

# Antioxidant Activity Assessment and Color Analysis of Skin from Different Peach Varieties Grown in South Carolina

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## Abstract

Peach skin is a byproduct from the further processing of fresh peaches with the potential to be recovered and utilized as a natural antioxidant. Color analysis, phenolic content and antioxidant activity of peach skin from 13 varieties of peaches grown in South Carolina were determined. Color analysis indicated that Norman, Cary Mac, Ruby Prince and Flame Prince differed from other varieties of peaches. Antioxidant activity of peach skin extracts were evaluated by the total phenolics (TP), DPPH free radical scavenging (DPPH), ferric reducing antioxidant power (FRAP) and ferrous ion chelating (FIC) assays. The range of total phenolics content was 8.38 - 18.81 (gallic acid equivalent mg/g dry weight). The total phenolic content was highly correlated to DPPH and FRAP activity in peaches ranging from 8 - 23 AE/mg and 5 - 12 AE/mg, respectively. Three peach varieties with skins having the greatest antioxidant capacity were Red Globe, Scarlet Prince, and O'Henry.

## Keywords

Peach Skin, Antioxidant, Total Phenolics, DPPH Assay, FRAP Assay

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## 1. Introduction

Peach (*Prunus persica*) is one of the most popular fruits worldwide partly due to its good taste and nutrient composition [1] [2]. In the early 1970s, the annual per capita consumption of peaches in the US reached a peak of 13 pounds. By 2008, the annual consumption had dropped to 8.8 pounds per person [3]. In 2012, the US peach total production reached 978,260 tons while utilized production reached 965,420 tons [4]. South Carolina

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is the second largest peach producing state in the United States. In 2012, South Carolina had a total of 17,000 acres of peach trees, total peach production reached 75,000 tons which was valued at over \$74 million. However for SC in 2012, unharvested peaches totaled 3700 tons and harvested but not sold peaches was 1050 tons ranking 1st and 2nd among all states, respectively [4]. Unutilized peaches could be further processed to recover valuable products including natural antioxidants. The possibility of extracting antioxidants from South Carolina peach skin byproducts has not been previously reported. Researchers have indicated that phenolic compounds play an important role in antioxidant activity of peaches [5]. Major phenolic compounds found in peaches include caffeic acid, chlorogenic acid, leucoanthocyanins, catechins and flavonols [5] [6]. Other antioxidants found in peaches include ascorbic acid and carotenoids; however, the content of ascorbic acid and carotenoids in peaches is relatively low [5]. Desmond and Daniele (2008) [7] pointed out that phenolic compounds were concentrated in peach skin.

Antioxidants act by several mechanisms; thus no one assay can capture the different modes of action and therefore several different methods are employed to measure antioxidant capacity. Since 1958, numerous *in vitro* antioxidant assays have been proposed including 2,2-diphenyl-1-picrylhydrazyl (DPPH) free radical scavenging assay [8], ferrous chelating capacity (FIC) [9], trolox equivalent antioxidant capacity assay (TEAC) [10], oxygen radical absorbing capacity (ORAC) [11], total radical trapping antioxidant parameter (TRAP) [12], ferric ion reducing antioxidant power assay (FRAP) [13], and 2,2'-azino-bis (3-ethylbenzothiazoline-6-sulphonic acid (ABTS) assay [14]. All of these antioxidant assays have been applied to test the antioxidant activity *in vitro*. Mostly, they are used to evaluate fruits, vegetables, plant extracts, beverages and nutritional supplements. Total phenolic assay, DPPH assay and FRAP assay are the most popular methods applied in research. The objective of this research was to determine the antioxidant (phenolic) content and antioxidant capacity of peach skin from various peach varieties grown in South Carolina.

## 2. Materials and Methods

### 2.1. Materials

2,2-diphenyl-1-picrylhydrazyl (DPPH), iron chloride hexahydrate ( $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ ), 3-(2-pyridyl)-5,6-diphenyl-1,2,4-triazine-p,p'-disulfonic acid monosodium salt hydrate (ferrozine), Folin & Ciocalteu's phenol reagent, sodium carbonate, iron chloride tetrahydrate ( $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$ ) were purchased from Sigma-Aldrich. 2,4,6-tris(2-pyridyl)-s-triazine (TPTZ) was purchased from Sigma-Fluka. L-ascorbic acid, gallic acid were purchased from sigma life science. Ethyl alcohol (absolute, anhydrous, ACS/USP Grade) was purchased from Pharmco-AAPER. Acetic acid glacial was purchased from BDH. Hydrochloric acid solution certified 0.1 N was purchased from Fisher Scientific.

### 2.2. Peaches

13 cultivars of peaches from South Carolina, namely Summer Gold, Contender, July Flame, Scarlet Prince, Fire Prince, Cary Mac, Ruby Prince, Red Globe, Norman, Bounty, Early August Prince, Flame Prince, O'Henry at harvest stage between July 12 and August 15, 2013, were obtained from a SC producer. These varieties were chosen since these were the most popular during the peak growing season in this region of the Southeast US.

