

PAHs in Sediments along the Semi-closed Areas of Alexandria, Egypt

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ABSTRACT

Sediment samples were collected from 49 sampling stations along the semi-closed areas of Alexandria coasts, Egypt. Total concentrations of 15 out of 16 EPA-PAHs in sediments were varied from 4.2 to 886 ng·g⁻¹ with an average value of 176 ng·g⁻¹ (dry wt). The average total organic carbon (TOC) percent was varied from 0.04 to 7.65%. Higher concentration of total pyrolytic hydrocarbons (Σ COMB) than total fossil hydrocarbons (Σ COMB), declared that atmospheric fall-out is the significant source of PAHs to marine sediments of the semi-closed area of Alexandria. The selected marked compounds and special PAHs compound ratios (phenanthrene/anthracene; fluoranthene/pyrene; Σ COMB/ Σ EPA-PAHs) suggest the pyrogenic origins, especially traffic exhausts, are the dominant sources of PAHs in most locations. Interferences of rather petrogenic and pyrolytic PAH contaminations were noticed in the harbours, especially marine area due to petroleum products deliveries and fuel combustion emissions from the ships staying alongside the quays.

Keywords: Surface Sediment, PAHs, Alexandria, Egypt, GC-MS

1. Introduction

Polycyclic aromatic hydrocarbons (PAHs) have been described as mutagenic, carcinogenic and teratogenic included in the US EPA and the EU priority pollutants list [1]. PAHs solubility decrease with increasing molecular weight [2]. Consequently, the uptake of a contaminant is governed by its bioavailability, and organisms are often enriched in the lower molecular weight PAHs relative to the sediment [3]. Once deposited in sediments, PAHs are less subjected to photochemical or biological oxidation, especially if the sediment is anoxic. Thus, sedimentary PAHs tend to be persistent and may accumulate to high concentrations [4]. Numerous research studies assessed the PAHs inputs in the North-western and Central Mediterranean [5]. Conversely, in the Eastern Mediterranean few data have been published on the presence of PAHs in coastal sediments close to point sources (municipal and river discharges, etc.) [6]. In addition, hydrocarbon budgets are available for the Western part of Mediterranean Sea [7], but there is a tremendous lack of information regarding the Southern Mediterranean [8].

This is the first systematic study to be undertaken along the semi-closed areas of Alexandria (Egyptian Mediterranean Sea coast) as a comparative study between four sectors of different identity. Two relevant criteria are used fairly to discriminate between the various natural and anthropogenic hydrocarbon inputs and to evaluate the extent of hydrocarbon pollution in the area and its link to combustion processes and/or releasing of unburned fossil fuels.

2. Materials and Methods

A total of 49 surface sediment samples were collected during January 2010 at sites shown in **Figure 1**. Sediments were collected utilizing a stainless-steel grab. Six grabs were taken from each location from which the top 3 cm were scooped into pre-cleaned wide-mouth glass bottles, frozen and transported to the laboratory and stored at -20°C until analysis. The samples were analyzed for PAHs following well established techniques [9].

To control the analytical reliability and assure recovery efficiency and accuracy of the results, 6 analyses

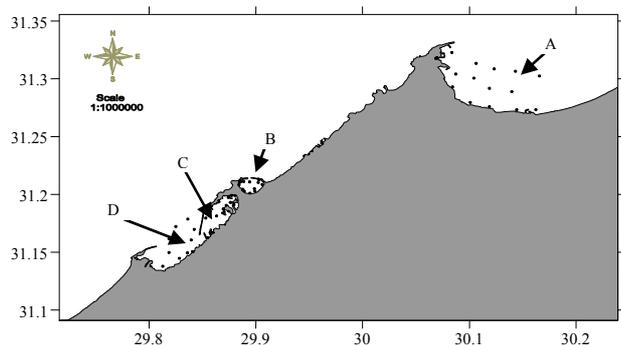


Figure 1. Sampling locations collected from the area of investigation: (A) Abu Qir Bay, (B) Eastern Harbour, (C) Western Harbour and (D) El Max Bay during 2010.

were conducted on PAH compound reference materials, HS-5 (sediments) provided by NRC-IMB of Canada and SRM-2974 (Freeze-dried mussels tissue) (*Mytilus edulis*) provided by NIST of USA as well as sediment samples of known PAH levels spiked with a mixture consisting of 2 μg each of PAHs were analyzed as above to validate the analytical method used in this study. The lowest DL was 0.01 $\mu\text{g}/\text{ml}$ for lower molecular mass compounds while indeno[1,2,3-cd]pyrene has the highest at 0.1 $\mu\text{g}/\text{ml}$. The recovery efficiency ranged from 92% to 111% for HS-5, 88% to 96% for SRM-2974 and 93% to 105% for the spiked samples. Samples were analyzed by Gas Chromatograph-Mass Spectrometer; GC-MS (Trace DSQ II MS) with fully scanned 50 - 650 Daltons per second mode and 70 eV electron energy for confirmation. GC/MS is equipped with split/split less injector and a fused silica capillary column; Thermo TR-35 MS (30 m, 0.25 mm, 0.25 μm) with 35% phenyl polysilphenylene-siloxane. Helium was used as carrier gas at 1.5 $\text{mL}\cdot\text{min}^{-1}$. The temperature was programmed from 60°C - 100°C with rate of 8°C·min⁻¹, then maintained at 100°C for 1min, and from 100 - 300°C with rate of 5°C·min⁻¹ and was maintained at 300°C for 20 min. The injector and detector temperatures were set at 280°C and 300°C, respectively. Three μL volume of each sample was injected in the split less mode and the purge time was 1 min. The response factor of individual PAH compounds to the internal standard was measured and calculated at least three times at the beginning, in the middle, and at the end for each batch of GC injections (15 samples).

