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Stability and Sensory Quality of Dried Papaya

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Abstract

Fruit dehydration is a way of supplying the population with healthy and nutritious foods. The shelf life of dried fruit can be defined by the evaluation of changes occurred in chemical characteristics during storage. This study aims to evaluate the sensory quality and the stability of papaya cv. Tainung n° 1 dehydrated by convective drying. Fresh and dried papaya were evaluated for color, moisture, pH, acidity, water activity, soluble solids, vitamin C, carotenoids, total extractable polyphenols (TEP) and antioxidant activity (ABTS). The sensorial acceptance of the dried papaya was evaluated using a structured nine-point hedonic scale. For the stability study, the analysis of moisture, pH, titratable acidity, water activity, total carotenoids and vitamin C were carried out every 30 days of storage until 120 days. During storage, the moisture content of dried papaya remained constant, but there were undesirable changes in color, increase of acidity and reduction of soluble solids. The degradation of total carotenoids and vitamin C followed the first order reaction, and the half-life time was 346 days for carotenoids, whereas for vitamin C it was only 29 days. In the sensory analysis, the dried papaya received grades between 5.0 and 6.0 for all evaluated attributes. Dried papaya is recommended to be consumed up to 30 days, since within this period a product with higher total carotenoids content, vitamin C and with satisfactory physicochemical and sensorial characteristics were obtained.

Keywords

Carica papaya, Carotenoids, Color, Dehydrated Fruit, Drying, Vitamin C

1. Introduction

The pulp of papaya (Carica papaya L.) has a pleasant aroma and taste, which are

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important qualities for sensorial acceptance for people of all ages. This fruit also presents interesting nutritional characteristics that include it in the select group of healthy foods. The chemical composition of papaya is in concordance with the cultivar, climate, cultural treatments and maturation stage. The Tainung n° 1 variety presents, in 100 g of pulp, 45 kcal; 11.06% carbohydrate; 25 mg of calcium; 17 mg of magnesium; 222 mg potassium; and 78.5 mg of vitamin C [1].

Papaya is also an important source of carotenoids (β -carotene, lycopene and β -cryptoxanthin) and phenolic (myricetin, fisetin, morin, quercetin, kaempferol and isorametina) [2] [3], which has an important antioxidant effect and contributes to the fruit sensory characteristics as color, taste and texture [4].

It is consumed preferentially fresh, but is a highly perishable fruit. It is estimated that losses occurring in the field, transportation and storage are 25% [5]. Post-harvest losses also include the degradation of bioactive compounds that are essential for fruit quality [6]. Papaya industrialization is an alternative that makes it possible to minimize these losses and to use the fruits that are out of standard for commercialization. The application of appropriate technologies to preserve papaya nutritional properties is one of the ways to supply the population with healthy and nutritious processed foods.

Among the various conservation methods, drying is a relatively simple process in which water activity in food is reduced, minimizing or inhibiting the chemical and enzymatic reactions and microbial growth that are responsible for its deterioration [7]. Although different dehydration processes are used by the food industry, convective drying is the most common one, due to its easiness of processing and accessible technology [8].

During the fruit drying process, changes in appearance and color may occur and compromise the product acceptance. These changes occur through the action of the polyphenol oxidases enzymes that are present in the fruits, leading to the formation of melanins (dark pigments) from the oxidation of phenols to o-quinones [9]. This dimming can be minimized by immersion of the fruit in solutions with enzymatic inhibitors such as citric, phosphoric, malic or ascorbic acids [10].

Loss of nutrients, such as vitamin C and carotenoids, may occur during storage, thus reducing the nutritional value of dried fruit. Degradation of these nutrients depends on factors such as pH, oxygen exposure, presence of light, metals, enzymes, temperature and the drying methods used [6] [10] [11].

In addition to the nutritional aspects, sensorial attributes must be evaluated in order to determine the quality of dried fruits. The aroma and flavor may be altered due to loss of volatiles during drying and storage of the dried fruits, leading to lower product acceptance [10].

The objective of this study was to evaluate the physicochemical characteristics, bioactive compounds, sensory acceptance and the stability of dried papaya obtained by convective drying.

