

Influence of the Fluoride Atoms Doping on the FeSe Superconductor

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ABSTRACT

It is reported the influence of the interstitial atoms doping on the FeSe superconductor. Polycrystalline samples with FeSeF_x and FeSeB_x nominal compositions were prepared by solid state reaction. An enhancement of the superconducting transition temperature was observed in the temperature dependence of the electrical resistivity curve to the $\text{FeSeF}_{0.015}$ sample. $R(T)$ data display superconducting behavior close to 12 K. The T_c increased with F doping by up to 50%. In contrast, boron doping no change the superconducting properties of the FeSe compound. As the $\text{FeSe}_{1-x}\text{Te}_x$ system the fluoride doping introduce a negative chemical pressure in the FeSe superconductor. This fact suggests that fluoride doping have changed the electronic properties of the FeSe phase.

Keywords: Superconductivity; Pnictides; Interstitial

1. Introduction

Since the surprisingly discovery of superconductivity in high temperature of 55 K to the fluorine-doped rare-earth iron oxyarsenides, $\text{REFeAs}_{1-x}\text{F}_x$, many research has been made on new superconducting materials [1-4]. The new materials class has Fe and share a common structural feature. In 2008, Hsu *et al.* [5] reported the superconductivity at 8 K in PbO-prototype FeSe compound knows as “11” family. FeSe system is one of the most attracting systems, because it has the simplest crystal structure among the iron-based superconductors [5,6]. McQueen has been shown that superconductivity in $\text{Fe}_{1+\delta}\text{Se}$ is destroyed by very small changes in stoichiometric [7]. They also showed that non-superconducting $\text{Fe}_{1+\delta}\text{Se}$ is not magnetically ordered down to 5 K and it is destroyed by very small changes in stoichiometry. Then, immediate instability against an ordered magnetic state should not be considered as intrinsic characteristic [7].

While a superconducting transition temperature T_c of FeSe at ambient pressure is ~ 8 K, the T_c has been raised to 27 K with the application of pressure [8-10]. Compared with the layered FeAs systems, FeSe has not only the same planar sublattices but also displays structure and magnetic instability [11-14]. Within the family of “11” binary iron chalcogenides the superconducting state exists over a wide range of Te-doping in the Fe (Se, Te) solid solution with maximum T_c of 15 K [15-22]. These substitutions generate chemical pressures, because S, Se and

Te have the same valence and the different ionic radius. Considering the huge positive pressure effect on T_c , compressing the lattice seems to be effective to enhance T_c . However the increase of T_c in $\text{FeSe}_{1-x}\text{S}_x$, which is chemically pressured FeSe, is only 2 K. On the other hand, the increase of T_c in $\text{FeSe}_{1-x}\text{Te}_x$, which is applied a negative chemical pressure, is the larger value of 5 K. In fact, the physical pressure and chemical pressure effects are not equivalent in FeSe system [15-22]. Here we report the influence of interstitial doping on the FeSe superconductor. We found that the $\text{FeSe} + 0.5\% \text{FeF}_3$ shows superconducting transition like Fe (SeTe) solid solution. On the other hand, the low content boron atoms doping do not change the electric and magnetic properties of FeSe superconductor.

2. Experimental Procedure

FeSeF_x were prepared using mixtures of appropriate amount of high purity Fe, Se, and $x\%$ FeF_3 ($x = 0.1, 0.3, \text{ and } 0.5$) powders. The mixtures were compacted in square shape of $10 \times 10 \text{ mm}^2$ and 2 mm thick, sealed in a quartz ampoule, and placed inside a box furnace model Lindberg/Blue M furnace at 1000°C for 10 hours, and cooled at 400°C and, then kept for 3 days. After this heat treatment, the samples were ground and homogenized in agate mortar, compacted again in the same dimensions mentioned before, and sintered at 400°C for 3 days. After this additional heat treatment, the samples were submitted to a rapid

quenching in water in order to trap the tetragonal phase. All samples were characterized by X-ray powder diffraction in a Shimadzu diffractometer (model XRD 6000) using Ni filter $\text{CuK}\alpha$ radiation. The simulation of the structure and refinement of the lattice parameters were carried out by using both the Powder Cell software and the Rietveld method. The simulation of the crystal structure was based on the literature data and the results were compared with the experimental diffractograms. The electrical resistance as a function of temperature, $R(T)$, were performed by conventional four-point probe method in the temperature interval between 2.0 and 300 K. In order to remove the thermal power effects arising in the silver-soldered samples terminal, the measurements were done by applying an alternating polarity of the dc current and taking the average voltage between the two measurements.