3. Results and Discussion

Individual and total concentrations, as well as characteristic ratios for the identification of PAH origins are given in **Tables 1** and **2**. Total PAHs (ΣPAHs) ranged from 4.2 to 886.08 with an average of 169.37 $\text{ng}\cdot\text{g}^{-1}$ dry weight. The highest concentration of total PAHs was recorded in

sediments collected from station 28 (Eastern Harbour), followed by that in stations 1, 26 and 34. Lower concentrations were detected in samples of stations 11 - 15. In this study, sediment samples collected near the sewage outlet, cities and harbor appeared to have extremely high concentrations of total PAHs. These suggest that PAHs accumulated in Mediterranean Sea sediments came from different sources such as sewage discharge from nearby human activities and fuel combustion emissions. Although the number of PAHs analyzed in any given study may differ, the 15 compounds analyzed herein comprise the vast majority of PAHs found in most estuarine or marine sediments. It has been demonstrated that the nature of the sediment influences the distribution and concentration of PAHs. Sediments with high organic carbon content were characterized with high values of PAHs [10]. However, a moderate correlation of $r = 0.646$ was found between ΣPAHs and total organic carbon (TOC) concentrations in the present sediments (**Table 1, Figure 2**). This may be suggesting that both of direct input and type of sediment found locally would determine the distribution and concentrations of PAHs in sediments. Moreover, the relationship between total PAHs and organic carbon was only significant for highly contaminated sites where total PAH concentration was greater than 2000 $\text{ng}\cdot\text{g}^{-1}$ dry wt. In this study, none of the sediment samples had total PAH concentrations more than 2000 $\text{ng}\cdot\text{g}^{-1}$ where the samples contain lower TOC.

Chiou *et al.* [10] found that the high partitioning of PAHs to sedimentary organic matter was mainly due to the significant aromatic fraction of the organic matter. All locations had concentrations lower than the Effects Range-Low (ERL) value (4022 $\text{ng}\cdot\text{g}^{-1}$) suggested by Long *et al.* [11]. They reported that the concentrations below the ERL value represent a minimal-effect range, *i.e.* adverse biological effect would rarely be observed below the ERL. On the other hand, if the concentration was higher than the Effects Range-Median (ERM) value (44792 $\text{ng}\cdot\text{g}^{-1}$), adverse effects on biological systems will frequently occur. The concentrations of individual PAH recorded in the present study ranged from 4 to 886 $\text{ng}\cdot\text{g}^{-1}$, and were much lower than both of the ERL and ERM limit (**Table 3** and **Figure 3**). In addition, similar observations were found relative to the Threshold Effect Level (TEL) and Probable Effect Level (PEL) [11], where Dibenzo(a,h)anthracene, Acenaphthylene and acenaphthene have average concentrations > TEL and < PEL (**Figure 4**). The individual PAH concentrations in this study were also lower than the national sediment quality criteria proposed by USEPA [12] for fluoranthene (3000 $\text{ng}\cdot\text{g}^{-1}$), acenaphthylene (2400 $\text{ng}\cdot\text{g}^{-1}$) and phenanthrene (2400 $\text{ng}\cdot\text{g}^{-1}$). Benzo(a)pyrene (BaP), the most potent carcinogenic PAHs, and the sum of six carcinogenic PAHs

Table 1. Concentrations (ng/g; dry wt) of PAHs in sediment samples collected from the studied area during 2010.