2. Material and Methods

2.1. Sample Preparation

Papaya (*Carica papaya* L.) fruits of Tainung n° 1 variety were purchased in local market and processed at maturity stage 4 (bark surface 50% to 75% yellow). The fruits were washed in water and immersed in solution with 100 mg/L of chlorine for 15 minutes. Fruits were peeled, the seeds removed and cut crosswise into 1.2 cm thick slices.

The slices were subdivided into pieces of 3.0×3.0 cm and submitted to the following treatments: T1—control treatment (without antioxidant treatment); T2—papaya immersion in solution containing 25 mg/L citric acid + 75 mg/L of ascorbic acid for 10 minutes, according to Reis *et al.* [12].

2.2. Convective Drying

The papaya slices (T1 and T2) were dehydrated at 60° C in a dryer with fixed air velocity of 1.5 m/s until reaching final humidity between 15% and 25% (w.b). The dried papayas were packed in polyethylene terephthalate/aluminum/polyethylene (PET/Al/PE) and stored for 120 days at room temperature (28° C \pm 2° C).

2.3. Physicochemical Analysis

The fresh and dried papayas were evaluated for color, moisture, pH, titratable acidity (TA), water activity (a_w) and soluble solids (SS).

The color was determined using the colorimeter (Konica Minolta, model CR400, CIELAB system, illuminant D65). A sample of ten pieces of each treatment was analyzed. The coordinates L* (lightness), a* (green/red intensity) and b* (blue/yellow intensity) and the color attributes C* (color intensity/chroma) and h* (hue/color angle). The color variation (ΔE^*) was also calculated by the Equation (1) [13].

$$\Delta E^* = \left[\left(L_0^* - L^* \right)^2 + \left(a_0^* - a^* \right)^2 + \left(b_0^* - b^* \right)^2 \right]^{1/2} \tag{1}$$

The analysis of moisture, pH, soluble solids and titratable acidity were performed according to the Adolfo Lutz Institute [14]. The water activity was determined at 25°C by direct reading in water activity determinant (Aqualab Series 3/3TE).

2.4. Bioactive Compounds and Antioxidant Activity

The total carotenoid contents were determined according to Rodriguez-Amaya and Kimura [15]. The extraction was carried out with acetone, followed by the petroleum ether partition and the UV-visible spectrophotometer reading at 450 nm. The results were expressed in µg of carotenoids per g of pulp.

The content of vitamin C was determined by the reaction of ascorbic acid with 2.6-dichlorophenol indophenol (DCFI), according to the procedure described by Oliveira [16], expressed in mg of ascorbic acid per 100 g of pulp.

For the determination of TEP and antioxidant activity, papaya extracts were prepared as described by Larrauri *et al.* [17], with modifications. The extraction was done with 50% methanol for a period of 20 minutes in an ultrasonic bath, followed by centrifugation at 11,000 rpm for 15 minutes. The residue was subjected to the second extraction in 70% acetone, following the same procedure. The extracts were mixed and transferred to the 50 mL volumetric flask, the volume was filled with distilled water. The TEP contents were determined using the Folin-Ciocalteau reagent (1:3) and the standard curve of gallic acid, according to the method described by Singleton *et al.* [18] and following the modifications proposed by Rufino *et al.* [19].

The total antioxidant activity was determined by capture of the radical ABTS $^+$ according the methodology proposed by RE *et al.* [20], with modifications. The decrease in absorbance was measured after 6 minutes of reaction and the reading at 734 nm. The results were expressed as μ M trolox/g of fruit, from trolox standard curve.

2.5. Stability of Dried Papaya

The color, moisture, pH, titratable acidity, water activity, total carotenoids and vitamin C were determinated at zero time and every 30 days of storage for a period of 120 days for the stability of the dried papaya. Degradation of the total carotenoids and vitamin C during storage followed the first-order reaction, and the data were evaluated using the following model:

$$C = C_0 e^{-kt}$$

where C_0 in C that represent component concentrations at zero and t time, respectively, and k is the reaction rate constant (time⁻¹).

