

# The Reaction of Cyanoacetylhydrazine with Furan-2-Aldehyde: Novel Synthesis of Thiophene, Azole, Azine and Coumarin Derivatives and Their Antitumor Evaluation

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## ABSTRACT

The reaction of cyanoacetylhydrazine **1** with furan-2-aldehyde **2** gives the hydrazide-hydrazone derivative **3**. The latter compound undergoes a series of heterocyclization reactions to give new heterocyclic compounds. The antitumor evaluation of the newly synthesized products against three cancer cell lines, namely breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268) were recorded. Some of the synthesized compounds show high inhibitory effects.

**Keywords:** Acetylhydrazine; Hydrazide-Hydrazone; Thiophene; Thiazole; Pyridine; Coumarin; Antitumor

## 1. Introduction

Substituted aminothioxomethyl hydrazides are important building blocks in synthetic heterocyclic chemistry. The S/N regioselective nucleophilic competition in the synthesis of heterocyclic systems by intermolecular and intramolecular cyclization, as well as the change in reaction conditions which might favor N-attack or S-attack or even attack on the substituted terminals are important factors for the diversity of the produced heterocyclic systems from the title reaction precursor. On the other hand, the synthetic potential and biological activity of several heterocyclic related to the named hydrazides have been explored to the maximum extent. Among the pharmacological profiles are their antimicrobial [1,2], antitubercular [3,4], anticonvulsant [5,6] anti-inflammatory [7, 8], antidepressant [9], antitumor [10], and analgesic activities [11]. Continuing our interest in developing new heterocyclic systems based on hydrazide-hydrazones [12, 13] as well as design and synthesis of heterocyclic with promising biological activities [14-17], we focused our work on developing novel polyfunctionalized heterocyclic compounds with potential bioactivity.

## 2. Results and Discussion

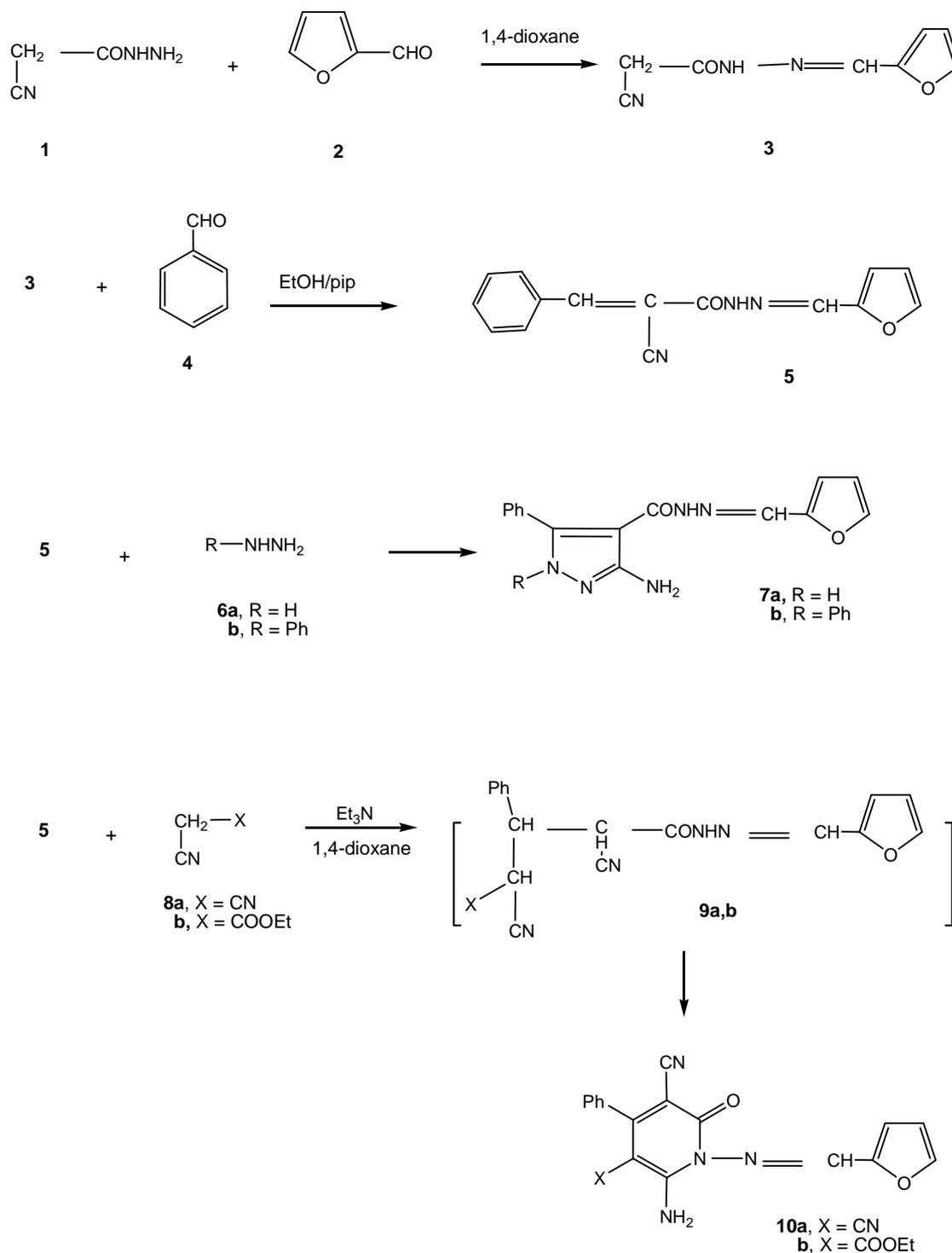
In this work we report the reaction of cyanoacetylhydra-

zine (**1**) with furan-2-aldehyde (**2**) to give the hydrazide-hydrazone derivative **3**, the structure of which was confirmed on the basis of analytical and spectral data. Thus, the <sup>1</sup>H NMR spectrum showed a singlet at  $\delta$  6.60 (2H) ppm corresponding to CH<sub>2</sub> group, a multiplet at  $\delta$  6.52 - 7.88 (3H) ppm corresponding to furan protons, a singlet at  $\delta$  7.54 (1H) ppm corresponding to CH group, and a singlet (D<sub>2</sub>O-exchangeable) at  $\delta$  11.62 (1H) ppm corresponding to NH group.

The formed hydrazide-hydrazone derivative **3** was subjected through a series of heterocyclization reactions to form products with expected pharmaceutical applications. Thus, condensation of **3** with benzaldehyde (**4**) in ethanolic/piperidine solution gave the benzylidene derivative **5**. The reaction of compound **5** with either hydrazine hydrate (**6a**) or phenylhydrazine (**6b**) gave the pyrazole derivatives **7a** and **7b**, respectively.

