

Realization of Potassium Chloride Sensor Using Photonic Crystal Fiber

Gopinath Palai¹, Nilambar Mudului², Santosh K. Sahoo¹, Sukanta K. Tripathy^{3*}, Sumanta K. Patanaik³

¹Gandhi Institute for Technological Advancement, Bhubaneswar, India; ²Gandhi Engineering College, Bhubaneswar, India; ³National Institute of Science and Technology, Berhampur, India. Email: *sukantakutripathy@yahoo.co.in

Received October 1st, 2013; revised November 5th, 2013; accepted November 12th, 2013

Copyright © 2013 Gopinath Palai *et al.* This is an open access article distributed under the Creative Commons Attribution License, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

ABSTRACT

We propose to measure the concentration of potassium chloride using photonic crystal fiber having circular air holes of diameter 400 nm. The principle of measurement is based on the linear variation of the transmitted field emerging from the PCF with respect to concentration of potassium chloride. Field distribution in photonic crystal structure is simulated using plane wave expansion (PWE) method. Simulation result reveals that the intensity of transmitted light varies linearly with respect to concentration of potassium chloride filled in the air holes.

Keywords: PCF; Potassium Chloride; PWE

1. Introduction

In the last decade, a great deal of work has been focused on the field of photonic crystal fiber (PCF) technology due to their optical properties. An interesting characteristics display by such structure is the selectivity of the electromagnetic field that can propagate along them, in other words the propagation of certain frequency is prohibited [1]. The operational principle of PCF is based on the photonic band gap. Photonic band gap of the fiber depends on the periodic variation of permittivity as a function of position. This periodic variation can occur in one, two or three dimensions and can be manipulated by controlling the magnitude, shape and period of permittivity. Apart from these parameters, photonic band gap of the same fiber also depends on the defect available in the PCF. The photonic band gap of PCF having defect, controls the electric field distribution in the same fiber. As far as the application of PCF is concerned, it is used almost in all fields including ultra low pass transmission sensor, power deliveries and nonlinear optics [2-4]. Recently, an optical sensor using 2D photonic crystal structure is theoretically designed for measurement of sugar, salt, alcohol and Cygel concentration [5,6]. In this paper, we have theoretically designed a photonic crystal fiber which is used to measure the concentration of potassium

*Corresponding author.

chloride by knowing transmitted intensities. In this case, the principle of concentration measurement is based on linear variation of electric field with concentration. The simulation for electric field distribution is obtained using plane wave expansion method (PWE).

Potassium chloride is a natural compound in the mineral sylite food and seawater. It is necessary for various body functions especially for the proper functioning of heart [7]. Potassium chloride is used in the prevention or treatment of low blood levels of potassium [8]. It is also used in the treatment of hypokalemia [9]. Potassium chloride is used in everyday life as a water softener instead of sodium chloride [10]. The other major use of potassium chloride is the manufacture of fertilizer [11].

The schematic diagram of the PCF is shown in **Figure 1**.

Figure 1 represents a square lattice PCF having defect at centre and silicon is taken as background material. The air holes in this fiber contain solution having different percentage of potassium chloride. The lattice constants and diameter of air holes are 1 µm and 0.4 µm respectively.

2. Mathematical Approach

The electric field distribution in photonic crystal fiber is computed using, Helmoltz equation, which is given by [11].

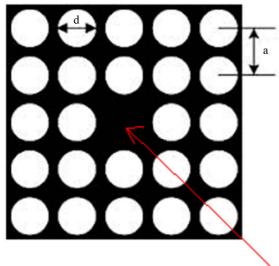




Figure 1. Schematic diagram of photonic crystal fiber.

$$\frac{1}{\in (r)} \nabla \times \{\nabla \times E(r)\} = \frac{\omega^2}{c^2} E(r)$$
(1)

The solution of Equation (1) is expressed as

$$E(r) = E_{k,r}(r) \cdot e^{i.k.r}$$
⁽²⁾

where $E_{k,r}$ is the periodic function with periodicity of lattice. The wave functions are represented interms of Bloch wave and expanded in to fourier series over lattice vector, which is expressed as

$$E_{kr}(x, y) = \sum E'_{kr}(G_r) \exp(i(k_x + G_x, r) \cdot x + (k_y + G_y, r) \cdot y)$$
(3)

Here $E'_{kr}(G_r)$ are eigen vectors to be found during the eigen problem solution. G_x and G_y are called Fourier coefficients for harmonics. Using above equations, we solved eigen value problem and compute the field distribution in the photonic crystal fiber. The transmitted intensity is expressed in terms of electric field as [12].

$$I = 0.5nc_0 \in \left|E\right|^2 \tag{4}$$

where "n" is refractive index modulation, $_0$ is the permittivity of free space, "c" is the velocity of light, E is the peak electric field at coordinate [2,5].

3. Result and Analysis

Table 1 shows the variation of dielectric constant with concentration of potassium chloride as [13].

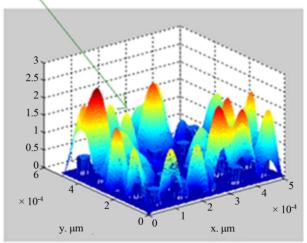
Taking data from **Table 1** and using PWE method, simulation is made to compute the electric field distribution in photonic crystal fiber having different concentrations of potassium chloride in solutions. The simulation results for KCl concentration of 2.5% and 20%, are

shown in **Figures 2(a)** and **(b)** respectively, simulations for other concentrations (5%, 7%, 10%) are done but not shown here.

Table 1. Variation of dielectric	constant	with	concentration
of potassium chloride.			