2.6. Sensory Analysis

The study was approved by the Research Ethics Committee of Maria Milza College, located in Governador Mangabeira, Bahia, Brazil (authorization n° 246/11). The sensory test was performed in individual cabins under white light by fifty not trained judges, with age between 18 and 60 years. The tests were carried out at 30-day storage. The acceptance test was made using a structured 9-points hedonic scale, in which 1 represents "dislike extremely" and 9 represents "like extremely". The attributes intensity of color, aroma, flavor and texture were evaluated using a nine-point intensity scale as described in Meilgaard *et al.* [21].

2.7. Statistical Analysis

All drying processes and all analysis were repeated three times. The one-way analysis of variance (ANOVA) was used to compare the two treatments of dried papaya (T1 and T2) at 5% significance level, by F test. Results were expressed as mean \pm standard deviation.

The study of stability was performed in the completely randomized design in a split plot system. Two treatments (T1 and T2) constituted the plots and five sto-

rage times (0, 30, 60, 90, 120 days) constituted the subplots. The data were analyzed by ANOVA at 5% significance level and regression analysis using Sisvar and R programs.

3. Results and Discussion

3.1. Physicochemical Characteristics of Fresh and Dried Papaya

The results of characterization of fresh and dried papaya (zero storage time) were shown in Table 1. The antioxidant treatment with citric acid and ascorbic acid (T2) used in papaya slices had no significant effect (p > 0.05) in physicochemical characteristics and in content of bioactive compounds when compared to the control treatment (Table 1). The dried papaya presented, on average, 23.42% moisture and water activity 0.57. In general, these low values of moisture and water activity are sufficient to guarantee the microbiological stability of dehydrated products. According Sagar and Kumar [10], the pathogenic bacteria do not grow in products with a_w below 0.85, while fungi and yeasts are more tolerant and, to inhibit them, the product must have a_w lower than 0.62. Therefore, the dried papaya obtained in this study can be considered as a microbiologically stable product.

Table 1. Physicochemical characteristics of fresh and dried papaya (zero storage time)—values on a wet basis.

N · l · l Cl · · · ·	Fresh	Dried Papaya			
Physicochemical Characteristics	Papaya	T1 ⁽¹⁾	T2 ⁽²⁾	Average ⁽³⁾	
Humidity (%)	88.82 ± 0.85	20.06 ± 5.89^{a}	26.78 ± 4.71 ^a	23.42 ± 4.75	
Titratable acidity, TA (% citric acid)	0.10 ± 0.01	0.94 ± 0.04^{a}	0.83 ± 0.09^{a}	0.89 ± 0.08	
pH	5.11 ± 0.13	5.15 ± 0.17^{a}	4.98 ± 0.22^{a}	5.06 ± 0.12	
Soluble solids, SS (°Brix)	10.78 ± 0.55	156.43 ± 26.01 ^a	123.92 ± 21.94 ^a	140.18 ± 22.98	
SS/TA	106.69 ± 14.02	166.47 ± 26.76^{a}	150.44 ± 33.12 ^a	158.46 ± 11.34	
$a_{ m w}$	0.94 ± 0.02	0.55 ± 0.03^{a}	0.58 ± 0.09^{a}	0.57 ± 0.02	
Total carotenoids (μg/g)	36.38 ± 1.61	219.99 ± 7.25^{a}	218.92 ± 20.95^{a}	219.46 ± 0.76	
Vitamin C (mg/100g)	53.84 ± 4.88	375.54 ± 18.27 ^a	269.96 ± 66.92 ^a	322.61 ± 74.85	
TEP (mg GAE/100g)	47.04 ± 1.17	$274.70 \pm 107.80^{\circ}$	263.50 ± 91.13 ^a	269.09 ± 7.92	
ABTS (μM trolox/g)	4.35 ± 0.34	8.27 ± 2.69^a	7.44 ± 2.58^{a}	7.86 ± 0.59	
L*	53.68 ± 2.38	55.08 ± 6.00^{a}	51.35 ± 4.43^{a}	53.22 ± 2.64	
a*	26.85 ± 0.48	37.35 ± 1.46^{a}	35.52 ± 3.71 ^a	36.44 ± 1.30	
b*	41.88 ± 2.36	49.98 ± 6.66^{a}	47.79 ± 4.29^{a}	48.89 ± 1.54	
C*	49.92 ± 1.86	62.68 ± 5.34^{a}	59.79 ± 5.22^{a}	61.24 ± 2.04	
h*	57.29 ± 1.79	52.97 ± 3.82^a	52.99 ± 1.02^a	52.98 ± 3.82	
ΔE^*	-	11.69 ± 0.74^{a}	12.51 ± 2.60^{a}	12.10 ± 0.60	

