

# Enhanced Ferroelectric Properties of Multilayer SBT-BTN Thin Films for NVRAM Applications

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## Abstract

Ferroelectric SrBi<sub>2</sub>Ta<sub>2</sub>O<sub>9</sub>-(Bi<sub>4</sub>Ti<sub>3</sub>)<sub>1-x</sub>Nb<sub>x</sub>O<sub>12</sub> (x = 0.02) (SBT-BTN) multilayer thin films with various stacking periodicity have been synthesized on Ir/Ti/SiO<sub>2</sub>/Si substrate by metal organic chemical vapor deposition technique (MOCVD). Tributylbismuth [Bi(C<sub>4</sub>H<sub>9</sub>)<sub>3</sub>], Strontium-bis[Tantal(pentan-ethoxy)(2-methoxyethoxid)] [Sr[Ta(OEt)<sub>5</sub>(OC<sub>2</sub>H<sub>4</sub>OMe)]<sub>2</sub>], Titanium Bis(isopropoxy)bis(1-methoxy-2-methyl-2-propoxide) [Ti(OiPr)<sub>2</sub>(mmp)<sub>2</sub>] and Niob-ethoxide [Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub>] were selected as precursors. X-ray diffraction patterns show that the multilayer films annealed at 800°C consist of fully formed perovskite phase with polycrystalline structure and plate-like grains with no crack. The remanent polarization (2·Pr) and coercive field (Ec) are 16.2  $\mu$ C/cm<sup>2</sup> and 230 kV/cm, respectively, which is much higher, compared to pure SBT film (2·Pr = 6.4  $\mu$ C/cm<sup>2</sup>, Ec = 154 kV/cm). In the films prepared above 700°C, postannealing increased the capacitor shorting rate; this was attributed to oxidizing of the top iridium layer. In this paper, the dependence of composition variation around stoichiometric on ferroelectric properties in SBT-BTN multilayer films is studied.

# Keywords

Ferroelectric, SBT, BTN, MOCVD, Multilayer

# **1. Introduction**

Aurivillius-type structure compounds have been widely investigated by their use in non-volatile ferroelectric memories (NVMs) where they are prepared as thin films [1] [2]. Recently, bismuth-layered-structure ferroelectric materials (BLSF), such as Bi<sub>4</sub>Ti<sub>3</sub>O<sub>12</sub> (BIT) and SrBi<sub>2</sub>Ta<sub>2</sub>O<sub>9</sub> (SBT), and different ions substitutions have been studied to improve their fatigue, ferroelectric properties and possibly low polarization switching voltage. In or-

der to improve their ferroelectric properties, the substitution technique, called "site engineering", was adopted from [3].

In a previous work, Park [2] found that  $Bi_{3.25}La_{0.75}Ti_3O_{12}$  (BLT) thin film annealed at 650°C exhibited a large remanent polarization  $P_r = 12 \ \mu\text{C/cm}^2$ . It was found that Nd doping of BIT film led to increased  $P_r$  value of 25  $\mu$ C/cm<sup>2</sup> due to decrease in oxygen vacancy concentration [4] [5]. Noguchi [6] pointed out that enhancement of remanent polarization was achieved when Ti<sup>4+</sup> ions in BiT were replaced by ions with higher valences (e.g. V<sup>5+</sup> and W<sup>5+</sup>). Shulman [7] and Villegas [8] reported reduction of conductivity of these films by doping with Nb or W. Such structures as Bi<sub>4</sub>Ti<sub>3</sub>O<sub>12</sub>-SrBi<sub>2</sub>Ti<sub>2</sub>O<sub>15</sub> (BIT-SBT), Bi<sub>3</sub>TiNbO<sub>9</sub>-Bi<sub>4</sub>Ti<sub>3</sub>O<sub>12</sub> (BTN-BIT), Bi<sub>2</sub>MoO<sub>6</sub>-Bi<sub>3</sub>TiNbO<sub>9</sub> (BM-BTN) and Bi<sub>2</sub>WO<sub>6</sub>-Bi<sub>3</sub>TiNbO<sub>9</sub> (BW-BTN) have attracted further interest due to their artificially enhanced dielectric, ferroelectric and pyroelectric properties [9]-[14]. Ortega [15] showed dramatically increased remanent polarization (~20.8  $\mu$ C/cm<sup>2</sup>) and dielectric constant (~373) in bilayered (SBT/SBN) structures.

All these works suggest that substitution is an effective way to improve the ferroelectric property. Although perovskite type thin films SBT, BIT, and BTN have been extensively studied, no work on multilayer SBT and Nb<sup>5+</sup>-substituted BiT (BTN) thin films has yet been carried out. Therefore, we deposited SBT/BTN multilayer thin films with different stacking periodicities utilizing metal organic chemical vapor deposition process (MOCVD), optimized in our previous work [16]-[18].