On the other hand, the reaction of compound **5** with either malononitrile (**8a**) or ethyl cyanoacetate (**8b**) gave the pyridone derivatives **10a** and **10b**, respectively. The reaction took place *via* 1,6-intramolecular dipolar cyclization of the intermediate **9a,b** (cf. **Scheme 1**), analytical and spectral data of the latter compounds were consistent with the assigned structures. Thus, the <sup>1</sup>H NMR spectrum of **10b** showed, a triplet at  $\delta$  1.29 (3H) ppm indicating the presence of the CH<sub>3</sub> group, a quartet at  $\delta$  4.24 (2H) ppm corresponding to CH<sub>2</sub> group, a multiplet at  $\delta$  6.52 - 7.88 (3H) ppm for furan protons, a singlet

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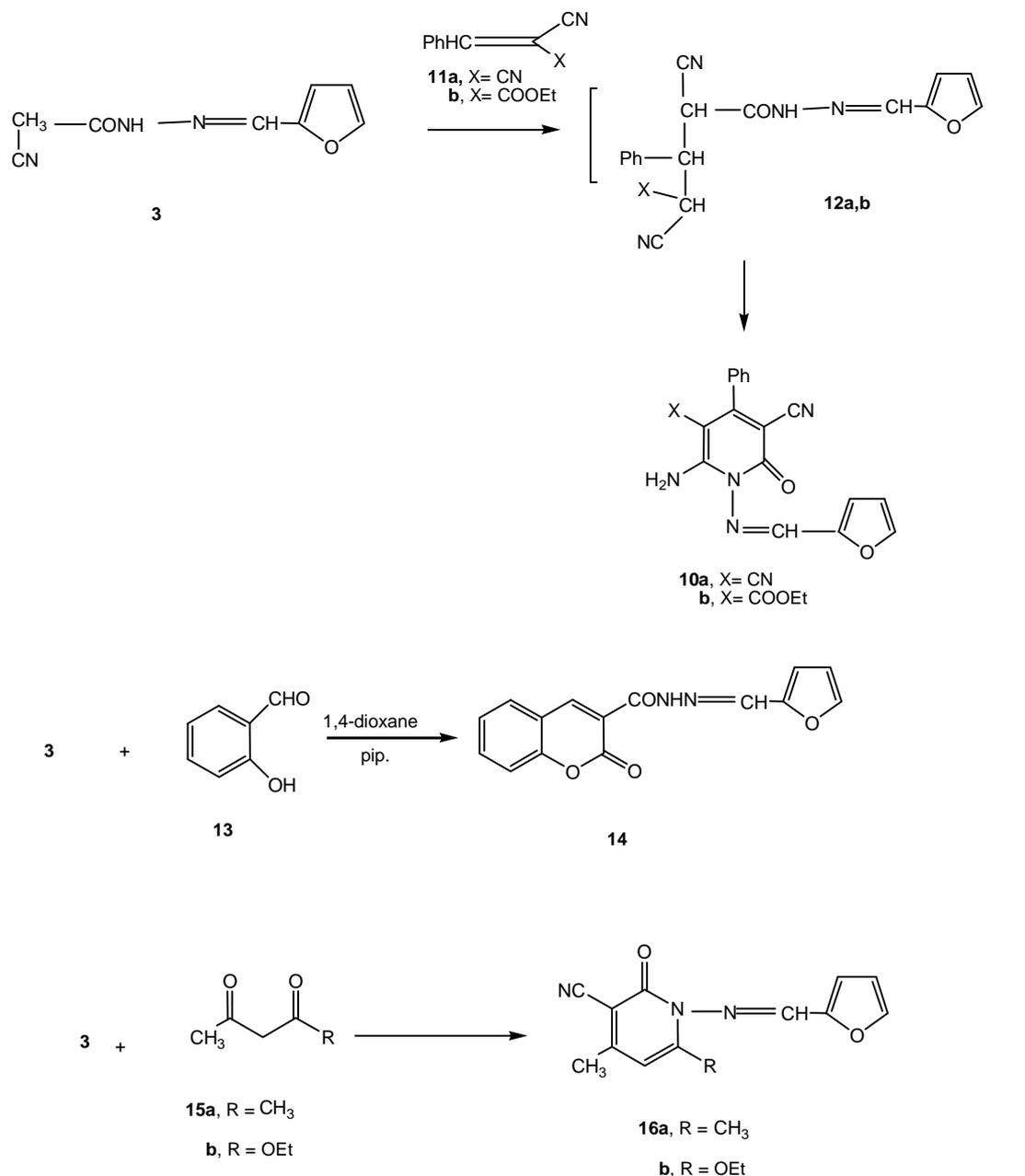


Scheme 1. Synthesis of compounds 3, 5, 7a,b and 10a,b.

(D<sub>2</sub>O-exchangeable) at  $\delta$  7.05 (2H) ppm equivalent to the NH<sub>2</sub> group, a multiplet at  $\delta$  7.17 - 7.40 (5H) ppm indicating the phenyl protons and a singlet at  $\delta$  7.56 (1H) ppm corresponding to the CH group.

Further confirmation of the structures of **10a,b** was achieved through an alternative synthetic route involving

the reaction of compound **3** with the cinnamionitrile reagents **11a,b** to afford the same pyridone derivatives **10a, b** (finger print IR, mp. and mixed mp.) with better yields (95%, 98%) than in their formation by the reaction of compound **5** and either malononitrile (75% yield) or ethyl cyanoacetate (74% yield) (cf. **Scheme 2**). On the



Scheme 2. Synthesis of compounds 10a,b, 14 and 16a,b.

other hand, compound **3** condensed with salicylaldehyde (**13**) to give the coumarin derivative **14**. The analytical and spectral data of compound **14** are in agreement with the proposed structure. Thus, the  $^1\text{H}$  NMR spectrum showed, a singlet at  $\delta$  6.95 (1H) ppm corresponding to coumarin H-4, a singlet at  $\delta$  7.52 (1H) ppm corresponding to CH group, a multiplet at  $\delta$  6.52 - 7.88 (3H) ppm corresponding to furan protons, a multiplet at  $\delta$  7.33 - 7.55 (4H) ppm corresponding to aromatic protons and a singlet ( $\text{D}_2\text{O}$ -exchangeable) at  $\delta$  11.61 (1H) ppm corresponding to NH group. The reactivity of compound **3** towards 1,3-dicarbonyl compounds was studied to afford pyridine

derivatives with wide range of biological activities thus, the reaction of **3** with either acetylacetone (**15a**) or ethyl acetoacetate (**15b**) gave the pyridine derivatives **16a** and **16b**, respectively (cf. Scheme 2). The structures of the latter products were established on the basis of analytical and spectral data. Thus,  $^1\text{H}$  NMR spectrum of **16b** showed a triplet at  $\delta$  1.35 (3H) ppm corresponding to  $\text{CH}_3$  group, a singlet at  $\delta$  2.34 (3H) ppm corresponding to  $\text{CH}_3$  group, a quartet at  $\delta$  4.22 (2H) ppm corresponding to  $\text{CH}_2$  group, a singlet at  $\delta$  5.79 (1H) ppm corresponding to pyridine proton, a singlet at  $\delta$  7.51 (1H) ppm corresponding to CH group and a multiplet at  $\delta$  6.52 - 7.88

(3H) ppm corresponding to furan protons. On the other hand, we studied the reactivity of compound **3** with cyanomethylene reagents thus, the reaction of **3** with either malononitrile (**8a**) or ethyl cyanoacetate (**8b**) gave the 2-pyridone derivatives **19a,b**. The formation of the latter products took place *via* the intermediate formation of **17a,b** (cf. **Scheme 3**). The <sup>1</sup>H NMR spectrum of **18b** as an example showed a singlet at  $\delta$  5.80 (1H) ppm corresponding to the pyridine proton, a multiplet at  $\delta$  6.52 - 7.88 (3H) ppm corresponding to furan protons, a singlet (D<sub>2</sub>O-exchangeable) at  $\delta$  7.00 (2H) ppm corresponding to NH<sub>2</sub> group, a singlet at  $\delta$  7.52 (1H) ppm corresponding to CH group, and a singlet (D<sub>2</sub>O-exchangeable) at  $\delta$  12.56 (1H) ppm corresponding to OH group.

The reaction of **3** with either acetophenone (**19**) or 4-methoxy acetophenone (**20**) in ammonium acetate gave the Knoevenagel condensated products **21** and **22** respectively.

Next, we studied the reactivity of the active methylene group present in compound **3** towards diazonium salts. Thus, the reaction of compound **3** with aryldiazonium chlorides **23a-23d** gave the phenylhydrazone derivatives **24a-24d**. Analytical and spectral data of the latter products are all consistent with the proposed structures (see experimental section). Compound **24a** reacted with ethyl cyanoacetate (**8b**) to give the pyridazine derivative **26**. The reaction took place *via* the intermediate formation of **25** (cf. **Scheme 3**). On the other hand, the reaction of **24a** with phenyl isothiocyanate (**27**) gave the triazine derivative **29** *via* the intermediate formation of **28** (cf. **Scheme 3**). Analytical and spectral data of the latter product are in agreement with the proposed structure.