Means followed by the same letter in the same row are not significantly different (p > 0.05) by F test. ⁽¹⁾Control Treatment; ⁽²⁾Immersion in aqueous solution containing 25 mg/L citric acid + 75 mg/L ascorbic acid for 10 minutes. ⁽³⁾Average values of treatments T1 and T2; a_w: water activity; TEP: total extractable polyphenols; ABTS = antioxidant activity by the ABTS method; $\Delta E * = total$ variation of color dried papaya compared to fresh papaya.

After drying process, the solids and organic acids of papaya were concentrated. The SS/TA increased 48.52%, which may positively influence in the acceptance of the dried papaya by consumers, since products with higher SS/TA have more pronounced sweetness. There were also increased levels of bioactive compounds in dried papaya, where values were approximately six times higher than those presented in fresh one. The increase in concentration of these compounds, promoted the increase of its antioxidant activity, which passed from 4.35 for fresh to 7.86 μ M trolox/g for dried papaya (T1 and T2 average) (**Table** 1).

The use of the citric acid and ascorbic acid combination did not have a significant effect (p > 0.05) on coloration of dried papaya (Table 1). By comparing the chromatic attributes of fresh and the dried papaya (T1 and T2 average), it was found that the lightness of papaya (L*) did not change during the drying process, indicating that there was no browning of the product. However, the dried papaya showed a more intense and more reddish color than the fresh papaya, which can be seen by the higher values for a* and C* and lower values for h*. These results may be associated with the concentration of papaya carotenoids after drying, especially lycopene, which is the main carotenoid on fresh papaya and responsible for the red color of the fruit [22]. Mishra et al. [5] also observed an increase in the values of a* coordinate in dehydrated papaya by using a combination of osmotic dehydration and infrared drying. However, different from that observed in this study, L* decreased and the product presented a reddish brown coloration. Different results from this study was also observed by Canizares and Mauro [23], in which papayas dehydrated by convective drying at 60°C and 70°C showed more yellowish color and less intense than the fresh papaya.

The total color variation, ΔE^* , was 12.10 (**Table 1**), a value close to that reported by Chong *et al.* [9], which was 10.36, in papaya dried by hot-cold air technology. Despite close values, in the study conducted by Chong *et al.* [9], papaya color changed from deep red to yellow-brown, an opposite result to the present study. According to these authors, the use of high drying temperatures, above 70°C, in spite of promoting the inactivation of the PPO enzyme, favor the oxidation of papain enzyme, releasing free amino acids that react with the reducing sugars of the papaya, triggering the Maillard reaction. Therefore, the drying temperature used in this study (60°C) was satisfactory, allowing color preservation and obtaining a dried product with high concentration of bioactive compounds.

3.2. Degradation of the Bioactive Compounds of Papaya after Drying

When comparing the contents of the bioactive compounds on dry basis (**Table 2**), some of these compounds were lost after drying. There was loss of 17.69% for TEP and dried papaya showed an average of 347.36 mg GAE/100g. Chong *et al.* [9], evaluate the effect of different papaya dehydration methods and found losses

Table 2. Degradation of the bioactive compounds of papaya after drying process (zero storage time)—values on a dry basis.