MOCVD methods are widely used because they show advantages, such as simple injection system, superior step coverage, high deposition rate and composition uniformity over a large area compared to other techniques like pulsed laser deposition (PLD) [19] [20], metal-organic decomposition (MOD) and sol-gel (SG) [21]-[23].

In this paper, the MOCVD synthesis and characterization of multilayered SBT/BTN thin films on (111)Ir/ Ti/SiO<sub>2</sub>/(100)Si substrates in a single-wafer reactor are reported.

### 2. Methods

The Bi<sub>4-x/3</sub>Ti<sub>3-x</sub>Nb<sub>x</sub>O<sub>12</sub> (BTN) (x = 0.02) and SrBi<sub>2</sub>Ta<sub>2</sub>O<sub>9</sub> layer stacks were deposited by MOCVD reactor with liquid delivery system on 150 mm silicon wafers.10nm Ti and thereafter 100nm Ir were deposited on SiO<sub>2</sub> as bottom electrode by e-beam vaporizer. Tributylbismuth [Bi(C<sub>4</sub>H<sub>9</sub>)<sub>3</sub>], Strontium-bis[Tantal(pentan-ethoxy)(2methoxyethoxid)] [Sr[Ta(OEt)<sub>5</sub>(OC<sub>2</sub>H<sub>4</sub>OMe)]<sub>2</sub>], Titanium Bis(isopropoxy)bis(1-methoxy-2-methyl-2-propoxide) [Ti(OiPr)<sub>2</sub>(mmp)<sub>2</sub>] and Niob-ethoxide [Nb(OC<sub>2</sub>H<sub>5</sub>)<sub>5</sub>] were selected as precursors. Sr-Ta, Bi and Ti precursors were dissolved in octane of 0.02 mole/L concentration, Nb was dissolved of 0.005 mole/L concentration. Previously we reported the application of these precursors for SBT films deposition [17] [24] [25]. The process temperature was varied from 400°C to 550°C, the pressure in the chamber was kept at about 5.0 mbar. The deposition rate of the BTN was about ~50 nm/h, of SBT ~ 100 nm/h at 450°C and 5 bar chamber pressure. The oxygen to nitrogen ratio in a chamber was set to 3:2 and 3:4 for BTN and SBT deposition, respectively.

The prepared layer stacks are shown schematically in **Figure 1**. The multilayer M2-A and M3-A follow the same sequence as M2-B (BTN/SBT) except that BTN is interchanged with SBT and vice versa.

The thickness of the deposited SBT-BTN-SBT, BTN-SBT and SBT-BTN layers was in the range of 150 - 250 nm. The metal-ferroelectric-metal (MFM) capacitors were fabricated by deposition 0.2 - 1 mm square top electrodes using PVD through a shadow mask. The films were annealed in atmosphere for 30 minutes at the several different temperatures for final crystallization. The film thickness was measured on a spectroscopic ellipsometer in the spectral range of 280 - 780 nm in 10 nm steps. The polarization versus electric field (P-E) hysteresis measurements were performed using a ferroelectric test module TF ANALYZER 1000 (aixACCT Systems GmbH). Results for SBT films are discussed with the precursor solution compositions expressed as x/y/2, where x and y represent the Sr and Bi atomic atomic contents, respectively, normalized with the Ta content fixed at 2. The phase formation and crystallographic orientation of the films were characterized by X-ray diffraction (XRD) using Cu-K $\alpha$  radiation, the chemical composition of the films was analyzed with X-ray photoelectron spectroscopy (XPS).

## 3. Results and Discussion

The relationship between process parameter and composition of SBT films has been studied in previous work of our group [16]-[18] [24]. The XPS measurements of SBT and BTN films indicated an atomic composition of these films at various processing parameters. The BTN films with various Ti/Bi mole ratios were deposited at  $450^{\circ}$ C by changing the input mass flow of Ti(OiPr)<sub>2</sub>(mmp)<sub>2</sub> to Bi(C<sub>4</sub>H<sub>9</sub>)<sub>3</sub> vapors (Figure 2(a)). As the mass flow



Figure 1. Schematic diagrams of multilayer thin films: (a) M-3A; (b) M2-A; (c) M2-B.



Figure 2. Ti/Bi mole ratio of BNT thin films as a function of (a) the Ti mass flow and (b) substrate temperature.

ratio of Ti precursor increased from 2 g/h to 10 g/h, Ti/Bi ratio went up all the way, while temperature increase from 400°C to 550°C reduced Ti/Bi ratio in the film from 0.88 to 0.53 respectively (Figure 2(b)). By Ti mass flow of ~6.6 g/h and process temperature of 450°C nearly stoichiometric composition ( $Bi_{4.17}Ti_{2.98}Nb_{0.02}O_{12\pm x}$ ) can be obtained.