3. Results and Discussions

X-ray powder diffraction data at room temperature for the $\text{FeSe} + x\% \text{FeF}_3$ ($x = 0.1; 0.3$ and 0.5) samples are displayed in **Figure 1**. All peaks can be indexed based on tetragonal structure space group $P4/nmm$, with lattice parameters $a = 3.36 \text{ \AA}$ and $c = 5.52 \text{ \AA}$. These results reveal that F atoms are been solubilized within the crystalline structure of the space group $P4/nmm$.

In order to determine what site fluoride doping is occupied the refinement of X-ray data was made. The initial analysis suggests that F atoms could occupy the “4d” Wyckoff position. While the increase F content from nominally 0.3% to 0.5% leads to relatively small changes in the lattice parameters. The anisotropic expansion of the crystalline structure can be observed on the (200) and (003) reflection. It is possible to observed in the (003) reflection that there is not significantly expansion/contraction of the lattice parameter. On the other hand the (200) reflection the small change could be related.

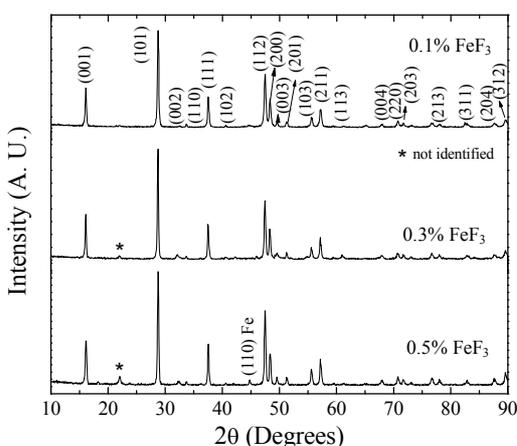


Figure 1. X-ray diffraction for the $\text{FeSe} + x\% \text{FeF}_3$ showing that the fluoride does not collapse the PbO prototype structure.

In order to study the transport properties of the $\text{FeSe} + x\% \text{FeF}_3$ samples, electrical resistance measurements as a function of temperature have been performed. $R(T)$ curve for the $\text{FeSe} + 0.1\% \text{FeF}_3$ reveals a typical metallic behavior between 15 and 150 K (**Figure 2**).

The superconducting transition at 10 K with transition broadening of 3 K can be observed in the measurement. This fact shows the fluoride doping does not destroy the superconductivity of the FeSe superconductor. On the other hand, the fluoride doping increases the superconducting temperature. In the inset of the **Figure 2** is displayed the measurement between 4.0 K and 25.0 K with different electrical currents.

In the **Figure 3** the $R(T)$ curve for the $\text{FeSe} + 0.3\% \text{FeF}_3$ reveals like metallic behavior between 15 and 300 K. On the other hand, the superconducting transition is close to 11.2 K. This fact suggests the fluoride is changing the transport properties of the FeSe superconductor.

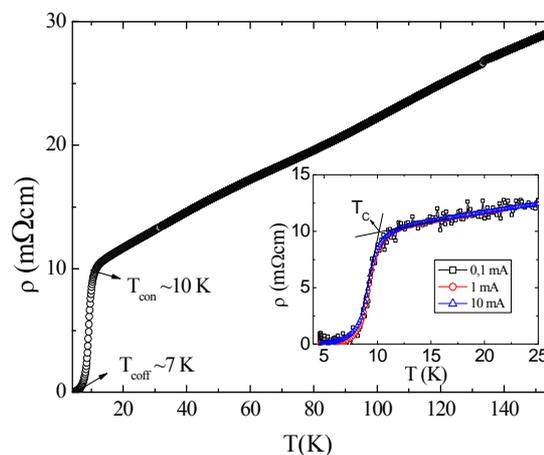


Figure 2. Electrical resistance as a function of temperature from 4 to 150 K to the $\text{FeSe} + 0.1\% \text{FeF}_3$ shows the superconducting transition at 10 K. The normal state is an electron-phonon scattering.