Bioactive compounds	Fresh Papaya –	Dried Papaya			0/ D 1 - 4: (4)	% Retention ⁽⁵⁾
		T1 ⁽¹⁾	T2 ⁽²⁾	Average ⁽³⁾	% Degradation ⁽⁴⁾	% Retention
TEP (mg GAE/100g)	422.01 ± 26.72	338.31 ± 111.07a	356.40 ± 106.18a	347.36 ± 12.79	17.69	82.31
Vitamin C (mg/100g)	481.23 ± 8.37	471.53 ± 42.33a	$365.63 \pm 72.45a$	418.58 ± 74.88	13.02	86.98
Carotenoids (μg/g)	329.53 ± 4.00	264.79 ± 17.25a	$305.23 \pm 5.92a$	285.00 ± 28.59	13.51	86.95
ABTS (μM trolox/g)	39.12 ± 5.22	10.42 ± 3.60a	10.32 ± 4.18a	10.37 ± 0.07	-	-

Means followed by the same letter in the same row are not significantly different (p > 0.05) by F test. (1) Control Treatment; (2) Immersion in aqueous solution containing 25 mg/L citric acid + 75 mg/L ascorbic acid for 10 minutes; (3) Average values of treatments T1 and T2; (4) Calculated for average value of treatment T1 and T2 in relation of fresh papaya; (5) Calculated by difference between 100 and the % of degradation.

from 7% to 69% for phenolic content, and the dried product had contents from 84.78 to 250 mg GAE/100g, lower values than the results of this study.

Degradation of vitamin C during the drying process was only 13.02% (**Table 2**), representing 86.98% of retention of vitamin C in the dried product (T1 and T2 average). These values were similar to those reported by Garcia *et al.* [24], which found a retention of 83.03% and 87.30%, in dehydrated papaya at 60°C and 70°C, respectively. Hawlader *et al.* [25] investigated the dehydration of guava and papaya using the same drying conditions of the present study and verified that 75% of vitamin C was retained in papaya and only 25% in guava. This result suggests that vitamin C degradation is dependent on the food matrix and that papaya may present a natural barrier to oxidation during drying [24]. Another factor that interferes on the degradation of bioactive compounds is the drying parameters used. Kurozawa *et al.* [26] observed that increasing air velocity promotes a greater retention of vitamin C in dried papaya at 60°C, a fact that is associated with the shorter time papaya is exposed to high temperatures. This may explain the high vitamin C retention of dried papaya in this study.

For carotenoids, the drying process employed promoted the 13.51% degradation. Factors such as high temperature, presence of light, oxygen, acids and metals can promote the degradation of these pigments during the drying process. However, Udomkun et al. [27], tested several drying temperature combinations (50°C, 60°C, 70°C and 80°C), air velocity (0.2, 0.5 and 0.7 m/s) and air specific humidity (10 and 25 g water/kg of dry air), in papaya dehydration process, and found losses of 5.7% to 75% of β -carotene content, and from 34.66% to 76.41% of lycopene content, but did not find a direct relationship between these variables and the degradation of the evaluated carotenoids. The low degradation of carotenoids during the papaya drying may be related to the moisture reduction of the product. According to Goula et al. [28], the kinetics of lycopene degradation depends on product moisture, and when the moisture content is high (55% - 95%), chemical reactions occur faster than in low moisture content (below 55%). This is due to the greater solubilization and mobility of the catalytic substances in the reaction. Di Scala and Crapiste [29] verified the same behavior during the drying of red peppers.

3.3. Stability of Dried Papaya

3.3.1. Physicochemical Changes during Storage

The stability of the dried papaya during storage can be observed in **Figures 1-3**. No significant difference (p > 0.05) was observed between the control treatments (T1) and with addition of 25 mg/L citric acid + 75 mg/L ascorbic acid (T2) during storage. However, the physicochemical caracteristics of dried papaya changed significantly (p < 0.05) during the storage (**Figure 1**). The acidity increased and the soluble solids content reduced, leading to SS/TA reduction. The SS/TA decreased from 158.46 to 53.30, thus compromising the quality of the dried papaya (**Figures 1(a)-(c)**). Probably this reduction was due to reaction between reducing sugars present in papaya and free amino groups resulting in the formation of intermediate compounds such as CO_2 and acids (Maillard reaction).

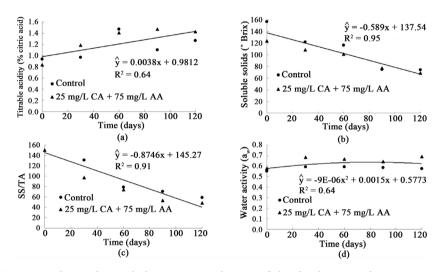


Figure 1. Physicochemical characteristics changes of the dried papaya during storage. Lines represent the linear model adjusted. CA: citric acid; AA: ascorbic acid.