Parameter	PAHs	Ph/An	Flu/Py	BaA/Chr	ΣCARC	%CARC	ΣTFPAH	%ΣTFPAH	ΣCOMP	%ΣCOMP	ΣTFPAH/ ΣCOMP	BkF/BaP	BbF/BaP	%TOC
Site														
1	460.96	3.05	0.33	0.71	59.42	6.41	3.27	1.67	457.69	5.65	0.01	5.18	0.55	7.65
2	44.57	0.05	1.00	0.29	6.62	0.71	0.99	0.51	43.58	0.54	0.02	5.47	2.29	0.62
3	31.88	0.11	1.00	4.00	10.79	1.16	3.86	1.97	28.02	0.35	0.14	2.52	2.19	1.13
4	80.22	1.62	3.00	1.00	4.70	0.51	1.60	0.82	78.62	0.97	0.02	1.09	0.14	1.11
5	62.16	1.62	1.00	1.00	2.45	0.26	0.59	0.30	61.57	0.76	0.01	1.08	0.85	0.23
6	23.96	0.08	0.00	1.00	4.46	0.48	1.16	0.59	22.79	0.28	0.05	2.29	2.00	0.31
7	29.13	0.06	1.00	2.00	1.15	0.12	0.67	0.34	28.46	0.35	0.02	8.83	0.17	0.49
8	35.83	0.06	0.33	0.50	7.28	0.78	0.29	0.15	35.53	0.44	0.01	2.17	1.83	0.12
9	102.18	1.62	1.00	0.50	2.77	0.30	1.49	0.76	100.69	1.24	0.01	2.89	1.56	0.43
10	17.08	0.06	0.00	1.00	4.30	0.46	0.70	0.36	16.38	0.20	0.04	1.60	2.80	0.04
11	9.30	1.67	0.00	0.00	1.70	0.18	0.30	0.15	9.00	0.11	0.03	12.40	2.00	0.35
12	12.60	1.00	NR	NR	6.60	0.71	0.50	0.26	12.10	0.15	0.04	1.96	1.56	0.31
13	4.20	2.00	NR	0.00	2.20	0.24	0.20	0.10	4.00	0.05	0.05	1.00	0.57	0.12
14	10.60	3.50	NR	0.20	3.90	0.42	0.20	0.10	10.40	0.13	0.02	2.00	0.32	0.23
15	7.40	2.33	NR	1.00	4.60	0.50	0.50	0.26	6.90	0.09	0.07	0.33	0.50	0.23
AV	62.14	1.26	0.79	0.94	8.20	0.88	1.09	0.56	61.05	0.75	0.04	3.39	1.29	0.89
Max	460.96	3.50	3.00	4.00	59.42	6.41	3.86	1.97	457.69	5.65	0.14	12.40	2.80	7.65
Min	4.20	0.05	0.00	0.00	1.15	0.12	0.20	0.10	4.00	0.05	0.01	0.33	0.14	0.04
16	56.9	0.71	14.00	0.04	11.37	1.23	0.58	0.30	56.37	0.70	0.01	3.70	0.29	1.56
17	268.9	1.62	0.37	0.02	4.52	0.49	1.32	0.67	267.58	3.30	0.00	6.57	1.43	2.24
18	183.7	3.44	24.50	0.09	8.30	0.89	5.40	2.76	178.30	2.20	0.03	3.00	0.86	2.13
19	130.20	1.62	23.00	0.33	18.33	1.98	1.86	0.95	128.33	1.58	0.01	1.74	0.32	0.59
20	78.00	0.03	1.31	0.33	8.94	0.96	0.98	0.50	77.01	0.95	0.01	3.68	0.27	0.45
21	36.40	0.03	0.95	2.00	3.91	0.42	2.94	1.50	33.46	0.41	0.09	0.87	0.61	0.21
22	192.79	1.62	1.36	0.63	7.02	0.76	2.08	1.06	190.71	2.35	0.01	17.44	1.19	0.44
23	215.11	0.02	1.42	0.04	40.46	4.36	0.70	0.36	214.41	2.65	0.00	1.72	0.02	0.69
24	182.12	4.87	28.00	0.50	3.93	0.42	1.28	0.65	180.84	2.23	0.01	3.61	0.48	0.50
25	170.94	1.61	0.00	0.50	1.36	0.15	2.82	1.44	168.12	2.07	0.02	9.00	2.33	0.49
26	339.13	4.86	1.31	1.33	27.17	2.93	1.19	0.61	337.94	4.17	0.00	1.84	0.19	1.02
27	196.88	0.04	0.01	0.02	50.15	5.41	1.75	0.90	195.13	2.41	0.01	2.06	0.08	1.22