### 2.3. Color Analysis

Color of peach skin was measured on a model CR-400 chroma meter (Minolta Co. Ltd., Japan) by placing the colorimeter orifice directly on the peach surface prior to skin removal. Four peaches of each variety were randomly picked, four locations on each peach were measured using the chroma meter. The color was expressed as CIE 1976  $L^*$ ,  $a^*$ ,  $b^*$ , chroma ( $C^*$ ), hue ( $h^*$ ), with  $L^*$  representing the lightness of the color ( $L^* = 0$  yields black and  $L^* = 100$  indicates diffuse white),  $a^*$  represents the redness to greenness of color ( $a^*$  negative values indicate green while positive values indicate magenta),  $b^*$  represents the yellowness to blueness of color ( $b^*$  negative values indicate blue and positive values indicate yellow).  $C^*$  represents the intensity or purity of color,  $h^*$  represents hue of color.

### 2.4. Preparation of Peach Skin Samples

Based on the size of different peach varieties, 10 to 15 peaches of each variety were picked for sampling. They

were subdivided into 4 groups randomly. Peach skins were knife-peeled by hand and placed in marked sample bags and sealed. Those bags were frozen at  $-80^{\circ}\text{C}$  until analyzed. For each peach variety, four antioxidant assays were performed with one bag of peach skins each replication. In total four replications of each antioxidant assays were analyzed.

## 2.5. Extraction Method

Extraction procedures were based on Lim [15]. For one replication, one frozen sample bag of each variety was held at room temperature for 10 mins prior to extraction. After that, the remaining flesh in peach skin was scraped quickly and 10 g of peach skin was randomly taken from each sealed sample bag. Sample weights were recorded on a model B204-S College Monobloc analytical balance (Mettler Toledo, Toledo, OH). Moisture content was detected at the same time with a model HB43-S Mettler Toledo. Peach skins were blended for 6 seconds with 200 ml 50% ethanol and then homogenized 30 s. Homogenization was performed using a model PT 10/35 polytron with a model PCU 11 power control unit (Kinematica, Swizerland). The homogenized solution was placed in an ultrasound unit (model 5510R-DTH Ultrasonic unit. (Output 42 KHz  $\pm 6\%$ , Branson Ultrasonics Corporation, Danbury, CT) and extracted for 30min at room temperature. Later, the extraction was centrifuged (Beckman Coulter Avanti J-26S XPI Centrifuge, Jersey city, NJ) at 15,008 g,  $5^{\circ}\text{C}$  for 15 minutes. In total, four replication of extraction were performed and supernatant was recovered carefully for antioxidant activity analysis.

## 2.6. Antioxidant Activity Evaluation

### 2.6.1. Total Phenolics Content

Total phenolic compounds in each peach skin extract were determined with Folin-Ciocalteu reagent according to the method of Singleton [16]. Gallic acid was used as a standard phenolic compound. Briefly, 0.04 ml of each peach skin extract was diluted with distilled water (3.16 ml). 0.2 ml of Folin-Ciocalteu reagent was added and then mixed thoroughly. Within 8 minutes, 0.6 ml of  $\text{Na}_2\text{CO}_3$  (20%) was added, mixed, and incubated 30 min in a  $40^{\circ}\text{C}$  water bath. The absorbance was measured at 765 nm with a spectrophotometer (Model 4001/4 Genesys 20 Thermo Fisher Scientific, Waltham, MA) Results were expressed as gallic acid equivalent (GAE) (mg gallic acid/g dried extract).

### 2.6.2. DPPH Free Radical Scavenging Assay

Radical scavenging activity of peach skin extract was measured according to the method of Molyneux [17]. Briefly, 0.4 ml of each peach skin extract was added to 2 ml of a DPPH solution (0.2 mM in 50% ethanol) and kept for 30 min at room temperature. The absorbance was measured at 517 nm with a spectrophotometer. (Model 4001/4 Genesys 20 spectrophotometer Thermo Fisher Scientific, Waltham, MA) L-Ascorbic acid (50  $\mu\text{M}$  - 400  $\mu\text{M}$ ) was used as a standard. Results were expressed as ascorbic acid equivalent (AE-DPPH) (mg ascorbic acid/g dried extract).

### 2.6.3. Ferric Reducing Antioxidant Power (FRAP) Assay

Reducing power of peach skin extract was determined by FRAP assay described by Benzie and Strains [13]. Briefly, 300 mM acetate buffer (pH 3.6), 10 mM TPTZ in 40 mM HCl, and 20 mM  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  were freshly prepared. pH was measured pH meter (by model S/N 004602, Orion Research Inc., city, state) Working FRAP reagent was prepared by mixing acetate buffer, TPTZ and  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  in the ratio of 10:1:1 at the time of use. 100  $\mu\text{l}$  of each peach skin extract was mixed with 3 ml of working FRAP reagent and then kept at  $37^{\circ}\text{C}$  water bath for 4 minutes. The absorbance was measured at 593 nm with a spectrophotometer (model 4001/4 Genesys 20 spectrophotometer, Thermo fisher scientific, Waltham, MA) Ascorbic acid (100  $\mu\text{M}$  - 1000  $\mu\text{M}$ ) was used as a standard. The reducing power was expressed as ascorbic acid equivalent (AE-FRAP) (mg Ascorbic Acid/g dried extract).