Recently our research group was involved through a comprehensive program dealing with studying the reaction of phenylisothiocyanate with active methylene reagents followed by heterocyclization with  $\alpha$ -halo ketones. The reactions lead to the formation of either thiazole or thiophene derivatives depending on the reaction conditions and the nature of the  $\alpha$ -halocarbonyl compounds.

In continuation of this program we report here the reaction of compound **3** with phenylisothiocyanate (**27**) in DMF solution containing KOH to form the intermediate potassium sulphide salt **30**. The latter intermediate underwent heterocyclization upon reaction with ethyl chloroacetate (**31**) to give the thiazole derivative **33**. Formation of the latter product took place *via* the intermediate formation of **32**. Similarly the reaction of **30** with  $\alpha$ -chloroacetone (**34a**) or phenacyl bromide (**34b**) gave the thiazole derivatives **35a** and **35b**, respectively (**Scheme 4**). Analytical and spectral data are consistent with the proposed structures. Thus, <sup>1</sup>H NMR spectrum of **35a** showed a singlet at  $\delta$  2.35 (3H) ppm corresponding to CH<sub>3</sub> group, a singlet at  $\delta$  5.81 (1H) ppm corresponding to thiazole proton, a multiplet at  $\delta$  6.25 - 7.20 (5H) ppm

corresponding to aromatic protons, a multiplet at  $\delta$  6.52 - 7.88 (3H) ppm corresponding to furan protons, a singlet at  $\delta$  7.52 (1H) ppm corresponding to CH group and a singlet (D<sub>2</sub>O-exchangeable) at  $\delta$  11.62 (1H) ppm corresponding to NH group.

## 2.1. Antitumor Activity

### 2.1.1. Material, Methods and Reagents

Fetal bovine serum (FBS) and L-glutamine, were from Gibco Invitrogen Co. (Scotland, UK). RPMI-1640 medium was from Cambrex (New Jersey, USA). Dimethyl sulfoxide (DMSO), doxorubicin, penicillin, streptomycin and sulforhodamine B (SRB) were from Sigma Chemical Co. (Saint Louis, USA). Samples Stock solutions of selected compounds from **3-34a,b** were prepared in DMSO and kept at -20°C. Appropriate dilutions of the compounds were freshly prepared just prior the assays. Final concentrations of DMSO did not interfere with the cell growth.

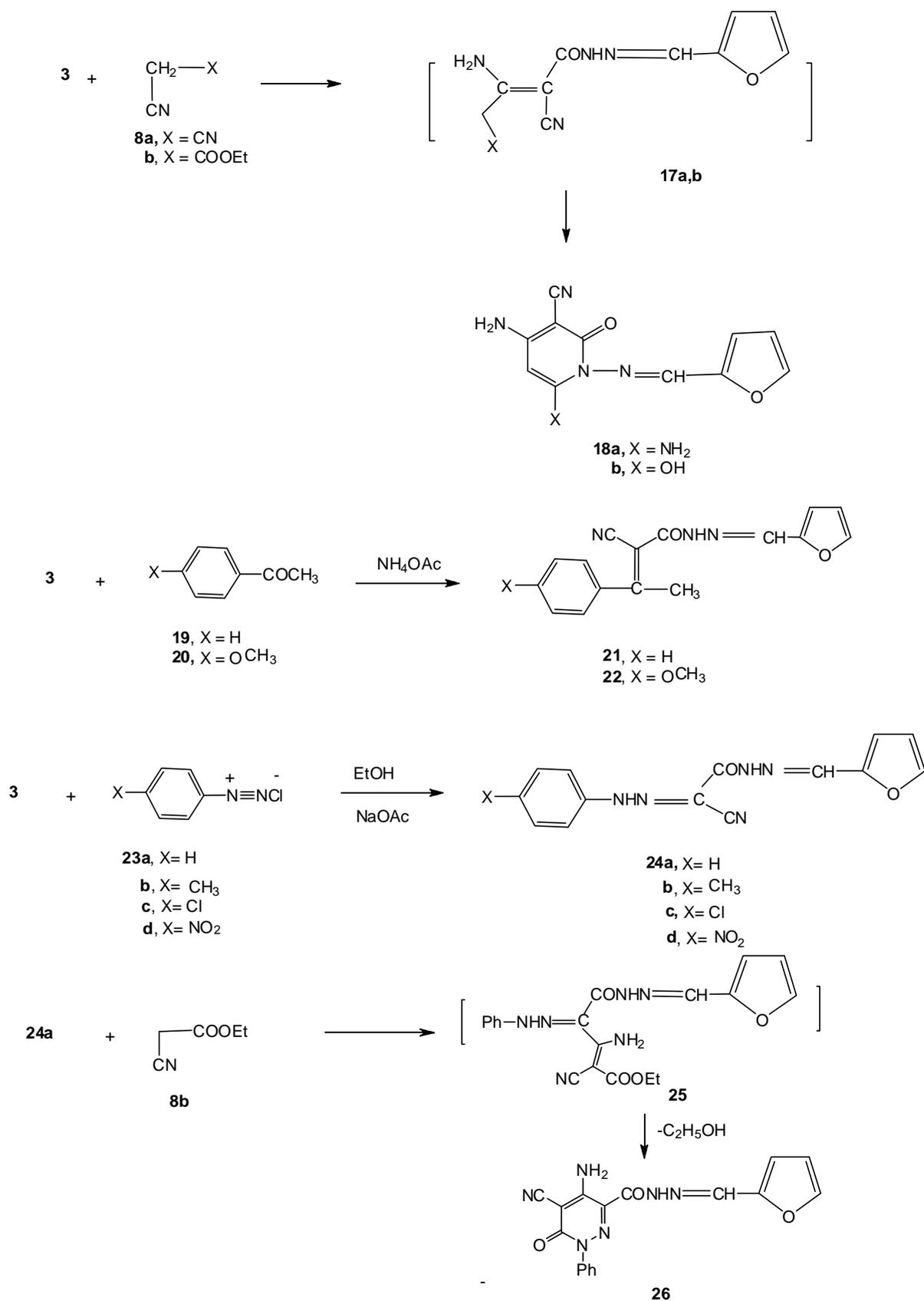
### 2.1.2. Cell Cultures

Three human tumor cell lines, MCF-7 (breast adenocarcinoma), NCI-H460 (non-small cell lung cancer), and SF-268 (CNS cancer) were used. MCF-7 was obtained from the European Collection of Cell Cultures (ECACC, Salisbury, UK) and NCI-H460 and SF-268 were kindly provided by the National Cancer Institute (NCI, Cairo, Egypt). They grow as monolayer and routinely maintained in RPMI-1640 medium supplemented with 5% heat inactivated FBS, 2 mM glutamine and antibiotics (penicillin 100  $\mu$ /mL, streptomycin 100  $\mu$ g/mL), at 37°C in a humidified atmosphere containing 5% CO<sub>2</sub>. Exponentially growing cells were obtained by plating  $1.5 \times 10^5$  cells/mL for MCF-7 and SF-268 and  $0.75 \times 10^4$  cells/mL for NCI-H460, followed by 24 h of incubation. The effect of the vehicle solvent (DMSO) on the growth of these cell lines was evaluated in all the experiments by exposing untreated control cells to the maximum concentration (0.5%) of DMSO used in each assay.