28	886.1	6.04	1.24	0.85	143.47	15.46	0.57	0.29	885.51	10.93	0.00	1.95	0.03	3.18
AV	225.9	2.0	7.5	0.5	25.3	2.7	1.8	0.9	224.1	2.8	0.0	4.4	0.6	1.1
Max	886.1	6.0	28.0	2.0	143.5	15.5	5.4	2.8	885.5	10.9	0.1	17.4	2.3	3.2
Min	36.4	0.0	0.0	0.0	1.4	0.1	0.6	0.3	33.5	0.4	0.0	0.9	0.0	0.2
29	266.4	1.62	4.20	0.04	23.68	2.55	2.70	1.38	263.74	3.25	0.01	2.22	0.25	2.67
30	292.8	2.79	1.40	0.23	46.50	5.01	14.80	7.56	278.00	3.43	0.05	2.75	0.43	6.00
31	47.8	0.06	1.00	1.50	4.92	0.53	1.98	1.01	45.79	0.57	0.04	3.50	1.35	0.36
32	356.9	0.62	1.26	11.94	78.00	8.41	0.90	0.46	356.00	4.39	0.00	2.59	1.55	2.71
33	184.8	0.03	1.20	4.17	45.80	4.94	3.40	1.74	181.40	2.24	0.02	2.12	0.51	3.96
34	542.3	0.53	4.88	0.08	11.20	1.21	2.90	1.48	539.40	6.66	0.01	3.60	1.03	2.89
35	205.3	0.01	1.08	0.12	7.29	0.79	1.40	0.71	203.89	2.52	0.01	8.62	2.15	1.82
36	56.7	0.02	4.25	0.83	13.00	1.40	2.20	1.12	54.50	0.67	0.04	1.63	0.41	0.63
37	719.0	0.68	9.71	0.80	153.80	16.58	2.40	1.23	716.60	8.84	0.00	2.33	0.07	3.58
38	157.6	42.11	3.63	7.85	36.60	3.94	1.90	0.97	155.70	1.92	0.01	14.05	2.74	2.81
39	181.2	0.06	3.26	0.07	21.19	2.28	2.39	1.22	178.81	2.21	0.01	2.07	0.08	3.09
AV	273.7	4.4	3.3	2.5	40.2	4.3	3.4	1.7	270.3	3.3	0.0	4.1	1.0	2.8
Max	719.0	42.1	9.7	11.9	153.8	16.6	14.8	7.6	716.6	8.8	0.1	14.1	2.7	6.0
Min	47.8	0.0	1.0	0.0	4.9	0.5	0.9	0.5	45.8	0.6	0.0	1.6	0.1	0.4
40	73.3	0.38	0.10	NR	2.98	0.32	24.38	12.46	48.90	0.60	0.50	3.33	1.44	0.49
41	281.9	1.62	1.00	NR	6.57	0.71	1.03	0.52	280.90	3.47	0.00	1.17	3.67	2.17
42	200.9	1.47	1.50	1.00	2.52	0.27	0.80	0.41	200.13	2.47	0.00	1.00	0.70	0.93
43	201.1	1.74	1.24	1.00	1.62	0.17	1.89	0.97	199.19	2.46	0.01	27.50	3.50	1.26
44	119.5	2.12	0.80	1.00	3.40	0.37	0.85	0.43	118.69	1.46	0.01	1.78	1.67	0.30
45	100.2	0.11	1.00	2.00	3.02	0.33	44.92	22.96	55.24	0.68	0.81	2.29	1.43	1.54
46	226.6	1.65	0.77	1.00	3.13	0.34	32.62	16.67	193.95	2.39	0.17	1.73	0.80	0.61
47	70.8	0.19	3.80	1.00	3.20	0.35	6.40	3.27	64.40	0.79	0.10	1.42	0.75	0.73
48	22.9	2.77	1.50	NR	3.54	0.38	1.27	0.65	21.65	0.27	0.06	1.00	0.33	0.30
49	121.8	3.42	2.00	1.00	3.93	0.42	4.77	2.44	117.06	1.44	0.04	1.63	4.75	0.42
AV	141.9	1.5	1.4	1.1	3.4	0.4	11.9	6.1	130.0	1.6	0.2	4.3	1.9	0.9
Max	281.9	3.4	3.8	2.0	6.6	0.7	44.9	23.0	280.9	3.5	0.8	27.5	4.8	2.2
Min	22.91	0.11	0.10	1.00	1.62	0.17	0.80	0.41	21.65	0.27	0.00	1.00	0.33	0.30

Abu Qir Sector from 1 - 15; Eastern Harbour sector: 16 - 28, Western Harbour sector: 29 - 39, El Max sector: 40 - 49, Av: average, Max: maximum, Min: minimum, PAHs: Sum of aromatic hydrocarbons, Phe: phenanthrene, An: anthracene, Flu: Fluoranthene, Py: pyrene, BaA: Benzo(a)anthracene, Chr: Chrysene, BkF: benzo(k)fluoranthene, BaP: benzo(a)pyrene, BbF: benzo(b)fluoranthene, Σ CARC: Sum of BaA + BbF + BkF + BaP + DBA + InP, Σ TFFPAH: Sum Acenaphthylene + Acenaphthene, Σ COMP: Sum of all PAHs except those calculated for Σ TFFPAH.

Table 2. Concentrations (ng/g; dry wt) of individual PAHs in sediment samples during 2010.