### 2.6.4. Ferrous Ion Chelating (FIC) Assay

Ferrous ion-chelating potential of peach skin extract was determined according to the method of Gulcin [18]. Briefly, 1 ml of peach skin extract was mixed with 0.2 ml of 2 mM  $\text{FeCl}_2$ . Then 2.4 ml 50% ethanol was added. The reaction was initiated by the addition of 0.4 ml of 5 mM ferrozine. The mixture was allowed to sit at room

temperature for 10 minutes before absorbance was measured at 562 nm with a spectrophotometer (Model 4001/4 Genesys 20 spectrophotometer Thermo fisher scientific, Waltham, MA). EDTA was used as positive control. The ability of the extract to chelate ferrous ion was calculated using the following equation

$$\text{Chelating ability (\%/mg/ml)} = \left[ \frac{A_0 - A_1}{A_0} \right] \times 100\% / \text{Concl}$$

where  $A_0$  is the absorbance of the control,  $A_1$  is the absorbance of peach skin sample or EDTA, Concl is the concentration of peach skin extracts.

### 3. Statistical Analysis

An analysis of variance (ANOVA) was conducted to compare the effect of peach varieties on peach skin color ( $L^*$ ,  $a^*$ ,  $b^*$ , hue, chroma), total phenolic content and antioxidant capacity (DPPH, FRAP, FIC), and when peach variety significantly ( $P < 0.05$ ) affected an individual peach property, the Tukey post hoc test was used to determine if there was a significant difference ( $P < 0.05$ ) among 13 varieties of peaches (SAS, 2011). The relationship of four antioxidant assays and peach skin color with phenolic content of peach skin was determined using the Pearson coefficient correlation.

## 4. Results and Discussion

### 4.1. Color Analysis

Lightness and hue values varied among the 13 varieties evaluated with Cary Mac, Ruby Prince, Flame Prince being significantly lighter ( $P < 0.05$ ) than all other varieties except for Contender (Figure 1). Also, the value of lightness of Norman is significantly lower ( $P < 0.05$ ) than other peaches excluding July Flame and Summer Gold. The Norman variety was significantly lower ( $P < 0.05$ ) from other peaches in chroma value while other peaches had similar color purity (Figure 2). For hue value, Cary Mac and Flame Prince were also significantly higher ( $P < 0.05$ ) than all other peach varieties except Ruby Prince and Contender (Figure 3). In summary, Norman, Cary Mac, Ruby Prince and Flame Prince differed in several color parameters compared to other peach varieties (Table 1).

### 4.2. Total Phenolic Content

Phenolic content differed across the varieties of peaches tested. Among all the varieties, Red Globe had the highest phenolic content while Ruby Prince had the lowest mean content of phenolics (Figure 4). The order from greatest to lowest average phenolic content was Red Globe, Scarlet Prince, O'Henry, Bounty, Fire Prince, Norman, July Flame, Cary Mac, Summer Gold, Flame Prince, Contender, Early August Prince, Ruby Prince. Red Globe had a significantly greater ( $P < 0.05$ ) phenolic content than July Flame, Cary Mac, Summer Gold, Flame Prince, Contender, Early August Prince and Ruby Prince while the Scarlet Prince variety had greater ( $P < 0.05$ ) phenolic content than Early August Prince and Ruby Prince. Red Globe and Summer Gold varieties

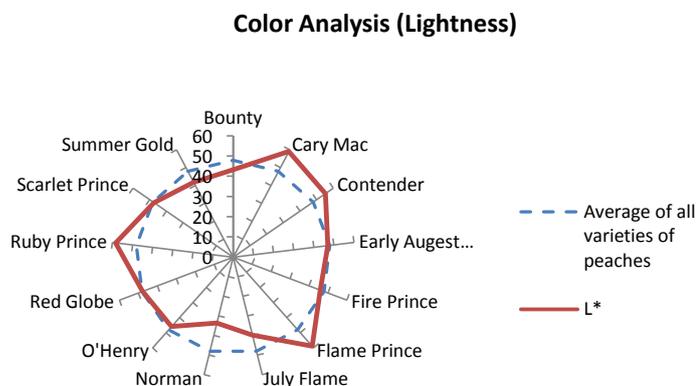


Figure 1. Color analysis ( $L^*$ ) of 13 cultivars of peaches.