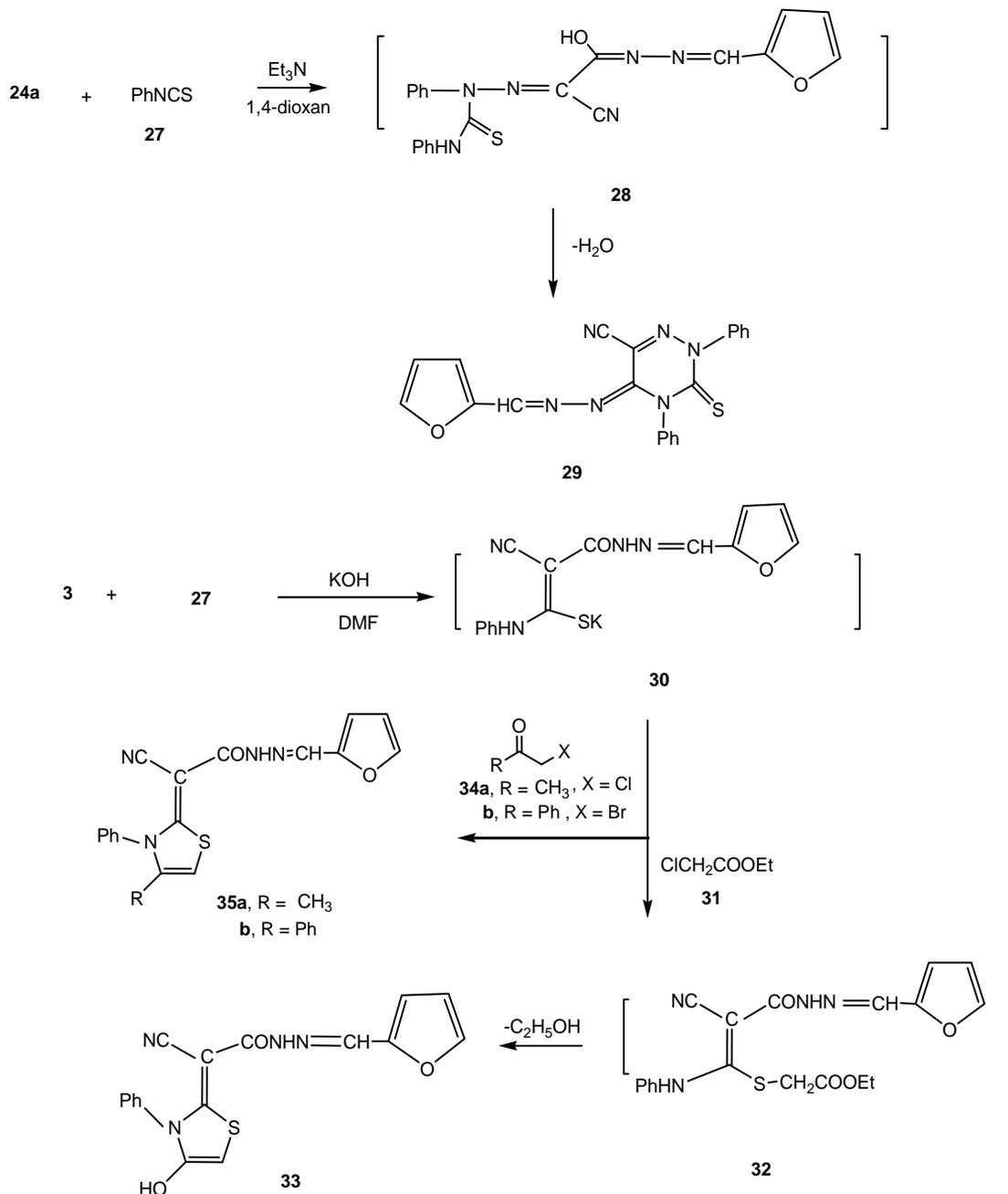
### 2.1.3. Effect on the Growth of Human Tumor Cell Lines

The effect of selected compounds from the newly synthesized products **3-35a,b** was evaluated on the *in vitro* growth of three human tumor cell lines representing different tumor types, namely, breast adenocarcinoma (MCF-7), non-small cell lung cancer (NCI-H460) and CNS cancer (SF-268), after a continuous exposure of 48 h. The evaluations of the newly synthesized products were measured using doxorubicin as the positive control and the results are summarized in **Table 1**.

All the compounds were able to inhibit the growth of the human tumor cell lines in a dose-dependent manner



Scheme 3. Synthesis of compounds 18a, 18b, 21, 22, 24a-24d and 26.



Scheme 4. Synthesis of compounds 29, 30, 33 and 35a,b.

(data not shown). The pyrazole derivative **7b** and the coumarine derivative **14** showed the best results, exhibiting an equivalent potency in all the three tumor cell lines which is still much lower than the gram positive control doxorubicin. On the other hand, compounds **3**, **7a**, **10a**, **18a**, **18b**, **21**, **22**, **24c**, **26** and **33** showed moderated growth inhibitory effect. Comparing the activities of **18a** and **18b** it is observed that the hydroxyl group in **18b** presents a stronger growth inhibitory effect than the amino substituent in **18a** although the results in NCI-H460 cell line are comparable. It is clear from **Table 1**

that some compounds like **10b**, **16a,b** and **35b** showed very low activity towards certain cell line, MCF-7, and moderate activity towards other cell lines.

Results are given in concentrations that were able to cause 50% of cell growth inhibition (GI<sub>50</sub>) after a continuous exposure of 48 h and show means  $\pm$  SEM of three-independent experiments performed in duplicate.

### 3. Conclusion

In conclusion, we have developed efficient methods for the synthesis of new hydrazone-hydrazone derivatives *via*

**Table 1.** Effect of the newly synthesized products on the growth of three human tumor cell lines.

| Compound           | GI <sub>50</sub> (μmol/L) |              |              |
|--------------------|---------------------------|--------------|--------------|
|                    | MCF-7                     | NCI-H460     | SF-268       |
| <b>3</b>           | 20.0 ± 0.2                | 30.6 ± 1.4   | 38.4 ± 0.6   |
| <b>5</b>           | 4.1 ± 0.7                 | 2.2 ± 0.8    | 2.4 ± 1.8    |
| <b>7a</b>          | 36.6 ± 12.2               | 30.6 ± 8.6   | 60.4 ± 14.8  |
| <b>7b</b>          | 0.04 ± 0.01               | 0.01 ± 0.008 | 0.03 ± 0.01  |
| <b>10a</b>         | 20.2 ± 0.4                | 26.3 ± 0.8   | 30 ± 0.8     |
| <b>10b</b>         | 70.6 ± 16.9               | 38.9 ± 10.8  | 50.8 ± 8.6   |
| <b>14</b>          | 0.1 ± 0.07                | 0.02 ± 0.008 | 1.2 ± 0.2    |
| <b>16a</b>         | 42.7 ± 17.5               | 20.2 ± 12.8  | 50.0 ± 9.01  |
| <b>16b</b>         | 50.1 ± 0.7                | 23.2 ± 4.8   | 18.4 ± 1.8   |
| <b>18a</b>         | 22.7 ± 17.5               | 36.2 ± 12.8  | 40.0 ± 9.01  |
| <b>18b</b>         | 12.7 ± 17.5               | 40.2 ± 12.8  | 50.0 ± 9.01  |
| <b>21</b>          | 10.8 ± 0.6                | 12.5 ± 0.8   | 16.7 ± 1.6   |
| <b>22</b>          | 20.0 ± 0.2                | 30.6 ± 1.4   | 38.4 ± 0.6   |
| <b>24a</b>         | 35.0 ± 1.8                | 44.0 ± 0.8   | 20.5 ± 1.1   |
| <b>24b</b>         | 11.9 ± 0.6                | 14.1 ± 0.6   | 20.3 ± 0.5   |
| <b>24c</b>         | 18.0 ± 0.6                | 20.0 ± 0.4   | 30.5 ± 8.0   |
| <b>24d</b>         | 10.8 ± 0.6                | 12.5 ± 0.8   | 16.7 ± 1.6   |
| <b>26</b>          | 22.4 ± 10.2               | 25.1 ± 0.8   | 16.9 ± 4.8   |
| <b>29</b>          | 33.2 ± 0.6                | 15.3 ± 1.4   | 20.3 ± 1.5   |
| <b>33</b>          | 20.0 ± 0.4                | 24.3 ± 0.8   | 32 ± 0.8     |
| <b>35a</b>         | 31.0 ± 1.8                | 40.0 ± 0.8   | 18.5 ± 1.1   |
| <b>35b</b>         | 70.6 ± 16.9               | 38.9 ± 10.8  | 50.8 ± 8.6   |
| <b>doxorubicin</b> | 0.04 ± 0.008              | 0.09 ± 0.008 | 0.09 ± 0.007 |