Compound		Concentration, ng/g														
Site	Naph	Acthy	Ace	Phe	Ant	Flu	Pyr	BaA	Chr	BbF	BkF	BaP	InP	DBA	B(ghi)P	
Abu Qir Bay	1	0.00	1.92	1.35	196.7	64.42	3.65	10.96	33.46	46.92	7.88	73.65	14.23	0.58	3.27	1.92
	2	0.00	0.49	0.49	1.28	25.00	0.10	0.10	0.20	0.69	3.85	9.19	1.68	0.40	0.49	0.59
	3	0.50	3.17	0.20	0.89	7.92	0.10	0.10	0.40	0.10	6.73	7.72	3.07	0.30	0.30	0.40
	4	0.10	1.30	0.20	42.66	26.37	0.30	0.10	0.10	0.10	0.50	3.80	3.50	0.20	0.40	0.60
	5	0.00	0.29	0.29	35.29	21.76	0.10	0.10	0.10	0.10	1.08	1.37	1.27	0.00	0.00	0.39
	6	0.10	0.58	0.48	1.07	13.87	0.00	0.10	0.19	0.19	2.72	3.10	1.36	0.00	0.19	0.00
	7	0.10	0.19	0.38	1.25	20.67	0.10	0.10	0.19	0.10	0.10	5.10	0.58	0.10	0.19	0.00
	8	0.00	0.19	0.10	1.36	21.07	0.10	0.29	0.10	0.19	4.27	5.05	2.33	0.10	0.49	0.19
	9	0.10	1.29	0.10	58.51	36.14	0.10	0.10	0.10	0.20	1.39	2.57	0.89	0.20	0.20	0.30
	10	0.10	0.20	0.40	0.60	9.69	0.00	0.10	0.10	0.10	2.80	1.60	1.00	0.20	0.20	0.00
	11	0.00	0.10	0.20	0.50	0.30	0.00	0.10	0.00	0.10	1.00	6.20	0.50	0.10	0.10	0.10
	12	0.00	0.00	0.50	0.20	0.20	0.00	0.00	0.20	0.00	3.90	4.90	2.50	0.00	0.00	0.20
	13	0.00	0.20	0.00	0.20	0.10	0.00	0.00	0.00	0.10	0.80	1.40	1.40	0.00	0.00	0.00
	14	0.10	0.00	0.10	0.70	0.20	0.10	0.00	0.10	0.50	0.80	5.00	2.50	0.30	0.20	0.00
	15	0.00	0.40	0.10	0.70	0.30	0.00	0.00	0.10	0.10	1.50	1.00	3.00	0.00	0.00	0.20
	16	0.10	0.29	0.19	0.97	1.36	2.72	0.19	0.39	10.88	2.14	27.31	7.39	0.39	1.07	1.55
	17	0.09	0.94	0.28	143.5	88.69	2.92	7.82	0.19	9.90	1.89	8.67	1.32	0.28	0.85	1.51
	18	0.20	0.40	4.80	117.1	34.00	4.90	0.20	0.20	2.20	3.00	10.50	3.50	0.40	1.20	1.10
19	0.10	0.59	1.18	52.55	32.45	2.25	0.10	0.20	0.59	3.92	21.47	12.35	0.59	1.27	0.59	
20	0.10	0.59	0.29	1.18	39.88	1.67	1.28	0.29	0.88	1.67	22.40	6.09	0.59	0.29	0.79	
Eastern Harbour	21	0.10	1.47	1.37	0.59	22.80	1.76	1.86	0.20	0.10	1.37	1.96	2.25	0.10	0.00	0.49
	22	0.10	0.89	1.09	88.93	54.94	5.93	4.35	0.49	0.79	1.88	27.57	1.58	0.79	2.27	1.19
	23	0.00	0.40	0.30	0.70	38.27	30.02	21.17	1.29	34.79	0.70	48.01	27.93	0.80	9.74	0.99
	24	0.10	0.79	0.39	136.9	28.09	2.75	0.10	0.20	0.39	1.08	8.15	2.26	0.20	0.20	0.49
	25	0.87	0.39	1.55	100.8	62.59	0.10	ND	0.10	0.19	0.68	2.62	0.29	0.10	0.19	0.49
	26	ND	0.69	0.49	193.9	39.92	20.65	15.81	0.40	0.30	4.15	39.62	21.54	0.49	0.59	0.59
	27	0.19	1.27	0.29	3.41	84.71	0.10	7.40	0.29	12.95	1.36	36.03	17.53	22.69	8.28	0.39
Western har- bour	28	0.09	0.28	0.19	379.9	62.88	61.84	49.91	69.89	82.48	1.61	104.5	53.50	1.42	17.05	0.47
	29	0.10	2.20	0.40	102.1	63.14	6.29	1.50	1.30	33.57	3.70	32.87	14.79	1.90	2.00	0.60
	30	0.30	0.50	14.00	70.90	25.40	14.30	10.20	12.40	53.60	8.60	55.00	20.00	1.10	4.40	2.10
	31	0.08	0.79	1.11	1.90	32.62	0.08	0.08	0.24	0.16	2.14	5.56	1.59	0.32	0.63	0.48

	32	0.10	0.40	0.40	77.20	125.5	10.20	8.10	21.50	1.80	31.80	53.10	20.50	2.70	1.50	2.10
	33	0.40	2.80	0.20	2.80	80.20	1.20	1.00	5.00	1.20	11.60	47.80	22.60	1.00	5.60	1.40
	34	0.10	1.90	0.90	174.0	329.1	3.90	0.80	0.50	6.40	3.10	10.80	3.00	1.90	2.70	3.20
	35	ND	1.10	0.30	1.70	158.7	5.29	4.89	1.70	14.27	2.79	11.18	1.30	1.00	0.50	0.60
	36	ND	0.40	1.80	0.60	25.50	1.70	0.40	0.50	0.60	2.90	11.60	7.10	0.80	1.70	1.10
	37	0.10	2.10	0.20	113.0	165.5	34.00	3.50	89.30	111.6	4.00	133.3	57.30	2.20	1.00	1.90
	38	0.10	1.60	0.20	75.80	1.80	6.90	1.90	25.90	3.30	5.20	26.70	1.90	2.50	1.10	2.70
	39	0.10	1.99	0.30	2.59	46.37	12.64	3.88	4.18	61.89	1.09	28.86	13.93	1.29	0.70	1.39
	40	ND	13.75	10.64	10.64	28.27	0.26	2.59	ND	ND	1.69	3.89	1.17	ND	0.13	0.26
	41	0.21	0.62	0.21	168.0	103.7	0.21	0.21	ND	0.21	4.52	1.44	1.23	0.41	0.41	0.62
	42	0.13	0.53	0.13	116.2	79.02	0.40	0.27	0.13	0.13	0.93	1.33	1.33	0.13	ND	0.27
	43	0.13	0.81	0.94	115.2	66.40	4.18	3.37	0.13	0.13	0.94	7.42	0.27	0.27	ND	0.81
El Max	44	0.12	0.49	0.24	68.45	32.28	5.22	6.55	0.12	0.12	1.82	1.94	1.09	ND	0.36	0.73
	45	0.16	44.44	0.32	4.76	44.13	0.16	0.16	0.32	0.16	1.59	2.54	1.11	ND	ND	0.32
	46	ND	24.71	7.91	109.7	66.60	4.88	6.35	0.20	0.20	1.17	2.54	1.46	0.20	0.10	0.59
	47	ND	6.02	0.38	8.83	46.73	2.43	0.64	0.13	0.13	1.15	2.18	1.54	0.13	0.26	0.26
	48	0.13	1.14	ND	10.51	3.80	0.38	0.25	ND	0.13	0.89	2.66	2.66	ND	ND	0.38
	49	ND	4.68	0.08	86.12	25.17	0.33	0.17	0.08	0.08	3.18	1.09	0.67	ND	ND	0.17

Table 3. ERL and ERM for PAHs recorded in sediment according to Long *et al.* [11].