### Color Analysis (Color Purity)

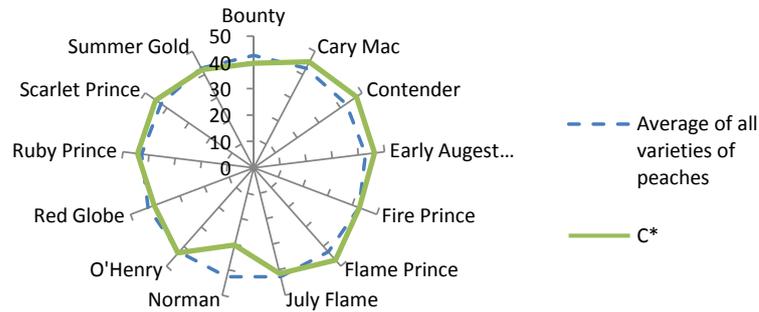


Figure 2. Color analysis (C\*) of 13 cultivars of peaches.

### Color Analysis (Hue)

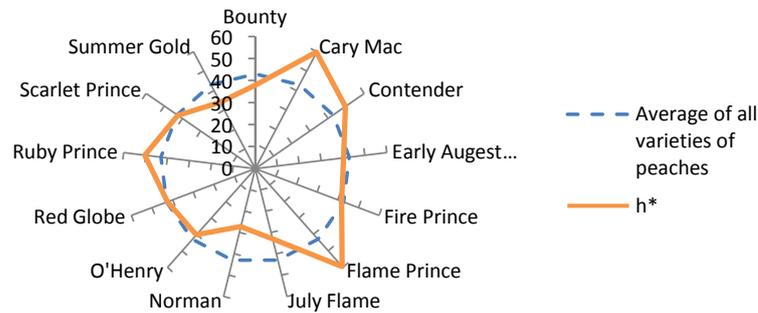


Figure 3. Color analysis (hue) of 13 varieties of peaches.

### Total Phenolics Content

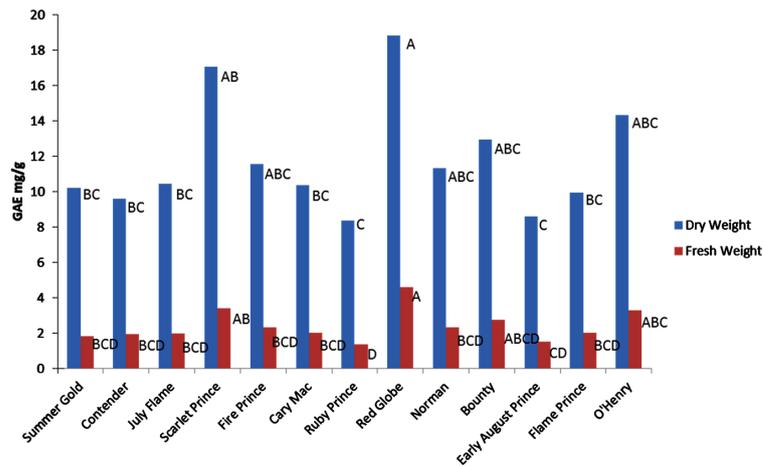


Figure 4. Antioxidant activity of total phenolics assay A-D means within the same color bar having the same subscript are not significantly different ( $P > 0.05$ ). Standard error (SE) = 1.54,  $n = 12$ .

had a large variation while the Scarlet Prince variety had both a high phenolic content and a relatively small range of variation in phenolic content (Figure 5).