the reaction of cyanoacetylhydrazine with furan-2-aldehyde. The produced product underwent a series of heterocyclizations. Their antitumor evaluations through the three cancer cell lines showed that compounds **7b** and **14** showed the maximum inhibitory effect among the tested compounds.

#### 4. Experimental

All melting points were determined in open capillaries and are uncorrected. Elemental analyses were performed on a Yanaco CHNS Corder elemental analyzer (Japan). IR spectra were measured using KBr discs on a Pye Unicam SP-1000 spectrophotometer. <sup>1</sup>H NMR spectra were measured on a Varian EM 390 - 200 MHz instrument in CD<sub>3</sub>SOCD<sub>3</sub> as solvent using TMS as internal

standard and chemical shifts are expressed as δ ppm. Mass spectra were recorded on Kratos (75 eV) MS equipment (Germany). Elemental analyses were carried out by the Micro-analytical Data Center at Cairo University and were performed on Vario EL III Elemental CHNS analyzer.

**2-Cyano-N'-(furan-2-ylmethylene)acetohydrazide (3).** To a solution of compound **1** (0.099 g, 1.0 mmol) in 1,4-dioxane (30 mL) and furfural (**2**) (0.09 g, 1.0 mmol) were added. The reaction mixture was heated under reflux for 3 hr, then was left until the reaction mixture be cooled. The formed solid product was collected by filtration. Yellow crystals from 1,4-dioxane, 98% (0.097 g) yield; mp. 190°C - 192°C, MS: m/z = 177 (M<sup>+</sup>), IR (KBr): ν/cm<sup>-1</sup> = 3200 (NH), 3059 (CH<sub>Ar</sub>), 2198 (CN), 1690 (CO), 1550 (C=N), 750 (C-O), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>): δ = 6.60 (s, 2H, CH<sub>2</sub>), 6.52 - 7.88 (m, 3H, furan-CH), 7.54 (s, 1H, CH), 11.62 (s, 1H, NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>8</sub>H<sub>7</sub>N<sub>3</sub>O<sub>2</sub> (177.16): C, 54.24; H, 3.98; N, 23.72%. Found: C, 54.11; H, 4.25; N, 23.59%.

**2-Cyano-N'-(furan-2-ylmethylene)-3-phenylacrylohydrazide (5).** To a solution of compound **3** (0.177 g, 1.0 mmol) in ethanol (30 mL), piperidine (3 drops) and benzaldehyde (**4**) (0.10 g, 1.0 mmol) were added. The reaction mixture was heated under reflux for 3 h, and then poured on an ice/water mixture containing a few drops of hydrochloric acid. The formed solid product was collected by filtration.

Yellow crystals from 1,4-dioxane, 95% (0.168 g) yield; mp. 140°C - 142°C, MS: m/z = 265 (M<sup>+</sup>), IR (KBr): ν/cm<sup>-1</sup> = 3200 (NH), 3159 (CH<sub>Ar</sub>), 2200, 1680 (2CN), 1580 (C = C), 1550 (C = N), 780 (C-O), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>): δ = 6.52 - 7.88 (m, 3H, furan-CH), 7.84 (s, 1H, CH), 7.33 - 7.60 (m, 5H, H<sub>Ar</sub>), 8.45 (s, 1H, CH), 11.61 (s, 1H, NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>15</sub>H<sub>11</sub>N<sub>3</sub>O<sub>2</sub> (265.27): C, 67.92; H, 4.18; N, 15.84%. Found: C, 68.26; H, 4.28; N, 15.93%.

#### Synthesis of pyrazole derivatives 7a,b.

**General procedure:** To a solution of compound **5** (0.265 g, 1.0 mmol) in 1,4-dioxane (25 mL) and dimethylformamide (10 mL), either hydrazine hydrate (**6a**) (0.05 g, 1.0 mmol) or phenylhydrazine (**6b**) (0.10 g, 1.0 mmol) was added. The reaction mixture in each case was heated under reflux for 3 h, and then poured on an ice/water mixture containing a few drops of hydrochloric acid. The formed solid product in each case was collected by filtration.

**3-Amino-N'-(furan-2-ylmethylene)-5-phenyl-1H-pyrazole-4-carbohydrazide (7a).** Brown crystals from 1,4-dioxane/dimethylformamide mixture, 87% (0.230 g) yield, mp. 250°C - 252°C, IR (KBr): ν/cm<sup>-1</sup> = 3380 (NH<sub>2</sub>), 3400, 3320 (2NH), 3200 (CH<sub>Ar</sub>), 1680 (CO), 1560 (C=N), 780 (C-O), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>): δ = 6.52 - 7.88 (m, 3H, furan-CH), 7.00, 7.20 (s, 2H, NH<sub>2</sub>, D<sub>2</sub>O-exchangeable),

7.41 - 7.79 (m, 5H, H<sub>Ar</sub>), 7.55 (s, 1H, CH), 11.62 (s, 1H, NH, D<sub>2</sub>O-exchangeable), 13.7 (s, 1H, pyrazole-NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>15</sub>H<sub>13</sub>N<sub>5</sub>O<sub>2</sub> (295.30): C, 61.01; H, 4.44; N, 23.72 %. Found: C, 61.34; H, 4.62; N, 23.91 %.

**3-Amino-1,5-diphenyl-N-(furan-2-ylmethylene)-1H-pyrazole-4-carb-ohydrazide (7b).** Yellowish brown crystals from 1,4-dioxane/dimethylformamide mixture, 90% (0.238 g) yield, mp. 148°C - 150°C, IR (KBr):  $\nu/\text{cm}^{-1}$  = 3370 (NH<sub>2</sub>), 3250 (NH), 2690 - 3400 (CH<sub>Ar</sub>), 1665 (CO), 1600 (C=N), 775 (C-O), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>):  $\delta$  = 6.52 - 7.88 (m, 3H, furan-CH), 7.41 - 7.79 (m, 10H, H<sub>Ar</sub>), 7.00, 7.20 (s, 2H, NH<sub>2</sub>, D<sub>2</sub>O-exchangeable), 7.55 (s, 1H, CH), 11.62 (s, 1H, NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>21</sub>H<sub>17</sub>N<sub>5</sub>O<sub>2</sub> (371.39): C, 67.91; H, 4.61; N, 18.86%. Found: C, 68.20; H, 4.81; N, 19.02%.

#### Synthesis of pyridine derivatives 10a,b.

**Method A:** To a solution of compound **5** (0.265 g, 1.0 mmol) either malononitrile (**8a**) (0.06 g, 1.0 mmol) or ethyl cyanoacetate (**8b**) (0.11 g, 1.0 mmol) was added in 1,4-dioxane (30 mL) containing triethylamine (3 drops). The reaction mixture was heated under reflux for 3 h. then poured on an ice/water mixture containing a few drops of hydrochloric acid. The solid product formed in each case was collected by filtration.