Component, ng/g	ERL	ERM
Acenaphthene	16	500
Acenaphthylene	44	640
Anthracene	85.3	1100
Fluorene	19	540
Naphthalene	160	2100
Phenanthrene	240	1500
BaA	261	1600
BaP	430	1600
Chrysene	384	2800
Dibenzo(a,h)Anthracene	63.4	260
Fluoranthene	600	5100
Pyrene	665	2600
Total PAHs	2967	20340

ERL = concentration at lower tenth percentage at which adverse observed, ERM = concentration at which adverse effects were observed at 50% of test organisms.

(Σ PAHCARC) [13] were highest at stations number 37 and 28 with 153.8 and 143.47 ng/g, respectively (**Table 1**). BaP was ranged from 0.27 at station 43 to 57.30 ng·g⁻¹ at station 37 with a mean of 7.63 ng·g⁻¹, falling in the concentration range between rural and urban areas [14].

One difficulty in identifying PAH origins is the possible coexistence of many contamination sources, and the transformation processes that PAHs can undergo before deposition in the analyzed sediments. Nevertheless, some compounds could exhibit comparable evolution kinetics that could be used to identify the origin of organic matter in the environment [15]. The compounds of pyrene, phenanthrene and benzo(b)fluoranthene are components of fossil fuels and a portion of them is associated with their combustion. Benzo(a)pyrene is usually emitted from catalyst and no catalyst automobiles. Benzo(a)anthracene and chrysene are often resulted from combustion of both diesel and natural gas [16]. The ratio of sum of major combustion specific compounds (Σ COMB, Flu, Pyr, BaA, Chr, BbF, BkF, BaP, InP and BghiP) to the sum of 16 EPA-PAHs (Σ COMB/ Σ PAHs) were ranged from 0.55 to 1.0 and the Σ COMB concentrations displayed values from 4 to 885.5 ng·g⁻¹ (**Table 1**), representing average of 96% of total anthropogenic PAHs. This ratio

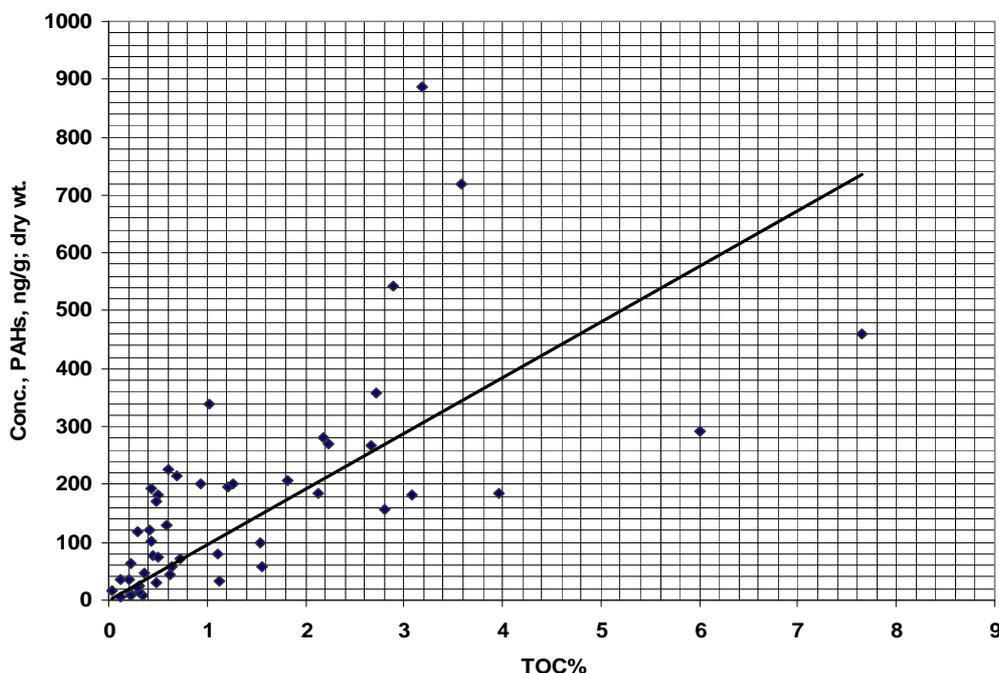


Figure 2. A plot of total PAHs concentration (ng/g; dry wt) vs. %TOC of the sediments collected from the studied sediments ($r = 0.646$).