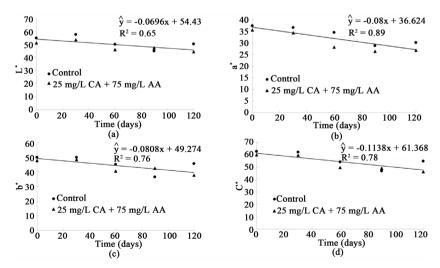


Figure 2. Color changes of the dried papaya during storage. Lines represent the linear model adjusted. CA: citric acid; AA: ascorbic acid.

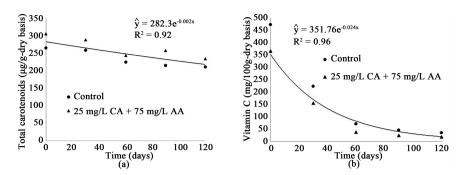


Figure 3. Degradation of total carotenoids (a) and vitamin C (b) as a function of storage time. Lines represent the exponencial model adjusted. CA: citric acid; AA: ascorbic acid.

Dried papaya moisture remained constant during storage, with a mean value of 23.20%, but water activity increased to 0.64 (Figure 1(d)). This may have occurred because, in products with low water activity, between 0.20 and 0.60, small amounts of absorbed water may reflect the increase in water activity, a fact that is represented by the sorption isotherm of dehydrated foods [30].

3.3.2. Color Changes during Storage

The dried papaya darkened during storage, probably due to the Maillard reaction, a fact that can be observed by the reduction in L* values (Figure 2(a)). One of the products of this reaction is hydroxymethylfurfural, which polymerizes with compounds containing amine groups, under less acidic conditions (pH > 5.0), giving rise to brown pigments called melanoidines. The increase in water activity also favors this reaction, which reaches the maximum between 0.6 and 0.7 [31]. Udomkun *et al.* [6] also observed the browning of the dried papaya after nine months of storage. Another factor that may have contributed to the browning of papaya is the dehydroascorbic acid oxidation which hydrolyzes to form 2,3-diacetogulonic acid. This substance undergoes polymerization with amino acids such as lysine, glutamic acid and others, promoting the dark pigment formation [9]. According to Hymavathi and Khader [32], the degradation of β -carotene can also results in the generation of compounds from the Maillard reaction, leading to the browning of the product.

The coordinates a^* and b^* and the C^* attribute also reduced significantly (p < 0.05) and the papaya presented a less intense reddish coloration (**Figures 2(b)-(d)**) after 120 days. On the other hand, the value of h^* remained constant during storage (data not shown). The reduction of these coordinates occurred due to the carotenoids degradation, since the papaya color is the result of β -carotene, α -carotene, cryptoxanthin and lycopene presence.

3.3.3. Degradation of the Total Carotenoids and Vitamin C during Storage

Degradation of the total carotenoids during storage followed the first-order reaction (Figure 3(a)). After 30 days, there was a reduction of approximately 6% of total carotenoid content and after 120 days this reduction was 21%. Probably

the beginning of the oxidative process occurred during the drying process and with the formation of free radicals the reaction was triggered even in the absence of light. According to Reis *et al.* [12], besides the oxidative process, the heating that occurs during the dehydration process can promote the redistribution of the cellular food constituents, with the release of acids. These conditions favor the isomerization of the carotenoids from the transform to the cis and these isomers may also undergo oxidation. In addition to color loss, the isomerization and oxidation process results in loss of the pro-vitamin A activity of some carotenoids, such as β -carotene [33]. The carotenoid content reduced from 282.3 μ g/g to 222.06 μ g/g and the half-life was 346 days, which indicates that the degradation of these pigments occurred very slowly. Therefore, the use of TEP packaging was efficient in preserving these pigments during storage.