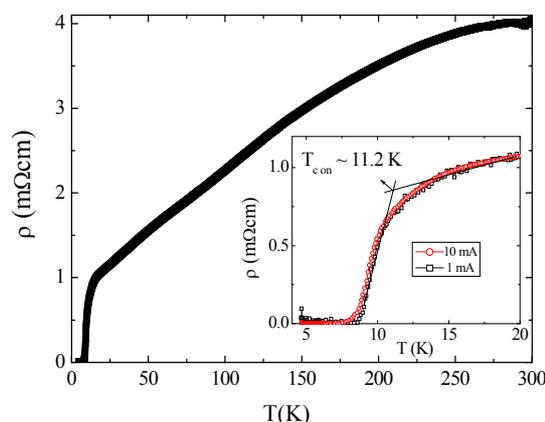


Figure 3. Electrical resistivity as a function of temperature for the $\text{FeSe} + 0.3\% \text{FeF}_3$ sample. Inset displays the superconducting transition starting at ~ 11.2 K.

In **Figure 4** is shown the $R(T)$ for the FeSe + 0.5% FeF₃. The FeSe + 0.5% FeF₃ compound superconducts at 13 K (onset temperature). This transition temperature is compared to FeSe_{0.5}Te_{0.5}. The transition broadening is due to impurity in the FeSe + 0.5% FeF₃ as is possible to observed in the X-ray diffraction data.

Figure 5 show the normalized electrical resistance as a function of temperature to the FeSeF_x. The $R(T)$ data is clear the F doping is changed the electrical behavior of the FeSe superconductor.

The low content boron atoms doping do not collapse the crystalline structure nor change the electrical and magnetic properties of the FeSe superconductor.

4. Conclusion

This work has investigated the effect of fluoride doping in the FeSe superconductor, which exhibits the simplest crystal structure among the iron-based superconductors.

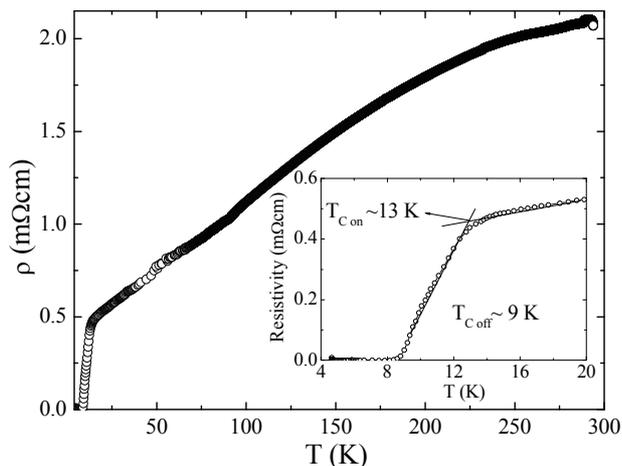


Figure 4. Electrical resistivity as a function of temperature for the FeSe + 0.5% FeF₃ sample. Inset displays a superconducting transition starting at ~13 K.

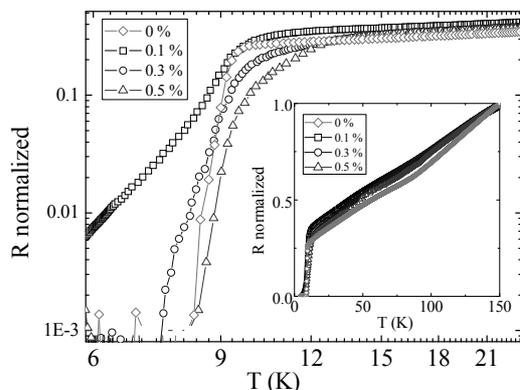


Figure 5. Electrical resistance normalized as a function of temperature to the FeSeF_x compounds. The FeSe + 0.5% FeF₃ compound superconducts at 13 K (onset temperature). This transition temperature is compared to FeSe_{0.5}Te_{0.5}.

The results show is possible to dope the FeSe superconductor with the interstitial atoms. The boron doping not changes the superconducting properties of the FeSe. On the other hand the fluoride doping applied a negative chemical pressure in the FeSe superconductor which increases the T_c like in the FeSe_{1-x}Te_x system.

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