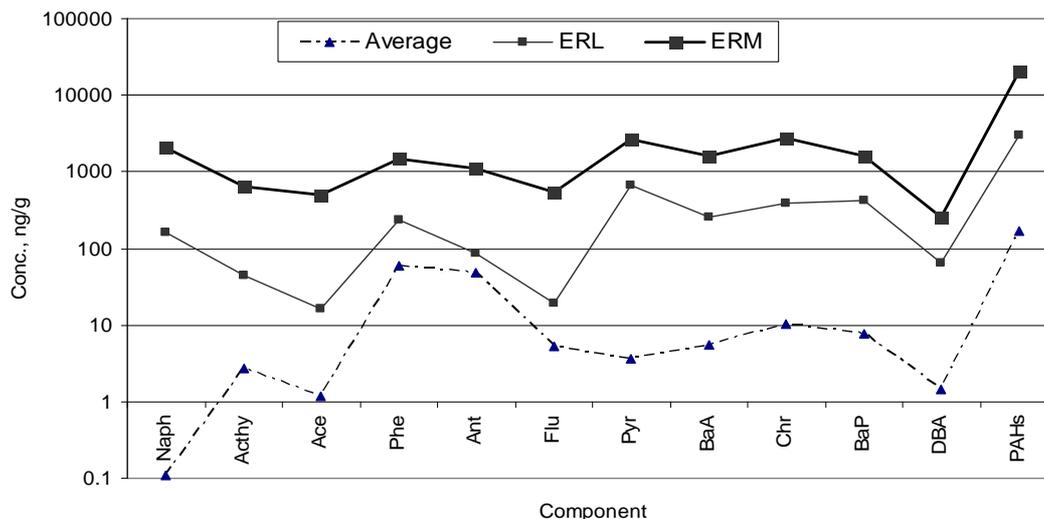


Figure 3. Diagram showed the average concentration ($\text{ng}\cdot\text{g}^{-1}$; dry wt) of PAH in Mediterranean Sea sediments relative to ERL and ERM.

was 1 at stations 17, 23, 26 and 28 (Eastern Harbour); stations 32, 37 (Western Harbour); stations 41 and 42 (El Max Bay), which indicated that the PAHs at these sites mainly come from combustion origin. The higher $\sum\text{COMB}/\sum\text{PAHs}$ ratio values further indicated that extensive combustion activities affected the PAHs in sediment samples. The sources of PAHs, whether from fuel-combustion (pyrolytic) or from crude oil (petrogenic)

contamination, may be identified by ratios of individual PAH compounds based on peculiarities in PAH composition and distribution pattern as a function of the emission source [17]. Ratio values such as phenanthrene/anthracene (Phe/Ant) and fluoranthrene/pyrene (Flu/Pyr) had been used by previous workers [18]. Petroleum often contains more phenanthrene relative to anthracene as phenanthrene is more thermodynamically stable

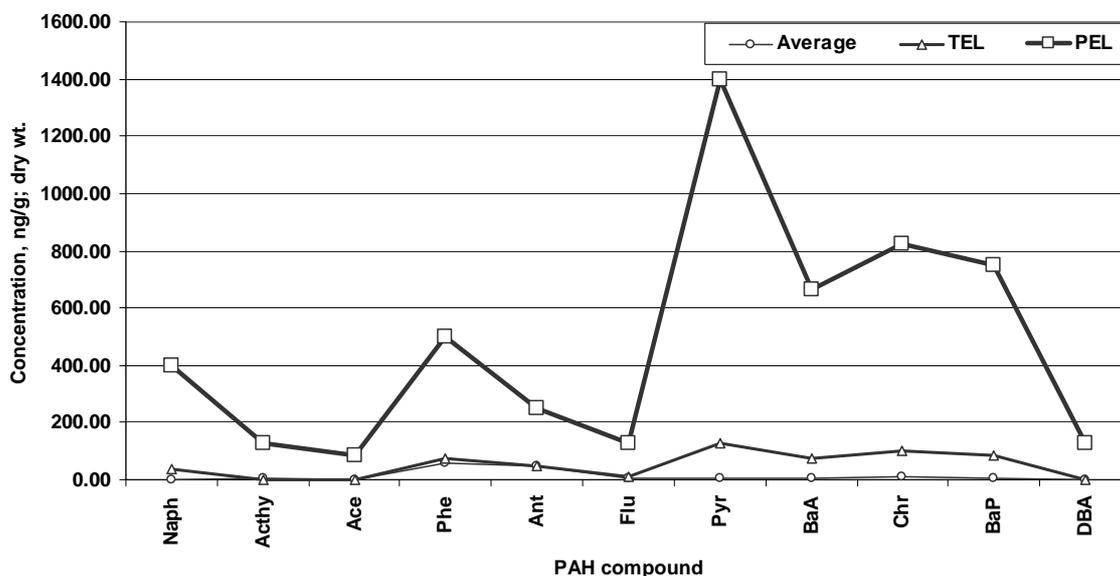


Figure 4. Diagram showed the average concentration ($\text{ng}\cdot\text{g}^{-1}$) of average PAH components in sediment samples Threshold Effect Level (TEL) and Probable Effect Level (PEL).