**Table 1.** Color analysis of 13 cultivars of peach skins.

Color	L*		C*		h*		a*		b*	
	Estimate means	Standard deviation	Estimate means	Standard deviation	Estimate means	Standard deviation	Estimate means	Standard deviation	Estimate means	Standard deviation
<b>Bounty</b>	43.32 <sup>C</sup>	7.06	39.69 <sup>B</sup>	7.72	37.76 <sup>BCD</sup>	12.03	30.24 <sup>ABC</sup>	6.40	24.48 <sup>DEF</sup>	9.12
<b>Cary Mac</b>	59.07 <sup>A</sup>	4.97	45.49 <sup>AB</sup>	2.16	59.82 <sup>A</sup>	11.14	22.48 <sup>D</sup>	7.74	38.61 <sup>AB</sup>	4.77
<b>Contender</b>	55.30 <sup>AB</sup>	6.40	47.52 <sup>A</sup>	3.40	49.72 <sup>AB</sup>	9.66	30.08 <sup>ABC</sup>	5.19	36.00 <sup>ABC</sup>	6.82
<b>Early August Prince</b>	47.03 <sup>BC</sup>	4.22	46.27 <sup>AB</sup>	3.75	40.37 <sup>BCD</sup>	4.66	35.05 <sup>A</sup>	2.80	30.00 <sup>ABCDE</sup>	4.41
<b>Fire Prince</b>	45.98 <sup>BC</sup>	9.19	42.92 <sup>AB</sup>	5.76	41.93 <sup>BC</sup>	13.69	30.43 <sup>ABC</sup>	5.58	28.62 <sup>BCDE</sup>	10.28
<b>Flame Prince</b>	58.83 <sup>A</sup>	8.34	46.92 <sup>AB</sup>	3.98	59.58 <sup>A</sup>	13.14	22.59 <sup>D</sup>	6.75	39.92 <sup>A</sup>	8.62
<b>July Flame</b>	39.73 <sup>CD</sup>	4.23	41.47 <sup>AB</sup>	5.39	32.98 <sup>CD</sup>	5.74	34.45 <sup>AB</sup>	3.66	22.73 <sup>EF</sup>	5.76
<b>Norman</b>	33.56 <sup>D</sup>	5.77	30.26 <sup>C</sup>	8.82	27.11 <sup>D</sup>	6.44	26.41 <sup>CD</sup>	6.61	14.43 <sup>F</sup>	6.69
<b>O'Henry</b>	45.89 <sup>BC</sup>	8.72	43.33 <sup>AB</sup>	4.83	40.18 <sup>BCD</sup>	11.50	32.04 <sup>ABC</sup>	3.99	27.95 <sup>CDE</sup>	9.03
<b>Red Globe</b>	47.73 <sup>BC</sup>	13.72	40.33 <sup>AB</sup>	9.69	42.36 <sup>BC</sup>	19.58	26.39 <sup>CD</sup>	6.68	27.62 <sup>CDE</sup>	15.07
<b>Ruby Prince</b>	58.30 <sup>A</sup>	7.38	44.31 <sup>AB</sup>	2.83	50.27 <sup>AB</sup>	11.05	27.63 <sup>BCD</sup>	5.94	33.67 <sup>ABCD</sup>	6.56
<b>Scarlet Prince</b>	47.48 <sup>BC</sup>	8.13	45.10 <sup>AB</sup>	6.61	42.55 <sup>BC</sup>	11.52	32.11 <sup>ABC</sup>	5.78	30.59 <sup>ABCDE</sup>	9.05
<b>Summer Gold</b>	41.81 <sup>CD</sup>	7.09	42.10 <sup>AB</sup>	7.40	34.05 <sup>CD</sup>	8.55	34.03 <sup>AB</sup>	4.35	24.10 <sup>DEF</sup>	8.45

A-E mean within the same column with the same subscript are not significantly different ( $P > 0.05$ ).  $n = 16$ .

### 4.3. DPPH Assay

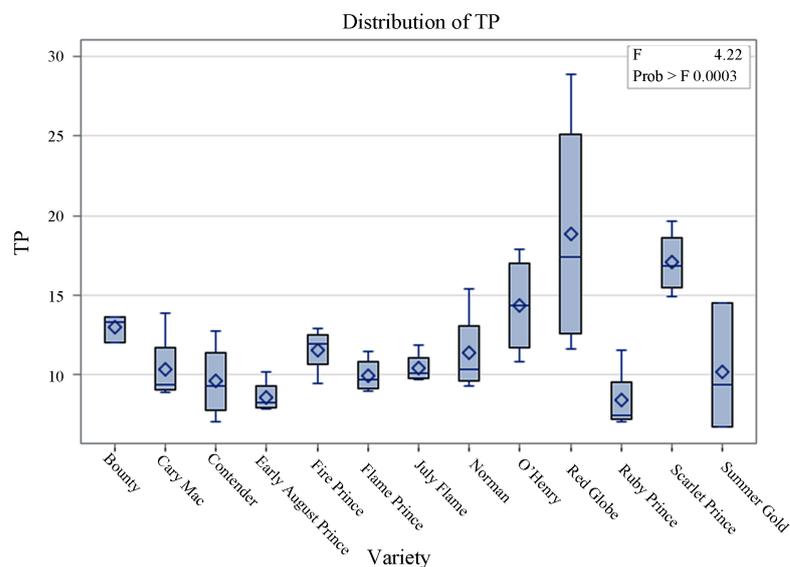
Ascorbic acid was used as the standard for the DPPH assay thus results were expressed as ascorbic acid equivalent (AE). The Red Globe variety had the greatest radical scavenging activity (Figure 6). The order from highest to radical scavenging activity was Red Globe, O'Henry, Scarlet Prince, Bounty, Norman, Flame Prince, Fire Prince, July Flame, Cary Mac, Summer Gold, Contender, Early August Prince, Ruby Prince. Red Globe peaches were not significantly different ( $P > 0.05$ ) from O'Henry, Scarlet Prince, Bounty and Norman but had greater ( $P < 0.05$ ) radical scavenging activity than other varieties. O'Henry had greater ( $P < 0.05$ ) scavenging activity than Early August Prince, Contender and Ruby Prince. Box and whisker plot revealed that Red Globe and O'Henry peach samples had more variation than others and that Scarlet Prince (as with phenolic content) had low variation and relatively high scavenging activity (Figure 7).