**Figure 3** shows Nb/Ti mole ratio in the BTN films as a function of Nb mass flow. It was shown in previous work by Song [26], that in BTN film  $P_r$  rises with increasing Nb content to reach a maximum at a Nb content of about 0.02. In our experiments it was unable to detect Nb in XPS spectrum as long as Nb-mass flow was below ~7.5 g/h. As Nb mass flow reached 7.5 g/h, we observed a small Nb-doublet in a XPS survey at ~208 eV, which corresponds 2.3 Nb at % in the film. It was noted that BTN thin film has higher  $P_r$  value of 4.9  $\mu$ C/cm<sup>2</sup> than BiT without Nb<sup>5+</sup> doping (~1.8  $\mu$ C/cm<sup>2</sup>).

**Figure 4(a)** shows Sr/Ta and Bi/Ta mole ratios in the SBT films deposited by changing the input gas flow of  $Bi(C_4H_9)_3$  to  $Sr[Ta(OEt)_5(OC_2H_4OMe)]_2$  vapors. As chamber temperature increased, Bi/Ta slightly increased, while Sr/Ta versus temperature was almost unchanged (**Figure 4(b)**). By Sr-Ta mass flow of ~5.4 g/h and process temperature of 450°C nearly stoichiometric composition in sense of Bi content ( $Sr_{0.58}Bi_{2.04}Ta_2O_{9\pm x}$  as determined by XPS) can be obtained. However Sr/Ta ratio was below stoichiometric, it was reported by Li [27] that Sr deficient films ( $Sr_{0.8}Bi_{2.2}Ta_2O_{9\pm x}$ ) yield higher remanent polarization than stoichiometric films ( $SrBi_2Ta_2O_9$ ), because of partial substitution of Bi for Sr ions results in the increase in 'rattling space'. On the other hand, as Bi concentration drops below stoichiometric in Sr deficient films, the exchange between Sr and Bi causes the serious degradation of the ferroelectric properties, also reported by Li [27] and Boyle [28].

Polarization–electric field (P-E) hysteresis loops were measured to examine the effect of Bi deficiency on the degradation of the ferroelectric properties in SBT films (shown in Table 1) under a maximum electric field of 500 kV/cm. As the Bi content decrease from 2.2 to 1.3,  $2 \cdot P_r$  drops from 7.4  $\mu$ C/cm<sup>2</sup> to 2.3  $\mu$ C/cm<sup>2</sup> respectively.

After successful depositing of single SBT and BTN films, multilayer stack was deposited alternating SBT and BTN layers (Figure 1). Figure 5 shows the XRD patterns of single SBT, bilayer M2-A, M2-B and multilayer M3-A thins films annealed at 800°C for 30 min in atmosphere. The constituent phase and the orientation of the SBT-BTN, SBT-BTN-SBT and BTN-SBT films were identified by X-ray diffraction (XRD) using Cu-K $\alpha$  radiation to distinguish the SBT, BIT and BTN phases, pyrochlore and fluorite phases. Pyrochlore and fluorite phases are contamination phases in the Sr-Bi-Ta-O and Bi-Ti-Nb-O system. It was ascertained that the ferroelectricity was hardly observed when the films was heat treated below 750°C. As the annealing temperature increase, annealing results in sharp peaks with high intensity, indicating better crystallinity. All films consisted mainly of the perovskite phase even thought a small amount of fluorite phase coexisted. The intensities of the SBT peaks are different, the possible reason is different grain size. The major XRD peaks corresponding to (115), (200) and (315) planes were more pronounced than (006) and (0010) lines, which indicated that preferred growth orientation along *c*-axis is suppressed. Two peaks located at about 36.0° and 57.0° correspond to fluorite phase. The Nb<sup>5+</sup> substitution had very little effect on the crystal orientation of the BiT film. We make attention to BTN as *m* 



Figure 3. Nb/Ti mole ratio of BNT thin films as a function of the Nb mass flow.



Figure 4. Sr/Ta and Bi/Ta mole ratio of SBT thin films as a function of (a) Bi mass flow (b) substrate temperature.

Table 1. Ferroelectric properties of Bi-defficient SBT films.			
Structure	$2 \cdot P_r (\mu C/cm^2)$ @500 kV/cm	E <sub>C</sub> , kV/cm	
$Sr_{0.68}Bi_{1.3}Ta_2$	2.3	89.9	
$Sr_{0.5}Bi_{1.65}Ta_2 \\$	3.6	84.9	
$Sr_{0.66}Bi_{1.74}Ta_2$	4.6	194.0	
$Sr_{0.58}Bi_{2.2}Ta_2$	7.4	184.9	

= 2 material with the lattice parameter of a = 0.540 nm, b = 0.545 nm, c = 2.516 nm and BIT as m = 3 material with the lattice parameter of a = 0.541 nm, b = 0.545 nm, c = 3.284 nm. We observe only one unit cell of BIT (114) in M2-A and M3-A layers and no peaks considered to BTN (0012) and (0018). At 800°C annealing temperature the pyrochlore phase appears and this would impair the ferroelectric properties.