Permittivity	
1.804	
1.824	
1.833	
1.846	
1.873	







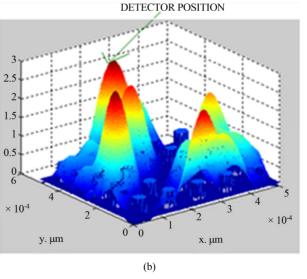


Figure 2. (a) Simulated field distribution diagram for 2.5% KCl concentration; (b) simulated field distribution diagram for 20% KCl concentration.

From **Figures 2(a)** and **(b)**, it is seen that electric field is 0.9 V/ μ m and 2.3 V/ μ m with respect to concentration of 2.5% and 20% at the position of (2,5) in x-y co-ordinate system.

Apart from this, it is also seen that electric field increases with the increase of potassium chloride concentration, which varies from 2.5% to 20%. The same variation of an electric field with respect to KCl concentration is shown in **Figure 3**.

Figure 4 represents the experimental set up to determine the concentration of KCl.

From **Figure 4**, it is seen that, light from a sodium vapor lamp ($\lambda = 590$ nm) is incident on the photonic crystal Fiber. Transmitted light from the photonic crystal fiber is then received by photo detector, which is connected to power meter (optical). A graph is then plotted between transmitted intensity (power measured in the power meter per unit area) and concentration of potassium chloride, which is shown in **Figure 5**.

From **Figure 4**, it is seen that, transmitted intensity increases with the increase of potassium chloride concentration, which varies from 2.5% to 20%. It is also seen that an excellent trend line is fitted with the variation of intensity and concentration.

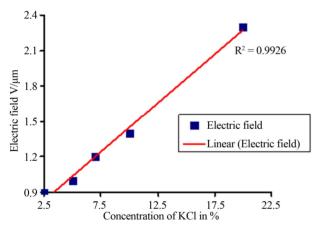


Figure 3. Variation of transmitted intensity with concentration of KCl.

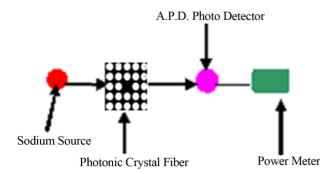


Figure 4. Experimental set up for determining potassium chloride concentration.

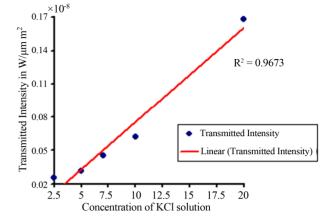


Figure 5. Variation of transmitted intensity with concentration of KCl.

4. Conclusion

Concentration of potassium chloride is estimated by measuring the intensity of transmitted light. Plane wave expansion method is used to determine the field distribution. This measurement is accurate due to its working principle operations in optical frequency. The photonic crystal fiber acts as a sensor to obtain the concentration of potassium chloride in its solution.

REFERENCES

- J. C. Knight, J. Broeng, T. A. Birks and P. St. J. Russell, "Photonic Band Gap Guidance in Optical Fibers," *Science*, Vol. 282, No. 5393, 1998, pp. 1476-1478. <u>http://dx.doi.org/10.1126/science.282.5393.1476</u>
- [2] J. Knight, T. Birks, B. Mangan and P. St. James Russell, "Photonic Crystal Fibers: New Solutions in Fiber Optics," *Optics & Photonic News*, Vol. 13, No. 3, 2002, pp. 26-30. <u>http://dx.doi.org/10.1364/OPN.13.3.000026</u>
- [3] J. Fini and R. Bise, "Progress in Fabrication and Modeling of Microstructured Optical Fibers," *Japanese Journal* of Applied Physics, Vol. 43, 2004, pp. 5717-5730. http://dx.doi.org/10.1143/JJAP.43.5717
- [4] T. A. Birks, J. C. Knight and P. St. J. Russell, "Endlessly Single-Mode Photonic Crystal Fiber," *Optics Letters*, Vol. 22, No. 13, 1997, pp. 961-963.
- [5] G. Palai and S. K. Tripathy, "A Novel Method for Measurement of Concentration Using Two Dimensional Photonic Crystal Structures," *Optics Communications*, Vol. 285, No. 10, 2012, pp. 2765-2768. http://dx.doi.org/10.1016/j.optcom.2012.01.072
- [6] G. Palai, S. K. Tripathy, N. Muduli, D. Patnaik and S. K. Patnaik, "A Novel Method to Measure the Strength of CygelTM by Using Two Dimensional Photonic Crystal Struct Ures," *AIP Conference Proceedings*, Vol. 1461, 2012, pp. 383-386. <u>http://dx.doi.org/10.1063/1.4736926</u>
- [7] http://www.medicinenet.com/potassium_chloride/article.h tm
- [8] D. R. Lide, "CRC Handbook of Chemistry and Physics,"

86th Edition, CRC Press, Boca Raton.

- [9] "Hypokalemia: Treatment & Medication," 2012. http://emedicine.medscape.com/
- [10] Greenwood, N. Norman and A. Earnshaw, "Chemistry of the Elements," Pergamon Press, Oxford, 1984.
- [11] G. Lorient and D. Linden, "New Ingredients in Food Processing: Biochemistry and Agriculture," CRC Press, Boca Raton, 1999, p. 357. <u>http://dx.doi.org/10.1201/9781439822760</u>
- [12] I. A. Sukhoivanov and I. V. Guryev, "Photonic Crystals: Physics and Practical Modeling," Springer, Berlin-Heidelberg, 2009, 242 p. http://dx.doi.org/10.1007/978-3-642-02646-1
- [13] D. P. Subedi, L. Zajickova, V. Bursikova and J. Janca, "Surface Modification of Polymers by Low Pressure Plasma and Its Characterization," *Proceedings of Week of Doctoral Students WDS*'00, Praha, 2000, pp. 303-308.