### 4.4. FRAP Assay

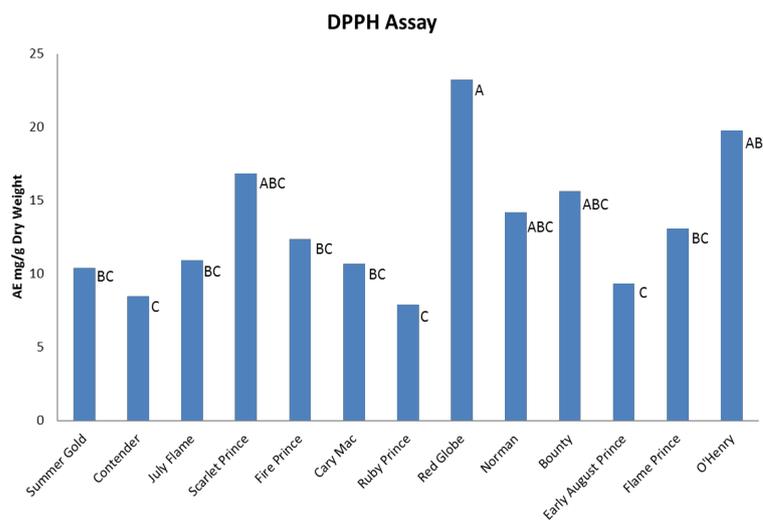
Ascorbic acid was also used as the standard reference for FRAP analysis therefore results were expressed as ascorbic acid equivalents (AE). The order from highest to least reducing ability of peach varieties was Red Globe, Scarlet Prince, Bounty, Norman, O'Henry, July Flame, Fire Prince, Summer Gold, Early August Prince, Cary Mac, Flame Prince, Contender, Ruby Prince (Figure 8). Red Globe had significantly ( $P < 0.05$ ) more reducing ability than Early August Prince, Cary Mac, Flame Prince, Contender and Ruby Prince. Ruby Prince was lower ( $P < 0.05$ ) in FRAP AE than O'Henry, Norman, Bounty, Scarlet Prince and Red Globe. Box whisker distribution of FRAP assay showed that the Red Globe peach variety had the most variation (Figure 9).

### 4.5. FIC Assay

Results of ferrous ion chelating (FIC) assay differed from the other antioxidant assays which may be due to the fact that FIC assay measures chelating ability, a different mode of antioxidant action compared to other assays used. The order of highest to lowest chelating ability was Early August Prince, Ruby Prince, Norman, Flame Prince, Summer Gold, Contender, Cary Mac, Bounty, July Flame, Scarlet Prince, Fire Prince, Red Globe,



**Figure 5.** Box and whisker of total phenolics assay (dry weight).



**Figure 6.** Antioxidant activity of DPPH assay A-C means with the same subscript are not significantly different ( $P > 0.05$ ). SE = 1.94, n = 12.

O’Henry (Figure 10). Early August Prince was significantly greater ( $P < 0.05$ ) than July Flame, Scarlet Prince, Fire Prince, Red Globe and O’Henry. O’Henry had significantly lower ( $P < 0.05$ ) chelating ability than Early August Prince, Ruby Prince, Flame Prince and Norman. For chelating ability, Ruby Prince peaches had the largest variation while Bounty, Flame Prince, Red Globe and Scarlet Prince varied the least (Figure 11).

#### 4.6. Correlation of Four Antioxidant Assays

There is no one assay that measures all aspects of antioxidant capacity. Therefore, antioxidant assays based on different modes of action were performed to evaluate overall antioxidant capacity. Correlation of different antioxidant assays was performed to verify the relationship between assays in evaluating different peach varieties. P value of all the Pearson correlation was less than 0.05. TP and DPPH assays were the most closely correlated with an R value of 0.92. Table 2 shows that total phenolics, DPPH and FRAP assays are all positively and significantly correlated while the FIC assay negatively correlated to total phenolics and DPPH and in non-significantly related to the FRAP assay.

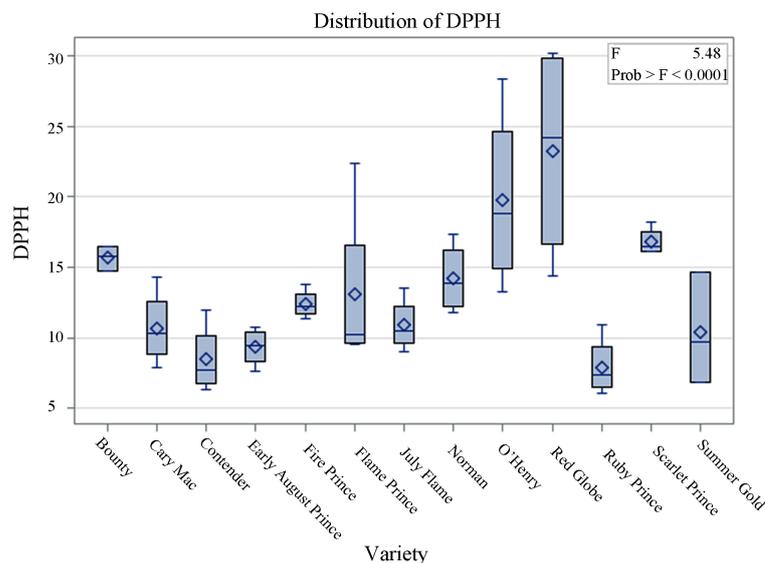


Figure 7. Box and whisker plot of DPPH assay.