**Method B:** To a solution of compound **3** (0.177 g, 1.0 mmol) either benzylidene malononitrile (**11a**) (0.154 g, 1.0 mmol) or benzylidene ethyl cyanoacetate (**11b**) (0.201 g, 1.0 mmol) was added in 1,4-dioxane (30 mL) containing triethylamine (3 drops). The reaction mixture was heated under reflux for 3h, and then poured on an ice/water mixture containing a few drops of hydrochloric acid. The solid product formed in each case was collected by filtration.

**6-Amino-1-(furan-2-ylmethyleneamino)-2-oxo-4-phenyl-1,2-dihydro-pyridine-3,5-dicarbonitrile (10a).** Dark red crystals from 1,4-dioxane, yield [75% (0.198 g), method A; 95% (0.168 g), method B], mp. 190°C - 192°C, MS:  $m/z$  = 329 (M<sup>+</sup>), IR (KBr):  $\nu/\text{cm}^{-1}$  = 3300 (CH<sub>Ar</sub>), 3220 (NH<sub>2</sub>), 2350, 2480 (2CN), 1665 (CO), 1580 (C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>):  $\delta$  = 6.52 - 7.88 (m, 3H, furan-CH), 7.05 (s, 2H, NH<sub>2</sub>, D<sub>2</sub>O-exchangeable), 7.17 - 7.60 (m, 5H, H<sub>Ar</sub>), 7.56 (s, 1H, CH). *Analysis Calcd for* C<sub>18</sub>H<sub>11</sub>N<sub>5</sub>O<sub>2</sub> (329.31): C, 65.65; H, 3.37; N, 21.27%. Found: C, 65.73; H, 3.59; N, 21.38%.

**Ethyl-2-amino-5-cyano-1-(furan-2-ylmethylene-amino)-6-oxo-4-phen-yl-1,6-dihydropyridine-3-carboxylate (10b).** Brown crystals from 1,4-dioxane, yield 74% (0.196 g), method A; 98% (0.173 g), method B], mp. 180°C - 182°C, MS:  $m/z$  = 376 (M<sup>+</sup>), IR (KBr):  $\nu/\text{cm}^{-1}$  = 3210 (NH<sub>2</sub>), 3180 - 3050 (CH<sub>Ar</sub>), 2380 (CN), 1660 (CO), 1565 (C=N), 1190 (C-O), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>):  $\delta$  = 1.29 (t, 3H, J = 7 Hz, CH<sub>3</sub>), 4.24 (q, 2H, J = 7 Hz, CH<sub>2</sub>), 6.52 - 7.88 (m, 3H, furan-CH), 7.05 (s, 2H, NH<sub>2</sub>, D<sub>2</sub>O-exch-

ngeable), 7.17 - 7.40 (m, 5H, H<sub>Ar</sub>), 7.56 (s, 1H, CH). *Analysis Calcd for* C<sub>20</sub>H<sub>16</sub>N<sub>4</sub>O<sub>4</sub> (376.37): C, 63.82; H, 4.28; N, 14.89%. Found: C, 63.90; H, 4.41; N, 14.58%.

**N-(Furan-2-ylmethylene)-2-oxo-2H-coumarin-3-carbohydrazide (14).** To a solution of compound **3** (0.177 g, 1.0 mmol) in 1,4-dioxane (20 mL), piperidine (3 drops) and salicylaldehyde (**13**) (0.12 g, 1.0 mmol) were added. The reaction mixture was heated under reflux for 2 h, and then poured on an ice/water mixture containing a few drops of hydrochloric acid. The solid product was collected by filtration. Yellow crystals from dimethylformamide, 95% (0.168 g) yield, mp. 280°C - 282°C, MS:  $m/z$  = 282 (M<sup>+</sup>), IR (KBr):  $\nu/\text{cm}^{-1}$  = 3400 (CH<sub>Ar</sub>), 3100 (NH), 1700, 1600 (2CO), 1550 (C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>):  $\delta$  = 6.95 (s, 1H, coumarin C<sub>4</sub>-H), 7.52 (s, 1H, CH), 6.52 - 7.88 (m, 3H, furan-CH), 7.33 - 7.55 (m, 4H, H<sub>Ar</sub>), 11.61 (s, 1H, NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>15</sub>H<sub>10</sub>N<sub>2</sub>O<sub>4</sub> (282.25): C, 63.83; H, 3.57; N, 9.92%. Found: C, 63.97; H, 3.73; N, 10.31%.

#### Synthesis of pyridine derivatives 16a,b.

**General procedure:** To a solution of compound **3** (0.177 g, 1.0 mmol) either acetylacetone (**15a**) (0.1 g, 1.0 mmol) or ethyl acetoacetate (**15b**) (0.13 g, 1.0 mmol) was added in 1,4-dioxane (25 mL) containing piperidine (3 drops). The reaction mixture was heated under reflux for 5 h, and then poured on an ice/water mixture containing a few drops of hydrochloric acid. The solid product formed in each case was collected by filtration.

**4,6-Dimethyl-1-(furan-2-ylmethyleneamino)-2-oxo-1,2-dihydropyridine-3-carbonitrile (16a).** Brown crystals from 1,4-dioxane, 85% (0.150 g) yield, mp. 80°C - 82°C, MS:  $m/z$  = 241 (M<sup>+</sup>), IR (KBr):  $\nu/\text{cm}^{-1}$  = 2942, 2932 (2 CH<sub>3</sub>), 2900 (CH<sub>Ar</sub>), 2300 (CN) 1620 (CO), 1565 (C=N), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>):  $\delta$  = 2.22, 2.34 (2s, 6H, 2CH<sub>3</sub>), 5.78 (s, 1H, pyridine H-3), 7.52 (s, 1H, CH=N), 6.52 - 7.88 (m, 3H, furan-CH). *Analysis Calcd for* C<sub>13</sub>H<sub>11</sub>N<sub>3</sub>O<sub>2</sub> (241.25): C, 64.72; H, 4.60; N, 17.42%. Found: C, 64.82; H, 4.44; N, 17.72%.

**6-Ethoxy-1-(furan-2-ylmethyleneamino)-4-methyl-2-oxo-1,2-dihydro-pyridine-3-carbonitrile (16b).** Reddish brown crystals from dimethylformamide, 82% (0.145 g) yield, mp. 240°C - 242°C, MS:  $m/z$  = 271 (M<sup>+</sup>), IR (KBr):  $\nu/\text{cm}^{-1}$  = 2940 (CH<sub>3</sub>), 2910 (CH<sub>Al</sub>), 2900 (CH<sub>Ar</sub>), 2280 (CN), 1630 (CO), 1560 (C=N), 1070 (C-O), <sup>1</sup>H NMR (DMSO-d<sub>6</sub>):  $\delta$  = 1.35 (t, 3H, CH<sub>3</sub>), 2.34 (s, 3H, CH<sub>3</sub>), 4.22 (q, 2H, CH<sub>2</sub>), 5.79 (s, 1H, pyridine H), 7.51 (s, 1H, CH=N), 6.52 - 7.88 (m, 3H furan-CH). *Analysis Calcd for* C<sub>14</sub>H<sub>13</sub>N<sub>3</sub>O<sub>3</sub> (271.27): C, 61.99; H, 4.83; N, 15.49%. Found: C, 62.28; H, 4.93; N, 15.72%.

#### Synthesis of pyridine derivatives 18a,b.

**General procedure:** To a solution of compound **3** (0.177 g, 1.0 mmol) either malononitrile (**8a**) (0.06 g, 1.0 mmol) or ethyl cyanoacetate (**8b**) (0.11 g, 1.0 mmol) was added in 1,4-dioxane (25 mL) containing triethylamine

(4 drops). The reaction mixture was heated under reflux for 3h, and then poured on an ice/water mixture containing a few drops of hydrochloric acid. The solid product formed in each case was collected by filtration.