Vitamin C is often used as a quality index during food processing and storage [24]. It is seen in Figure 3(b) that degradation of this vitamin during storage following the first-order reaction. The vitamin C content of the dried product reduced from 351.76 mg to 19.75 mg vitamin C/100g at 120 days, which represents a reduction of about 94%. The half-life was 29 days, that is, after this period the vitamin C content was halved. Therefore, dried papaya should be consumed within 30 days of storage to be considered a good nutritional quality product. This result corroborates the values reported by Udomkum *et al.* [6] in which the half-life time was approximately 30 days in osmotically dehydrated papaya followed by convective drying at 70°C. In a study carried by Canizares and Mauro [23], the vitamin C contents retention in dehydrated papayas at temperatures of 60°C and 70°C, was 30.28% and 48.34% respectively, after 30 days of storage, values close to the present study.

3.4. Sensory Acceptance of Dried Papaya

There was no significant difference (p > 0.05) between the two treatments (T1 and T2) regarding sensorial acceptance (**Table 3**). Sensory acceptance was

Table 3. Sensory acceptance and attribute intensity of dried papaya.

Attributes	Acceptance Test(1)			Attribute Intensity ⁽²⁾		
	T1 ⁽³⁾	T2 ⁽⁴⁾	Average ⁽⁵⁾	T1 ⁽³⁾	T2 ⁽⁴⁾	Average ⁽⁵⁾
Color	5.82 ± 1.79a	5.34 ± 2.03a	5.58 ± 0.34	6.18 ± 1.77a	6.66 ± 1.98a	6.42 ± 0.34
Aroma	6.12 ± 1.61a	5.44 ± 1.51a	5.78 ± 0.48	$5.04 \pm 2.05a$	$5.48 \pm 2.18a$	5.26 ± 0.31
Flavor	5.42 ± 1.95a	5.94 ± 1.88a	5.68 ± 0.37	5.30 ± 1.90a	5.42 ± 1.99a	5.36 ± 0.08
Texture	$4.90 \pm 2.17a$	$5.34 \pm 2.22a$	5.12 ± 0.31	6.82 ± 1.73 a	$6.20 \pm 1.67a$	6.51 ± 0.44
Overall	5.70 ± 1.78a	5.36 ± 1.84a	5.53 ± 0.24	-	-	-

Means followed by the same letter in the same row are not significantly different (p > 0.05) by F test. ⁽¹⁾mean hedonic score (n = 50) according nine-point scale, with the terms "dislike extremely" (1) and "like extremely" (9). ⁽²⁾Attribute intensity according to nine-point scale: very light color (1) to very dark (9); weak aroma (1) to strong aroma (9); weak flavor (1) to strong flavor (9); very soft texture (1) very hard texture (9). ⁽³⁾Control Treatment; ⁽⁴⁾Immersion in aqueous solution containing 25 mg/L citric acid + 75 mg/L of ascorbic acid for 10 minutes. ⁽⁵⁾Average of treatments T1 and T2.

intermediate and the dried papaya was classified as "neither like nor dislike" and "liked slightly". A similar result was obtained Canizares and Mauro [23] for papayas coated with pectin and dehydrated at 60°C, which received a mean score of 5.30, and papayas coated with pectin added vitamin C that received a mean score of 5.36, for the overall acceptance attribute.

Regarding intensity scale, dried papaya was considered dark (6.42), a fact confirmed by the values of C^* , h^* , a^* and ΔE (**Table 1**). The aroma and taste were considered intermediate (neither strong nor weak), probably due to the loss of part of the volatile compounds during the drying process. The note for the texture was 6.51, indicating that the dried papaya presented a hard texture. Such results may explain the intermediate acceptability of dried papaya.

4. Conclusions

The degradation of total carotenoids and vitamin C followed the first order reaction, and the half-life time was 346 days for carotenoids and 29 days for vitamin C.

After 120 days of storage, the dried papaya darkened and presented less intense reddish coloration. The vitamin C content reduced 94% and the total carotenoids content reduced 21%.

Sensory acceptance was intermediate and the dried papaya was classified as "neither like nor dislike" and "liked slightly". The aroma and taste were considered intermediate (neither strong nor weak) and the dried papaya presented a hard texture.

Dried papaya must be consumed within 30 days of storage to be considered a good nutritional quality product.

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