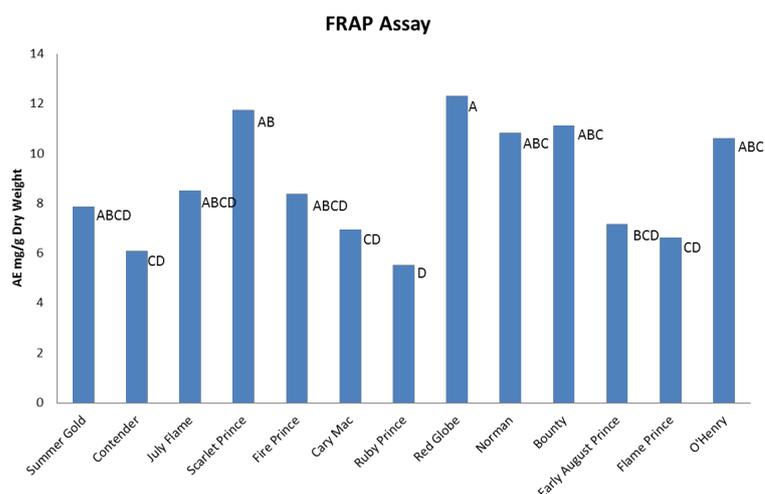
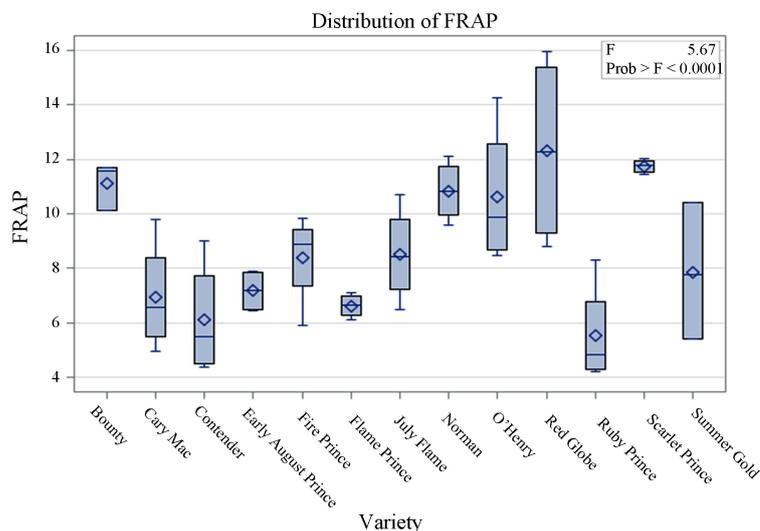


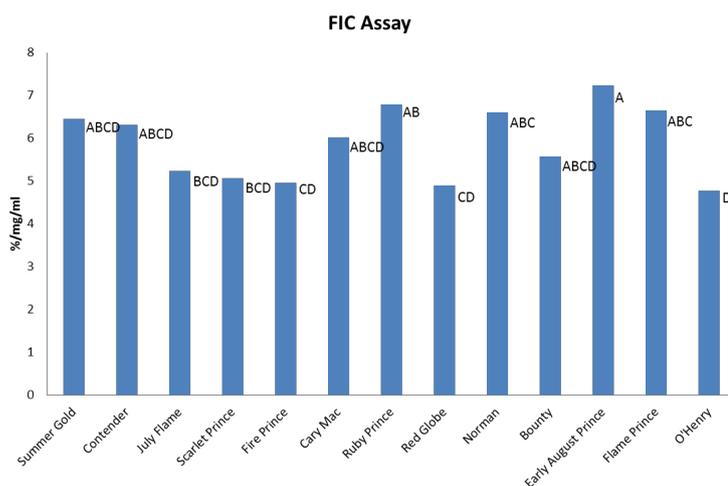
Figure 8. Antioxidant activity of FRAP assay A-D means with the same sub-script are not significantly different ( $P > 0.05$ ). SE = 0.96, n = 12.

Table 2. Correlation of four antioxidant assays.