**4,6-Diamino-1-(furan-2-ylmethyleneamino)-2-oxo-1,2-dihydropyridine-3-carbonitrile (18a).** Red crystals from dimethylformamide, 95% (0.168 g) yield, mp. 250°C - 252°C, MS:  $m/z = 243$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3150, 3200$  ( $2NH_2$ ), 2150 (CN), 1650 (CO), 1550 (C=N),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 5.80$  (s, 1H, pyridine CH), 6.53 - 7.88 (m, 3H, furan-CH), 7.00, 7.19 (2s, 4H,  $2NH_2$ ,  $D_2O$ -exchangeable), 7.51 (s, 1H, CH). *Analysis Calcd for*  $C_{11}H_9N_3O_2$  (243.22): C, 54.32; H, 3.73; N, 28.79%. Found: C, 54.52; H, 3.93; N, 28.83%.

**4-Amino-1-(furan-2-ylmethyleneamino)-6-hydroxy-2-oxo-1,2-dihydropyridine-3-carbonitrile (18b).** Yellow crystals from 1,4-dioxane, 98% (0.173 g) yield, mp. 80°C - 83°C, MS:  $m/z = 244$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3450$  (OH), 3150 ( $NH_2$ ), 3050 ( $CH_{Ar}$ ), 2130 (CN), 1650 (CO), 1550 (C=N),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 5.80$  (s, 1H, pyridine-CH), 6.52 - 7.88 (m, 3H, furan-CH), 7.00 (s, 2H,  $NH_2$ ,  $D_2O$ -exchangeable), 7.52 (s, 1H, CH), 12.56 (s, 1H, OH,  $D_2O$ -exchangeable). *Analysis Calcd for*  $C_{11}H_8N_4O_3$  (244.21): C, 54.10; H, 3.30; N, 22.94%. Found: C, 54.37; H, 3.42; N, 23.18%.

#### Synthesis of hydrazide derivatives 21 and 22.

**General procedure:** To a mixture of compound 3 (0.177 g, 1.0 mmol) either acetophenone (19) (0.12 g, 1.0 mmol) or *p*-methoxyacetophenone (20) (0.15 g, 1.0 mmol) a catalytic amount of ammonium acetate (0.05 g) was added. The reaction mixture was fused at 140°C for 15 min, then left to cool. The solid product formed after boiling in dimethylformamide was collected by filtration.

**2-Cyano- $\bar{N}$ -(furan-2-ylmethylene)-3-phenylbut-2-enhydrazide (21).** Dark brown crystals from dimethylformamide, 77% (0.136 g) yield, mp. 230°C - 232°C, MS:  $m/z = 279$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3250$  ( $CH_{Ar}$ ), 3190 (NH), 3050 ( $CH_3$ ), 2180 (CN), 1660 (CO), 1560 (C=C),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 2.42$  (s, 3H,  $CH_3$ ), 6.52 - 7.88 (m, 3H, furan-CH), 7.33 - 7.77 (m, 5H,  $H_{Ar}$ ), 7.00 (s, 1H, NH,  $D_2O$ -exchangeable), 7.53 (s, 1H, CH). *Analysis Calcd for*  $C_{16}H_{13}N_3O_2$  (279.29): C, 68.81; H, 4.69; N, 15.05%. Found: C, 69.73; H, 4.72; N, 14.88%.

**2-Cyano- $\bar{N}$ -(furan-2-ylmethylene)-3-(4-methoxyphenyl) but-2-enhydrazide (22).** Dark orange crystals from 1,4-dioxane, 75% (0.132 g) yield, mp. 178°C - 180°C, MS:  $m/z = 309$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3280$  (NH), 3200 ( $CH_{Ar}$ ), 3050, 3080 (2  $CH_3$ ), 2180 (CN), 1650 (CO), 1580 (C=C), 1300, 1000 (2 C-O),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 2.42$  (s, 3H,  $CH_3$ ), 3.83 (s, 3H,  $CH_3$ ), 6.52 - 7.88 (m, 3H, furan-CH), 7.33 - 7.87 (m, 4H,  $H_{Ar}$ ), 7.00 (s, 1H, NH,  $D_2O$ -exchangeable), 7.52 (s, 1H, CH). *Analysis Calcd for*  $C_{17}H_{15}N_3O_3$  (309.32): C, 66.01; H, 4.89; N, 15.52%. Found: C, 65.83; H, 4.63; N, 15.72%.

#### Synthesis of phenylhydrazone derivatives 24a-24d.

**General procedure:** To a cold solution (0°C - 5°C) of compound 3 (0.177 g, 1.0 mmol) in ethanol (15 mL) an equimolar amount of diazotized benzenediazonium chloride (23a), 4-methylbenzenediazonium chloride (23b), 4-chlorobenzenediazonium chloride (23c) and 4-methoxybenzenediazonium chloride (23d) were gradually added under stirring. The solid product separated during stirring in an ice bath was collected by filtration and washed thoroughly with water.

**2-[2-(Furan-2-ylmethylene)hydrazinyl]-2-oxo- $\bar{N}$ -phenylacetohydraz-onoylcyanide (24a).** Yellow crystals from 1,4-dioxane, 92% (0.162 g) yield, mp. 130°C - 132°C, MS:  $m/z = 281$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3280$  (NH), 3100 ( $CH_{Ar}$ ), 2160 (CN), 1680 (CO), 1550 (C=N),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 6.52 - 8.44$  (m, 8H,  $5H_{Ar}$ , furan-3CH), 7.52 (s, 1H, CH), 10.55, 11.61 (2s, 2H, 2NH,  $D_2O$ -exchangeable). *Analysis Calcd for*  $C_{14}H_{11}N_5O_2$  (281.27): C, 59.78; H, 3.94; N, 24.90%. Found: C, 59.92; H, 4.12; N, 24.69%.

**2-(2-(Furan-2-ylmethylene)hydrazinyl)-2-oxo- $\bar{N}$ -p-tolylacetohydraz-onoylcyanide (24b).** Pale yellow crystals from 1,4-dioxane, 95% (0.168 g) yield, mp. 160°C - 162°C, MS:  $m/z = 295$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3250$  (NH), 3100 ( $CH_{Ar}$ ), 3080 ( $CH_3$ ), 2150 (CN), 1670 (CO), 1550 (C=N),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 2.34$  (s, 3H,  $CH_3$ ), 7.53 (s, 1H, CH), 6.52-8.44 (m, 7H,  $4H_{Ar}$ , furan-3CH), 10.55, 11.61 (2s, 2H, 2NH,  $D_2O$ -exchangeable). *Analysis Calcd for*  $C_{15}H_{13}N_5O_2$  (295.30): C, 61.01; H, 4.44; N, 23.72%. Found: C, 60.82; H, 4.31; N, 23.99%.

**$\bar{N}$ -(4-Chlorophenyl)-2-[2-(furan-2-ylmethylene)hydrazinyl]-2-oxoacetohydrazonoylcyanide (24c).** Orange crystals from 1,4-dioxane, 95% (0.168 g) yield, mp. 168°C - 170°C, MS:  $m/z = 317$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3200$  (NH), 3090 ( $CH_{Ar}$ ), 2180 (CN), 1600 (CO), 1550 (C=N), 1100 (C-Cl),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 7.52$  (s, 1H, CH=N), 6.52 - 8.44 (m, 7H,  $4H_{Ar}$ , furan-3H), 10.54, 11.62 (2s, 2H, 2NH,  $D_2O$ -exchangeable). *Analysis Calcd for*  $C_{14}H_{10}ClN_5O_2$  (315.71): C, 53.26; H, 3.19; N, 22.18%. Found: C, 53.37; H, 3.28; N, 22.09%.

