bandia	Limestone	and Diac	k Basalt
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	Cleanliness (%)	Absorption (%)	Los Angeles (%)	Deval (%)	ρs (g/cm³)	Applating (%)
Limestone	2.96	2.4	40.17	9.77	2.49	3
Basalt	0.32	0.36	12.21	2.31	3.05	5

#### 4.4. Proctor Tests

Proctor tests were carried out using the C-Mould of dimensions: a diameter of 250 mm and a height without extension of 200 mm. The experimental procedure used is described in detail in the work of [23]. The ability to compact ballast is assessed through the modified Proctor test, in accordance with the French standard NF P 94-093 (1993). The Proctor test makes it possible to know the maximum possible density of a material, and its water content necessary to achieve this density, in order to be able to assess the permanent deformations of materials.

The obtained Proctor curves are shown in **Figure 4**.

**Figure 4** shows that basalt has a higher dry density than limestone and has a lower water content. This sensitivity of Bandia Limestone by rap-port to water is the result of its plasticity and cohesive structure. Studies by [29] have revealed that this material has a double porosity and that a large par-tie of water is retained by matrix suction and intergranular porosity. This-while, the insensitivity of the basalt to water gives it a good drainability of surface water.

## **5.** Conclusions

Comparative study between limestone and basalt ballasts shows that basalt is more suitable as ballast than limestone. Indeed, basalt has good density; it is insensitive to water and has the best characteristics of hardness. From a compositional point of view, the presence of silica minerals gives it a higher silica percentage than limestone. Because of a chemical and biochemical nature, limestone has poor intrinsic characteristics. Its use as ballast is not recommended. This conclusion is in line with that of the work [22] [23].

Materials in question have long been used in public works in Senegal but this article gives the novelty of comparing their physical-chemical characteristics in order to use them as ballast. In the specific context of FerroVer ballast in Senegal, this study gives the advantage of being able to directly recommend Diack Basalt at the expense of Bandia Limestone. The scientific interest of this study is to contribute to the identification and choice of quality materials that can be used



Figure 4. Compaction curves of materials studied.

as ballast.

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## **Conflicts of Interest**

The authors declare no conflicts of interest regarding the publication of this paper.

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