Pearson correlation coefficients (R), N = 13				
Prob >  r  under $H_0$ : $\rho = 0$				
		DPPH assay	FRAP assay	FIC assay
TP assay	R	0.92	0.88	-0.76
	P	<0.0001	<0.0001	0.0024
DPPH assay	R		0.87	-0.69
	P		<0.0001	0.0084
FRAP assay	R			-0.66
	P			0.014



**Figure 9.** Box and whisker plot of FRAP assay.



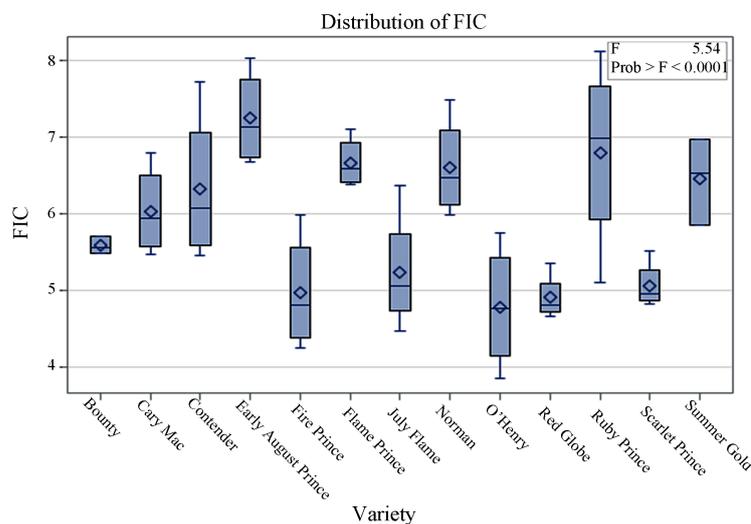
**Figure 10.** Antioxidant activity of FIC assay A-D means with the same subscript are not significantly different ( $P > 0.05$ ). SE = 0.36, n = 12.

#### 4.7. Correlation of Color and Total Phenolic Content

Correlation of color and total phenolic content was performed to determine the relationship of color and phenolic content. P values of all Pearson correlations were greater than 0.05 and R values were all close to 0 (**Table 3**) therefore, peach skin color measurements were not significantly correlated to total phenolic content of peaches.

The total phenolic content and antioxidant activity of peach skin were affected by peach cultivar. The range of total phenolics content was 8.38 - 18.81 (Gallic Acid Equivalent mg/g dry weight). Babbar [19] reported total phenolics content of six fruit residues which are kinnow seed 3.68 mg GAE/g dry weight, kinnow peel 17.5 mg GAE/g dry weight, litchi seed 17.9 mg GAE/g dry weight, litchi pericarp 24.6 mg GAE/g dry weight, grape seed 37.4 mg GAE/g dry weight, banana peel 3.8 mg GAE/g dry weight. Though the total phenolic content of peach skins were lower than grape seed, all the peach skins from different varieties were higher than kinnow seed and banana peel. Some varieties like Red Globe and Scarlet prince were similar to kinnow peel and litchi seed. In general, peach skin possessed a high antioxidant capacity. The order of antioxidant capacity strength of TP, DPPH and FRAP assays followed a similar trend due to cultivar while the result of FIC assay results did not match the other assays.

Previous research has showed that total phenolic content of O'Henry peach skin evaluated by HPLC-DAD



**Figure 11.** Box and whisker plot of FIC assay.

**Table 3.** Correlation of color and total phenolic content.

Pearson correlation coefficients (R), N = 13						
Prob >  r  under H <sub>0</sub> : Rho = 0						
		L*	a*	b*	C*	h*
Total phenolic	R	-0.2365	-0.0212	-0.2045	-0.2234	-0.19
	P	0.4365	0.9451	0.5027	0.4632	0.5341

method was 120.2 mg/100g fresh weight [20]. In the current study, total phenolics content (Gallic acid equivalent) of O'Henry was 328.7 mg/100g fresh weight. Since total phenolic content was expressed directly in previous research but expressed indirectly as Gallic acid equivalent in current research, they cannot be compared directly. Factors which may influence the total phenolic contents of peaches included growing environment, weather conditions, time of harvest and soil conditions.

Correlations among TP, DPPH and FRAP assays were all high while the correlation coefficient of FIC assay to the other assays was not as high. This is case since the FIC assay measures chelating ability while the other assays measure phenolics and reducing or radical scavenging which are closely related to phenolic structure. While both DPPH assay and FRAP assay measure reducing reactions, the antioxidant capacity evaluated of DPPH assay was generally higher than FRAP assay. DPPH assay is based on the transfer of hydrogen atom, while FRAP assay depends on the electron transfer, particularly reducing of Fe(III) [21]. Not all antioxidants can reduce the Fe(III), and this could be the reason the FRAP assay gave a lower value than the DPPH assay.

## 5. Conclusion

Since peach variety affected antioxidant capacity and phenolic content of skin, use of skin to produce a natural antioxidant additive would require quality control to ensure consistent antioxidant product. Further studies may include HPLC analysis of peach skin components and antioxidant activity test *in vivo*. Also, since peach skin extracts have strong antioxidant activity, production of a natural additive for use as a food additive could be an economic benefit to the peach industry.

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