tricyclic aromatic isomer than anthracene, so a Phe/Ant ratio is observed to be very high in PAH petrogenic pollution, but low ratio in pyrolytic contamination cases. Crude oil had a Phe/Ant ratio of around 50, and motor vehicle exhaust had a ratio of around four [19]. Low Phe/Ant ratio values (<10) indicated the major PAH input was from combustion of fossil fuel. Sediments with $\text{Phe}/\text{Ant} > 10$ were mainly contaminated by petrogenic inputs and $\text{Phe}/\text{Ant} < 10$ was typical of pyrolytic sources. In the present study, the ratio of Phe/Ant compounds was < 10 , reflecting pyrolytic derived PAHs except for station 38 (repairing ship station) which recorded Phe/Ant ratio > 10 (ratio = 42.11), indicating petrogenic derived PAHs in such station. In other sediments the Phe/Ant ratio values were around 0.01 - 6.04 (Table 1), suggesting that they were pyrolytic-derived PAHs. On the other hand fluoranthene/pyrene (Flu/Pyr) ratio also indicated the origin of PAHs. Sicre et al. [20] suggested that a Flu/Pyr ratio of less than 1 was attributed to petrogenic sources and values greater than 1 were obviously related to a pyrolytic origin. Combustion of coal and wood gave Flu/Pyr ratios of 1.4 and 1, respectively, while crude oil and fuel oil had values of 0.6 - 0.9 [20]. In the present study, most sites had Flu/Pyr ratio values more than 1 (Table 1).

The PAHs whose concentrations are susceptible of co-varying in the environment were identified in this study on the basis of the correlation factor values (Table 4). This statistical approach is based on the fact that each pollution source produces a characteristic PAH pattern; so, the correlations of all the individual PAHs can give

an idea whether they all originate from the same source or not. A lack of correlation was noticed between anthracene and other PAHs ($r = 0.116$ to 0.304). Significant correlations were noted between Flu-Phe ($r = 0.602$), Flu-Pyr ($r = 0.882$), Flu-BaA ($r = 0.759$), Pyr-Phe ($r = 0.645$), %TOC-Phe ($r = 0.458$), %TOC-BaA ($r = 0.534$), %TOC-Chr ($r = 0.629$), %TOC-BKf ($r = 0.669$), %TOC-BghiP ($r = 0.697$). These indicated that Flu, Phe, Pyr, BaA and Chr might originate from the same sources. Chrysene and benzo(a)anthracene are derived from processes of organic matter combustion at high temperature, with values of Chr/BaA ratio lower than 1. In contrast, low maturation of organic matter during burial in the sedimentary matrix could lead to an inversion of this tendency: $\text{Chr}/\text{BaA} = 1$ [18]. It has been shown that chrysene derivatives are more stable than benzantracenic ones because of the possibility of the latter ones to convert to chrysene compounds.

4. Conclusions

The present work represents the detailed study of the distribution and origin of petroleum hydrocarbons in 49 sediment samples collected from the semi-closed areas of Alexandria, Egypt. The studied samples were less contaminated by petroleum hydrocarbons. Concentrations of PAHs in Mediterranean Sea sediments are shown to be substantially lower than those from other coastal areas and are generally comparable to levels encountered in the other Mediterranean Sea coast. The most contaminated site ($886 \text{ ng}\cdot\text{g}^{-1}$) was the Eastern Harbor (station 28). A mixture of pyrolytic

Table 4. Correlation coefficient matrix for sediment individual PAHs (n = 49).

Compound	Naph	Acthy	Ace	Phe	Ant	Flu	Pyr	BaA	Chr	BbF	BkF	BaP	InP	DBA	B(ghi)P	%TOC
Naph	1															
Acthy	-0.028	1.000														
Ace	0.084	0.245	1.000													
Phe	0.065	-0.065	0.046	1.000												
Ant	0.072	0.027	-0.042	0.406*	1.000											
Flu	-0.073	-0.084	0.032	0.602***	0.208	1.000										
Pyr	-0.087	-0.055	0.072	0.645***	0.116	0.882***	1.000									
BaA	-0.022	-0.060	-0.023	0.517**	0.274	0.759***	0.547***	1.000								
Chr	-0.016	-0.082	0.115	0.409*	0.275	0.775***	0.561***	0.828***	1.000							
BbF	0.106	-0.079	0.067	0.080	0.229	0.073	0.079	0.223	0.046	1.000						
BkF	0.015	-0.126	0.048	0.472**	0.304	0.794***	0.625***	0.880***	0.865***	0.338	1.000					
BaP	0.016	-0.113	0.010	0.442**	0.269	0.869***	0.688***	0.827***	0.843***	0.261	0.945***	1.000				
InP	0.085	-0.059	-0.050	-0.035	0.204	0.036	0.128	0.077	0.129	0.061	0.227	0.238	1.000			
DBA	0.055	-0.114	0.033	0.455**	0.174	0.747***	0.852***	0.470**	0.530**	0.075	0.629***	0.699***	0.394*	1.000		
B(ghi)P	0.084	-0.100	0.172	0.320	0.566***	0.248	0.132	0.374*	0.372*	0.440**	0.485**	0.321	0.086	0.193	1.000	
%TOC	0.145	-0.009	0.270	0.458**	0.385*	0.358*	0.338	0.534**	0.629***	0.419*	0.669***	0.503**	0.106	0.378*	0.697***	1.000

Correlations are significant at *p < 0.05 (low significant), **p < 0.01 (moderate significant) and ***p < 0.001 (high significant).

and petrogenic PAHs were observed with a major pyrolytic predominance.

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