Polarization-electric field (P-E) hysteresis loops were measured to examine the effect in the ferroelectric properties of the combination of the SBT and BTN multilayer with different stacking periodicity under a maximum electric field of 500 kV/cm (Figure 6). All the films exhibit a well-defined and saturated hysteresis loops.



Figure 5. XRD patterns of (a) single SBT, (b) bilayer M2-A and (c) multilayer M3-A annealed at 800°C for 30 min.

**Figure 6** shows the P-E hysteresis curve measured on the bilayer (M2-A, M2-B), multilayer (M3-A) and the single layer SBT films. The measured remanent polarization values of the bilayer M2-A (~15.4  $\mu$ C/cm<sup>2</sup>) is comparable with multilayer M3-A (~16.2  $\mu$ C/cm<sup>2</sup>), higher than M2-B (~9.8  $\mu$ C/cm<sup>2</sup>) and much higher than that of the single layer SBT (~6.4  $\mu$ C/cm<sup>2</sup>), which is consistent with other reports [26] [29].

Although, the coercive field of the bilayer (e.g. M2-A,  $E_C = 184 \text{ kV/cm}$ , M2-B,  $E_C = 165 \text{ kV/cm}$ ) and multilayer (e.g. M3-A,  $E_C = 230 \text{ kV/cm}$ ) is higher than that of single SBT ( $E_C = 154 \text{ kV/cm}$ ). The Nb<sup>5+</sup> substitution hasn't induced the large inhomogeneity in the thin film due to the close valence of Nb<sup>5+</sup> and Ti<sup>4+</sup>, so  $E_C$  almost no change. The P<sub>r</sub> value and the squareness of the hysteresis loops increased with increasing postannealing temperature (now shown). Both the remanent polarizations P<sub>r</sub> and the coercive electric field  $E_C$  of multilayer M3-A film increase with the increase of annealing temperature up to 800°C, so at 750°C 2·P<sub>r</sub> values reached 6.8  $\mu$ C/cm<sup>2</sup> ( $E_C = 194 \text{ kV/cm}$ ) and 16.2  $\mu$ C/cm<sup>2</sup> ( $E_C = 229 \text{ kV/cm}$ ) at 800°C. The increase of the P<sub>r</sub> and E<sub>C</sub> value with increasing postannealing temperature is mainly due to increase of the crystallinity of the film and decrease



**Figure 6.** P-E hysteresis loop for (a) multilayer SBT/BTN, (b) bilayer M2-A, (c) bilayer M2-B and (d) single layer SBT thin films at room temperature.

<b>Table 2.</b> Ferroelectric properties of SB1/
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Structure	$2 \cdot P_r$ , $\mu C/cm^2$	E <sub>c</sub> , kV/cm
SBT	6.4	154
M2-A (SBT-BTN)	15.4	184
M2-B (BTN-SBT)	9.8	165
M3-A (SBT-BTN-SBT)	16.2	230

of the vacancies of Bi and oxide ions in films, also reported elsewhere [6] [30]. On the films processed above 700°C, postannealing markedly increased the capacitor shorting rate, this was attributed to oxidizing of the top Ir layer observed in other reports [30]. Table 2 summarizes the ferroelectric properties of the investigated films. The ferroelectric properties of multilayer thin films are different from that of the pure thin films due to the presence of the interface region. The interface stress plays a significant role for close lattice matched systems [31]. In the case of multilayers, the combining effect of both the layers is reflected in the properties by improving the fatigue and with a high polarization [9]-[14]. In our SBT/BTN multilayer systems, ferroelectric properties are the result of the contributions from both SBT and BTN.

#### 4. Conclusion

Multilayer structures with alternated layer of SBT and BTN have been prepared on Ir/Ti/SiO<sub>2</sub>/Si substrate using MOCVD technique. On the films processed above 700°C, postannealing increased the capacitor shorting rate. The multilayer structure exhibited well-defined hysteresis loop, with a higher remanent polarization of ~16.2  $\mu$ C/cm<sup>2</sup> and a coercive field of 230 kV/cm under a maximum electric field of 500 kV/cm and 100 Hz. These results are considerably better compared to SBT single layer samples. The results reveal that the multilayer structure is a promising material combination for ferroelectric memory applications.

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