**2-[2-(Furan-2-ylmethylene)hydrazinyl]- $\bar{N}$ -(4-nitrophenyl)-2-oxoacetohydrazonoylcyanide (24d).** Reddish brown crystals from 1,4-dioxane, 96% (0.169 g) yield, mp. 150°C - 152°C, MS:  $m/z = 326$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3280$  (NH), 3100 ( $CH_{Ar}$ ), 2160 (CN), 1680 (CO), 1550 (C=N), 1530, 1350 ( $NO_2$ ),  $^1H$  NMR (DMSO- $d_6$ ):  $\delta = 7.52$  (s, 1H, CH), 6.52 - 8.44 (m, 7H,  $4H_{Ar}$ , furan-3CH), 10.55, 11.61 (2s, 2H, 2NH,  $D_2O$ -exchangeable). *Analysis Calcd for*  $C_{14}H_{10}N_6O_4$  (326.27): C, 51.54; H, 3.09; N, 25.76%. Found: C, 51.67; H, 3.38; N, 25.83%.

**4-Amino-5-cyano- $\bar{N}$ -(furan-2-ylmethylene)-6-oxo-1-phenyl-1,6-dihydro-2-pyridazine-3-carbohydrazide (26).** To a solution of compound 24a (0.281 g, 1.0 mmol) ethyl cyanoacetate (8b) (0.11 g, 1.0 mmol) was added in 1,4-

dioxane (25 mL) containing triethylamine (4 drops) the reaction mixture was heated under reflux for 4 h, then poured on an ice/water mixture containing a few drops of hydrochloric acid. The solid product formed was collected by filtration.

Dark brown crystals from dimethylformamide, 78% (0.219 g) yield, mp. 250°C - 252°C, MS:  $m/z = 348$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3320$  (NH<sub>2</sub>), 3180 (NH), 2220 (CN), 1700, 1650 (2 CO), 1550 (C=N), <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta = 6.52 - 8.48$  (m, 8H, 5H<sub>Ar</sub>, furan-3CH), 7.10 (s, 2H, NH<sub>2</sub>), 7.50 (s, 1H, CH), 11.62 (s, 1H, NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>17</sub>H<sub>12</sub>N<sub>6</sub>O<sub>3</sub> (348.32): C, 58.62; H, 3.47; N, 24.13%. Found: C, 58.52; H, 3.62; N, 24.40%.

**5-[(Furan-2-ylmethylene)hydrazono]-2,4-diphenyl-3-thioxo-2,3,4,5-tetrahydro-1,2,4-triazine-6-carbonitrile (29).** To a solution of compound **24a** (0.281 g, 1.0 mmol) in 1,4-dioxane (25 mL) containing triethylamine (4 drops) phenyl isothiocyanate (**27**) (0.13 g, 1.0 mmol) was added. The reaction mixture was heated under reflux for 3 h, and then poured on an ice/water mixture containing a few drops of hydrochloric acid. The solid product formed was collected by filtration. Dark orange crystals from ethanol, 63% (0.177 g) yield, mp. 60°C - 62°C, IR (KBr):  $\nu/cm^{-1} = 3050$  (CH<sub>Ar</sub>), 2250 (CN), 2150 (C=S), 1550 (C=N), <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta = 6.25 - 8.50$  (m, 10H, 10H<sub>Ar</sub>), 6.52 - 7.88 (m, 3H, furan-3CH), 7.51 (s, 1H, CH). *Analysis Calcd for* C<sub>21</sub>H<sub>14</sub>N<sub>6</sub>OS (398.44): C, 63.30; H, 3.54; N, 21.09%. Found: C, 63.51; H, 3.22; N, 21.27%.

#### Synthesis of thiazole derivatives **33** and **35a,b**.

**General procedure:** To a solution of compound **3** (0.177 g, 1.0 mmol) in dimethylformamide (20 mL) containing potassium hydroxide (0.05 g, 1.0 mmol), phenyl isothiocyanate (**27**) (0.13 g, 1.0 mmol) was added. The reaction mixture was stirred over night, either ethyl chloroacetate (**31**) (0.12 g, 1.0 mmol) or chloroacetone (**34a**) (0.09 g, 1.0 mmol) or phenacyl bromide (**34b**) (0.19 g, 1.0 mmol) was added with continuation of stirring over night. Then the reaction mixture in each case was poured on an ice/water mixture containing a few drops of hydrochloric acid, and then was left over night. The solid product formed in each case was collected by filtration.

**2-Cyano-*N*-(furan-2-ylmethylene)-2-(4-hydroxy-3-phenylthiazol-2(3*H*)-ylidene)acetohydrazide (33).** Reddish brown crystals from ethanol/water, 67% (0.118 g) yield, mp. 67°C - 69°C, MS:  $m/z = 352$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3390$  (OH), 3280 (NH), 3150 (CH<sub>Ar</sub>), 2150 (CN), 1680 (CO), 1638 (C=C), 1580 (C=N), <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta = 5.79$  (s, 1H, thiazole-CH), 6.25 - 7.20 (m, 5H, C<sub>6</sub>H<sub>5</sub>), 6.52 - 7.88 (m, 3H, furan-3CH), 7.51 (s, 1H, CH), 11.62 (s, 1H, NH, D<sub>2</sub>O-exchangeable), 12.55 (s, 1H, OH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>17</sub>H<sub>12</sub>N<sub>4</sub>O<sub>3</sub>S (352.37): C, 57.95; H, 3.43; N, 15.90; S, 9.10%. Found:

C, 58.28; H, 3.59; N, 16.04; S, 9.39%.

**2-Cyano-*N*-(furan-2-ylmethylene)-2-(4-methyl-3-phenylthiazol-2(3*H*)-ylidene)acetohydrazide (35a).** Dark brown crystals from acetic acid, 63% (0.111 g) yield, mp. 80°C - 82°C, MS:  $m/z = 350$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3220$  (NH), 2980 (CH<sub>3</sub>), 2200 (CN), 1670 (CO), 1640 (C=C), 1600 (C=N), <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta = 2.35$  (s, 3H, CH<sub>3</sub>), 5.81 (s, 1H, thiazole-CH), 6.25 - 7.20 (m, 5H, H<sub>Ar</sub>), 6.52 - 7.88 (m, 3H, furan-3CH), 7.52 (s, 1H, CH) 11.62 (s, 1H, NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>18</sub>H<sub>14</sub>N<sub>4</sub>O<sub>2</sub>S (350.39): C, 61.70; H, 4.03; N, 15.99; S, 9.15%. Found: C, 61.95; H, 3.87; N, 16.21; S, 8.88%.

**2-Cyano-2-(3,4-diphenylthiazol-2(3*H*)-ylidene)-*N*-(furan-2-ylmethylene)acetohydrazide (35b).** Dark green crystals from acetic acid, 60% (0.106 g) yield, mp. 70°C - 72°C, MS:  $m/z = 412$  ( $M^+$ ), IR (KBr):  $\nu/cm^{-1} = 3220$  (NH), 3150 (CH<sub>Ar</sub>), 2190 (CN), 1690 (CO), 1600 (C=C), 1620 (C=N), <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta = 5.81$  (s, 1H, thiazole CH), 6.25 - 7.20 (m, 5H, H<sub>Ar</sub>), 6.52 - 7.88 (m, 3H, furan-3CH), 7.52 (s, 1H, CH), 7.71 - 7.69 (m, 5H, H<sub>Ar</sub>), 11.62 (s, 1H, NH, D<sub>2</sub>O-exchangeable). *Analysis Calcd for* C<sub>23</sub>H<sub>16</sub>N<sub>4</sub>O<sub>2</sub>S (412.46): C, 66.97; H, 43.91; N, 13.58; S, 7.77%. Found: C, 67.29; H, 3.66; N, 13.72; S, 8.01%.

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